

an incision was made in the fourth right intercostal space and the pericardium was incised. A Beckman polarographic PO₂ microelectrode (in a Riley needle) was placed in the coronary sinus via the superior vena cava. Compound, dissolved in isotonic saline, was administered as a bolus either intravenously (jugular) or intraduodenally through an indwelling catheter.

Acknowledgment. The authors are grateful to Dr. D. L. Garmaise for his help and encouragement. They are also indebted to Dr. S. Hanessian of the University of Montreal, Canada, for many helpful discussions during the progress of this work.

References and Notes

- (1) (a) Organic Chemical Research; (b) Department of Pharmacology and Medicinal Chemistry.
- (2) (a) P. Somani, H. D. Brondyk, R. A. Hill, and H. H. Stein, *Fed. Proc., Fed. Am. Soc. Exp. Biol.*, **32**, 3254 (1973); (b) H. H. Stein, H. D. Brondyk, R. N. Prasad, A. Fung, and K. Tietje, 166th National Meeting of the American Chemical Society, Chicago, Ill., August 26–31, 1973, MEDI 41.
- (3) P. Somani and H. D. Brondyk in ref 2b, MEDI 42.
- (4) H. H. Stein, *J. Med. Chem.*, **16**, 1306 (1973).
- (5) H. H. Stein, P. Somani, and R. N. Prasad, *Ann. N.Y. Acad. Sci.*, **255**, 380 (1975).
- (6) R. E. Harmon, C. V. Zenarosa, and S. K. Gupta, *Chem. Ind. (London)*, 1141 (1969).
- (7) R. R. Schmidt, U. Schloz, and D. Schwille, *Chem. Ber.*, **101**, 590 (1968).
- (8) G. P. Moss, C. B. Reese, K. Schofield, R. Shapiro, and L. Todd, *J. Chem. Soc.*, 1149 (1963).
- (9) H. Ungar-Waron, E. Hurwitz, J.-C. Jaton, and M. Sela, *Biochim. Biophys. Acta*, **138**, 513 (1967).
- (10) M. Sela and H. Ungar-Waron, *Methods Enzymol.*, **12B**, 900 (1968).
- (11) P. J. Harper and A. Hampton, *J. Org. Chem.*, **35**, 1688 (1970).
- (12) H.-J. Fritz, R. Machat, and R. R. Schmidt, *Chem. Ber.*, **105**, 642 (1972).
- (13) R. R. Schmidt and H.-J. Fritz, *Chem. Ber.*, **103**, 1867 (1970).
- (14) E. Fauland, W. Kampe, M. Thiel, K. Dietman, and W. Juhram, German Patent No. 2 034 784 (Jan 1972).
- (15) H. G. Shoepke, T. D. Darby, and H. D. Brondyk, *Pharmacologist*, **8**, 204 (1966).
- (16) I. Merits and D. J. Anderson, *Xenobiotica*, **3**, 381 (1973).
- (17) A. Bloch, M. J. Robins, and J. R. McCarthy, Jr., *J. Med. Chem.*, **10**, 908 (1967).
- (18) NMR studies, reported herein, were done under the direction of Dr. Richard S. Egan of Abbott Laboratories, North Chicago, Ill.
- (19) P. A. Hart and J. P. Davis, *J. Am. Chem. Soc.*, **91**, 512 (1969).
- (20) Mass spectra were determined under the direction of Dr. Milton Levenberg of Abbott Laboratories, North Chicago, Ill.

Analogues of 8-Azaguanosine

Robert D. Elliott and John A. Montgomery*

Kettering-Meyer Laboratory, Southern Research Institute, Birmingham, Alabama 35205. Received March 29, 1976

Two routes for the synthesis of 6-substituted 8-azaguanosine analogues are described. 2,5,6-Triamino-4(3H)-pyrimidinethione (1) was converted by methylation, nitrosation, and acetylation to *N*-acetyl-7-(methylthio)-3H-1,2,3-triazolo[4,5-*d*]pyrimidin-5-amine (5). The reaction of 5 with 2,3,5-tri-*O*-acetyl-D-ribofuranosyl chloride gave a mixture of the 7-, 8-, and 9-(2,3,5-tri-*O*-acetyl-β-D-ribofuranosyl)-8-azapurines 4a–c which was converted to 8-azaguanosine (7c) and the corresponding 7- and 8-substituted isomers 7a and 7b. 4a–c were also converted with NaOMe to 6-*O*-methyl-8-azaguanosine (8c) and to the corresponding 7- and 8-substituted isomers 8a and 8b. The preferred route, however, to 6-substituted 8-azaguanosine analogues is an unambiguous synthesis through *N*²-acetyl-6-(benzylthio)-*N*⁴-(2,3-*O*-isopropylidene-β-D-ribofuranosyl)-5-nitro-2,4-pyrimidinediamine (13), prepared from the reaction of the chloropyrimidine 10 with the aminoribose 11. Catalytic hydrogenation of 13 gave the aminopyrimidine 14, which was converted with nitrous acid to the nucleoside β-20. Treatment of β-20 with dilute acid gave 7-(benzylthio)-3-β-D-ribofuranosyl-3H-1,2,3-triazolo[4,5-*d*]pyrimidin-5-amine (19). Replacement of the benzylthio group of 19 with various nucleophilic reagents gave 8-aza-6-thioguanosine 17 and analogues 8c, 15, and 16. The thione 17 rearranges in aqueous solution to the thiadiazolopyrimidine 21. The parent [1,2,3]thiadiazolo[5,4-*d*]pyrimidine-5,7-diamine (24a) was prepared by nitrosation of the triaminopyrimidine (23a). Rearrangement of 24a in the presence of base gave a high yield of the thione 25a which could be rearranged with heat to 24a. Compounds 8a–c, 15–19, 24a, and 25a were evaluated in the H.Ep.-2 cell culture screen and compounds 8c, 15–19, 24a, and 25a in the L1210 mouse leukemia screen. Only one compound, 8c, showed high cytotoxicity and borderline L1210 activity resulting from its enzymatic conversion to 8-azaguanosine.

The anticancer activity of 8-azaadenosine (3-β-D-ribofuranosyl-3H-1,2,3-triazolo[4,5-*d*]pyrimidin-7-amine)¹ and 8-azainosine (3,6-dihydro-3-β-D-ribofuranosyl-7H-1,2,3-triazolo[4,5-*d*]pyrimidin-7-one)^{1,2} in experimental animals systems coupled with the clinical efficacy of the 6-thiopurines against human leukemias^{3,4} led us to prepare alkoxy- and sulfur-containing congeners of these nucleosides. These compounds proved to be substrates for adenosine kinase, to be cytotoxic to cell lines possessing this enzyme but not to lines deficient in it, and to show some activity against murine leukemias.^{1,5} These initially encouraging results and the knowledge of the anticancer activity of 6-thioguanine³ and 8-azaguanine (5-amino-3,6-dihydro-7H-1,2,3-triazolo[4,5-*d*]pyrimidin-7-one)⁶ suggested the preparation and reevaluation of the known 8-aza-6-thioguanine (5-amino-3,6-dihydro-7H-1,2,3-triazolo[4,5-*d*]pyrimidine-7-thione) (25a)⁷ and the synthesis

of its ribonucleoside 17 since, although 6-thioguanosine is rapidly phosphorylated to 6-thioguanine,³ 8-azapurine ribonucleosides such as 8-azainosine are not and show activity not possessed by the parent heterocycles.^{1,2}

The successful synthesis of 7-(methylthio)-3-(2,3,5-tri-*O*-acetyl-β-D-ribofuranosyl)-3H-1,2,3-triazolo[4,5-*d*]pyrimidine⁸ by the molecular sieve catalyzed reaction of 7-(methylthio)-3H-1,2,3-triazolo[4,5-*d*]pyrimidine with 2,3,5-tri-*O*-acetyl-D-ribofuranosyl chloride and the reactivity of the methylthio group of this nucleoside to nucleophilic displacement reactions^{8,9} suggested a similar sequence for the preparation of 8-aza-6-thioguanosine (5-amino-3,6-dihydro-3-β-D-ribofuranosyl-7H-1,2,3-triazolo[4,5-*d*]pyrimidine-7-thione, 17).

