Cardiac Glycosides. 6. Gitoxigenin C16 Acetates, Formates, Methoxycarbonates, and Digitoxosides. Synthesis and Na⁺,K⁺-ATPase Inhibitory Activities

Toshihiro Hashimoto,[†] Hargovind Rathore,[†] Daisuke Satoh,[‡] George Hong,[†] Jane F. Griffin,[#] Arthur H. L. From,[§] Khalil Ahmed,[§] and Dwight S. Fullerton^{*†}

College of Pharmacy, Oregon State University, Corvallis, Oregon 97331, 2-2-17, Kamikotoen, Nishinomiya City, Japan, Veterans Administration Medical Center and University of Minnesota, Minneapolis, Minnesota 55417, and Medical Foundation of Buffalo, Inc., Buffalo, New York 14203. Received July 30, 1985

A series of 17 gitoxigenin 16 β -formates, acetates, and methoxycarbonates was synthesized, including their 3β -acetates, formates, and digitoxosides. A 16 β -formate group was generally found to increase activity 30 times, a 16 β -acetate group 9-12 times, while a 16β -methoxycarbonate decreased activity by two-thirds. 3β -Formates and acetates had little effect on activity by themselves, but sometimes reduced the activity-increasing properties of 16β -formates and acetates. A 3β -digitoxoside increases the activity of gitoxigenin by 15 times, but the effect is less if the 16β -group is esterified. And finally, a 16-one decreases activity dramatically. These data suggest an important role for C16 esters and possibly the presence of a separate binding site on Na⁺,K⁺-ATPase corresponding to the cardenolide C16 position.

Gitaloxigenin (6, gitoxigenin 16β -formate) and its tridigitoxoside gitaloxin are naturally occurring cardenolide 16 β -formates found in Digitalis purpurea.³ Interest in gitaloxin has recently increased, and it has even been proposed that this "forgotten cardiac glycoside of Digitalis purpurea" may be responsible for most of purpurea's therapeutic activity.^{4a} In studies with a guinea pig heart Na⁺,K⁺-ATPase assay system, Depover and Godfraind found that gitaloxigenin (6) is 5 times more potent than digitoxigenin and 41 times more potent than gitoxigenin (2).^{4b} We have been studying the relationship of cardenolide structure and Na⁺,K⁺-ATPase inhibitory activity and have found an excellent correlation with C17 side group carbonyl oxygen position obtained from crystallographic results.^{1,5-8} Thus, we wished to extend these studies to examine in detail the effects of 16β -substitution on cardenolide activity.

In the present study, we report the synthesis of a series of gitoxigenin 16β -formates, acetates, methoxycarbonates, 16-ones, and their biological activity (inhibition of Na⁺,K⁺-ATPase). Since previous studies have also shown that the sugar directly attached to the genin has the greatest role in sugar site binding,^{1,9,10} we also made the monodigitoxoside 24 (the monodigitoxoside analogue of gitaloxin) and its acetate analogue 22. Preliminary accounts of the crystallographic and computer graphics analyses of some of these and related compounds have recently been published.² A detailed account of these investigations will be published separately in the near future.

Chemistry. The synthesis of acetates 3, 4, 5, and β -Ddigitoxoside 19 were modifications of methods previously reported by Satoh and co-workers.¹¹⁻¹⁴

As shown in Table I, several experiments were conducted to delineate conditions for selective acetylation of the 3β - or 16β -hydroxyl groups. Selective acetylation of gitoxigenin 2 to make 16β -acetate 5 was best achieved at 17 °C with acetic anhydride and pyridine. Yields were typically 70%. Reaction for more than 1 h results in increasing amounts of the 3β , 16β -diacetate 3. In contrast, formic acetic anhydride in pyridine produces the 3β , 16β diformate 8 even at 0-5 °C. The desired 16β-formate 6 could be made instead with formic acetic anhydride and sodium formate in DMF, albeit not as selectively (45% of

6, 13% of the 3β -formate 7, and 19% of the 3β , 16β -diformate 8. Selective acetylation of 2 to produce 3β -acetate 4 (in 56.9% yield) was best with acetic anhydride in acetic acid and chloroform. Alternatively, the 3β , 16β -diacetate 3 could be selectively hydrolyzed to 4 in 65% yield in methanolic potassium bicarbonate. These same methods were then used to produce the mixed esters 9 and 10.

In all cases, the 3β -esters could be easily distinguished from the 16 β -esters by chemical shifts of the C16 and C3 protons.¹⁵ Proton NMR shifts were consistent with those previously reported by Tori and Aono.¹⁵ A new and interesting finding is that the C15 protons are diasterotropic. In gitoxigenin 16β -acetate (5) and gitoxigenin 3β , 16β -diacetate (3) the C15 protons appear at δ 2.75 (dd, J = 16, 10 Hz, 1 H, C15 H¹) and 1.82 (m, 1 H, AM part of an AMC system superimposed on the methylene protons of the rest of the steroid ring system, $C15 H^2$). These assignments were made with two-dimensional ¹H/¹H and ¹³C/¹H spectra, both obtained on a Brucker 400-MHz instrument. The C15 H¹ proton at δ 2.75 is coupled to the C15 H² at δ 1.82. Both the C15 H¹ and H² protons are coupled to

