

min. ED_{+300%} values are recorded in Table I. ED_{+300%} is defined as the dose at which the average sleeping time of the animals in a test group is increased by 300% in comparison to that of a control group.

References

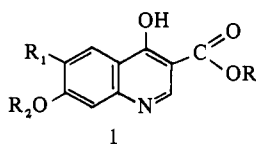
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Structure and Anticoccidial Activity of a New Series of 4-Hydroxyquinoline-3-carboxylates

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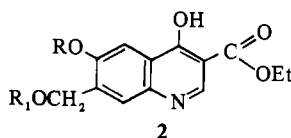
A new group of broad-spectrum coccidiostats, the 4-hydroxyquinoline-3-carboxylates (**1**), was first described by Spencer, *et al.*¹ The activity of these compounds was later confirmed in several publications.²⁻⁴ Decoquinatone³ (**1a**),



1a , decoquinatone	1b , nequinatone	1c , buquinolone
R = Et	R = Me	R = Et
R ₁ = OC ₁₀ H ₂₁	R ₁ = <i>n</i> -Bu	R ₁ = <i>i</i> -BuO
R ₂ = Et	R ₂ = CH ₂ C ₆ H ₅	R ₂ = <i>i</i> -Bu

nequinatone² (**1b**), and buquinolone¹ (**1c**) are among the most effective coccidiostats known at present. These compounds all have an alkoxy substituent in position 7 and an alkyl or alkoxy substituent in position 6.

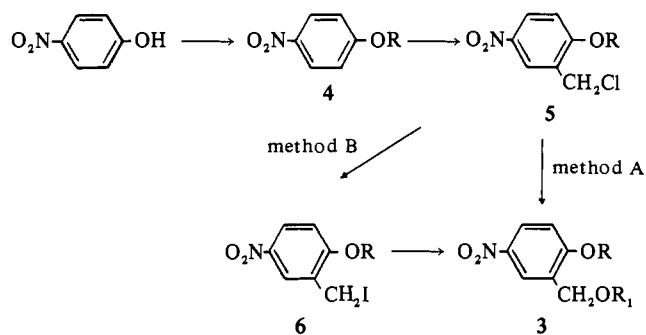
In contrast, the compounds **2** described here are ethyl 6-alkoxy-4-hydroxyquinoline-3-carboxylates with an alkoxy-methyl or an aralkoxymethyl substituent in the 7 position. These compounds are also potent coccidiostats.



Chemistry. Nitro compounds **3** were used to initiate the synthetic pathways leading to the formation of compounds

2. These nitro compounds themselves are synthesized as outlined in Scheme I. The reaction of *p*-nitrophenol with

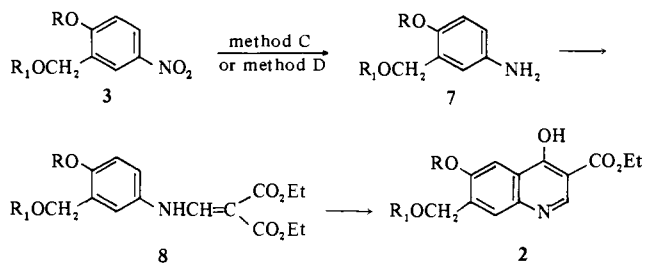
Scheme I



an appropriate alkyl halide in the presence of a base, e.g., NaOH or NaOMe, yields the *p*-nitrophenoxyalkanes **4**. These alkoxybenzenes may be chloromethylated using paraform, ZnCl₂, and gaseous HCl to produce the benzyl chlorides **5**. The desired nitro compounds **3** may be formed by the reaction of sodium alcoholates with benzyl chlorides (method A) although various side products and tars are frequently obtained. Therefore, the more usual procedure is to have the chlorides react with NaI in acetone to obtain benzyl iodides **6** (method B). High yields of the desired benzyl ethers **3** may be readily obtained by addition of appropriate alcohols to the system. Most of the benzyl halides, summarized in Table I, are novel.

