

liquid, which was further dissolved in Et₂O and filtered to remove an insoluble material. Removal of Et₂O from the filtrate afforded 0.65 g of liquid [ν max (neat) 1700 (broad), 1565, and 1355 cm⁻¹].

5-Nitro-2-thiadiazole Derivatives.—Standard techniques or methods² were used for the preparation of the compounds described below and the yields are based on the amount of **5** used.

3-[(5-Nitro-1,3,4-thiadiazol-2-yl)methylene]amino}-2-oxazolidinone (6) was obtained in 6–15% yield and recrystallized from Me₂CO–EtOH as yellow crystals, mp 250–255°. *Anal.* (C₈H₅N₅O₃S) C, H, N, S.

1-[(5-Nitro-1,3,4-thiadiazol-2-yl)methylene]amino}-2-imidazolidinone (7) was obtained in 13–23% yield and recrystallized from 50% aq EtOH as yellow crystals, mp 230–233°; nmr (DMSO-*d*₆): τ 2.1 (s, 1 H, CH=N), 2.3 (s, 1 H, NH), 5.8–6.7 (m, 4 H, CH₂CH₂). *Anal.* (C₈H₆N₆O₃S) C, H, N, S.

5-Nitro-1,3,4-thiadiazole-2-carboxaldehyde thiosemicarbazone (8) was obtained in 22% yield as a red solid; no suitable sol-

vent for recrystallization was found, mp > 290°. *Anal.* (C₈H₄N₆O₃S₂) C, N, S, H: calcd 1.73; found 2.70.

2-Amino-5-(5-nitro-1,3,4-thiadiazol-2-yl)-1,3,4-thiadiazole (9).—The method reported previously² was used: 46% yield from **8**, recrystallized from EtOH–DMF, yellow crystals, mp 240° dec. *Anal.* (C₄H₂N₆O₃S₂) C, H, N, S.

Acknowledgment.—We wish to thank Dr. T. L. Chang (Stamford Laboratories, American Cyanamid Co.) for the mass spectral data and interpretation, Dr. G. A. Kemp and staff for *in vitro* and *in vivo* antibacterial assays, Mr. A. C. Dornbush and staff (Lederle Laboratories) for the *in vitro* antifungal assays, and Mr. G. S. Redin and staff (Lederle Laboratories) for their *in vivo* antibacterial assays.

New Compounds

Some Indole Derivatives¹

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In connection with other work in progress in this laboratory it was necessary to prepare the compounds described in Tables I and II for screening purposes,

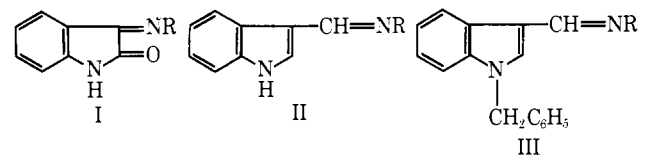
TABLE I

R ₂ CO	Mp, °C ^a	Yield, %	Formula	Analyses ^{b,c}
Androstanolone benzoate	161–162	90	C ₃₆ H ₄₈ N ₃ O ₃	H, N ^d
Estrone 3-methyl ether	225–227	10 ^e	C ₂₉ H ₃₈ N ₃ O ₂	H, N ^f
Pregnenolone	252–254 ^g	66 ^h	C ₃₁ H ₄₁ N ₃ O ₂	C, H
Testosterone benzoate	165–167	67 ⁱ	C ₃₆ H ₄₁ N ₃ O ₃	N
2-Nitrobenzaldehyde	174–175 ^j	85	C ₁₇ H ₁₄ N ₃ O ₃	C, H, N
4-Chloro-2-nitrobenzaldehyde	203–204	83	C ₁₇ H ₁₃ ClN ₃ O ₃	C, H
Indole-3-carboxaldehyde	228–230 ^k	77	C ₁₀ H ₈ N ₃ O	C, H, N

^a Recrystallized from EtOH unless otherwise noted. ^b Analyses indicated within 0.3%. ^c All compounds exhibited expected spectra. ^d Calcd: C, 76.43. Found: C, 75.72. ^e Yield (93%) based on recovered steroid. ^f Calcd: C, 76.45. Found: C, 75.70. ^g Not recrystallized. ^h Yield (94%) based on recovered steroid. ⁱ Reaction time increased to 5 hr. ^j Reaction product is C₁₇H₁₄N₃O₃·C₂H₅OH, (mp 111–112°, analyses: C, H); the product was heated to 130° to give the product in the Table. ^k A. Alemany, M. Bernabe, C. Elorriaga, E. F. Alvarez, M. Lora-Tamayo, and O. Nieto [*Bull. Soc. Chim. Fr.*, 2486 (1966)] report mp 193°.

by condensing indole-3-acetic acid hydrazide with carbonyl compounds and by condensing isatin, indole-3-carboxaldehyde, and 1-benzylindole-3-carboxalde-