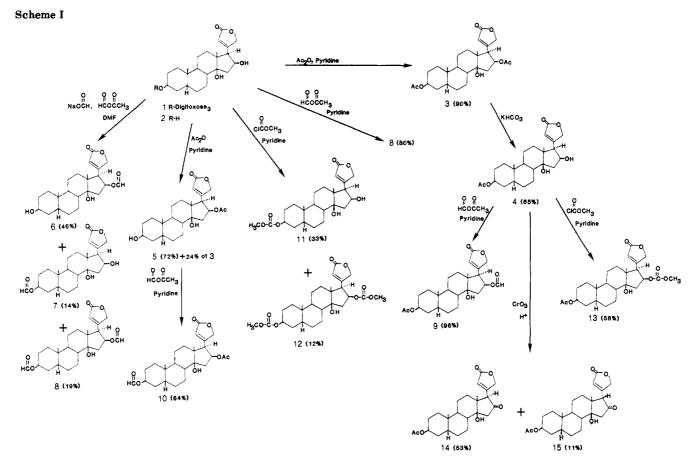
- (1) Paper 5 in this series: Rohrer, D. C.; Kihara, M.; Deffo, T.; Rathore, H.; Ahmed, K.; From, A. H. L.; Fullerton, D. S. J. Am. Chem. Soc. 1984, 106, 8269.
- (a) Griffin, J. F.; Rohrer, D. C.; Ahmed, K.; From, H. L.; Fullerton, D. S. The Effect of 16β-Substitution on the Structure and Activity of Digitoxigenin: Evidence for an Additional Binding Interaction with Na⁺,K⁺-ATPase; Abstracts, 25th Annual Medicinal Chemistry Symposium, State University of New York at Buffalo, June, 1984. (b) Ahmed, K.; Fullerton, D. S.; Rohrer, D. C.; Griffin, J. F.; From, A. H. L. Fed. Proc., Fed. Am. Soc. Exp. Biol. 1985, 44, 842. (c) Fullerton, D. S.; Griffin, J. F.; Rohrer, D. C.; From, A. H. L.; Ahmed, K. Trends Pharmacol. Sci. 1985, 6, 279.
- (3) Angliker, E.; Barfuss, F.; Renz, J. Chem. Ber. 1958, 41, 479.
- (4) (a) Woodcock, B. G.; Rietbrock, N. Trends Pharmacol. Sci. 1985, 273. (b) DePover, A.; Godfraind, T. Naunyn-Schmiedeberg's Arch. Pharmacol. 1982, 321, 135.
- (5) Ahmed, K.; Rohrer, D. C.; Fullerton, D. S.; From, A. H. L. J. Biol. Chem. 1983, 258, 8092.
- (6) From, A. H. L.; Fullerton, D. S.; Deffo, T.; Rohrer, D. C.; Ahmed, K. J. Mol. Cell. Cardiol. 1984, 16, 835.
- Fullerton, D. S.; Kitatsuji, E.; Deffo, T.; Rohrer, D. C.; Ahmed, (7)K.; From, A. H. L. Curr. Top. Membr. Transp. 1983, 19, 257.
- (8) Kihara, M.; Yoshioka, K.; Deffo, T.; Fullerton, D. S.; Rohrer,
- D. C. Tetrahedron 1984, 40, 1121.
 (9) Fullerton, D. S.; Yoshioka, K.; Rohrer, D. C.; From, A. H. L.; Ahmed, K. Mol. Pharmacol. 1980, 17, 43.
- (10) Yoda, A. Ann. N.Y. Acad. Sci. 1974, 242, 598.
- (11) Satoh, D. Showa Yakka Daigabu Kiyo 1964, 14, 14.
- (12) Morita, J.; Satoh, D. Chem. Pharm. Bull. 1968, 16, 1056.
- (13) Satoh, D.; Morita, J. Japan. Patent 450 665, 1965.
- (14) Satoh, D.; Aoyama, K. Chem. Pharm. Bull. 1970, 18, 94.
- (15) Tori, K.; Aono, K. Showa Yakka Daigaku Kiyo 1965, 15, 1.

[†]Oregon State University.

[‡]2-2-17, Kamikoten.

[§] Veterans Administration Medical Center and University of Minnesota.

Medical Foundation of Buffalo, Inc.



the C16 proton at δ 5.45. In the ${}^{13}C/{}^{1}H$ spectrum, two protons at δ 2.75 and 1.82 are coupled to C15 at 41.0 ppm.

To learn if the electronic character of the oxygen functionality at C16 was an important determinant of activity, we prepared the analogues series of methoxycarbonates, as well as the 16-one 14. The 3β -methoxycarbonate 11 and 3β ,16 β -dimethoxycarbonate 12 were prepared from gitoxigenin (2), by treatment with methoxycarbonyl chloride in pyridine. The mixed ester 13 was similarly prepared from 3β -acetate 4 in 58% yield. Jones oxidation of 4 gave the 16-one analogue 14, along with its 17α -isomer 15.

Using a modification of the procedures we have previously used for synthesis of digitoxigenin 3β -digitoxoside,⁸ we made the bisdigitoxoside 18 via dialdehyde 16 and diol 17 as shown in Scheme II. Repeat of the same series of steps (periodate oxidation, sodium borohydride reduction to the diol, and acid hydrolysis) gave monodigitoxoside 19 in 70% yield from 18.