The pathway starting with benzyl ethers **3**, as outlined in Scheme II, is used for the final synthesis of compounds

Scheme II



2. The nitro compounds **3** may be reduced to their corresponding anilines **7** by two different methods: either catalytic reduction with PtO₂ in ethanol (method C) or reduction with ammonium chloride and iron (method D). These anilines are condensed with diethyl methoxymethylmalonate in boiling ethanol or 2-propanol.

Only a limited number of condensation products **8** were isolated, the majority being used in their crude form for the last stage of the synthesis. Ring closure of compounds **8** was effected by heating in diphenyl ether or diphenylmethane, both solvents being equally suitable (Table II).

Chemotherapy. For screening purposes, 18-day-old male Hisex chickens weighing between 100 and 120 g were housed in individual cages for the duration of the experiment. Feed, known not to contain coccidiostat, was available at will. On day 0, the chickens were divided into three groups: four noninfected, nontreated birds; four infected, nontreated birds; two infected, treated birds. Coccidiosis was induced by inoculation of the test animals with approximately two million sporulated oöcysts of *Eimeria Acervulina*. For 6 days, the treated birds were given the compounds **2** at will at a dose of 0.01% of their feed. On the seventh day, the medicated feed was replaced by normal feed for five subse-

Table I

Compd	R	X	Yield, purified, %	Mp or bp (mm), °C	Formula	Analyses ^a
9	Me	Cl	84	78.5	C ₈ H ₈ ClNO ₃ ^b	Cl
10	Me	I	81	103	C ₈ H ₈ INO ₃ ^c	I, N
11	Et	Cl	73	69.5	C ₉ H ₁₀ ClNO ₃ ^d	Cl
12	Et	I	85	88	C ₉ H ₁₀ INO ₃	I, C, H, N
13	<i>n</i> -Bu	Cl	87	170–173 (1)	C ₁₁ H ₁₄ ClNO ₃	Cl, C, H, N
14	<i>n</i> -Bu	I	72	57	C ₁₁ H ₁₄ INO ₃	I, C, H, N
15	<i>n</i> -C ₇ H ₁₅	Cl	63	215–219 (2, 5)	C ₁₄ H ₂₀ ClNO ₃ ^e	Cl
16	<i>n</i> -C ₁₀ H ₂₁	Cl	87	^f	C ₁₇ H ₂₆ ClNO ₃	Cl
17	<i>n</i> -C ₁₀ H ₂₁	I	83	63	C ₁₇ H ₂₆ INO ₃	I, C, H, N

^aGc analysis gave a purity of at least 94% for all compounds. ^bR. Quelet and H. Coudanne, *C. R. Acad. Sci., Paris, Ser. C*, **252**, 894 (1961). ^cG. Bendz, *et al.*, *J. Chem. Soc.*, 1130 (1950). ^dM. Wakae, *et al.*, *Chem. Abstr.*, **54**, 10921h (1960). ^eR. Collins and M. Davis, *J. Chem. Soc. C*, 873 (1966). ^fCrude residue was used in the next step.

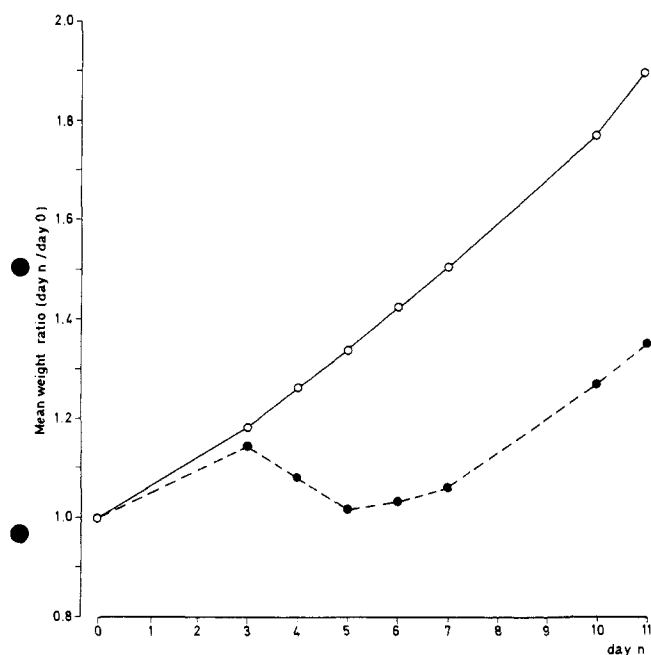


Figure 1. Growth rate for noninfected (○-○) and for infected (●-●) chicks.