2,5,6-Triamino-4(3H)-pyrimidinethione¹⁰ (1) was alkylated with methyl iodide to give a 72% yield of 6-(methylthio)-2,4,5-pyrimidinetriamine (2). This route to

Table I. Uv Spectra of Substituted 8-Azaguanine

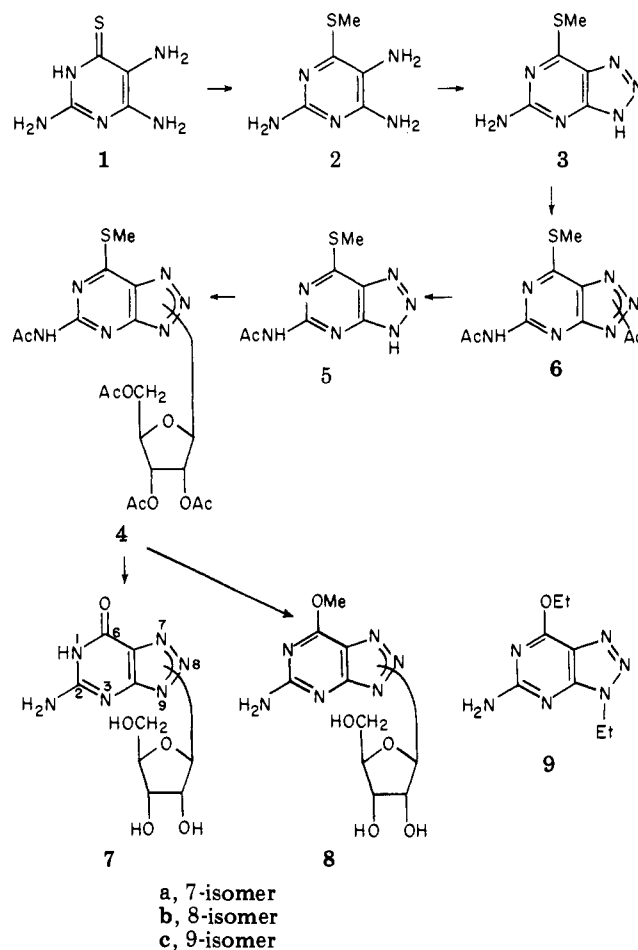
8-Azaguanine ^a	λ_{\max} , nm ($\epsilon \times 10^{-3}$)		
	0.1 N HCl	pH 7	0.1 N NaOH
1-Methyl ^b	252 (5.65), 264 (sh) (4.16)		
3-Methyl ^c	205 (25.7), 264 (7.39)	213 (7.56), 249 (9.85), 267 (11.2)	249 (9.76), 267 (11.2)
7-Methyl ^d	210 (23.5), 270 (5.37)	210 (25.7), 240 (7.12), 296 (5.23)	219 (20.1), 245 (sh) (5.46), 297 (5.89)
8-Methyl ^d	206 (26.8), 269 (8.84)	211 (28.7), 241 (6.69), 292 (6.32)	218 (22.7), 250 (4.97), 296 (7.98)
7a	210, 273	211, 240 (sh), 300	219, 245 (sh), 298
7b	208, 277	215, 240 (sh), 304	221, 257, 308
9- β -D-Ribofuranosyl ^e	255 (13.4), 275 (sh) (9.34)	256 (12.9), 275 (sh) (9.00)	221 (23.3), 278 (11.7)

^a Purine numbering. ^b See ref 15. ^c See ref 16. ^d See ref 17. ^e Sample prepared according to ref 14.

2 proved to be more convenient than that described by Daves et al.¹¹ Nitrosation of 2 gave a 90% yield of the known 7-(methylthio)-3H-1,2,3-triazolo[4,5-d]pyrimidin-5-amine¹² (3). Attempts to convert 3 to the ribonucleoside by silylation with hexamethyldisilazane and treatment of the silylated derivative with triacetylribofuranosyl chloride or the corresponding bromide gave an impractically low yield of a mixture of nucleosides. The reaction of 3 with Ac₂O gave an 89% yield of a diacetyl derivative 6, which was partially hydrolyzed in dilute base to give a 98% yield of *N*-acetyl-7-(methylthio)-3H-1,2,3-triazolo[4,5-d]pyrimidin-5-amine (5), the infrared spectrum of which contained an absorption band at 1660 cm⁻¹, confirming the position of the acetyl group. Treatment of 5 with triacetylribofuranosyl chloride in refluxing 1,2-dichloroethane containing Linde 4A molecular sieve¹³ gave a low yield of a mixture of three nucleosides which was partially separated by column chromatography into one pure compound and a mixture of the other two. These nucleosides were identified by treatment with H₂O₂ in glacial HOAc followed by deblocking with methanolic NaOMe to give the corresponding crude 8-azaguanine ribonucleosides 7a-c. The site of ribosylation was determined by comparison of the uv spectra of 7a-c with the spectra of 8-azaguanosine¹⁴ (7c) and the known *N*-methyl-8-azaguanines (see Table I). A comparison of the peak ratios and shapes of the uv curves left no doubt as to the identity of 7a-c and, accordingly, 4a-c, the nucleosides obtained in the sieve-catalyzed reaction. Treatment of 4a-c with methanolic NaOMe gave the corresponding methoxy analogues 8a-c. A uv spectrum of 8c was similar to that reported for 7-ethoxy-3-ethyl-3H-1,2,3-triazolo[4,5-d]pyrimidin-5-amine¹⁸ (9).