Protection of the 3'- and 4'-hydroxyls of 19 as the acetonide (using methods we have previously reported for other cardenolide digitoxose acetonides⁸), followed by esterification of the 16β -hydroxyl as described above for the genin analogues, and acid hydrolysis removal of the acetonide group¹⁶ gave 22, the 16β -acetate analogue of 19, and 24, the 16β -formate analogue of 19.

Biology. As in our previous studies, a hog kidney Na^+,K^+ -ATPase preparation was used for studying the activity of cardiac steroids.^{5,6} In brief, the inhibition was measured under type I binding conditions (i.e., with Mg²⁺, Na⁺, and ATP as the binding ligands, 15-min preincubation for genins and 2 h for glycosides). All assays were carried out under equilibrium drug binding conditions and during the linear phase of the ATPase reaction.⁵ The

resulting I_{50} values (concentration required for 50% inhibition of the Na⁺,K⁺-ATPase) are shown in Table II. Each I_{50} was confirmed at least three times and the results did not vary by more than 5% in any case.

Effect of the 16\beta-OH. As can be seen in Table II, gitoxigenin is only 20% as active as digitoxigenin. This is consistent with the activity-decreasing effect of a 16 β -OH reported by Depover and Godfraind.^{4b}

Effect of a 16 β -Formate. A 16 β -formate increases genin activity about 30-fold. Examples include gitoxigenin 16 β -formate (6) vs. gitoxigenin (2), 30 times increase; the corresponding 3 β -acetates 9 vs. 4, 35 times increase.

Effect of a 16 β -Acetate. A 16 β -acetate increases genin activity less than a 16 β -formate, only 9–12-fold. Examples include gitoxigenin 16 β -acetate (5) vs. gitoxigenin (2), 12 times increase, and the corresponding 3β ,16 β -diacetate 3 vs. 3β -acetate 4, 8 times increase.

Effect of a 3β -Formate. A 3β -formate has little effect on activity and, in the presence of a 16β -ester, may decrease activity slightly. Examples include gitoxigenin (2) vs. 3β -formate 7, no effect; 3β , 16β -diformate 8 vs. 16β formate 6, no significant change; and 3β -formate 16β acetate 10 vs. 16β -acetate 5, slight decrease.

Effect of a 3β -Acetate. A 3β -acetate has little effect on activity and, in the presence of a 16β -ester, may decrease activity slightly. Examples include gitoxigenin (2) vs. 3β -acetate 4, no effect; 3β , 16β -diacetate 3 vs. 16β -acetate 5, slight decrease; and 3β -acetate, 16β -formate 9 vs. 16β formate 6, no significant change.

Effect of a 16 β -Methoxycarbonate. A 16 β -methoxycarbonate decreases activity, as shown with 12 vs. gitoxigenin (2), loss of two-thirds of activity.

Effect of a 3β -Digitoxoside. With digitoxigenin analogues, as reported in the previous paper in this series, a 3β -digitoxoside increases activity by an average of 8.9 times. However, a different relationship exists with gitoxigenin analogues. If the 16β -OH is unsubstituted, the

⁽¹⁶⁾ Kaiser, F.; Voigtlaender, W. Ger. Offen. 2306580; Chem. Abstr. 1974, 81, 136458k.