quent days. During the experiment, all birds were weighed eight times, their fecal consistency was recorded three times, and a fecal count for oocysts was carried out once. All birds were sacrificed on the eleventh day of the experiment. Comparison of the growth rates of the control groups (Figure 1) shows a marked decrease in weight gain for the infected chickens. The main weight ratio (day 5/day 0) for these birds is 1.02. The anticoccidial activity of a compound can be determined by comparing the mean weight ratio (day *n*/day 0) from day 3 to day 7 with that of the two control groups. The results obtained on day 5 are presented in Table III, together with those of three leading coccidiostats.

A more detailed investigation into the activity of different concentrations of compound **88** was conducted with the aid of three different strains of *Eimeria*. Administration of the active compound was initiated on the day of inoculation (*i.e.*, simultaneous treatment) or two days before (*i.e.*, prophylactic treatment). Apart from this, the experimental procedure was as previously described.

The mean weight ratios are given in Table IV together with data on compound **1b** (nequinat) for comparative

purposes. The results indicate that compound **88** is at least as effective as nequinat and has therefore been selected for further investigation.

Experimental Section[†]

2-Butoxy- α -chloro-5-nitrotoluene (13). Gaseous HCl was passed through a mixture of 50.7 g (0.26 mol) of 4-O₂N-C₆H₄-OBU, 28.2 g (0.94 mol) of (CH₂O)₃, and 61.3 g (0.45 mol) of anhydrous ZnCl₂ for 3 hr at a temperature of 80°. The reaction mixture was cooled and poured into H₂O and CHCl₃. The organic layer was separated, washed with NaHCO₃ solution and with H₂O, dried (MgSO₄), and evaporated. The residue was distilled giving 54 g (87%) of **13**, bp 170–173° (1 mm). *Anal.* (C₁₁H₁₄ClNO₃) C, H, N, Cl.

2-Butoxy- α -iodo-5-nitrotoluene (14). A mixture of 24.3 g (0.1 mol) of **13**, 17 g (0.113 mol) of NaI, and 125 ml of Me₂CO was refluxed for 1 hr, after which time the reaction mixture was poured into H₂O and extracted with *i*-Pr₂O. The organic layer was dried (MgSO₄) and evaporated. Crystallization from EtOH yielded 24 g (72%) of **14**, mp 57°. *Anal.* (C₁₁H₁₄INO₃) C, H, N, I.

Method A. 2-Butoxy- α -heptyloxy-5-nitrotoluene (57). A solution of 4.2 g (0.18 g-atom) of Na in 120 ml of *n*-C₇H₁₅OH was added dropwise, at a temperature of 40°, to a mixture of 29.3 g (0.12 mol) of **13** in 40 ml of C₇H₁₅OH. The mixture was stirred for 72 hr at room temperature, after which time 200 ml of Et₂O was added. The organic layer was washed with H₂O, dried (MgSO₄), and evaporated. The residue was distilled to give 25 g (64%) of **57**, bp 184–188° (0.2 mm). *Anal.* (C₁₈H₂₅NO₄) C, H, N.

Method B. 2-Ethoxy- α -undecyloxy-5-nitrotoluene (48). A solution of 6.9 g (0.3 g-atom) of Na in 160 g of *n*-C₁₁H₂₁OH was added dropwise, at a temperature between 45 and 50°, to a stirred mixture of 61 g (0.2 mol) of **12** in 80 g of *n*-C₁₁H₂₁OH. Stirring was continued for 2 days at room temperature. The reaction mixture was poured into H₂O and extracted with Et₂O. The extract was dried (MgSO₄) and evaporated. The solid residue was crystallized from petroleum ether yielding 41.3 g (59%) of **48**, mp 50°. *Anal.* (C₂₀H₃₃NO₄) C, H, N.