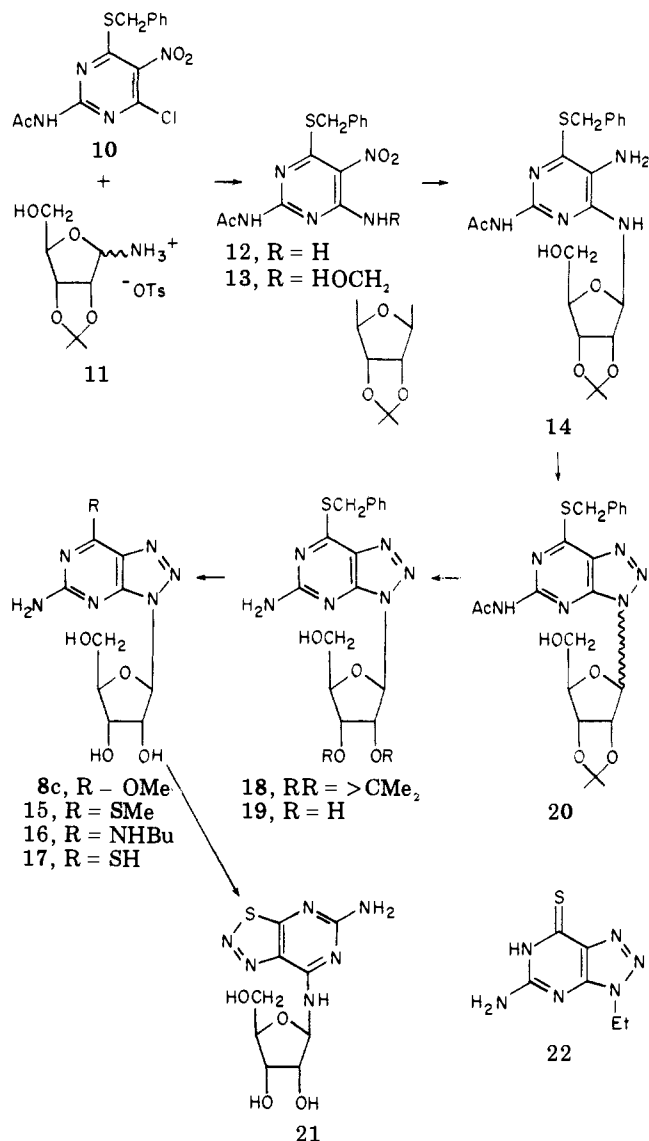
Unambiguous syntheses of 8-azapurine nucleosides have been effected by reaction of 4-chloro-5-nitropyrimidines with substituted glucopyranosylamines, followed by hydrogenation of the nitro group and cyclization with nitrous acid.^{19,20} An attempt by Rokos and Pfeleiderer²¹ to apply this reaction to ribosylamine by treatment of 1-amino-2,3,5-tri-*O*-benzoyl-1-deoxy-D-ribofuranose with 4-chloro-5-nitropyrimidines was unsuccessful because of apparent instability of the aminoribose. The recent disclosure of a simple synthesis of the more stable 2,3-*O*-isopropylidene-D-ribofuranosylamine tosylate²² (11) suggested its use in an unambiguous synthesis of 8-aza-6-thioguanosine (17).

The reaction of 11 with the chloropyrimidine²³ (10) in EtOH containing NaHCO₃ gave a 44% yield of anomeric pure *N*²-acetyl-6-(benzylthio)-*N*⁴-(2,3-*O*-isopropylidene- β -D-ribofuranosyl)-5-nitro-2,4-pyrimidinediamine (13). An initial attempt to carry out this reaction in the presence of Et₃N as the acid acceptor gave a 30% yield of 13 and a 12% yield of the aglycon 12, probably resulting from base decomposition of 11 with liberation of NH₃. Hydrogenation of 13 in EtOH containing 5% Pd on



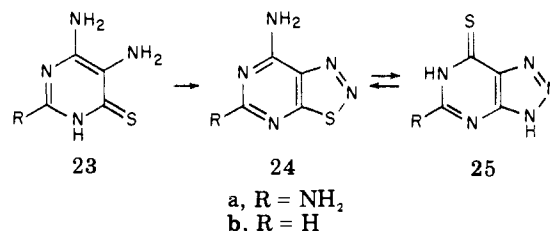
charcoal gave an 82% yield of the anomerically pure ribosylaminopyrimidine (14). The use of Raney nickel in this reduction gave a lower (43%) yield of 14. Addition of NaNO₂ to a cold solution of 14 in dilute AcOH gave a 4:1 mixture of *N*-acetyl-7-(benzylthio)-3-(2,3-*O*-isopropylidene- β -D-ribofuranosyl)-3H-1,2,3-triazolo[4,5-d]pyrimidin-5-amine (β -20) and the corresponding α -anomer (α -20). Although β -20 was found to be stable in AcOH, the amine 14 racemized at an appreciable rate under these conditions. To minimize formation of the α -anomer, a dry mixture of 14 and NaNO₂ was added to a cold solution of dilute AcOH to give an 89% yield of anomerically pure β -20. An examination of the ¹H NMR spectra of α -20 and β -20 confirmed their identity. The observed difference in the chemical shifts ($\Delta\delta$) in Me₂SO-*d*₆ of the isopropylidene methyl groups is δ 0.10 for α -20 and δ 0.19 for β -20, which is consistent with the values established for isopropylidene derivatives of D-ribonucleosides.²⁴ The coupling constant of β -20 is <2.0 Hz, indicative of a β -ribonucleoside.²⁵ In addition, the H₁ chemical shift of β -20 is upfield from the

same bands of the α -anomer α -**20** in keeping with published observations.^{26,27} The acetyl and isopropylidene blocking groups of β -**20** were removed by hydrolysis in a solution of EtOH and 0.5 N H₂SO₄. Under optimum conditions, the hydrolysis gave a 48% yield of analytically pure deblocked nucleoside **19**, a 15% yield of crude isopropylidene nucleoside **18**, and a 10% yield of crude 8-azaguanosine (**7c**). The 8-azaguanosine was identified by comparison with an authentic sample,¹⁴ and **18** was identified by a uv spectrum similar to **19**, a molecular ion of m/e 430, and a negative metaperiodate-Schiff test. Attempts to improve the yield of **19** by increasing or decreasing the reaction time gave higher yields of **7c** and **18**, respectively, with a resultant decrease in the yield of **19**.



The benzythio group of **19** could be easily displaced with a variety of nucleophiles at room temperature. Treatment of **19** with NaOMe in MeOH gave a 72% yield of 7-methoxy-3- β -D-ribofuranosyl-3H-1,2,3-triazolo[4,5-d]pyrimidin-5-amine (**8c**). In a similar reaction with MeSH in MeOH containing NaOMe, **19** gave a 75% yield of 7-(methylthio)-3- β -D-ribofuranosyl-3H-1,2,3-triazolo[4,5-d]pyrimidin-5-amine (**15**). The butylamino analogue **16** was prepared in 86% yield by treatment of **19** with excess *n*-butylamine. The reaction of **19** with NaSH in ethanol gave a precipitate of 8-aza-6-thioguanosine (**17**) as the sodium salt. An aqueous solution of this salt was acidified

to pH 6 to give a precipitate of pure **17** in 66% yield. The structure of **17** was confirmed by comparison of its uv spectrum with that of 8-aza-9-ethyl-6-thioguanine (**22**).¹⁸ A solution of **17** in water under N₂ rearranged slowly to give a 73% yield of *N*⁷- β -D-ribofuranosyl[1,2,3]thiadiazolo[5,4-*d*]pyrimidine-5,7-diamine (**21**). A ¹H NMR spectrum of **21** showed only one anomer which is assumed to be β since anomerization would be expected to give a mixture. The peak for H₁ is the predicted multiplet which forms a doublet upon exchange of the ribosyl NH with D₂O. Other evidence for this structure is its elemental analysis, a molecular ion of m/e 300 in its mass spectrum, and a uv spectrum similar to the parent [1,2,3]thiadiazolo[5,4-*d*]pyrimidine-5,7-diamine (**24a**), which was prepared by nitrosation of 2,5,6-triamino-4(3*H*)-pyrimidinethione¹⁰ (**23a**) with NaNO₂ in 0.5 N HCl. This procedure has been used for the synthesis of [1,2,3]thiadiazolo[5,4-*d*]pyrimidin-7-amine (**24b**) from 5,6-diamino-4(3*H*)-pyrimidinethione²⁸ (**23b**). Structural assignment of **24a** is based on elemental analysis, NMR data, insolubility in base, a molecular ion of m/e 168, absence of a molecular ion of m/e 334 for the disulfide of **23a** (field desorption mass spectrum), and negligible difference in the uv absorption at pH 7 and 13, indicative of the lack of an acidic proton on the chromophore.



