2,3,4,4a,5,9b-Hexahydro-lH-pyrido[4,3-b]indole Journal of Medicinal Chemistry, 1979, Vol. 22, No. 6 **677**

The number of animals alive at the end of 48 h was recorded and the LD_{50} value estimated according to the method of Litchfield and Wilcoxon.⁹ If none were dead after 48 h at 300 mg/kg, the LD_{50} value was estimated to be greater than 300 mg/kg and no further dosing was required. If numerous deaths occurred at 300 mg/kg, the compound was dosed at 100 mg/kg and the LD_{50} value estimated from the number of deaths at that dose.

Dose-Response SHR Assay. The procedures for the dose-response SHR assay were identical to the primary level procedure, except rats were dosed at 100, 30, 10, and 3 mg/kg. Five rats were used for each dose, and heart rate and blood pressure were recorded.

References and Notes

- (1) M. L. Cannon, R. P. Tannenbaum, and M. LaFranco, *Am. Pharm.,* NS18, 250 (1978).
- (2) U.S. Veterans Administration, Cooperative Study Group on Antihypertensive Agents, *J. Am. Med. Assoc,* 213,1143 (1970).
- (3) T. R. Dober, A. Kagen, and W. B. Kunnel, "Habits and Coronary Heart Disease", The Framingham Heart Study, U.S. Department of Health, Education, and Welfare, Public

Health Service, National Heart Institute, 1966.

- (4) E. D. Freis, *Physiol. Rev.,* 40, 27 (1960).
- (5) E. D. Frohlich, R. O. Tarazi, and H. P. Dustan, *Am. J. Med. Sci.,* 257, 9 (1969).
- (6) J. Koch-Weser, *Am. J. Cardiol.,* 32, 499 (1973).
- (7) J. Koch-Weser, *N. Engl. J. Med.,* 295, 320 (1976).
- (8) J. Koch-Weser, *Am. Heart J.,* 95, 1 (1978).
- (9) J. T. Litchfield, Jr., and F. Wilcoxon, *J. Pharmacol. Exp. Ther.,* 96, 99 (1949).
- (10) W. W. Paudler, and T. K. Chen, *J. Heterocycl. Chem., 7,* 767 (1970).
- (11) N. Rabjohn, *Org. React.,* 5, 331 (1949).
- (12) J. Wagner, J. W. Faigle, P. Imhof, and G. Liehr, *Arzneim.-Forsch.,* 27 (II), 2388 (1977).
- (13) C. Hansen, J. E. Quinlan, and G. L. Lawrence, *J. Org. Chem.,* 33, 347 (1968).
- (14) C. Hansch, E. J. Lien, and F. Helmer, *Arch. Biochem. Biophys.,* 128, 319 (1968).
- (15) L. P. Hammett, "Physical Organic Chemistry", McGraw-Hill, New York, 1940, Chapter 7.
- (16) C. K. Hancock and C. P. Falls, *J. Am. Chem. Soc,* 83, 4214 (1961).

Synthesis of 2,3,4,4a,5,9b-Hexahydro-1H-pyrido[4,3-b]indole Derivatives and Their Central Nervous System Activities

Yasutaka Nagai,* Akira Irie, Yoshinobu Masuda, Makoto Oka, and Hitoshi Uno

Research Laboratories, Dainippon Pharmaceutical Co., Ltd., 33-94, Enoki-cho, Suita, Osaka, Japan. Received December 11, 1978

The synthesis and some pharmacological effects of *cis-* and traras-2-substituted 2,3,4,4a,5,9b-hexahydro-lWpyrido[4,3-b]indole derivatives are described. In these derivatives, the substituents of the 2, 5 and 8 position, together with the relative configuration of the 4a and 9b position, influenced the potency of the central nervous system activities. A $cis-2-[3-(p-fluorobenzoyl)propyl]$ analogue $(5k)$ of carbidine (1) possessed not only thymoleptic-like biological activity but had more potent neuroleptic activity than the parent drug.