Method C. 3-(Octyloxymethyl)-*p*-phenetidine (43). A mixture of 47 g (0.15 mol) of **42**, 300 ml of EtOH, and 1 g of PtO₂ was hydrogenated at normal pressure and room temperature. After the calculated volume of H₂ was taken up, the catalyst was removed by filtration and the filtrate concentrated. The residue was distilled to give 36.5 g (87%) of **43**, bp 164–168° (0.15 mm). *Anal.* (C₁₇H₂₉NO₂) N.

Method D. 3-(Heptyloxymethyl)-*p*-phenetidine (40). 39 (101 g, 0.34 mol) was added gradually to a stirred and refluxing mixture of 65 g (1.16 g-atom) of Fe and 650 ml (0.78 *N*) of NH₄Cl solution. The reaction mixture was refluxed for an additional 8 hr, cooled, and extracted with PhMe; the organic layer was dried (MgSO₄) and evaporated. The residue was distilled giving 70 g (79%) of **40**, bp 155–158° (0.4 mm). *Anal.* (C₁₆H₂₇NO₂) N.

[†]All melting points were measured with a "Mettler FP 1" melting point apparatus and are uncorrected. Where analyses are indicated by symbols of the elements, analytical results obtained for those elements were within $\pm 0.4\%$ of the theoretical values unless otherwise stated.