Table I. Acetylation and Formylation of Gitoxigenin

reagents	temp, °C	time, h	products	recovered 2, %
$Ac_2O(2)$ mL)	17	1	72% 5	
pyridine (10 mL)			24% 3	
Ac ₂ O (50 mL) pyridine (60	17	14	90% 3	
mL) Ac ₂ O (70	60	50 min	50% 4	48
Ac ₂ O (100	60	2	trace 5 30% 4	45
11112)			16% 3	
Ac_2O (100 mL)	0–5	1	26% 4	59
H_2SO_4 (half			3% 3	
Ac ₂ O (160	60–70	2	5% 5 57% 4	
AcOH (8 mL)			18% 3	
mL)	0.5	1	96.4	59
mL)	0-0	1		09
mg) ZnCl ₂ (300			3% 3	
HCOOCOC- H ₃ (5 mL)	0–5	1	80% 8	
pyridine (10 mL)				
H ₃ (23	17	1	14% 7	
HCOONa			46% 6	
DMF (70 mL)			14% 7	
			19% 8	
CICOOCH		6		27
(6 mL) pyridine (30	0–5	-	12% 1 2	
	$\begin{array}{c} Ac_2O \ (2 \\ mL) \\ pyridine \ (10 \\ mL) \\ Ac_2O \ (50 \\ mL) \\ pyridine \ (60 \\ mL) \\ Ac_2O \ (70 \\ mL) \\ Ac_2O \ (70 \\ mL) \\ Ac_2O \ (100 \\ mL) \\ AcOH \ (8 \\ mL) \\ CHCl_3 \ (100 \\ mL) \\ AcOH \ (8 \\ mL) \\ CHCl_3 \ (100 \\ mL) \\ AcOH \ (60 \\ mg) \\ ZnCl_2 \ (300 \\ mg) \\ HCOOCOC- \\ H_3 \ (5 \\ mL) \\ HCOOCOC- \\ H_3 \ (23 \\ mL) \\ HCOONa \\ \ (300 \ mg) \\ DMF \ (70 \\ mL) \\ ClCOOCH_3 \\ \ (6 \\ mL) \end{array}$	$\begin{array}{ccccccc} Ac_2O & (2 & 17 & \\ mL) & \\ pyridine & (10 & \\ mL) & \\ Ac_2O & (50 & 17 & \\ mL) & \\ pyridine & (60 & \\ mL) & \\ Ac_2O & (70 & 60 & \\ mL) & \\ Ac_2O & (70 & 60 & \\ mL) & \\ Ac_2O & (100 & 0-5 & \\ mL) & \\ Ac_2O & (100 & 0-5 & \\ mL) & \\ Ac_2O & (100 & 0-5 & \\ mL) & \\ Ac_2O & (100 & 0-5 & \\ mL) & \\ Ac_2O & (100 & 0-5 & \\ mL) & \\ Ac_2O & (100 & 0-5 & \\ mL) & \\ Ac_2O & (100 & 0-5 & \\ mL) & \\ AcOH & (8 & \\ mL) & \\ AcOH & (60 & \\ mg) & \\ ZnCl_2 & (300 & \\ mg) & \\ AcOH & (60 & \\ mg) & \\ ZnCl_2 & (300 & \\ mg) & \\ HCOOCOCC & 0-5 & \\ H_3 & (5 & mL) & \\ pyridine & (10 & \\ mL) & \\ HCOONa & \\ & (300 & mg) & \\ DMF & (70 & \\ mL) & \\ ClCOOCH_3 & \\ & (6 & mL) & \\ pyridine & (30 & 0-5 & \\ \end{array}$	$\begin{array}{cccccccc} Ac_2O & (2 & 17 & 1 & \\ mL) & \\ pyridine & (10 & & \\ mL) & \\ Ac_2O & (50 & 17 & 14 & \\ mL) & \\ pyridine & (60 & & \\ mL) & \\ Ac_2O & (70 & 60 & 50 & \\ mL) & \\ Ac_2O & (100 & 60 & 2 & \\ mL) & \\ Ac_2O & (100 & 0-5 & 1 & \\ mL) & \\ Ac_2O & (100 & 0-5 & 1 & \\ mL) & \\ H_2SO_4 & (half & & \\ drop) & \\ Ac_2O & (160 & 60-70 & 2 & \\ mL) & \\ AcOH & (8 & & \\ mL) & \\ AcOH & (8 & & \\ mL) & \\ AcOH & (8 & & \\ mL) & \\ AcOH & (8 & & \\ mL) & \\ AcOH & (8 & & \\ mL) & \\ AcOH & (8 & & \\ mL) & \\ AcOH & (60 & & \\ mg) & \\ ZnCl_2 & (300 & & \\ mg) & \\ HCOOCOC- & 0-5 & 1 & \\ H_3 & (5 & mL) & \\ pyridine & (10 & & \\ mL) & \\ HCOOCOC- & 17 & 1 & \\ H_3 & (23 & & \\ mL) & \\ HCOONa & & \\ & (300 & mg) & \\ DMF & (70 & & \\ mL) & \\ ClCOOCH_3 & 6 & \\ & (6 & mL) & \\ pyridine & (30 & 0-5 & \\ \end{array}$	$\begin{array}{cccccccccc} Ac_2O & (2 & 17 & 1 & 72\% 5 \\ mL) \\ pyridine (10 & 24\% 3 \\ mL) \\ Ac_2O & (50 & 17 & 14 & 90\% 3 \\ mL) \\ pyridine (60 & \\mL) \\ Ac_2O & (70 & 60 & 50 \min & 50\% 4 \\ mL) & & \\ Ac_2O & (100 & 60 & 2 & 30\% 4 \\ mL) & & \\ Ac_2O & (100 & 60 & 2 & 30\% 4 \\ mL) & & \\ Ac_2O & (100 & 0-5 & 1 & 26\% 4 \\ mL) & & \\ H_2SO_4 & (half & 3\% 3 \\ drop) & & \\ S\% 5 \\ Ac_2O & (160 & 60-70 & 2 & 57\% 4 \\ mL) & & \\ AcOH & (8 & 18\% 3 \\ mL) & \\ AcOH & (8 & 18\% 3 \\ mL) & \\ AcOH & (8 & 18\% 3 \\ mL) & \\ AcOH & (60 & 5\% 5 \\ mg) & \\ ZnCl_2 & (300 & 3\% 3 \\ mg) & \\ HCOOCOCC & 0-5 & 1 & 26 4 \\ mL) & \\ AcOH & (60 & 5\% 5 \\ mg) & \\ ZnCl_2 & (300 & 3\% 3 \\ mg) & \\ HCOOCOCC & 17 & 1 & 14\% 7 \\ H_3 & (5 mL) & \\ pyridine & (10 \\ mL) & \\ HCOOCOC & 17 & 1 & 14\% 7 \\ H_3 & (23 \\ mL) & \\ HCOONa & 46\% 6 \\ & (300 mg) & \\ DMF & (70 & 14\% 7 \\ mL) & \\ HCOOCH_3 & 6 & 33\% 11 \\ & (6 mL) & \\ pyridine & (30 & 0-5 & 12\% 12 \\ \end{array}$

effect with gitoxigenin analogues is slightly larger—about a 15-fold increase, e.g., digitoxoside 19 vs. gitoxigenin (2).