It has been reported that carbidine (cis-2,8-dimethyl- $2,3,4,4a,5,9b$ -hexahydro-1H-pyrido $[4,3-b]$ indole dihydrochloride, 1) is an effective antipsychotic agent with thymoleptic properties.^{1,2} In animal studies, 1 not only depresses the central nervous system (CNS) but also enhances the stereotyped behavior induced by methamphetamine, $1-3$ a property which is characteristic of thymoleptics.2,4 However, the CNS-depressing activities of 1 were found to be extremely weak when tested on the basis of criteria for the typical neuroleptics, as will be shown. So, an attempt was made to prepare novel hexahydro-1H-pyrido $[4,3-b]$ indole derivatives which were as potent as the existing neuroleptics in CNS-depressing activities but which, like 1, still possessed the ability to potentiate the methamphetamine-induced stereotypy. It was found that cis-2-[3-(p-fluorobenzoyl)propyl]-8 methyl-2,3,4,4a,5,9b-hexahydro-1H-pyrido $[4,3-b]$ indole (5k) possessed such pharmaceutical properties.

Chemistry. Hydrogenolysis of 2-benzyl-2,3,4,5-tetrahydro-1H-pyrido[4,3-b]indoles 2 on palladium/carbon gave the debenzylated compounds 3, which were catalytically hydrogenated in dilute hydrochloric acid on platinum dioxide to afford (4a,9b-cis)-2,3,4,4a,5,9b-hexahydro derivatives 4 (Scheme I). The alkylation of 4 afforded the 2-substituted products 5. The alkylation or acylation of the butyrophenone derivative 5k gave the 5-substituted derivatives 6.

Reaction of 2-benzyl-8-methyl-2,3,4,5-tetrahydro-1Hpyrido[4,3-b]indole (2c) with $BH₃-THF$, followed by acid hydrolysis and subsequent neutralization, gave the trans-2,3,4,4a,5,9b-hexahydro derivative 7b. The same treatment of the 8-fluoro derivative 2b, however, gave the mixture of trans (7a) and cis compounds. This mixture was treated with acetic anhydride for separation to give the 5-acetyl derivatives 8b and 8a in a ratio of 6:1. Compound 8b was hydrolyzed to 7a, while 8a gave **4b** by the same hydrolysis followed by catalytic hydrogenolysis. Hydrogenolysis of 7 gave the debenzylated compounds 9, which were converted into the 2-substituted derivatives lOa-e and the 2,5-disubstituted ketones **11** in similar procedures mentioned above.

Compounds **4c** and 9b were benzoylated and then treated with formaline, followed by catalytic hydrogenation, to give the 2-benzoyl-5-methyl derivatives 13, which were quarternized with methyl iodide to give the 5,5-dimethyl derivatives **14a** and **14b,** respectively (Scheme II). In the NMR spectra, the difference in chemical shifts of the methyl group owing to the anisotropic shielding of the benzene ring⁶ is 10 Hz for 14a and 40 Hz for **14b,** which suggest the former to be the cis form and the latter to be the trans form, as was expected from the reduction procedures.⁶

Pharmacological Results and Discussion. The CNS-depressing property of hexahydropyrido[4,3-6]indole derivatives obtained in this study was examined by their effect on locomotor, muscle relaxating, and cataleptic activities in mice; the results are shown in Table I. It seems that the CNS-depressing property of these derivatives is affected by the substituent in the 2, 5, and 8 positions. Among the cis compounds **(5a-h** and 5k) with a methyl group in the 8 position similar to 1, the compounds (5g, **5h,** and 5k) with the bulky substituent group, such as 3-(2-chlorophenothiazin-10-yl)propyl, 4,4-bis(pfluorophenyl)butyl, and 3-(p-fluorobenzoyl)propyl, in the

Scheme I

Scheme II

2 position showed a potent activity in the locomotor test and only 5k was highly effective in all tests. As to the substituent in the 8 position of the cis-butyrophenone analogues $5i$ -m, the order of increasing potency was Et \leq $H \leq F \leq MeO \leq Me$. In the active butyrophenone derivatives 5k, introduction of substituents in the 5 position decreased the activities and 5k possessed the most potent activities in the cis series.