Table II

R_1O X
 R_2OCH_2

Compd	R ₁	R ₂	X	Yield, %	Method ^a	Mp or bp (mm), ^b °C	Formula	Analyses
18	Me	Et	NO ₂	70	A	66.6	C ₁₀ H ₁₃ NO ₄ ^c	N
19	Me	Et	NH ₂	47	D	109 (0.4)	C ₁₀ H ₁₅ NO ₂	N
20	Me	Et	NHCH=C(CO ₂ Et) ₂	90		47	C ₁₈ H ₂₅ NO ₆	C, H, N
21	Me	<i>i</i> -Bu	NO ₂	57	B	35	C ₁₇ H ₁₉ NO ₄	C, H, N
22	Me	<i>i</i> -Bu	NH ₂	90	C		C ₁₂ H ₁₉ NO ₂	
23	Me	<i>i</i> -Bu	NHCH=C(CO ₂ Et) ₂	90			C ₂₀ H ₂₉ NO ₆	
24	Me	<i>n</i> -C ₇ H ₁₅	NO ₂	51	A	50	C ₁₅ H ₂₃ NO ₄	C, H, N
25	Me	<i>n</i> -C ₇ H ₁₅	NH ₂	53	D	163-164 (0.5)	C ₁₅ H ₂₅ NO ₂	N
26	Me	<i>n</i> -C ₇ H ₁₅	NHCH=C(CO ₂ Et) ₂	91			C ₂₃ H ₃₅ NO ₆	N
27	Me	<i>n</i> -C ₈ H ₁₇	NO ₂	82	B	54	C ₁₆ H ₂₅ NO ₄	C, H, N
28	Me	<i>n</i> -C ₈ H ₁₇	NH ₂	27	D	174-176 (0.5)	C ₁₆ H ₂₇ NO ₂	N
29	Me	<i>n</i> -C ₈ H ₁₇	NHCH=C(CO ₂ Et) ₂	90			C ₂₄ H ₃₇ NO ₆	N
30	Me	CH ₂ C ₆ H ₅	NO ₂	82	B	83	C ₁₅ H ₁₅ NO ₄	C, H, N
31	Me	CH ₂ C ₆ H ₅	NH ₂	95	C		C ₁₅ H ₁₇ NO ₂	
32	Me	CH ₂ C ₆ H ₅	NHCH=C(CO ₂ Et) ₂	95			C ₂₃ H ₂₉ NO ₆	
33	Et	<i>i</i> -Bu	NO ₂	79	B	84.5	C ₁₃ H ₁₉ NO ₄	C, H, N
34	Et	<i>i</i> -Bu	NH ₂	80	C		C ₁₃ H ₂₁ NO ₂	
35	Et	<i>i</i> -Bu	NHCH=C(CO ₂ Et) ₂	95			C ₂₁ H ₃₁ NO ₆	
36	Et	<i>n</i> -C ₆ H ₁₃	NO ₂	87	B		C ₁₅ H ₂₃ NO ₄	N
37	Et	<i>n</i> -C ₆ H ₁₃	NH ₂	68	D	163-166 (1.2)	C ₁₅ H ₂₅ NO ₂	N
38	Et	<i>n</i> -C ₆ H ₁₃	NHCH=C(CO ₂ Et) ₂	95			C ₂₃ H ₃₅ NO ₆	N
39	Et	<i>n</i> -C ₇ H ₁₅	NO ₂	84	B		C ₁₆ H ₂₅ NO ₄	N
40	Et	<i>n</i> -C ₇ H ₁₅	NH ₂	79	D	155-158 (0.4)	C ₁₆ H ₂₇ NO ₂	N
41	Et	<i>n</i> -C ₇ H ₁₅	NHCH=C(CO ₂ Et) ₂	90		46.5	C ₂₄ H ₃₇ NO ₆	C, H, N
42	Et	<i>n</i> -C ₈ H ₁₇	NO ₂	88	B	198-200 (0.8)	C ₁₇ H ₂₇ NO ₄	N
43	Et	<i>n</i> -C ₈ H ₁₇	NH ₂	87	C	164-168 (0.15)	C ₁₇ H ₂₉ NO ₂	N
44	Et	<i>n</i> -C ₈ H ₁₇	NHCH=C(CO ₂ Et) ₂	80		64	C ₂₅ H ₃₉ NO ₆	C, H, N
45	Et	<i>n</i> -C ₉ H ₁₉	NO ₂	67	B	37	C ₁₈ H ₂₉ NO ₄	C, H, N
46	Et	<i>n</i> -C ₉ H ₁₉	NH ₂	70	C		C ₁₈ H ₃₁ NO ₂	
47	Et	<i>n</i> -C ₉ H ₁₉	NHCH=C(CO ₂ Et) ₂	83			C ₂₆ H ₄₁ NO ₆	
48	Et	<i>n</i> -C ₁₁ H ₂₃	NO ₂	59	B	50	C ₂₀ H ₃₃ NO ₄	C, H, N
49	Et	<i>n</i> -C ₁₁ H ₂₃	NH ₂	73	C	69	C ₂₀ H ₃₅ NO ₂ ·HCl	C, H, N, Cl
50	Et	<i>n</i> -C ₁₁ H ₂₃	NHCH=C(CO ₂ Et) ₂	91			C ₂₈ H ₄₅ NO ₆	