The reversible rearrangement of **24b** to the thione **25b** reported by Albert²⁸ also occurs with **24a**. Treatment of **24a** with hot base gave an 87% yield of the thione **25a**. Evidence for the structure of **25a** is based on elemental analysis, solubility in base, a uv spectrum similar to **17** and **22** at pH 1, a molecular ion of m/e 168, and the expected ¹H NMR absorption. This compound has essentially the same physical properties as those described for a sample of **25a** prepared from 8-azaguanine.⁷ The reverse rearrangement was demonstrated by heating dry **25a** at 165° to give a low yield of **24a**.

Both ¹H and ¹³C NMR spectra support the structures assigned to **24a** and **25a**. Assignments for the ¹³C NMR spectra were made by comparison with 2,6-diaminopurine, 6-mercaptopurine, and related compounds and by intensity changes observed on addition of D₂O.²⁹ The ¹H NMR spectrum of **24a** shows two absorption peaks of about the same intensity and shape, attributed to the two NH₂ groups. In contrast, the spectrum of **25a** shows a sharp absorption peak assigned to the NH₂ group and very broad absorption assigned to the two NH groups.

Biologic Evaluation. Compounds **8a-c**, **15-19**, **24a**, and **25a** were evaluated in the H.Ep.-2 cell culture screen³¹ and compounds **8c**, **15-19**, **24a**, and **25a** in the L1210 mouse leukemia screen³⁰ (see Table II). Only compounds **8c** and **25a** were highly cytotoxic. Compound **8c** proved to be a good substrate for adenosine deaminase indicating that its cytotoxicity is a result of its conversion to 8-azaguanosine (**7c**). In contrast to **8c**, the positional isomers **8a** and **8b** show a very low order of toxicity. The low toxicity of the remaining 8-azaguanosine analogues in Table II is also contrasted with the high toxicity of the corresponding 8-azainosine analogues.¹ The lack of L1210 activity and low toxicity of the thio and alkylthio analogues **15**, **17**, and **19** indicate that simple hydrolysis to **7c** or **25a**

Table II. Cytotoxicity and L1210 Screening Data

Compd no.	Cytotoxicity to H.Ep.-2 cells in culture, ED ₅₀ , μM ^a	L1210 ^b	
		Dose, mg/kg	% ILS ^c
7c	2	25	32 ^d
8a	>70		
8b	>70		
8c	2	100	29
15	>60	400	12
16	>60	400	7
17	>100	200	4
18	>50	400	1
19	>50	400	11
24a	>100	400	0
25a	4	100	0

^a The concentration required to inhibit the growth of cells to 50% of controls. See ref 31. ^b L1210 leukemic cells (10⁵) were implanted intraperitoneally in mice and treated with a single dose of compound on day 1. See ref 30. ^c Increase in life span at optimum dose. ^d See ref 32.

does not occur in cell culture or in vivo and that the presence of the amino group in 15 prevents it from functioning as a substrate for adenosine kinase, the enzyme that activates 8-aza-6-(methylthio)purine ribonucleoside.¹ In contrast, the presence of the amino group of 8c does not interfere with its ability to function as a substrate for adenosine deaminase (see above).

Of the compounds tested against L1210 leukemia in mice, only 8c showed borderline activity.

Experimental Section

Melting points were determined on a Kofler Heizbank or, when indicated, on a Mel-Temp apparatus. Absence of melting point data indicates an indefinite melting point. The ultraviolet absorption spectra were determined with a Cary Model 17 spectrophotometer. Each compound was dissolved in the solvent indicated in parentheses and diluted tenfold with 0.1 N HCl, pH 7 buffer, and 0.1 N NaOH. The ¹H NMR spectra were determined in 3–7% w/v solutions in Me₂SO-*d*₆ with a Varian XL-100-15 spectrometer operating at 100 MHz or T-60A operating at 60 MHz (internal Me₄Si). The relative peak areas are given to the nearest whole number, and chemical shifts quoted in the case of multiples are measured from the approximate center. The ¹³C NMR spectra were determined in 12% w/v solutions in Me₂SO-*d*₆ with a Varian XL-100-15 spectrophotometer operating at 25.160 MHz and equipped with a Digilab FTS NMR-3 pulser and data system. Mass spectral data were taken with a Varian MAT 311A instrument equipped with a combination E1/F1/FD ion source. Where analyses are indicated only by symbols of the elements, analytical results obtained for those elements were within ±0.4% of the theoretical values.

6-(Methylthio)-2,4,5-pyrimidinetriamine (2). A mixture of 2,5,6-triamino-4(3*H*)-pyrimidinethione¹⁰ (1, 8.40 g, 36.5 mmol), 2 N NaOH (56.5 ml), and methyl iodide (2.39 ml, 38.4 mmol) was stirred vigorously for 15 min at 0° and 45 min at 25°. The reaction mixture was adjusted to pH 8.6 with 6 N HCl and cooled in an ice bath to give a red precipitate which was collected, washed with cold H₂O, and dried in vacuo (P₂O₅): yield 4.52 g (72%); mp 190° (lit.¹¹ mp 191–192°).

***N*-Acetyl-7-(methylthio)-3*H*-1,2,3-triazolo[4,5-*d*]pyrimidin-5-amine (5).** A solution of 3¹² (2.50 g, 13.8 mmol), prepared in 90% yield from 2, was heated in refluxing acetic anhydride (28 ml) for 2 h, cooled to 25°, and diluted with Et₂O (60 ml). The crystalline diacetyl derivative 6 was collected, washed with Et₂O, and dried in vacuo; yield 3.26 g (89%); mp ~240° dec. Anal. (C₉H₁₀N₆O₂S) C, H, N. The diacetyl derivative (3.26 g, 12.3 mmol) was added in small portions to a stirred 1 N NaOH solution (24.5 ml) and the resulting mixture stirred for 20 h, filtered, treated with 1 N HCl (24.5 ml), and cooled in an ice bath. The crystalline product was collected, washed with H₂O, and dried in vacuo

(P₂O₅): yield 2.69 g (98%); mp 251° dec; λ_{max}, nm (ε × 10⁻³) (EtOH), (0.1 N HCl) 227 (13.4), 252 (16.2), 277 (10.8), 302 (13.6); λ_{max} (pH 7) 238 (21.4), 307 (12.3); λ_{max} (0.1 N NaOH) 237 (18.1), 308 (11.3). Anal. (C₇H₈N₆OS) C, H, N.