The compounds which showed marked activities in the above tests were examined for their effect on the methamphetamine-induced stereotyped behavior in rats. The results are shown in Table II. The 8-alkyl derivatives with a hydrogen or a lower alkyl group in the 5 position and a 3-(p-fluorobenzoyl)propyl group in the 2 position potentiated the stereotyped behavior similar to 1, but the 8-unsubstituted, 8-fluoro, and 8-methoxy analogues 5i, 5j, and 5m and the 2-[3-(2-chlorophenothiazin-10-yl)propyl] derivatives 5g exhibited no effect on this behavior or weakly antagonized it. The 2-[3-(p-fluorobenzoyl) weakly alitagolitzed it. The 2-₁₀-(p-fluorobelizoyi)-
propyll-8-methyl-2,3,4,5-tetrahydro derivative 15⁷ also showed no effect. Compound 5k showed a potentiating effect as potent as 1 on the stereotyped behavior at 2-10 mg/kg ip, which was the most potent among the compounds tested.

In addition, 5k enhanced apomorphine-induced gnawing at 0.03-0.7 mg/kg sc in rats. At higher doses (Table III), however, 5k, like typical neuroleptics, blocked the methamphetamine-induced hyperactivity in mice and apomorphine-induced gnawing in rats. On the other hand, 1 showed only potentiating effects on the stereotypy induced by both agents.

The neuroleptic activity of 5k was examined in more detail and the results obtained are shown in Table III. Compound 5k exhibited an inhibition of locomotion, conditioned avoidance response, and self-stimulation and

^a Analyses were obtained for C, H, N and, when those elements were presented, for F or S. The results obtained for these elements were within $\pm 0.4\%$ of the theoretical values.
^b No. of positive effects/no. tested. 2-chlorophenothiazin-10-ylpropyl.

H.

^a The symbols have the following meanings: ---, marked antagonism (>75%); --, moderate antagonism (50-75%), -, slight antagonism (25-50%); 0, no interaction (0-25%); +, slight potentiation (25-50%); ++, moderate potenti benzoyl)propyl]-8-methyl-2,3,4,5-tetrahydro-1H-pyrido[4,3-b]indole hydrochloride.

caused catalepsy similar to the existing neuroleptics. These activities were slightly less potent than those of chlorpromazine. Thus, 5k is a new compound having not only the neuroleptic properties, which are considerably more potent than 1, but also some thymoleptic-like properties.

As to the trans compounds, the butyrophenone derivatives 10d, 10e, 11a, and 11b showed marked activities in

the primary tests but strongly antagonized the methamphetamine-induced stereotyped behavior at 10 mg/kg sc. It is interesting that the cis $(5k)$ and trans isomers $(10e)$ exhibited the opposite effect on the stereotyped behavior. These trans-butyrophenone derivatives showed potent CNS-depressing activities and, in particular, exhibited more potent inhibition of self-stimulation in comparison

Table **III.** Comparative CNS Activities of Butyrophenone Derivatives

 ED_{so} , mg/kg

a 95% confidence limits. *^b* Antimeth = antimethamphetamine.

with the cis compounds 5k and **6a.** Introduction of a methyl group in the 5 position reduced the cataleptic activity, which is regarded as a measure of extrapyramidal side effects in clinical use;⁸ so 11a is a neuroleptic approximately as potent as chlorpromazine but with less cataleptic potential.

On the basis of these results, 5k and 11a are worth further studying and, in particular, the former has a unique pharmacological profile as a new neuroleptic. Further studies of this compound will be published elsewhere.

Experimental Section

Melting points were determined on a Yanagimoto melting point apparatus and are uncorrected. NMR spectra were taken in $Me₂SO-d₆$ solution with a Varian HA-100 spectrometer using Me4Si as an internal standard. Where the analyses are indicated only by the symbols of the elements, the analytical results were within $\pm 0.4\%$ of theoretical values. Organic extracts were dried over MgS04.

 8- **Ethyl-2,3,4,5-tetrahydro-1***H***-pyrido[4,3-***b***]indole (3d) Hydrochloride.** A mixture of 2d-HCl (16.3 g, 0.05 mol) and 5% Pd/C (2 g) in 70% EtOH (150 mL) was catalytically hydrogenated at 70 °C under ordinary pressure. After the theoretical amount of H_2 was absorbed, the catalyst was filtered. The filtrate was cooled, and the resulting precipitates were collected and recrystallized from 70% EtOH to give 3d-HCl (8.8 g, 75%), mp 256-258 °C. Anal. $(C_{13}H_{16}N_2 \cdot H\bar{C}I)$ C, H, Cl, N.

