for this compound; Virazole is the ICN Pharmaceuticals, Inc., trademark.

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Synthesis and Anti-DNA Virus Activity of the 5'-Monophosphate and the Cyclic 3',5'-Monophosphate of $9-(\beta$ -D-Xylofuranosyl)guanine

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9- $(\beta$ -D-Xylofuranosyl)guanine (xylo-G) was converted chemically to the 9- $(\beta$ -D-xylofuranosyl)guanine 5'-monophosphate (xylo-GMP) and 9- $(\beta$ -D-xylofuranosyl)guanine cyclic 3',5'-monophosphate (c-xylo-GMP). These compounds were tested against a variety of DNA viruses in tissue culture in parallel with 9- $(\beta$ -D-arabinofuranosyl)adenine (ara-A). This evaluation revealed that xylo-G, xylo-GMP, and c-xylo-GMP were all moderately active but less effective than ara-A. When the four compounds were administered intracerebrally as a treatment for herpes virus, type 1 induced encephalitis in mice, c-xylo-GMP exhibited superior activity to that shown by the other three. When administered intraperitoneally, c-xylo-GMP was found to have a therapeutic index of about 4, which is less than that for ara-A (\sim 30) in the same system.

Nucleosides possessing either D-arabinofuranose¹ or D-xylofuranose² in place of D-ribofuranose moieties have received increasing attention in recent years as antimetabolites. Of particular importance in antiviral and antitumor studies of such nucleoside analogues are 1-(β -Darabinofuranosyl)cytosine (ara-C),³ 9-(β -D-arabinofuranosyl)adenine (ara-A),⁴ and 9-(β -D-xylofuranosyl)adenine.⁵ The observation that most of the nucleoside analogues must be converted to nucleotides^{1,6} before they are biologically active, coupled with the increased water solubility of such nucleotides over the corresponding nucleosides,⁷ has prompted considerable activity toward the synthesis and biological evaluation of phosphorylated compounds of the above class.⁸

The importance of guanine nucleotide metabolism in a variety of microbiological and mammalian systems has been comprehensively reviewed.⁹ Antimetabolites have proved to be unique biochemical tools in probing enzymatic transformations. The biological resistance to purine and pyrimidine antimetabolites is ascribed to high levels of a deaminase^{4b} or lack of enzymatic phosphorylation of the nucleosides.¹⁰ This problem could be overcome by using 5'-monophosphates of the nucleosides. However, the free nucleotides at physiological pH carry two negative

charges and in general¹¹ penetrate the cell as an intact nucleotide in very small amounts. The exogenous adenosine cyclic 3',5'-monophosphate (c-AMP)^{12a} and guanosine cyclic 3',5'-monophosphate (c-GMP)¹³ may exert specific biological effects of c-AMP or c-GMP, respectively, on the cell membrane^{12b} and the metabolic pathways inside the cell. The interesting biological activity reported for 9-(β -D-xylofuranosyl)purines¹⁴ suggested the synthesis of 9-(β -D-xylofuranosyl)guanine 5'-monophosphate and the cyclic 3',5'-monophosphate as potential antiviral agents.



For the synthesis of 9-(β -D-xylofuranosyl)guanine (1), the method reported by Lee and co-workers¹⁵ via the

Table I. Comparative in Vitro Antiviral Activity of Ara-A, Xylo-G, Xylo-GMP, and c-Xylo-GMP

| | Virus ratings ^a | | | Min KB cell |
|------------|----------------------------|--------|-------|----------------|
| Compd | Herpes | Herpes | Vac- | toxic |
| | virus ^b | virus | cinia | dose, |
| | type 1 | type 2 | virus | μg/ml |
| Ara-A | 0.9 | 0.8 | 0.8 | 3.2 |
| Xylo-G | 0.6 | 0.2 | 0.5 | 10.0 |
| Xylo-GMP | 0.6 | c | 0.4 | 3.2 |
| c-Xylo-GMP | 0.5 | 0.4 | 0.5 | 32.0 |

^a The virus rating (VR) was determined by comparing CPE development in drug-treated cells (T) and virus control cells (C). The CPE value (0-4) assigned to T for each drug level was subtracted from that of C, and the differences (C - T) were totaled. If partial toxicity was evident at any drug level, the C - T of that level was divided by 2. The sum of all C - T values was then divided by ten times the number of test cups used per drug level. ^b Minimum inhibitory concentrations for compounds in this experiment are as follows: ara-A, $3.2 \mu g/ml$; xylo-G, $10 \mu g/ml$; xylo-GMP, $32 \mu g/ml$. ^c Not determined.

coupling of the trimethylsilyl derivative of 2-amino-6chloropurine with 2,3,5-tri-O-acetyl-D-xylofuranosyl bromide followed by the conversion of the condensed product, 2-amino-6-chloro-9-(2,3,5-tri-O-acetyl-β-D-xylofuranosyl)purine, with mercaptoethanol in the presence of sodium methoxide in methanol was used. The enzymatic phosphorylation of 1 using nucleoside phosphotransferase (from Pseudomonas trifolii¹⁶) and p-nitrophenyl phosphate as the phosphate donor has been reported¹⁷ to yield 9-(β -D-xylofuranosyl)guanine 5'monophosphate (2), isolated as the ammonium salt. However, the direct chemical phosphorylation^{8c,18} of 1 using phosphoryl chloride in trimethyl phosphate at -10° for 2 h followed by hydrolysis furnished a mixture of 2 and 9-(β -D-xylofuranosyl)guanine cyclic 3',5'-monophosphate (3), which was separated by ion-exchange chromatography (Dowex 1) and isolated in free acid form. The purity of 2 and 3 was assured by the homogeneity in several thin-layer systems and on paper electrophoresis (phosphate buffer, pH 7.2, and borate buffer, pH 9.2).