Nucleosides from the Reaction of 5 with 2,3,5-Tri-*O*-acetyl-*D*-ribofuranosyl Chloride. A mixture of 2,3,5-tri-*O*-acetyl-*D*-ribofuranosyl chloride³³ (8.6 mmol), 5 (1.81 g, 8.08 mmol), Linde 4A molecular sieve¹³ (40 g), and 1,2-dichloroethane (450 ml) was stirred at 65° for 5 days. The mixture was cooled at 25°, filtered, and evaporated to dryness in vacuo. A solution of the residue in benzene (50 ml) was filtered and evaporated to give 3.25 g of mixed nucleosides. A solution of the mixture in a minimum of CHCl₃ was absorbed on a 28-mm diameter column of E. Merck silica gel 60 (230–400 mesh, 150 g) equilibrated with CHCl₃ and developed with CHCl₃-MeOH (99:1). The faster moving fraction was evaporated to a foam (3.25 g) consisting of a mixture of *N*-acetyl-7-(methylthio)-2-(2,3,5-tri-*O*-acetyl)-β-*D*-ribofuranosyl-2*H*-1,2,3-triazolo[4,5-*d*]pyrimidin-5-amine (4b) and the corresponding 3-substituted isomer 4c. A slower moving fraction was evaporated to a foam (470 mg) which was identified as the crude 1-substituted isomer 4a: λ_{max}, nm (EtOH), (0.1 N HCl and pH 7) 248, 323.

5-Amino-1,6-dihydro-1-β-*D*-ribofuranosyl-7*H*-1,2,3-triazolo[4,5-*d*]pyrimidin-7-one (7a). A solution of crude 4a (100 mg) in AcOH (1 ml) was stirred with 30% H₂O₂ (0.12 ml) for 18 h and evaporated to a gum in vacuo. A solution of the residue in CHCl₃ was washed with H₂O and evaporated to dryness, and the residue was stirred with NaOMe (78 mg) in MeOH (4 ml) for 18 h. The solution was neutralized with Amberlite IR 120 cation-exchange resin and evaporated to give 27 mg of crude 7a: λ_{max}, see Table I.

8-Azaguanosine (7c) and 5-Amino-2,6-dihydro-2-β-*D*-ribofuranosyl-7*H*-1,2,3-triazolo[4,5-*d*]pyrimidin-7-one (7b). The mixture of 4b and 4c (100 mg) was oxidized with H₂O₂ and hydrolyzed as above to give a mixture of 7b and 7c (55 mg). Extraction of the mixture with H₂O (2 ml) left crude 7b as an insoluble white powder (13 mg): λ_{max}, see Table I. The extract was evaporated and the residue again extracted with water (1 ml) and evaporated in vacuo to give 24 mg of crude 7c. A TLC and uv spectrum of this material was identical with that of previously prepared 7c.¹⁴

7-Methoxy-1-β-*D*-ribofuranosyl-1*H*-1,2,3-triazolo[4,5-*d*]pyrimidin-5-amine (8a). A solution of crude 4a (200 mg) in MeOH (6 ml) containing NaOMe (78.3 mg, 1.45 mmol) was stirred for 18 h and a white precipitate of 8a was collected by filtration, washed with MeOH, and dried in vacuo (P₂O₅): yield 51 mg; mp ~173°. Additional product, 24 mg, mp ~173°, was obtained from the filtrate by neutralization with Amberlite IR-120 cation-exchange resin, evaporation, and crystallization from EtOH: λ_{max}, nm (ε × 10⁻³) (EtOH), (0.1 N HCl) 215 (sh) (18.9), 245 (sh) (7.23), 294 (6.65); λ_{max} (pH 7) 211 (24.7), 313 (4.53); λ_{max} (0.1 N NaOH) 310 (4.76). Anal. (C₁₀H₁₄N₆O₅·0.5H₂O) C, H, N.

7-Methoxy-2-β-*D*-ribofuranosyl-2*H*-1,2,3-triazolo[4,5-*d*]pyrimidin-5-amine (8b). A solution of the mixture of 4b and 4c (338 mg) in MeOH (10 ml) containing NaOMe (132 mg, 2.45 mmol) was stirred for 18 h, neutralized with IR-120 cation-exchange resin, and evaporated to dryness. The product mixture was separated by chromatography on a 2-mm Merck silica gel 60 F-254 preparative TLC plate using CHCl₃-MeOH (5:1) as the developing solvent. The faster moving band was extracted with MeOH to give 29 mg of crude 8c; this compound was found to be identical (TLC, uv) with the analytical sample described below. The slower moving band was extracted with MeOH to give 74 mg of crude 8b which was recrystallized from EtOH to give 18 mg of pure product: mp 190°; λ_{max}, nm (ε × 10⁻³) (H₂O), (0.1 N HCl) 204 (22.9), 243 (6.30), 292 (13.3); λ_{max} (pH 7) 215 (3.13), 311 (8.33); λ_{max} (0.1 N NaOH) 216 (2.67), 255 (sh) (4.41), 307 (8.57). Anal. (C₁₀H₁₄N₆O₅) C, H, N.

7-Methoxy-3-β-*D*-ribofuranosyl-3*H*-1,2,3-triazolo[4,5-*d*]pyrimidin-5-amine (8c). A solution of 19 (1.00 g, 2.56 mmol) and NaOMe (415 mg, 7.68 mmol) in anhydrous MeOH (200 ml) was stirred for 4 days exposed to air through a drying tube containing soda lime and Drierite. The solution was neutralized with Amberlite IR-120 cation-exchange resin and filtered, and the resin was washed thoroughly with MeOH and H₂O. The combined filtrate and wash was lyophilized and the residue washed

with Et₂O to remove benzyl disulfide. A solution of the dried residue in water (200 ml) was washed with Et₂O and lyophilized. The residue (650 mg) was crystallized from hot H₂O (charcoal, Celite) to give white needles which were dried at 100° in vacuo: yield 546 mg (72%); mp 209°; λ_{\max} , nm ($\epsilon \times 10^{-3}$) (H₂O), (0.1 N HCl) 213 (22.0), 244 (5.10), 283 (10.7); λ_{\max} (pH 7) 214 (22.2), 246 (5.36), 287 (10.6); λ_{\max} (0.1 N NaOH) 244 (5.40), 287 (10.7). Anal. (C₁₀H₁₄N₆O₅) C, H, N.

N²-Acetyl-6-(benzylthio)-N⁴-(2,3-O-isopropylidene)- β -D-ribofuranosyl-5-nitro-2,4-pyrimidinediamine (13). A suspension of *N*-acetyl-4-(benzylthio)-6-chloro-5-nitro-2-pyrimidineamine²³ (10, 5.00 g, 14.8 mmol), 2,3-*O*-isopropylidene- β -D-ribofuranosylamine *p*-toluenesulfonate²² (11, 10.6 g, 29.5 mmol), and NaHCO₃ (2.48 g, 29.5 mmol) in EtOH (75 ml) was stirred in an oil bath at 59° for 17 h, cooled to 25°, and stirred for 1 h. The white precipitate was collected by filtration, washed with a small quantity of EtOH, dried in vacuo, stirred with H₂O (50 ml) for 1 h, collected, and dried in vacuo (P₂O₅): yield 3.20 g (44%); mp 192°; λ_{\max} , nm ($\epsilon \times 10^{-3}$) (EtOH), (0.1 N HCl) 219 (23.4), 261 (27.2), 352 (13.5); λ_{\max} (0.1 N NaOH) 260 (15.7), 357 (17.4). Anal. (C₂₁H₂₅N₅O₇S) C, H, N. Substitution of Et₃N for NaHCO₃ in this reaction gave, after fractional crystallization from EtOH, a 30% yield of 13 and a 12% yield of N²-acetyl-6-(benzylthio)-5-nitro-2,4-pyrimidinediamine (12): λ_{\max} , nm ($\epsilon \times 10^{-3}$) (EtOH), (0.1 N HCl) 259 (20.3), 345 (12.9); λ_{\max} (pH 7) 215 (20.6), 256 (20.6), 348 (13.4); λ_{\max} (0.1 N NaOH) 256 (12.6), 351 (17.2). Anal. (C₁₃H₁₃N₅O₃S) C, H, N.