Table II. Na ⁺ ,K ⁺ -ATPase Inhibition Studies	Table	϶II.	Na ⁺ ,	,K+-A	ATPase	Inhi	bition	Studies
--	-------	------	-------------------	-------	--------	------	--------	---------

In contrast, the effect of a β -digitoxose is apparently less if the 16 β group is esterified. Examples include 16 β formate 6 vs. its digitoxoside 24, about a 7-fold increase; 16 β -acetate 5 vs. its digitoxoside 22, about a 7-fold increase; and gitoxin 16 β -acetate (26) vs. gitoxigenin 16 β -acetate (5), about a 5-fold increase.

These data thus show a consistent pattern of activity enhancement for small C16 β -esters, particularly a C16 β formate. As other have also observed,⁴ a C16β-OH decreases digitoxigenin's activity. In preliminary crystallographic and computer graphics studies, we found that the decreased activity of gitoxigenin can be explained by its C17 side group carbonyl oxygen position.² (The gitoxigenin 16-one analogues have not yet been studied in detail.) On the other hand, observations with gitoxigenin C16-esters² show that their altered C17 side group carbonyl oxygen positions do not explain their enhanced biological activities. An additional factor is clearly important. One possibility could be a separate C16 ester binding site on the receptor. Alternatively, as suggested by DePover and Godfraind, C16-esters might bind to the same receptor site as the C17 side group.⁴ Studies are in progress to help resolve these two possibilities.

A new and unexpected finding is that modification of the steroid portion of a given cardenolide glycoside (especially $C16\beta$) may modify the effect of altering the C3-OH substituent. It is possible that some interaction via the C16 binding site may induce a conformational effect on the sugar binding site of the receptor. It is also conceivable that if the intrinsic activity of a genin is quite high, then the maximum apparent potentiation caused by addition of a C3-OH sugar would be correspondingly reduced.

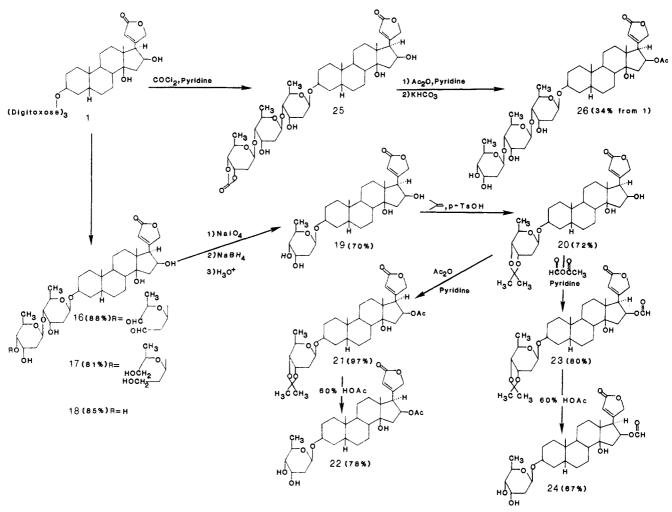
Experimental Section

Melting points were determined with a Thomas-Hoover melting point apparatus and are uncorrected. Elemental analyses were performed by MHW Laboratories, Phoenix, AZ. The 80-MHz ¹H NMR and 400-MHz ¹H and ¹³C spectra were taken at the Oregon State University NMR Spectroscopy Laboratory, Department of Chemistry. The 400-MHz spectra were taken on a Bruker AM 400 spectrometer. IR spectra were run as KBr pelles on a Beckman Model Acculab 7 spectrophotometer. Optical rotations in methanol were taken on a Perkin-Elmer 141 polarimeter. TLC was performed on 0.25-mm EM silica gel 60 F-254 glass plates. Flash column chromatography used silica gel 60, 230-400 mesh (EM Merck), in a 4 × 20 cm column.

 $(3\beta,5\beta,14\beta,16\beta,17\beta)$ -3,14,16-**Trihydroxycard**-20(22)-enolide (2) (Gitoxigenin). Gitoxin (1), 30 g (0.0384 mol), in 1.5 L of reagent methanol was heated on a steam bath for 15 min and then

analogue	hog kidney Na ⁺ , K ⁺ -ATPase inhbn data: K_{50} , M	approx rel act.
digitoxigenin	1.2×10^{-7}	5
gitoxigenin (2)	6.03×10^{-7}	1
gitoxigenin 3β -digitoxoside (19)	3.80×10^{-8}	15
gitoxigenin 3β -bisdigitoxoside (18)	5.19×10^{-8}	12
gitoxin (1)	6.99×10^{-8}	10
gitoxigenin 16β -formate (6)	2.00×10^{-8}	30
gitoxigenin 16 β -formate 3 β -digitoxoside (24)	3.13×10^{-9}	200
gitoxigenin 16 β -formate 3 β -acetate (9)	1.70×10^{-8}	35
gitoxigenin 3β , 16β -diformate (8)	1.48×10^{-8}	40
gitoxigenin 3β -formate (7)	6.76×10^{-7}	1
gitoxigenin 3β -acetate (4)	6.76×10^{-7}	1
gitoxigenin 16β -acetate	4.9×10^{-8}	12
gitoxigenin 3β , 16β -diacetate (3)	7.41×10^{-8}	8
gitoxigenin 16 β -acetate 3 β -formate (10)	9.55×10^{-8}	6
gitoxigenin 16 β -acetate 3 β -digitoxoside (22)	7.24×10^{-9}	85
gitoxigenin 3β -methoxycarbonate (11)	6.03×10^{-7}	1
gitoxigenin 3β , 16β -dimethoxycarbonate (12)	3.89×10^{-6}	0.2
gitoxin 16β -acetate (26)	1.02×10^{-8}	60
gitoxigenin-16-one 3β -acetate (14)	8.7×10^{-6}	0.1
17α -gitoxigenin-16-one 3β -acetate (15)	7.85×10^{-6}	0.1