51	Et	CH ₂ CH ₂ C ₆ H ₅	NO ₂	26	B	220-225 (2)	C ₁₇ H ₁₉ NO ₄	N
52	Et	CH ₂ CH ₂ C ₆ H ₅	NH ₂	89	C		C ₁₇ H ₂₁ NO ₂	
53	Et	CH ₂ CH ₂ C ₆ H ₅	NHCH=C(CO ₂ Et) ₂	92			C ₂₅ H ₃₁ NO ₆	N
54	<i>n</i> -Bu	<i>n</i> -Pr	NO ₂	76	A		C ₁₄ H ₂₁ NO ₄	N
55	<i>n</i> -Bu	<i>n</i> -Pr	NH ₂	81	C		C ₁₄ H ₂₃ NO ₂	
56	<i>n</i> -Bu	<i>n</i> -Pr	NHCH=C(CO ₂ Et) ₂	91			C ₂₂ H ₃₃ NO ₆	N
57	<i>n</i> -Bu	<i>n</i> -C ₇ H ₁₅	NO ₂	64	A	184-188 (0.2)	C ₁₈ H ₂₉ NO ₄	C, H, N
58	<i>n</i> -Bu	<i>n</i> -C ₇ H ₁₅	NH ₂	88	C		C ₁₈ H ₃₁ NO ₂	
59	<i>n</i> -Bu	<i>n</i> -C ₇ H ₁₅	NHCH=C(CO ₂ Et) ₂	64			C ₂₆ H ₄₁ NO ₆	N
60	<i>n</i> -Bu	<i>n</i> -C ₈ H ₁₇	NO ₂	71	A	192-194 (0.2)	C ₁₉ H ₃₁ NO ₄	C, H, N
61	<i>n</i> -Bu	<i>n</i> -C ₈ H ₁₇	NH ₂	91	C		C ₁₉ H ₃₃ NO ₂	
62	<i>n</i> -Bu	<i>n</i> -C ₈ H ₁₇	NHCH=C(CO ₂ Et) ₂	50			C ₂₇ H ₄₃ NO ₆	N
63	<i>n</i> -Bu	<i>n</i> -C ₉ H ₁₉	NO ₂	64	A	204-208 (0.2)	C ₂₀ H ₃₃ NO ₄	C, H, N
64	<i>n</i> -Bu	<i>n</i> -C ₉ H ₁₉	NH ₂	87	C		C ₂₀ H ₃₅ NO ₂	
65	<i>n</i> -Bu	<i>n</i> -C ₉ H ₁₉	NHCH=C(CO ₂ Et) ₂	67			C ₂₈ H ₄₅ NO ₆	N
66	<i>n</i> -C ₇ H ₁₅	Et	NO ₂	49	A	173-176 (0.4)	C ₁₆ H ₂₅ NO ₄	N
67	<i>n</i> -C ₇ H ₁₅	Et	NH ₂	91	C		C ₁₆ H ₂₇ NO ₂	
68	<i>n</i> -C ₇ H ₁₅	Et	NHCH=C(CO ₂ Et) ₂	82			C ₂₄ H ₃₇ NO ₆	N
69	<i>n</i> -C ₇ H ₁₅	CH ₂ C ₆ H ₅	NO ₂	67	A	243-244 (0.7)	C ₂₁ H ₂₇ NO ₄	N
70	<i>n</i> -C ₇ H ₁₅	CH ₂ C ₆ H ₅	NH ₂	83	C		C ₂₁ H ₂₉ NO ₂	
71	<i>n</i> -C ₇ H ₁₅	CH ₂ C ₆ H ₅	NHCH=C(CO ₂ Et) ₂	87			C ₂₉ H ₃₉ NO ₆	N
72	<i>n</i> -C ₁₀ H ₂₁	Et	NO ₂	67	A	196 (0.3)	C ₁₉ H ₃₁ NO ₄	N
73	<i>n</i> -C ₁₀ H ₂₁	Et	NH ₂	81	C		C ₁₉ H ₃₃ NO ₂	
74	<i>n</i> -C ₁₀ H ₂₁	Et	NHCH=C(CO ₂ Et) ₂	91			C ₂₇ H ₄₃ NO ₆	
75	<i>n</i> -C ₁₀ H ₂₁	<i>n</i> -Pr	NO ₂	80	B	33	C ₂₀ H ₃₃ NO ₄	C, H, N
76	<i>n</i> -C ₁₀ H ₂₁	<i>n</i> -Pr	NH ₂	83	C		C ₂₀ H ₃₅ NO ₂	
77	<i>n</i> -C ₁₀ H ₂₁	<i>n</i> -Pr	NHCH=C(CO ₂ Et) ₂	89			C ₂₈ H ₄₅ NO ₆	
78	<i>n</i> -C ₁₀ H ₂₁	<i>n</i> -C ₅ H ₁₁	NO ₂	82	B		C ₂₂ H ₃₇ NO ₄	
79	<i>n</i> -C ₁₀ H ₂₁	<i>n</i> -C ₅ H ₁₁	NH ₂	91	C		C ₂₂ H ₃₉ NO ₂	
80	<i>n</i> -C ₁₀ H ₂₁	<i>n</i> -C ₅ H ₁₁	NHCH=C(CO ₂ Et) ₂	93			C ₃₀ H ₄₉ NO ₆	