N²-Acetyl-6-(benzylthio)-N⁴-(2,3-O-isopropylidene)- β -D-ribofuranosyl-2,4,5-pyrimidinetriamine (14). A solution of 13 (3.65 g, 7.42 mmol) in EtOH (600 ml) was hydrogenated at atmospheric pressure in the presence of 5% Pd on charcoal (5.5 g) for 6 h. The solution, which took up 700 ml (28.8 mmol) of H₂, was warmed to 80° on a water bath and filtered and the catalyst extracted with boiling EtOH (500 ml). The combined filtrate and extract was evaporated in vacuo leaving a residue of white solid which was used for the preparation of β -20: yield 2.81 g (82%); mp 171°. Recrystallization of a portion from EtOH gave an analytical sample: mp 175°; λ_{\max} , nm ($\epsilon \times 10^{-3}$) (EtOH), (0.1 N HCl) 218 (18.6), 243 (21.1), 290 (8.00), 330 (9.00); λ_{\max} (pH 7) 213 (26.5), 273 (br) (9.57), 310 (11.2); λ_{\max} (0.1 N NaOH) 310 (10.7). Anal. (C₂₁H₂₇N₅O₅S) C, H, N.

7-(Methylthio)-3- β -D-ribofuranosyl-3*H*-1,2,3-triazolo[4,5-*d*]pyrimidin-5-amine (15). A solution of NaOMe (138 mg, 2.56 mmol) in MeOH (100 ml) was saturated with MeSH, treated with 19 (500 mg, 1.28 mmol), allowed to stand in a stoppered flask for 5 days, and diluted dropwise with H₂O (100 ml). The solution was neutralized with Amberlite IR-120 cation-exchange resin, filtered, and lyophilized. The residual powder was triturated with Et₂O (3 \times 25 ml), dried in vacuo, and recrystallized from hot H₂O (5 ml) to give white crystals, which were dried at 56° in vacuo (P₂O₅): yield 302 mg (75%); mp 209°; λ_{\max} , nm ($\epsilon \times 10^{-3}$) (H₂O), (0.1 N HCl) 221 (14.2), 247 (9.72), 279 (9.19), 315 (12.5); λ_{\max} (pH 7) 221 (14.6), 248 (10.6), 277 (8.64), 316 (12.6); λ_{\max} (0.1 N NaOH) 221 (13.2), 244 (10.6), 277 (8.64), 316 (12.6). Anal. (C₁₀H₁₄N₆O₄S) C, H, N.

N⁷-Butyl-3- β -D-ribofuranosyl-3*H*-1,2,3-triazolo[4,5-*d*]pyrimidine-5,7-diamine (16). A solution of 19 (100 mg, 0.256 mmol) and *n*-butylamine (0.5 ml) in MeOH (0.5 ml) was allowed to stand at 25° for 4 days and evaporated in vacuo. The residue was triturated with Et₂O (2 \times 2 ml), dried, and triturated with CHCl₃ (1 ml). The resulting white crystalline precipitate was collected, washed with CHCl₃, and dried at 56° in vacuo (P₂O₅): yield 75 mg (86%); mp 135°; λ_{\max} , nm ($\epsilon \times 10^{-3}$) (H₂O), (0.1 N HCl) 213 (21.7), 261 (13.2), 285 (10.7); λ_{\max} (pH 7) 233 (20.4), 291 (14.1); λ_{\max} (0.1 N NaOH) 232 (21.2), 291 (14.3). Anal. (C₁₃H₂₁N₇O₄) C, H, N.

8-Aza-6-thioguanosine [5-Amino-3,6-dihydro-3- β -D-ribofuranosyl-7*H*-1,2,3-triazolo[4,5-*d*]pyrimidine-7-thione Hydrate (17)]. A suspension of powdered Fisher purified sodium sulfhydrate (1.00 g) and 19 (500 mg, 1.28 mmol) in EtOH (20 ml) was stirred in a stoppered flask for 2 days. The resulting sodium salt of 17 was collected by filtration under N₂, washed with a small quantity of cold EtOH, and dried in vacuo: yield 572 mg. A portion of the salt (100 mg) in 2 ml of H₂O was filtered, acidified with 1 N HCl to pH 6, and cooled in an ice bath. The white precipitate of 17 was collected by filtration, washed with cold H₂O,

and dried in vacuo (P₂O₅): yield 47 mg (66%); mp 139–140° (Mel-Temp); λ_{\max} , nm ($\epsilon \times 10^{-3}$) (CH₃OH), (0.1 N HCl) 211 (21.5), 230 (sh) (11.8), 263 (8.90), 338 (20.3); λ_{\max} (pH 7) 212 (18.0), 258 (10.4), 334 (17.6); λ_{\max} (0.1 N NaOH) 257 (10.9), 292 (7.82), 328 (17.1). Anal. (C₉H₁₂N₆O₄S·H₂O) C, H, N.

7-(Benzylthio)-3-(2,3-*O*-isopropylidene)- β -D-ribofuranosyl-3*H*-1,2,3-triazolo[4,5-*d*]pyrimidin-5-amine (18) and 7-(Benzylthio)-3- β -D-ribofuranosyl-3*H*-1,2,3-triazolo[4,5-*d*]pyrimidin-5-amine (19). A solution of β -20 (500 mg, 1.06 mmol) in EtOH (20 ml) and 0.5 N H₂SO₄ (5 ml) was stirred in an oil bath at 44° for 41 h, cooled to 25°, neutralized with 4 N NaOH to pH 7, filtered, and evaporated to dryness in vacuo. The residue was washed with benzene (2 \times 5 ml), dried, and extracted with EtOH. This extract was filtered, concentrated in vacuo, and chromatographed on four 2-mm Merck silica gel 60F-254 preparative TLC plates using 9:1 CHCl₃-MeOH as the developer. A faster moving minor band was extracted with MeOH to give 69 mg of crude 18: λ_{\max} , nm (EtOH), (0.1 N HCl) 249, 280, 318; λ_{\max} (pH 7) 250, 279, 318. A nonmoving minor band was extracted with MeOH to give 30 mg of crude 8-azaguanosine (7c), which was identical (TLC and uv) with a previously prepared sample.¹⁴ The major band was extracted with MeOH to give 291 mg of crude 19, which was recrystallized from hot 5:1 H₂O-EtOH (30 ml): yield 199 mg (48%); mp 162°; λ_{\max} , nm ($\epsilon \times 10^{-3}$) (EtOH), (0.1 N HCl) 216 (19.0), 249 (9.36), 280 (9.14), 317 (13.6); λ_{\max} (pH 7) 216 (19.3), 249 (9.82), 278 (9.00), 317 (13.6); ¹H NMR δ 3.54 (m, H₅), 3.97 (m, 1, H₄), 4.27 (m, 1, H₃), 4.63 (s, SCH₂), 4.75 (m, H₂), 5.17 (d, 1, O₃H), 5.52 (d, 1, O₂H), 6.00 (d, 1, J_{1,2} = 5.0 Hz, H₁), 7.4 (m, 7, NH₂, C₆H₅). Anal. (C₁₆H₁₈N₆O₄S) C, H, N.