1.0 L of 0.08 N H₂SO₄ was added and the solution heated for an additional 2.5 h. The reaction was followed by TLC (CH₂Cl₂-acetone, 2:1). The flask was cooled in ice, 200 mL of 5% NaHCO₃ was added, and the resulting suspension was concentrated to 500 mL in vacuo and placed in a freezer at -40 °C overnight. The resulting crstals of **2** were recrystallized in MeOH and CHCl₃, yield 10.16 g (68%), mp 223-225 °C (lit.¹⁷ mp 224-226 °C).

(3β,5β,14β,16β,17β)-3,16-Diacetoxy-14-hydroxycard-20-(22)-enolide (3) (Gitoxigenin 3β,16β-Diacetate). Acetic anhydride (50 mL) was added to a stirred solution of gitoxigenin (2) (10.05 g) in dry pyridine (60 mL) at room temperature. After stirring for 14 h, the reaction mixture was poured into water and stirred for 5 h at 0–5 °C. The crystals were filtered and recrystallized from MeOH-EtOAc to yield 3 (11.53 g, 90%): mp 247-250 °C (lit.¹⁷ mp 248-250 °C); $[\alpha]^{22}_{D}$ -28.57° (c 1.05, pyridine); IR (KBr) 3550 (OH), 1780, 1745, 1725 (C=O), 1635, 1655 (C=C) cm⁻¹; UV λ_{max} (MeOH) 219 nm (ϵ 13744); ¹H NMR (acetone-d₆) δ 5.93 (1 H, m, $W_{h/2} = 2$ Hz, C₂₂-H), 5.45 (1 H, ddd, J = 10, 10,2 Hz, C₁₆-H) 5.05 (1 H, m, C₃-H), 4.81, 5.09 (2 H, dd, J = 16, 2Hz, C₂₁-H), 3.23 (1 H, d, J = 10 Hz, C₁₇-H), 2.75 (1 H, dd, J =16, 10 Hz, C₁₅-H), 1.82 (1 H, m, C₁₅-H), 1.94, 2.03 (6 H, s, 2 OCOCH₃), 0.98 (6 H, s, C₁₈-H and C₁₉-H). Anal. (C₂₇H₃₈O₇) C, H.

 $(3\beta,5\beta,14\beta,16\beta,17\beta)$ -3-Acetoxy-14,16-dihydroxycard-20-(22)-enolide (4) (Gitoxigenin 3 β -Acetate). (a) Method A: Hydrolysis of $3\beta,16\beta$ -Diacetate 3 with KHCO₃. To a solution of 3 (9.0 g) in MeOH (2.7 L) was added a solution of KHCO₃ (5.4 g) in MeOH (2.16 L)-H₂O (0.54 L), and the mixture was stirred at room temperature for 4 days. HCl (0.05 N) was added dropwise to the stirred solution until pH 6.5 at 0-5 °C. The solution was concentrated in vacuo and extracted with CHCl₃. The CHCl₃ layer was washed with H₂O, dried, and evaporated in vacuo. The residue (8.73 g) was chromatographed on a silica gel column (CH₂Cl₂, with an increasing acetone content). Acetate 4 was eluted with acetone–CH₂Cl₂ (2:8) and recrystallized from MeOH: 5.33 g (65%); mp 228–232 °C (lit.¹³ mp 228–233 °C); [α]²²_D –3.49° (c 0.86, pyridine); IR (KBr) 3560, 3510, 3420, 3350 (OH), 1790, 1765, 1740 (C=O), 1635 (C=C) cm⁻¹; UV λ_{max} (MeOH) 220 nm (ϵ 14 136); ¹H NMR (CDCl₃) δ 5.90 (1 H, s, C₂₂-H), 5.03 (1 H, m, C₃-H), 4.98 (2 H, d, J = 16 Hz, C₂₁-H), 3.63 (1 H, d, J = 8 Hz, C₁₇-H), 2.85, 2.93 (2 H, s, C₁₄-OH and C₁₆-OH), 2.03 (3 H, s, OCOCH₃), 0.98 (6 H, s, C₁₈-H and C₁₉-H). Anal. (C₂₅H₃₆O₆) C, H.

(b) Method B: Acetylation of Gitoxigenin (2) with Acetic Anhydride and Acetic Acid. To a solution of 2 (8.00 g) in CHCl₃ (100 mL) was added Ac₂O (160 mL) and acetic acid (8 mL). It was stirred at 60–70 °C for 2 h, poured into ice water, and extracted with CHCl₃ (300 mL). The CHCl₃ was washed successively with H₂O, 5% NaHCO₃, and H₂O, dried, and evaporated in vacuo. The residue (8.10 g) was chromatographed (silica gel (200 g, with acetone–CH₂Cl₂ as eluent with acetone content increasing to 15%). After crystallization (MeOH), 1.72 g (17%) of diacetate 3 was obtained, mp 247–250 °C (lit.¹³ mp 248–250 °C).