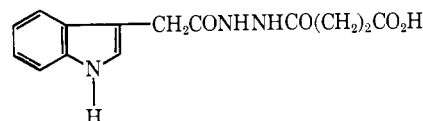
TABLE II



Type	RNH ₂	Mp, °C ^a	Yield, %	Formula	Analyses ^{b,c}
I	Sulfanilamide ^f	276–278	83	C ₁₄ H ₁₁ N ₃ O ₃ S	C, H
I	Sulfisoxazole ^f	237–239 ^d	60	C ₁₀ H ₁₀ N ₄ O ₄ S	C, H
I	Sulfathiazole ^f	335–338	93	C ₁₇ H ₁₂ N ₄ O ₃ S ₂	C, H
I	Sulfaguanidine ^f	296–297	44	C ₁₅ H ₁₃ N ₅ O ₃ S	H, N ^e
I	<i>p</i> -Sulfobenzamide hydrazide ^f	254–256	74	C ₁₅ H ₁₁ N ₃ O ₃ S	C, H
I	<i>p</i> -Toluenesulfon- <i>p</i> -hydrazide	207–209 ^f	90	C ₁₅ H ₁₃ N ₃ O ₃ S	C, H
I	3-(2-Aminoethyl)-indole	174–176	66	C ₁₈ H ₁₆ N ₃ O	H ^g
II	<i>p</i> -Sulfobenzamide hydrazide	240–241	88	C ₁₆ H ₁₃ N ₃ O ₄ S	C, H
II	2-Nitrophenylhydrazine	224–226 ^d	93	C ₁₅ H ₁₂ N ₃ O ₂	H ^h
II	<i>D</i> -Cycloserine	223–224 ⁱ	56	C ₁₂ H ₁₁ N ₃ O ₂	H, N ⁱ
III	Sulfathiazole	192–193 ^b	51	C ₂₃ H ₂₀ N ₄ O ₃ S ₂	C, H
III	Sulfanilamide	156–158	53	C ₂₂ H ₁₉ N ₃ O ₃ S	C, H

^a Recrystallized from EtOH unless otherwise noted. ^b Analyses indicated within 0.3%. ^c All compounds exhibited expected spectra. ^d Not recrystallized. ^e Calcd: C, 52.47. Found: C, 51.99. ^f M. P. Cava, R. O. Little, and D. R. Napier [*J. Amer. Chem. Soc.*, **80**, 2257 (1958)] report mp 190–200°. ^g Calcd: C, 74.72. Found: C, 74.11. ^h Calcd: C, 64.28. Found: C, 63.78. ⁱ Calcd: C, 62.87. Found: C, 62.18. ^j Triturated with hot EtOH–EtOAc. ^k From EtOAc. ^l Inactive (*T/C* = 83 – 102%) at 400 mg/kg against L-1210 lymphoid leukemia.

hyde with various amines. Reaction of indole-3-acetic acid hydrazide with succinic anhydride² gave I while



I

reaction of 3-aminocarbazole with 4-[bis(2-chloroethyl)-amino]-*o*-tolualdehyde gave the expected imine.³

(1) This work was supported by a research grant (CA 10345) from the National Cancer Institute, U. S. Public Health Service.

(2) F. W. Short and L. M. Long, *J. Heterocycl. Chem.*, **6**, 707 (1969).

(3) F. D. Popp, *J. Med. Chem.*, **7**, 210 (1964).

Experimental Section[†]

Condensations with Indole-3-acetic Acid Hydrazide.—Equimolar quantities of indole-3-acetic acid hydrazide and the appropriate carbonyl compounds were dissolved in a minimum of EtOH and heated on a steam bath for 30 min. After cooling, and in some cases standing for several days the products described in Table I were obtained by filtration.

Condensations with Amines.—In a similar manner equimolar quantities of isatin, indole-3-carboxaldehyde, or 1-benzylindole-3-carboxaldehyde were allowed to react in EtOH with the appropriate amines to give the compounds in Table II.

Indole-3-acetic Acid Hydrazide and Succinic Anhydride.—A mixture of 1.89 g (0.01 mole) of indole-3-acetic acid hydrazide and 1.00 g (0.01 mole) of succinic anhydride in Me₂CO (5 ml) was refluxed for 15 min and allowed to stand overnight at room temperature. Filtration gave 2.30 g (80%) of I, mp 203–204° from EtOH; ir(KBr): 3400, 3240, 2945, 1700 (broad), 1610 cm⁻¹. *Anal.* (C₁₄H₁₅N₃O₄): C, H.

Tolualdehyde Mustard and 3-Aminocarbazole.—A mixture of 1.82 g (0.01 mole) of 3-aminocarbazole and 2.60 g (0.01 mole) of 4-[bis-(2-chloroethyl)amino]-*o*-tolualdehyde was refluxed in EtOH to give 3.22 g (76%) of imine, mp 188° from EtOH. *Anal.* (C₂₄H₂₃Cl₂N₃): N.