***N*-Acetyl-7-(benzylthio)-3-(2,3-*O*-isopropylidene)- β -D-ribofuranosyl-3*H*-1,2,3-triazolo[4,5-*d*]pyrimidin-5-amine (β -20).** A mixture of 14 (2.81 g, 6.09 mmol) and NaNO₂ (740 mg, 10.7 mmol) in a 50-ml flask was cooled in an ice bath and treated rapidly with a cold (0°) solution of AcOH (28 ml) and H₂O (2.8 ml). The resulting mixture, after stirring for 30 min at 0° and 30 min at 25°, was poured into ice water (280 ml) and stirred for 5 min. The white precipitate was collected, washed with cold H₂O, and dried in vacuo (P₂O₅): yield 2.57 g (89%); melting point indefinite; λ_{\max} , nm ($\epsilon \times 10^{-3}$) (EtOH), (0.1 N HCl) 231 (14.5), 257 (16.7), 283 (sh) (11.3), 307 (16.6); λ_{\max} (pH 7) 231 (14.6), 257 (16.9), 283 (sh) (11.3), 307 (16.7); ¹H NMR δ 1.36, 1.55 (d, 6, CM₂), 2.31 (s, 3, COCH₃), 4.24 (m, 1, H₄), 5.17 (m, 1, H₃), 5.62 (m, 1, H₂), 6.41 (d, 1, J_{1,2} = <2.0 Hz, H₁), 7.42 (m, 5, C₆H₅). Anal. (C₂₁H₂₄N₆O₅S) C, H, N.

In a preliminary experiment (see discussion) a small quantity of the α -anomer (α -20) was isolated by separation of a 4:1 mixture of β -20 and α -20 on a 2-mm Merck silica gel 60 F-254 preparative TLC plate using CHCl₃-MeOH (97:3) as developing solvent: λ_{\max} , nm (EtOH), (0.1 N HCl and pH 7) 233, 256, 283 (sh), 306; ¹H NMR δ 1.11, 1.21 (d, 6, CM₂), 2.32 (s, 3, COCH₃), 3.66 (d, 2, H₅), 6.71 (d, 1, J_{1,2} = 5.0 Hz, H₁), 7.42 (m, 5, C₆H₅).

N⁷- β -D-ribofuranosyl[1,2,3]thiadiazolo[5,4-*d*]pyrimidine-5,7-diamine (21). A solution of 17 (15 mg) in H₂O (2 ml) was stirred in a stoppered vial under N₂ at ~45° for 5 days. The mixture was cooled in an ice bath and the white precipitate collected and dried in vacuo (P₂O₅): yield 11 mg (73%); mp 247° dec; λ_{\max} , nm ($\epsilon \times 10^{-3}$) (Me₂SO-H₂O, 2:23), (0.1 N HCl) 248 (21.8), 275 (11.1), 285 (sh) (10.6); λ_{\max} (pH 7) 247 (14.9), 311 (11.6); λ_{\max} (0.1 N NaOH) 248 (14.3), 313 (11.2); ¹H NMR δ 3.47 (m, 2, H₅), 3.75 (m, 1, H₄), 4.07 (m, 2, H_{2,3}), 4.73 (m, 1, O₅H), 4.88 (d, 1, O₂ or ₃H), 5.07 (d, 1, O₂ or ₃H), 5.89 (m, 1, H₁), 7.22 (s, 2, NH₂), 9.06 (d, 1, NH). Anal. (C₉H₁₂N₆O₄S) C, H, N.

[1,2,3]Thiadiazolo[5,4-*d*]pyrimidine-5,7-diamine (24a). (a) A solution of 23a¹⁰ (10.0 g, 43.4 mmol) in hot 0.5 N HCl (700 ml) was cooled to 15–20° and treated dropwise with a solution of NaNO₂ (4.49 g, 65.2 mmol) in H₂O (20 ml). The solution was stirred at 25° for 5 h, filtered, neutralized with 50% NaOH, and cooled in an ice bath. The precipitate was collected, washed with H₂O, and stirred with 1 N NaOH (100 ml) for 30 min. The crude product was collected, washed with H₂O, dried, triturated with 1 N NaOH (20 ml), collected, washed with H₂O, and dried in vacuo (P₂O₅): yield 2.77 g (38%); mp 256° dec; λ_{\max} , nm ($\epsilon \times 10^{-3}$) (Me₂SO-MeOH, 2:23), (0.1 N HCl) 235 (17.0), 274 (sh) (8.44), 283 (8.62); λ_{\max} (pH 7) 243 (12.4), 310 (10.1); λ_{\max} (0.1 N NaOH) 243 (12.5), 310 (10.1); ¹H NMR δ 6.98 (s, 2, NH₂), 8.03 (s, 2, NH₂); ¹³C NMR δ (± 0.02) 162.54 (C₅), 172.48 (C_{3a}), 133.21 (C_{7a}), 158.00

(C₇). Anal. (C₄H₄N₆S) C, H, N.

(b) Powdered **25a** (20 mg) in a stoppered vial under N₂ was heated in an oil bath at 165° for 3 h. The resulting solid was triturated with 1 N NaOH (1 ml), collected, washed with H₂O, and dried in vacuo (P₂O₅): yield 6.1 mg (31%). This product was the same (TLC, uv) as that described in (a).

5-Amino-3,6-dihydro-7H-1,2,3-triazolo[4,5-d]pyrimidine-7-thione (25a). A suspension of **24a** (500 mg, 2.98 mmol) in 1 N NaOH (30 ml) was heated under N₂ at ~98° until complete solution occurred (20 min). The solution was heated an additional 10 min, cooled to 25°, filtered under N₂, and acidified with 6 N HCl to pH 6. The white precipitate was collected, washed with cold H₂O, and dried in vacuo (P₂O₅): yield 434 mg (87%); mp 258° dec (Mel-Temp); λ_{max}, nm (ε × 10⁻³) (Me₂SO-MeOH, 2:23), (0.1 N HCl) 224 (14.1), 258 (7.41), 338 (17.3); λ_{max} (pH 7) 231 (19.5), 257 (5.04), 282 (sh) (34.7), 343 (16.8); λ_{max} (0.1 N NaOH) 224 (19.6), 279 (8.40), 328 (14.9); ¹H NMR δ 6.95 (s, 2, NH₂), 10-16 (2, NH); ¹³C NMR δ (±0.02) 154.02 (C₅), 148.90 (C_{3a}), 132.82 (C_{7a}), 177.62 (C₇). Anal. (C₄H₄N₆S) C, H, N.