cis-2,3,4,4a,5,9b-Hexahydro-1 ff-pyrido[4,3- *b* **Jindoles 4a-d and 4e.** Into a mixture of concentrated HCl (370 mL) and H_2O (250 mL) were added 3 (0.25 mol) and $PtO₂$ (1.3 g), and the mixture was submitted to catalytic hydrogenation at 70 °C under ordinary pressure. After the theoretical amount of H_2 was absorbed, the catalyst was filtered. The filtrate was made alkaline with NaOH and extracted with benzene. The extract was dried and concentrated. Solid residues (crude 4a and 4d) were recrystallized from benzene and n -hexane. Oily residues were converted into the oxalates (4b and 4c) and recrystallized from EtOH or purified by chromatography on silica gel with $CHCl₃–MeOH (30:1)$ (4e). 4a: mp 99–101 °C; yield 54%. Anal. $(C_{11}H_{14}N_2)$ C, H, N. 4b-dioxalate: mp 169-171 °C dec; yield 34%. Anal. $(C_{11}H_{13}FN_22C_2H_2O_4)$ C, H, F, N. 4c-dioxalate: mp 188-189 °C; yield 74%. Anal. $(C_{12}H_{16}N_2.2C_2H_2O_4)$ C, H, N. 4d: mp 82–85 °C; yield 65% . Anal. $(C_{13}H_{18}N_2)$ C, H, N. 4e: yield 52% . Anal. (C12H16N20) C, **H,** N.

cis- **or trans-2-Substituted** 2,3,4,4a,**5,9b-hexahydrolH-pyrido[4,3-b]indole Derivatives 5 and 10. Procedure A.** A mixture of 4 (or 9) (0.01 mol), appropriate alkyl halide (or methyl benzenesulfonate) (0.012 mol), and triethylamine (0.015 mol) in toluene (50 mL) was heated under reflux for 1-15 h. After being cooled, the reaction mixture was washed with H_2O , dried, and concentrated in vacuo. The residue was converted into the hydrochloride with ethanolic HC1 and recrystallized from a suitable solvent.

Procedure B. A mixture of 4 (or 9) (0.01 mol), appropriate alkyl halide (0.013 mol), potassium carbonate (0.03 mol), and potassium iodide (0.01 mol) in methyl ethyl ketone (100 mL) was heated under reflux for 10-25 h. The insoluble matter was removed by filtration and the filtrate was concentrated. The residue was converted into the hydrochloride with ethanolic HC1 and recrystallized from a suitable solvent or purified by chromatography on silica gel with CHCl₃-EtOH (20:1) in 5g.

Procedure C. A mixture of 4c (or **9b)** (2 g, 0.011 mol) and methyl vinyl ketone (0.9 g, 0.013 mol) in benzene (50 mL) was refluxed for 1 h. After the reaction mixture was concentrated, the residue was converted into salt and recrystallized from a suitable solvent.

cis- **or trans-8-Substituted 2-[3-(p-fluorobenzoyl) propyl]-5-methyl-2,3,4,4a,5,9b-hexahydro-li?-pyrido[4,3- 6]indole (6a, 11a, and lib).** A solution of 5k (or **lOd** or **lOe)** (0.01 mol) in a mixture of 37% HCHO $(1 \text{ g}, 0.012 \text{ mol})$, HCOOH $(1.2 \text{ g}, 0.026 \text{ mol})$, and $H₂O (10 \text{ mL})$ was heated under reflux for 15 min. The reaction mixture was made alkaline with dilute NaOH and extracted with benzene. The extract was washed with H20, dried, and concentrated. The residue was purified by chromatography on silica gel (20 g) with $CHCl₃-EtOH$ (200:1), converted into salt, and recrystallized from a suitable solvent.

cis-2-[3-(p-Fluorobenzoyl)propyl]-5-ethyl- (or benzyl) 8-methyl-2,3,4,4a,5,9b-hexahydro-1H-pyrido[4,3-b]indole (6b **and** 6c) **Dihydrochloride.** A mixture of 5k (3.5 g, 0.01 mol), ethyl iodide (or benzyl bromide) (0.012 mol), and triethylamine (1.3 g, 0.013 mol) in toluene (50 mL) was heated under reflux for 15 h. The reaction mixture was washed with $H₂O$, dried, and concentrated. The residue was converted into the dihydrochloride with ethanolic HC1 and recrystallized from EtOH.

cis-2-[3-(p-Fluorobenzoyl)propyl]-5-acetyl-8-methyl-2,3,4,4a,5,9b-hexahydro-1H-pyrido[4,3-b]indole (6b) Fumarate. A mixture of 5k (3.5 g, 0.01 mol) and Ac₂O (2.6 g, 0.026 mol) in benzene (30 mL) was heated under reflux for 1.5 h. The reaction mixture was washed with dilute NaOH and H_2O , dried, and concentrated. The residue was converted into the fumarate and recrystallized from EtOH to give 6d-fumarate (2.8 g, 64%), mp 164-166 °C.