The eluate with 20% acetone– CH_2Cl_2 contained 3 β -acetate 4, which was recrystallized from MeOH, yielding 5.046 g, (56%), mp 229–232 °C (lit.¹³ mp 228–235 °C).

(c) Method C: Acetylation of 2 with Acetic Anhydride. A solution of 2 (0.1 g, 2.6 mmol) in Ac₂O (10 mL) was stirred at 60 °C for 2 h. The mixture was concentrated to one-third the original volume in vacuum at 40 °C, H₂O (100 mL) was added, and the mixture was extracted with CH₂Cl₂. The combined extract was washed with 5% aqueous NaHCO₃ and H₂O, dried over MgSO₄, and concentrated to an amorphous powder. The crude amorphous powder was chromatographed, eluant 10% MeOH-CH₂Cl₂. Acetate 4 (R_1 0.55) was isolated and recrystallized by MeOH to afford 0.05 g (47%) of a white crystalline solid: mp 228-233 °C (lit.¹⁹ mp 228-232 °C). The minor product 3, R_1 0.75,

⁽¹⁷⁾ Hunger, A.; Reichstein, T. Helv. Chim. Acta 1950, 33, 76.

was isolated by eluting with 3% MeOH–CH₂Cl₂ and recrystallized by MeOH to afford 0.015 g (10%) of pure 3; mp 247–259 °C (lit.¹⁹ mp 248–250 °C).

(38,58,148,168,178)-16-Acetoxy-3,14-dihydroxycard-20-(22)-enolide (5) (Gitoxigenin 16β -Acetate). To a solution of 2 (1.018 g) in dry pyridine (10 mL) was added acetic anhydride (2 mL), and the mixture was stirred at room temperature for 1 h, poured into water, and extracted with CHCl₃. The CHCl₃ was washed with 1 N HCl, H₂O, 5% NaHCO₃, and H₂O, dried, and evaporated in vacuo. The residue (1.05 g) was chromatographed on a silica gel (50 g) column (MeOH-CH₂Cl₂ as eluent with an increasing MeOH content). The product (obtained in the MeOH--CH₂Cl₂ (2.5:7.5 fractions) was recrystallized from MeOH, yielding gitoxigenin 3 β ,16 β -diacetate (3) (294 mg, 23%), mp 247-250 °C (lit.¹⁷ mp 248-250 °C). The fractions with MeOH-CH₂Cl₂ (5:9.5) were recrystallized from EtOAc, yielding gitoxigenin 16β-acetate (5) (812 mg, 72%): mp 227-229 °C (lit.^{13,14} mp 226-229 °C); [α]²²_D -26.1° (c 1.11, pyridine); IR (KBr) 3525, 3390 (OH), 1795, 1755 (C=O), 1617 (C=C) cm⁻¹; UV λ_{max} (MeOH) 219 nm (ϵ 12710); ¹H NMR (acetone- d_6) δ 5.90 (1 H, m, $W_{h/2}$ = 2 Hz, C_{22} -H), 5.45 (1 H, ddd, J = 9, 9, 2 Hz, C_{16} -H), 4.94 (2 H, dd, H = 16, 2 Hz, C_{21} -H), 4.03 (1 H, m, C_{3} -H), 3.21 (1 H, d, J = 9 Hz, C_{17} -H), 2.98 (1 H, s, 14-OH), 2.78 (1 H, dd, J = 16, 10 Hz, C_{15a} -H), 1.81 (1 H, dd, J = 16, 9 Hz, C_{15b} -H), 1.93 (3 H, s, OOCCH₃), 0.98 $(3 \text{ H}, \text{ s}, \text{ C}_{19}\text{-}\text{H}), 0.93 (3 \text{ H}, \text{ s}, \text{ C}_{18}\text{-}\text{H}).$ Anal. $(\text{C}_{25}\text{H}_{36}\text{O}_6) \text{ C}, \text{ H}.$

(3\$\beta,5\$\beta,14\$\beta,16\$\beta,17\$\beta)-3,14-Dihydroxy-16-(formyloxy)card-20(22)-enolide (6) (Gitoxigenin 16\$-Formate, Gitaloxigenin), (3\$\beta,5\$\beta,14\$\beta,16\$\beta,17\$\beta)-14,16-Dihydroxy-3-(formyloxy)card-20-(22)-enolide (7) (Gitoxigenin 3β -Formate), and (3\$\beta, 5\$\beta, 14\$\beta, 16\$\beta, 17\$\beta)-3, 16-Bis(formyloxy)-14-hydroxycard-20-(22)-enolide (8) (Gitoxigenin 3β , 16β -Diformate). a. Acetic Formic Anhydride. To a dry 1-L, three-necked, round-bottomed flask were added 100 g of sodium formate and 200 mL of anhydrous ether. To this stirred mixture was added 90 mL of distilled acetyl chloride for 5 min, while the temperature was maintained at 23-27 °C. After the addition was complete, the mixture was stirred for 6 h at 23-27 °C and filtered, the solid residue rinsed with 100 mL of dry ether, and the ether removed in vacuo at 0-5 °C. The crude product was distilled to yield 73.5 g (65%) of acetic formic anhydride as a colorless liquid: bp 27–28 °C (10 mmHg) (lit.¹⁸ bp 25–28 °C); IR (neat) 1210, 1750 (C=O), 1060 (C—O—C) cm⁻¹; ¹H NMR (CDCl₃) δ 11.7 (s, 1 H, CH=O), 2.1 (s, 3 H, CH₃).