^aMethods refer to the Experimental Section. ^bWhen no physical data are given, compounds were used in their crude form. ^cR. Quelet and H. Coudanne, *Bull. Soc. Chim. Fr.*, 2445 (1963).

Diethyl [3-(Heptyloxymethyl)-*p*-phenetidinomethylene]malonate (41). A solution of 13.3 g (0.05 mol) of 40 and 11.9 g (0.055 mol) of diethyl methoxymethylenemalonate in 100 ml of *i*-PrOH was refluxed for 24 hr. After evaporation of the reaction mixture, the residue was crystallized from petroleum ether to give 19.5 g (90%)

of 41, mp 46.5°. *Anal.* (C₂₄H₃₇NO₆) C, H, N.

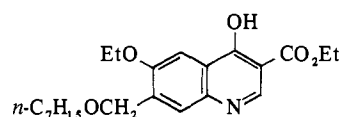
Ethyl 6-Ethoxy-7-heptyloxymethyl-4-hydroxy-3-quinoline-carboxylate (88). 41 (1.3 g, 0.03 mol) was added to 100 g of Ph₂O preheated to 230°. The resulting mixture was heated at 245° for 15 min. After cooling to 70° Me₂CO was added. The precipitate was

Table III

Compd	R ₁	R ₂	Yield purified, %	Mp, °C	Formula ^a	Weight ratio (day 5/day 0) ^b		Oocyst count (day 7) ^c		Fecal score (day 5) ^d	
						0.01%	0.001%	0.01%	0.001%	0.01%	0.001%
81	Me	Et	23	293.5	C ₁₆ H ₁₉ NO ₅	1.18		1;1;0;0	1;1	1;1;1;1	3;3
82	Me	<i>i</i> -Bu	17	248	C ₁₈ H ₂₃ NO ₅	1.17		1;1;1;1		2;1;2;3	
83	Me	<i>n</i> -C ₇ H ₁₅	27	+300	C ₂₁ H ₂₉ NO ₅	1.25		1;1;1;1	2;2;2;2	0;1;1;1	2;2;3;2
84	Me	<i>n</i> -C ₈ H ₁₇	25	230	C ₂₂ H ₃₁ NO ₅	1.25		0;0;0;0	3;2;2	1;1;1;1	2;2;1;3
85	Me	CH ₂ C ₆ H ₅	29	231	C ₂₁ H ₂₁ NO ₅	1.17		1;2;1;2		2;2;1;2	
86	Et	<i>i</i> -Bu	17	246	C ₁₉ H ₂₅ NO ₅	1.24		1;2;2	2;2	1;1;1;0	1;2
87	Et	<i>n</i> -C ₆ H ₁₃	31	222	C ₂₁ H ₂₉ NO ₅	1.20		1;1;1;1	1;1;3;3	1;1;1;1	3;2;3;3
88	Et	<i>n</i> -C ₇ H ₁₅	47	227	C ₂₂ H ₃₁ NO ₅	1.24	1.23	0;0;1;1	0;0;2;2	1;0;1;1	1;1;1;1
89	Et	<i>n</i> -C ₈ H ₁₇	31	226	C ₂₃ H ₃₃ NO ₅	1.26	1.23	0;0;0;0	0;0;1;1	0;1;1;1	1;1;2;1
90	Et	<i>n</i> -C ₉ H ₁₉	12	225	C ₂₄ H ₃₅ NO ₅	1.22		1;1;0;0	0;0;2;2	1;1;0;0	1;1;3;3
91	Et	<i>n</i> -C ₁₁ H ₂₃	31	222	C ₂₆ H ₃₉ NO ₅	1.21		1;1;1;1	1;1;1;1	1;1;1;1	1;1;1;2
92	Et	CH ₂ CH ₂ C ₆ H ₅	56	229	C ₂₃ H ₂₅ NO ₅	1.25		1;1;2;2	1;1;2;2	0;0;2;1	2;1;1;2
93	<i>n</i> -Bu	<i>n</i> -Pr	33	227	C ₂₂ H ₂₇ NO ₅	1.19		1;1;1;1	1;1;2;2	1;2;1;1	1;1;2;1
94	<i>n</i> -Bu	<i>n</i> -C ₇ H ₁₅	28	217	C ₂₄ H ₃₅ NO ₅	1.20		1;1;1;1	1;1;2;2	2;1;1;1	2;2;1;1
95	<i>n</i> -Bu	<i>n</i> -C ₈ H ₁₇	32	215	C ₂₅ H ₃₇ NO ₅	1.20		0;0;1;1	1;1;1;1	0;1;1;0	1;1;1;3
96	<i>n</i> -Bu	<i>n</i> -C ₉ H ₁₉	25	213	C ₂₆ H ₃₉ NO ₅	1.17		0;0;1;1	1;1;2;2	1;1;1;1	1;1;1;1
97	<i>n</i> -C ₇ H ₁₅	Et	32	221	C ₂₂ H ₃₁ NO ₅	1.18		1;2;2;2	1;1	1;3;1;1	3;3
98	<i>n</i> -C ₇ H ₁₅	CH ₂ C ₆ H ₅	27	219	C ₂₇ H ₃₃ NO ₅	1.18		1;1;1;1	1;1;1;1	1;2;0;1	1;1;2;3
99	<i>n</i> -C ₁₀ H ₂₁	Et	31	217	C ₂₅ H ₃₇ NO ₅	1.30		1;1;1;1	1;1;1;1	0;0;1;1	1;1;1;1
100	<i>n</i> -C ₁₀ H ₂₁	<i>n</i> -Pr	43	213	C ₂₆ H ₃₉ NO ₅	1.19		2;1	3;3	3;1	3;3
101	<i>n</i> -C ₁₀ H ₂₁	<i>n</i> -C ₅ H ₁₁	23	205	C ₂₈ H ₄₃ NO ₅	1.20		2;2;1;1	1;1;1;1	2;1;1;2	1;1;2;2
1a	Decoquinatate (ethyl 6- <i>n</i> -decyloxy-7-ethoxy-4-hydroxy-3-quinoline-carboxylate)					1.24		0;0;0;0	0;0;1;1	1;0;1;1	1;1;1;0
1b	Nequinatate (methyl 7-benzyloxy-6- <i>n</i> -butyl-4-hydroxy-3-quinoline-carboxylate)					1.24	1.19	0;0;1;0	0;1;2;1	0;1;1;0	0;0;2;1
1c	Amquinatate (methyl 7-diethylamino-4-hydroxy-6- <i>n</i> -propyl-3-quinolinecarboxylate)					1.22		0;1;1;1	1;1;3;1	0;0;1;0	1;2;3;1