 $trans-2-Benzv1-8-methyl-2,3,4,4a,5,9b-hexahvdro-1H$ **pyrido[4,3-b]indole (7b) Dihydrochloride.** To a cooled mixture of **2c** (34 g, 0.123 mol) and sodium borohydride (9.36 g, 0.246 mol) in tetrahydrofuran (THF; 300 mL) was added a solution of boron trifluoride etherate (46.9 g, 0.33 mol) in THF (60 mL) under N_2 during 1 h. The reaction mixture was stirred for 30 min at room temperature and then refluxed for 4 h. To the cooled mixture was added 6 N HC1 (200 mL), and THF was removed in vacuo. To the residual solution was added dioxane (150 mL), and the mixture was heated under reflux for 1 h and concentrated. The residue was made alkaline with dilute NaOH and extracted with CHC13. The extract was dried and concentrated. The residue was converted to the hvdrochloride and recrystallized from 80% EtOH to give 7b.HCl (23.2 g, 54%), mp 250-255 °C.

trans- **or cis-2-Benzyl-8-fluoro-2,3,4,4a,5,9b-hexahydro-** $1H$ -pyrido[4,3-*b*]indole (8b and 8a). Crude 7a (27 g) was prepared from 2b (30 g) by the same procedure for the preparation of 7_b .

A mixture of the crude 7a $(27 g)$ and Ac₂O $(25 g)$ in benzene (60 mL) was refluxed for 1.5 h. The reaction mixture was washed with dilute NaOH and H₂O and dried. The solvent was removed in vacuo and the residue was recrystallized from EtOH to give 8b (8.5 g, 24.4%). The mother liquor was concentrated and the residue was chromatographed on silica gel (60 g). The product obtained from the earlier fraction of CHCl₃ elution was recrystallized from EtOH to give 8a (1.4 g, 4%). 8a: mp 139-140 ${}^{\circ}$ C. Anal. (C₂₀H₂₁FN₂O) C, H, N. 8b: mp 157~159 ${}^{\circ}$ C. Anal. $(C_{20}H_{21}FN_2O)$ C, H, N.

A solution of 8a (1.3 g, 0.004 mol) in a mixture of concentrated HCl (4.2 mL) and $H_2O(2 \text{ mL})$ was heated under reflux for 1.5 h. The reaction mixture was made alkaline with NaOH and extracted with benzene. The extract was concentrated and to the residue was added Pd/C (0.2 g) and 50% EtOH (20 mL). The mixture was catalytically hydrogenated at 60 °C under ordinary pressure. After the theoretical amount of H_2 was absorbed, the catalyst and the solvent were removed and the residue was recrystallized from benzene and n-hexane to give **4b** (0.51 g, 67%),

trass-2-BenzyI-8-fluoro-2,3,4,4a,5,9b-hexahydro-l.ff $pyrido[4,3-b]indole(7a)$. A solution of 8b $(2.6 g, 0.008 mol)$ in a mixture of concentrated HCl (9.5 mL) and $H₂O$ (5 mL) was heated under reflux for 2 h. The solution was made alkaline with dilute NaOH and extracted with benzene. The extract was dried and concentrated. The residue was recrystallized from n -hexane to give 7a (1.7 g, 75%), mp 92-94 °C.