Antiviral Evaluation. Inhibition of the virus-induced cytopathic effect (CPE) was used as the initial indicator of antiviral activity. CPE was observed in human carcinoma of the nasopharynx (KB) cells after infection with type 1 (HV/1) or type 2 (HV/2) herpes virus or vaccinia virus (VV). In this system, monolayers (18-24 h) of cells were exposed to 320 CCID_{50} of virus and concentrations of each compound ranging in one-half log dilutions from 1000 to 1 μ g/ml were added within 15 min. The degree of CPE inhibition and compound cytotoxicity were observed microscopically after 72 h of incubation at 37° and scored numerically in order to calculate a virus rating (VR) as previously described.¹⁹ Significance of antiviral activity in terms of VR's has been assigned as follows: <0.5, slight or no activity; 0.5–0.9, moderate activity; and \geq 1.0, marked activity. The results of a single experiment in parallel with ara-A are shown in Table I. Of the compounds tested, ara-A possessed the best activity against all viruses with the xylopurines having comparable slight to moderate antiviral activity against HV/1 and VV but none had appreciable activity against HV/2.

The moderate in vitro anti-DNA virus activity observed for this class of synthetic compounds led us to the evaluation of their efficacy as antiviral agents in vivo. Encephalitis was induced in 18–20-g Swiss mice by intracerebral (ic) inoculation of an LD₇₅ of HV/1, strain 123.

Table II. Effect of Intracerebrally Administered Xylo-G, Xylo-GMP, c-Xylo-GMP, or Ara-A on Herpes Virus Type 1 Induced Encephalitis in Mice

| Compd | Dosage, mg/kg Survivors/total | | | | |
|--------------|-------------------------------|-------------------|--|--|--|
| Experiment 1 | | | | | |
| Saline | 0.03° | 6/24 | | | |
| Xylo-G | 5.0 | 4/10 | | | |
| Xylo-GMP | 1.25 | 4/10 | | | |
| c-Xylo-GMP | 1.25 | $9/10^{a}$ | | | |
| Ara-A | 3.2 | 5/9 | | | |
| Experiment 2 | | | | | |
| Saline | - | 3/20 | | | |
| c-Xylo-GMP | 1.25 | 6/10 ^b | | | |
| - | 0.31 | 5/10 ^b | | | |

^a Probability <0.01, Fisher's exact test. ^b Probability <0.05, Fisher's exact test. ^c ml.

 Table III.
 Effect of Intraperitoneally^a Administered

 c-Xylo-GMP on Herpes Virus Type 1 Induced

 Encephalitis in Mice

| Drug dosage, mg/kg/day | Toxicity controls (survivors/total) | Infected (survivors/total) |
|---------------------------|---|-------------------------------|
| Saline | | 2/21 |
| 104 | 4/5 | 2/10 |
| 52 | 5/5 | 5/10 ^b |
| 26 | с | 3/9 ^d |

^a b.i.d. \times 9, starting 4 h postvirus inoculation. ^b Probability (p) < 0.02, Fisher's exact test. ^c Not determined. ^d This dose extended survival by 2.4 days (p < 0.02, t test).

A single treatment with the maximum tolerated dose (MTD) of each drug was administered ic at 6 h postvirus inoculation.^{8d} c-Xylo-GMP increased the number of surviving animals to the greatest extent with xylo-G, xylo-GMP, and ara-A being similar (see Table II, experiment 1). These observations suggest that the cyclic nucleotide could be the active form, since xylo-G and xylo-GMP would have the same potential for conversion to other nucleotides (a di- or triphosphate). The cxylo-GMP activity was further verified in a second experiment with the compound being active at its MTD and MTD/4 (0.31 mg/kg). Thus, in this experiment c-xylo-GMP demonstrated a marked ability to protect mice from HV/1 encephalitic deaths with the compound having a therapeutic index (TI = MTD/minimum inhibitory dose) of ≥ 4 . This activity stimulated an additional experiment to determine the effect of intraperitoneal administration of c-xylo-GMP on the encephalitic deaths. c-Xylo-GMP was found to have a narrow range of effectiveness (Table III). The only dose (52 mg/kg/day) which increased survivors caused a marked weight loss in both toxicity and infected animals. In addition, the lowest dose significantly increased survival time. The TI of about 4, observed for c-xylo-GMP, was much lower than that usually found for ara-A $(\sim 30)^{8e}$ and probably resulted from the greater toxicity. Since c-xylo-GMP was more toxic and comparatively less active than other potential anti-DNA virus agents,^{8c} no further studies have been planned.