^aAll compounds were analyzed for C, H, and N. Compound 97, C: -0.54; 99, C: -0.64; 100, C: +0.47. ^bRefer to chemotherapy. ^c0, no oocysts in feces; 1, 0-5 × 10⁴ oocyst/g of feces; 2, 5 × 10⁴-1 × 10⁵ oocyst/g of feces; 3, 10⁵-2 × 10⁵ oocyst/g of feces; 4, more than 2 × 10⁵ oocyst/g of feces. ^d0, normal feces; 1, soft to normal feces; 2, fluid droppings with some mucous casts; 3, slimy, greyish, mucoid diarrhea.

Table IV. Chemotherapeutical Results of



<i>Eimeria</i> strain	Treatment	Mean weight ratio				
		Noninfected chicks	Infected chicks	Infected chicks treated with 0.01% of 88	Infected chicks treated with 0.001% of 88	Infected chicks treated with 0.001% of 1b
<i>Acervulina</i> ^a	Simultaneous	1.33	1.04	1.24	1.23	1.19
	Prophylactic	1.60	1.16		1.50	1.50
<i>Brunetti</i> ^b	Simultaneous	1.42	1.14	1.40	1.42	1.41
	Prophylactic	1.65	1.14		1.66	1.62
<i>Tenella</i> ^c	Simultaneous	1.34	1.19	1.53	1.51	1.49
	Prophylactic	1.78	1.33	1.79	1.74	1.72

^aResults of the 5th day. ^bResults of the 6th day. ^cResults of the 7th day.

filtered and triturated with Me₂CO for 1 hr. The precipitate was collected and dried *in vacuo* to give 5.5 g (47%) of 88, mp 227°. *Anal.* (C₂₂H₃₁NO₅) C, H, N.

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Synthesis and Antiviral Activity of Homologs of Noformycin

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In the past two decades, several papers have appeared on the antiviral activity of noformycin¹⁻⁷ (1) obtained from a culture of *Nocardia formica*. Among the viruses reportedly