*trans***-8-Fluoro- (or methyl) 2,3,4,4a,5,9b-hexahydro-**1H-pyrido[4,3-b]indole (9a and 9b). A mixture of 7 (0.02 mol) and 5% Pd/C $(1 g)$ in 50% EtOH (100 mL) was submitted to catalytic hydrogenation at 60 °C under ordinary pressure. After the theoretical amount of H_2 was absorbed, the catalyst and the solvent were removed. To the residue was added dilute $NH₄OH$, and the solution was extracted with benzene. The extract was dried and concentrated. The residue was recrystallized from a suitable solvent. **9a**: mp 121-123 °C (benzene-n-hexane); yield 82%. Anal. $(C_{11}H_{13}FN_2)$ C, H, F, N. 9b: mp 100-102 °C (ether-n-hexane); yield 78% . Anal. $(C_{12}H_{16}N_2)$ C, H, N.

cis- **or** *trans***-2-Benzoyl-8-methyl-2,3,4,4a,5,9b-hexahydro-1H-pyrido[4,3-b]indole (12a and 12b).** To a solution of 4c (or **9b)** (10 g, 0.053 mol) and triethylamine (10.7 g, 0.106 mol) in benzene (200 mL) was added benzoyl chloride (8.19 g, 0.059 mol) dropwise at room temperature; after the addition was completed, the reaction mixture was stirred for 1 h. The precipitate was filtered and the filtrate was extracted with dilute HC1. The acidic layer was made alkaline with $Na₂CO₃$ and extracted with benzene. The extract was dried and concentrated. The residue was chromatographed on silica gel (40 g) and elution with CHCl3-EtOH (100:1) gave crude **12a** (or **12b).** Crude **12a** was converted into the hydrochloride with ethanolic HC1 and recrystallized from EtOH. Crude **12b** was recrystallized from benzene and n -hexane.

12a-HCl: mp 185-190 °C; yield 58%. Anal. $(C_{19}H_{20}N_2O\text{-HCl})$ C, H, Cl, N. 12b: mp 169-171 °C; yield 51%. Anal. $(\tilde{C}_{19}H_{20}N_2O)$ C, H, N.

cis- or trans-2-Benzoyl-5,8-dimethyl-2,3,4,4a,5,9b-hexa**hydro-lff-pyrido[4,3-b]indole (13a and 13b).** After adding 37% aqueous HCHO (3.24 g, 0.04 mol) to a solution of 12 (2.9 g, 0.01 mol) in EtOH (60 mL), the resulting solution was stirred at room temperature for 30 min, then treated with Raney Ni (0.2 g), and finally catalytically hydrogenated at ordinary temperature and pressure. After the theoretical amount of H_2 was absorbed, the catalyst and the solvent were removed. The residue was recrystallized from AcOEt. **13a:** mp 150-152 °C; yield 45%. Anal. $(C_{20}H_{22}N_2O)$ C, H, N. 13b: mp 165-167 °C; yield 75%. Anal. $(C_{20}H_{22}N_2O)$ C, H, N.

cis- or trans-2-Benzoyl-5,8-dimethyl-2,3,4,4a,5,9b-hexa**hydro-lH-pyrido[4,3-ft]indole Methiodide (14a and 14b).** A solution of 13 (3.1 g, 0.01 mol) and an excess of methyl iodide in AcOEt (40 mL) was heated under reflux for 8 h. The separated quarternary ammonium salt was collected and recrystallized from a suitable solvent. **14a:** mp 169-171 °C (EtOH); yield 31%; NMR δ 3.41, 3.52 (s, each 3 H, N-CH₃). Anal. (C₂₀H₂₂N₂O·CH₃I) C, H, I, N. **14b:** mp 234-235 °C (dilute EtOH); yield 76%; NMR δ 3.23, 3.63 (s, each 3 H, N-CH₃). Anal. (C₂₀H₂₂N₂O-CH₃I) C, H, I, N.

Pharmacology Methods. Animals and Materials. Adult male STD-ddy mice and male Wister HLA rats were employed in the experiments. Test compounds were dissolved or suspended in 0.5% aqueous tragacanth and administered. All doses of the compounds are expressed as the form (salt or base) indicated in Table I.

Statistics. The ED_{50} values were calculated according to the method of Litchfield and Wilcoxon.⁹

Effect on Locomotor Activity in Mice. The effect of the compounds on locomotor activity was examined in mice using an Animex activity meter according to the modified method of Svensson and Thieme.¹⁰ A group of five mice was used for each dose of the compounds. At an appropriate time after oral administration of test compounds, each mouse was put into the cage on Animex and the locomotor activity was measured for 3 min. Average suppression of the locomotor activity by each dose of test compounds was expressed as percent inhibition of control, and the ED_{50} of each compound, defined as the dose which caused 50% inhibition, was calculated.