Reaction of 8c with Adenosine Deaminase. A solution of 22 units of a crystalline suspension of adenosine deaminase from calf intestinal mucosa in 0.01 ml of 3.2 M (NH₄)₂SO₄ (Sigma Chemical Co.) was added to a solution of **8c** (4.1 mg) in 50 ml of 0.05 M phosphate buffer at pH 7.5. A uv spectrum of this solution after 16 min showed complete conversion to **7c**.

Acknowledgment. This investigation was supported by the Division of Cancer Treatment, National Cancer Institute, National Institutes of Health, Department of Health, Education and Welfare, Contract No. NO1-CM-43762. The authors are indebted to Dr. W. C. Coburn, Jr., and Mrs. M. C. Thorpe who interpreted NMR data and to other members of the Molecular Spectroscopy Section of Southern Research Institute, who performed most of the microanalytical and spectral determinations reported. The authors also thank Dr. Adrian Albert for samples of 7- and 8-methyl-8-azaguanine and to Dr. Leroy Townsend for a sample of 3-methyl-8-azaguanine for uv comparisons.

References and Notes

- (1) J. A. Montgomery, R. D. Elliott, and H. J. Thomas, *Ann. N.Y. Acad. Sci.*, **255**, 292 (1975).
- (2) L. L. Bennett, Jr., M. H. Vail, P. W. Allan, and W. R. Laster, Jr., *Cancer Res.*, **33**, 465 (1973).
- (3) A. R. P. Paterson and D. M. Tiddin in "Handbook of Experimental Pharmacology", Vol. 28, A. C. Sartorelli and D. G. Johns, Ed., Springer-Verlag, Berlin, 1975, p 304.
- (4) R. B. Livingston and S. K. Carter, "Single Agents in Cancer Chemotherapy", IFI/Plenum, New York, N.Y., 1970, p 173.
- (5) L. L. Bennett, Jr., and J. A. Montgomery, *Methods Cancer Res.*, **3**, 549 (1967).

- (6) G. W. Kidder, V. C. Dewey, R. E. Parks, and G. L. Woodside, *Science*, **109**, 511 (1949).
- (7) C. T. Bahner, D. E. Bilanco, and E. M. Brown, *J. Am. Chem. Soc.*, **76**, 1370 (1954).
- (8) J. A. Montgomery and R. D. Elliott, *J. Chem. Soc.*, 1279 (1972).
- (9) W. Huttenlaub, R. L. Tolman, and R. K. Robins, *J. Med. Chem.*, **15**, 879 (1972).
- (10) J. J. McCormack and H. G. Mautner, *J. Org. Chem.*, **29**, 3370 (1964) (thiation method).
- (11) D. Daves, Jr., C. W. Noell, R. K. Robins, H. C. Koppel, and A. G. Beaman, *J. Am. Chem. Soc.*, **82**, 2633 (1960).
- (12) K. Weiss, R. K. Robins, and C. W. Noell, *J. Org. Chem.*, **25**, 765 (1960).
- (13) Pellets (¹/₁₆ in.) can be ordered from Union Carbide Corp., Houston, Texas 77027.
- (14) J. Davoll, *J. Chem. Soc.*, 1593 (1958).
- (15) C. W. Noell, L. B. Townsend, and R. K. Robins, *Synth. Proced. Nucleic Acid Chem.*, **1**, 44 (1968).
- (16) L. B. Townsend and R. K. Robins, *Synth. Proced. Nucleic Acid Chem.*, **1**, 18 (1968).
- (17) A. Albert and H. Taguchi, *J. Chem. Soc., Perkin Trans. 1*, 449 (1972).
- (18) Y. F. Shealy, R. F. Struck, J. D. Clayton, and J. A. Montgomery, *J. Org. Chem.*, **26**, 4433 (1961).
- (19) T. Kishikawa and H. Yuki, *Chem. Pharm. Bull.*, **14**, 1360 (1966).
- (20) W. Pfeleiderer and E. Buhler, *Chem. Ber.*, **99**, 3022 (1966).
- (21) H. Rokos and W. Pfeleiderer, *Chem. Ber.*, **104**, 748 (1971).
- (22) N. J. Cusack, B. J. Hildick, D. H. Robinson, P. W. Rugg, and G. Shaw, *J. Chem. Soc., Perkin Trans. 1*, 1720 (1973).
- (23) R. D. Elliott, C. Temple, Jr., J. Frye, and J. A. Montgomery, *J. Med. Chem.*, **18**, 492 (1975).
- (24) J. -L. Imbach, J. -L. Barascut, B. L. Kam, B. Rayner, C. Tamby, and C. Tapiero, *J. Heterocycl. Chem.*, **10**, 1069 (1973).
- (25) R. U. Lemieux and J. W. Lown, *Can. J. Chem.*, **41**, 889 (1963).
- (26) J. A. Montgomery and H. J. Thomas, *J. Am. Chem. Soc.*, **87**, 5402 (1965).
- (27) T. Nishimura and B. Shimizu, *Chem. Pharm. Bull.*, **13**, 803 (1965).
- (28) A. Albert, *J. Chem. Soc. C*, 152 (1969).
- (29) M. C. Thorpe and W. C. Coburn, Jr., unpublished results.
- (30) R. I. Geran, N. H. Greenberg, M. M. Macdonald, A. M. Schumacher, and B. J. Abbott, *Cancer Chemother. Rep., Part 3*, **3** (no. 2) (1972).
- (31) L. L. Bennett, Jr., M. K. Vail, S. Chumley, and J. A. Montgomery, *Biochem. Pharmacol.*, **15**, 1719 (1966).
- (32) J. A. Montgomery, F. M. Schabel, Jr., and H. E. Skipper, *Cancer Res.*, **22**, 504 (1962).
- (33) J. Davoll, B. Lythgoe, and A. R. Todd, *J. Chem. Soc.*, 967 (1948).

Inhibitors of tRNA Methyltransferases. S-Adenosylsulfonium Salts

Margaret E. Gnegy and Frederick J. Lotspeich*

Department of Biochemistry, West Virginia University Medical Center, Morgantown, West Virginia 26506.

Received September 4, 1975

Three new compounds have been synthesized and tested as in vitro inhibitors of normal and tumor tRNA methyltransferases. These compounds are 5'-methylpropyl(5'-adenosyl)sulfonium chloride (MEAS), 5'-methylpropyl(5'-adenosyl)sulfonium chloride (MPAS), and 5'-ethylpropyl(5'-adenosyl)sulfonium chloride (EPAS). They were prepared by reacting an alkyl iodide with the appropriate alkyladenosyl thioether. Inhibition assays revealed all three compounds to be inhibitors of normal and tumor tRNA methyltransferases. The propyl compounds were slightly better inhibitors of the tumor tRNA methyltransferases. MPAS, EPAS, and MEAS had K_i's of 58.5, 61.7, and 24.5, respectively, for the normal tRNA methyltransferases and 15.3, 13.8, and 44.3, respectively, for the tumor tRNA methyltransferases.

The discovery by Magee and Farber¹ that the chemical carcinogen dimethylnitrosamine would methylate tRNA to a greater extent than DNA led Borek² to postulate that

chemical alkylation might result in aberrant methylation in tumor cells. In a wide variety of tumor and oncogenic systems the tRNA methyltransferase activity is increased