Experimental Section

Melting points were taken on a Thomas-Hoover capillary melting point apparatus and are uncorrected. Specific rotations were measured in a 1-dm tube with a Perkin-Elmer Model 141 automatic digital readout polarimeter. Nuclear magnetic resonance (¹H NMR) spectra were recorded at 60 MHz on a Hitachi Perkin-Elmer R-20A spectrometer in Me₂SO-d₆ as well as in D₂O-NaOD using DSS as an internal standard. Ultraviolet spectra (uv, s = shoulder) were recorded on a Cary Model 15 spectrophotometer. Elemental analyses were performed by Galbraith Laboratories, Inc., Knoxville, Tenn., and the results are within $\pm 0.4\%$ of the theoretical values. Thin-layer chromatography was run on a silica gel F-254 (EM Reagents) plates and developed with either solvent system: A, 2-propanol-concentrated ammonium hydroxide-water (7:1:2, v/v); B, 2propanol-concentrated ammonium hydroxide-water (5:5:1.0:35, v/v); C, 1-butanol-acetic acid-water (5:2:3, v/v); and D, acetonitrile-0.2 M aqueous ammonium chloride (7:3, v/v). Evaporations were carried out under reduced pressure with bath temperature below 30°.

9-(β -D-Xylofuranosyl)guanine 5'-Monophosphate (2) and 9-(\$-D-Xylofuranosyl)guanine Cyclic 3',5'-Monophosphate (3). To a solution of phosphorus oxychloride (2.9 ml) in freshly distilled trimethyl phosphate (20.0 ml) cooled in an ice-salt bath (-10°) was added 9- $(\beta$ -D-xylofuranosyl)guanine¹⁵ (1, 3.0 g, 0.0105 mol, dried at 80° for 15 h over P₂O₅ under vacuum). The contents of the stoppered flask was stirred at -10°. Within 25 min a clear solution was obtained. Stirring was continued for 2 h before the resulting, slightly orange-colored solution was poured into icewater (150 ml) containing sodium hydrogen carbonate (2.5 g) with stirring and external cooling. The mixture was occasionally stirred in an ice bath for 1 h, and the pH was monitored at 5-6 by adding solid sodium hydrogen carbonate when needed. The pH stabilized solution was extracted with ether $(3 \times 50 \text{ ml})$ and the aqueous phase was concentrated in vacuo until salts began to crystallize. Enough water was added to achieve solution; the pH was adjusted to 6-7, before it was applied to a column containing Dowex 1 X2 (100-200 mesh, formate form, 100 ml). The resin was washed with water (31.) to remove unreacted 1 and the inorganic salts before using gradient elution (0.5 M formic acid to water). The eluent containing the product was pooled, frozen, and lyophilized to yield 2.95 g of a cream-colored solid which was found to be a mixture of 2 and 3.

The above mixture was dissolved in water (25 ml) and the pH was checked (6-7) before it was rechromatographed on a freshly generated Dowex 1 X2 resin column (100-200 mesh, formate form, 100 ml). The column was first washed with water (2 l.) before the nucleotides were eluted using a gradient (0.25 M formic acid to water). The first fraction was pooled and concentrated to ~ 15 ml. Ethanol (50 ml) was added and refrigerated overnight. The white solid that separated was collected, washed with ethanol, and dried over P2O5. It was crystallized from aqueous ethanol as needles to yield 2.45 g (63.7%) of 9-(β -D-xylofuranosyl)guanine cyclic 3',5'-monophosphate (3), mp >235° dec, which was homogeneous on paper electrophoresis and on paper chromatography (solvent Å): $[\alpha]^{25}D - 48.5^{\circ}$ (c 1.0, water); ¹H NMR (D₂O) δ 7.98 (s, C₈H), 5.99 (s, C₁H); uv λ max (pH 1) 255 nm (ϵ 13460), 275 s (9400); λ max (pH 7) 252 nm (ϵ 15140), 269 s (11100); λ max (pH 11) 256-268 nm (\$\epsilon 12800). Anal. (C10H12N5O7P.H2O, 363.22) C, H, N.

The second fraction containing the homogeneous product was pooled and concentrated to ~3 ml. Ethanol (25 ml) was added and refrigerated overnight. The white amorphous solid that separated was collected, washed with ethanol, and dried over P₂O₅ at 60° to yield 0.35 g (8.0%) of 9-(β -D-xylofuranosyl)guanine 5'-monophosphate (2), mp >250° dec, which was homogeneous on paper chromatography (solvent A): [α]²⁵D -40.3° (c 1.0, water); ¹H NMR (Me₂SO-d₆-D₂O) δ 7.90 (s, C₈H), 5.85 (d, J = 2.5 Hz, C₁H); uv λ max (pH 1) 255 nm (ϵ 12700), 274 s (8600); λ max (pH 7) 252 nm (ϵ 13800), 270 s (9700); λ max (pH 11) 256-267 nm (ϵ 11960). Anal. (C₁₀H₁₄N₅O₈P·3H₂O, 417.26) C, H, N.

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