Muscle Relaxation. Test compound, 100 mg, was administered po to a group of five mice and the effect was evaluated 20 min after the administration according to the modified method of Courvoisier et al.¹¹ When forepaws of a mouse are placed on a taut horizontal wire, the normal response is for the mouse to draw the hind paws up to the wire. The reaction was considered to be positive if the mouse failed to grasp the wire with the hind paws in less than 5 s at least once in three trails.

Catalepsy in Mice. Groups of five mice were tested for catalepsy according to the modified method of Courvoisier et al.¹¹ Each mouse was forced to put its forepaws on a rubber cap of 2.8 cm in height. The mice which failed to remove their paws from the cap within 30 s were considered to be cataleptic. The ED_{50} in this test is the dose which causes catalepsy in 50% of the mice.

Effect on Methamphetamine-Induced Stereotyped Behavior. This test was carried out by a modified method of Nayler and Costall.¹² Groups of six rats (200–250 g) were injected ip with graded doses of test compounds, followed 20 min later by an ip injection of methamphetamine, 2 mg/kg, and they were observed for stereotyped behavior at 30-min intervals for 3 h and then 1-h intervals for a further 3 h. The intensity of the stereotypy was evaluated by assessing the components of this behavior, i.e., sniffing, exploratory activity, head movement, gnawing, licking, and back locomotion, with an arbitrary score from 0 to 2, "0" referring to absent, "1" to moderate, and "2" to marked manifestation of each component. The sums of the scores for these components of stereotypy were used to quantitatively evaluate the effect of test compounds on the stereotyped behavior induced by methamphetamine. The results are expressed as the symbols representing the percent potentiation or antagonism and **are** given in Table **II.**

Effect on Apomorphine-Induced Gnawing in Rats. Groups of six rats were injected iv with 1 mg/kg of apomorphine hydrochloride 1 h after sc injection of test compounds, and at 5,10, and 20 min after apomorphine hydrochloride the rats were observed for gnawing movement for 1 min. Absence of the typical gnawing movement during at least one of the three observation periods was regarded as a positive effect.¹³ The ED_{50} in this test is defined as the dose which suppresses the gnawing movement in 50% of the rats. Furthermore, the ability of 5k to enhance the action of apomorphine hydrochloride was examined. Rats were injected iv with 0.05 mg/kg of apomorphine hydrochloride, a dose too small to induce gnawing by itself, 1 h after a sc injection of 5k and observed for the incidence of gnawing for 20 min.

Effect on Methamphetamine-Induced Hyperactivity in **Mice.** The suppression of methamphetamine-induced hyperactivity was examined according to the modified method of Ueki et al.¹⁴ At 100 min after oral administration of test compounds, five groups of three mice were injected with methamphetamine (5 mg/kg, ip), and 10 min thereafter the locomotor activity was measured with a photocell activity counter for 20 min. The ED_{50} in this test is the dose which reduces the average counts of the treated mice to half the count of control animals.

Effect on Active Avoidance in Mice. Effect of test compounds on one-way active avoidance in mice was examined using a box with two compartments, darkened and lighted, as described previously.¹⁵ Mice were previously trained to avoid electroshocks from the floor by moving from the darkened compartment to the lighted one in response to a warning stimulus. This test was carried out using for each group 10 mice which could avoid the shock at a rate more than 80% in 20 trials. The results are expressed as the ED_{50} values, defined as the dose that causes a 50% inhibition in the rate of the avoidance response.

Effect on Self-stimulation in Rats. Effect of test compounds on intracranial self-stimulation behavior was studied in rats with chronic electrodes implanted in the lateral posterior hypothalamus.¹⁶ The rats which showed a constant lever-pressing response of 50-100/ min were selected for the test. Groups of eight rats were used. In the test, self-stimulation rates were counted for 10 min before and 1, 2, 4, 6, 8, and 24 h after po or ip administration of test compounds. The ED_{50} in this test is the dose which causes 50% or more inhibition of the rate of the selfstimulation in 50% of the rats.

Acknowledgment. We thank Dr. M. Shimizu, the

director of these laboratories, and Dr. H. Nishimura for their encouragement and Dr. S. Minami for his helpful advice throughout the course of this work. Thanks are also due to Dr. T. Karasawa for the pharmacological tests and the members of Analytical Center of these laboratories for the elemental analyses and spectral measurements.