[†] Analyses by Spang Microanalytical Laboratory, Ann Arbor, Mich. All melting points were taken in capillaries and are corrected.

Studies of the Chemistry of Azole Derivatives.

XII. Possible Anticonvulsant
Thiazolo [3,2-*a*]benzimidazoles

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In view of the potent pharmacological activity of a large number of heterocyclic thioureas¹⁻³ additional thioureidothiazolo[3,2-*a*]benzimidazoles were synthe-

Experimental Section

2-Aminothiazolo[3,2-*a*]benzimidazol-3-(2*H*)-one.—A solution of thiazolo[3,2-*a*]benzimidazol-3-(2*H*)-one⁴ (5 g) in AcOH (20 ml) was slowly added at 0° to a solution of PhN₂Cl with stirring. The mixture was kept for 1 hr at (0–5°) and the product obtained was crystallized from EtOH. The azo compound (5 g) was dissolved in hot EtOH (25 ml). A solution of Na₂S₂O₄ (25 g) in H₂O (50 ml) was added and the mixture was refluxed for 30 min and then cooled. The amino compound obtained was recrystallized from EtOH, yield 57%, mp 185°. *Anal.* (C₁₂H₇N₃S): N, S.

Synthesis of Thioureas.—Equimolecular quantities of 2-aminothiazolo[3,2-*a*]benzimidazol-3-(2*H*)-one and an aryl isothiocyanate were refluxed in abs EtOH for 5 hr and cooled. The precipitated thioureas were crystallized (C₆H₆). The hydrochlorides were prepared in Et₂O solution.

Acknowledgment. The author is thankful to Dr. Kartar Singh, Director, Defence Science Laboratory for encouragement and Dr. H. K. Acharya for providing facilities.

† Part XI, J. M. Singh, *J. Med. Chem.*, **12**, 962 (1969).

A Reinvestigation of the Reaction of Monosodium
Urea with Various Substituted
Pvrazinecarboxylate Esters

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We have reinvestigated the reaction of monosodium urea with methyl 3,5-diamino-6-chloropyrazinecarboxylate¹ as well as the 5-methylamino analog and found that a small amount of the desired *N*-carbamoylpyr-

TABLE I
2-AMINOTHIAZOLO[3,2-*a*]BENZIMIDAZOL-3-(2*H*)-ONETHIOUREA HYDROCHLORIDES

No.	R	Formula	Mp, °C	Yield, %	Activity	LD ₅₀ (toxicity)
1	Ph	C ₁₆ H ₁₃ ClN ₄ S ₂	220–221	60	++	200
2	<i>o</i> -MePh	C ₁₇ H ₁₃ ClN ₄ S ₂	195–197	65	++	260
3	<i>p</i> -MePh	C ₁₇ H ₁₃ ClN ₄ S ₂	175–177	59	++	240
4	<i>m</i> -MePh	C ₁₇ H ₁₃ ClN ₄ S ₂	198–200	58	++	280
5	<i>o</i> -BrPh	C ₁₆ H ₁₂ ClBrN ₄ S ₂	210	60	+++	300
6	<i>p</i> -BrPh	C ₁₆ H ₁₂ ClBrN ₄ S ₂	165–166	65	+++	280
7	<i>m</i> -BrPh	C ₁₆ H ₁₂ ClBrN ₄ S ₂	189	58	+++	290
8	<i>o</i> -ClPh	C ₁₆ H ₁₂ Cl ₂ N ₄ S ₂	202–204	62	++++	300
9	<i>p</i> -ClPh	C ₁₆ H ₁₂ Cl ₂ N ₄ S ₂	211	59	++++	350
10	<i>m</i> -ClPh	C ₁₆ H ₁₂ Cl ₂ N ₄ S ₂	215	61	++++	330

^a All new compounds were analyzed for N,S and the analytical values were within ± 0.4% of the calculated values. ^b Mice were used for the experiments for anticonvulsant activity following the method in Putnam and H. H. Merritt, *Science*, **85**, 525 (1937). A + + + + rating was given if the convulsive threshold is elevated more than 60 ma. + + + + is raised to 60 ma. + + + is raised by 40 ma, + + is raised by 15–20 ma and + is raised by 10–15 ma, 3.5 hr after treatment.

sized. These compounds have been tested for anticonvulsant activity (Table I).

(1) L. Goldman, U. S. Patent 2,617,804, *Chem. Abstr.*, **48**, 2124 (1954).

(2) T. N. Ghosh and A. R. Chaudhuri, *J. Indian Chem. Soc.*, **28**, 268 (1951).

(3) H. P. Kautmann and P. Schultz, *Arch. Pharm.*, **273**, 22 (1935).

azinecarboxamide is produced in each case. The products were isolated by liquid-liquid partition chromatography (llpc) and shown to be the desired compounds by comparison of ir, mass spectrum, tlc,

(4) J. William Hamilton, Rosemary Capuzzi, and Elliott Cohen, *J. Med. Chem.*, **12**, 1102 (1969).