References and Notes

- (1) W. M. Herrmann and J. Fabricius, *Dis. Nerv. Syst.,* 35 (7, section 2), 28 (1974).
- (2) N. K. Barkov, A. Geller, and M. E. Jarvik, *Psychopharmacologia,* 21, 82 (1971).
- (3) N. K. Barkov, *Farmakol. Toksikol.,* 34, 647 (1971).
- (4) L. Valzelli, S. Consolo and C. Morpurgo in "Antidepressant Drugs", S. Garattini and M. N. G. Dures, Ed., Excerpta Medica, Amsterdam, 1966, p 61.
- (5) A. Smith, *J. Chem. Soc,* 1 (1970).
- (6) J. G. Berger, S. R. Teller, C. D. Adams, and L. F. Guggenberger, *Tetrahedron Lett.,* 1807 (1975).
- (7) R. H. Johnson and J. P. Oswald (to Abbott Laboratories), U.S. Patent 3466 293 (1969).
- (8) O. Hornykiewicz, C. H. Markham, W. G. Clark, and R. M. Fleming in "Principles of Psychoparmacology", W. G. Clark, Ed., Academic Press, New York, 1970, p 585.
- (9) J. T. Litchfield and F. Wilcoxon, *J. Pharmacol. Exp. Ther.,* 96, 99 (1949).
- (10) T. H. Svensson and G. Thieme, *Psychopharmacologia,* 14, 157 (1969).
- (11) S. Courvoisier, R. Ducrot, and L. Julou in "Psychotropic Drugs", S. Garattini and V. Ghetti, Ed., 1975, p 373.
- (12) R. J. Naylor and B. Costall, *Life Sci.,* 10, 909 (1971).
- (13) P. A. J. Janssen, C. J. E. Niemegeers, and K. H. L. Schellekens, *Arzneim.-Forsch.,* 15, 104 (1955).
- (14) S. Ueki, H. Sugano, N. Sato, J. Iwaki, H. Yonemura, T. Fukuda, H. Yamada, S. Nurimoto, G. Hayashi and Y. Kowa, *Nippon Yakurigaku Zasshi,* 61, 421 (1965).
- (15) M. Oka and M. Shimizu, *Jpn. J. Pharmacol,* 25,121 (1975).
- (16) C. Kamei, Y. Masuda, and M. Shimizu, *Jpn. J. Pharmacol.,* 24, 613 (1974).

Glycerides as Prodrugs. 1. Synthesis and Antiinflammatory Activity of 1.3-Bis(alkanoyl)-2-(O-acetylsalicyloyl)glycerides (Aspirin Triglycerides)¹

Gerard Y. Paris,* David L. Garmaise,² Denis G. Cimon,

Organic Chemical Research, Abbott Laboratories, Ltd., Montreal, Quebec, H3C 3K6

Leo Swett, George W. Carter, and Patrick Young

Department of Medicinal Chemistry and Department of Pharmacology, Abbott Laboratories, North Chicago, Illinois 60064. Received November 22, 1978

A series of l,3-bis(alkanoyl)-2-(0-acetylsalicyloyl)glycerides (aspirin triglycerides) having aspirin at the 2 position of glycerol and fatty acids at the 1 and 3 positions was prepared. The compounds were administered orally and tested for efficacy in the rat paw edema test, and the stomachs were examined for the presence of lesions. The results showed that the members of this series in which the fatty acids are of intermediate chain length (C_4-C_{12}) do not cause gastric lesions and have essentially all the systemic activity associated with aspirin.

Despite the enormous amount of work that has been done on development of antiinflammatory drugs, the classical remedy aspirin remains the drug of first choice in the treatment of arthritis. Although traditional and familiar, aspirin is not an innocuous substance. It causes **a** variety of side effects which limit its usefulness, often obliging the patient to switch to other medications which frequently have a similar profile of side effects. The primary unwanted effect is direct gastric irritation, due to contact of solid aspirin (or of concentrated solutions of aspirin) with the gastric mucosa. It has been reported recently³ that aspirin in antiarthritic doses causes endoscopically observable gastric erosion in 39% of the patients tested over a 4-week period. Many attempts have been made to develop aspirin derivatives and formulations which will yield adequate blood levels after oral administration without causing gastric irritation. We wish to report on a novel approach to this problem, in which