b. Synthesis of Compounds 6-8. To a solution of 3.017 g of 2 in 70 mL of dry dimethylformamide were added 22.5 mL of acetic formic anhydride and 300 mg of sodium formate. After stirring at room temperature for 1 h, the reaction mixture was poured into ice-H₂O, extracted with CHCl₃, dried over MgSO₄, and evaporated in vacuo to give an oil (3.10 g). The crude oil was chromatographed (silica gel, 200 g) with acetone-CH₂Cl₂. The eluant of 15% acetone-CH₂Cl₂ was evaporated. Recrystallization (EtOAc) gave 652 mg (18%) of 8 as colorless needles: mp 222-224 °C (lit.¹⁹ mp 221-224 °C); [α]²²_D-21.85° (c 1.19, pyridine); UV λ_{max} (MeOH) 218 nm (ϵ 15.332); IR (KBr) 3500 (OH), 1795, 1740, 1725, 1700 (C=O), 1632 (C=C) cm⁻¹; ¹H NMR (pyridine-d₅) δ 8.18, 8.35 (2 H, s, CHO), 6.30 (1 H, m, $W_{h/2} = 2$ Hz, C₂₂-H), 5.88 (1 H, m, C₃-H), 5.65 (1 H, m, C₁₆-H), 5.28 (2 H, m, C₂₁-H), 3.43 (1 H, d, J = 9 Hz, C₁₇-H), 2.84 (1 H, dd, J = 16, 9 Hz, C₁₅-H), 0.89 (3 H, s, C₁₈-H), 1.09 (3 H, s, C₁₉-H). Anal. (C₂₅H₃₄O₇) C, H.

The eluant with 20% acetone– CH_2Cl_2 was evaporated. Recrystallization (EtOAc) gave 445 mg (13%) of 7 as colorless needles: mp 207–210 °C (lit.¹⁹ mp 207–210 °C); $[\alpha]^{22}_D$ –5.1° (c 0.98, pyridine); UV λ_{max} (MeOH) 220 nm (ϵ 14 888); IR (KBr) 3550, 3500, 3425, 3340 (OH), 1790, 1765, 1730, 1715 (C=O), 1632 (C=C) cm⁻¹; ¹H NMR (CDCl₃) δ 8.03 (1 H, s, C₃-OOCH), 5.93 (1 H, s, C₂₂-H), 5.20 (1 H, m, C₃-H), 4.98 (2 H, m, C₂₁-H), 4.50 (1 H, m, C₁₆-H), 3.60 (1 H, d, J = 8 Hz, C₁₇-H), 2.81, 2.93 (2 H, s, C₁₄-OH and C₁₆-OH), 2.43 (1 H, dd, J = 16, 7 Hz, C₁₅-H), 0.98 (6 H, s, C₁₈-H and C₁₉-H). Anal. (C₂₄H₃₄O₆) C, H.

The eluant of 25% acetone– $CH_2\dot{Cl}_2$ was evaporated. Recrystallization gave 1.476 g (45%) of **6** as colorless plates: mp 214–216 °C (lit.¹⁹ mp 215–217 °C); $[\alpha]^{22}_{D}$ –16.47° (c 0.85, pyridine); IR

(KBr) 3550, 3430 (OH), 1795, 1740, 1710 (C=O), 1630 (C=C) cm⁻¹; UV λ_{max} (MeOH) 218 nm (ϵ 14 844); ¹H NMR (acetone- d_6) δ 7.93 (1 H, s, OCHO), 5.90 (1 H, m, $W_{h/2} = 2$ Hz, C_{22} -H), 5.58 (1 H, ddd, J = 9, 9, 2 Hz, C_{16} -H), 4.80, 5.05 (2 H, dd, J = 16, 2 Hz, C_{21} -H₂), 4.05 (1 H, m, C_3 -H), 3.28 (1 H, d, J = 9 Hz, C_{17} -H), 2.98, 3.23 (2 H, s, C_3 -OH and C_{14} -OH), 2.80 (1 H, dd, J = 16, 9 Hz, C_{16} -H), 0.95 (3 H, s, C_{18} -H), 1.00 (3 H, s, C_{19} -H). Anal. ($C_{24}H_{34}O_6$) C, H.

c. Alternate Synthesis of 8. To a solution of gitoxigenin (2) (1.017 g) in dry pyridine (10 mL) was added acetic formic anhydride (5 mL), and the mixture was stirred at 0-5 °C for 1 h. The reaction mixture was poured into water, stirred for 30 min, and filtered, and the crude crystals were recrystallized (MeOH-EtOAc) to yield 932 mg (80%) of 8 as colorless needles, mp 222-224 °C (lit.¹⁹ mp 221-224 °C).

 $(3\beta,5\beta,14\beta,16\beta,17\beta)$ -3-Acetoxy-16-(formyloxy)-14hydroxycard-20(22)-enolide (9) (Gitoxigenin 3\beta-Acetate 16_β-Formate). Method A. To a stirred mixture of 4 (0.3 g, 0.7 mmol), 2,6-bis(methylamino)pyridine (0.1 g), and Et₃N (0.6 mL) in CH₂Cl₂ (5 mL) was slowly added formic acetic anhydride (2 mL, 23 mmol) at 0 °C. The mixture was stirred for 20 min (0-10 °C), added to ice- H_2O , and extracted with CH_2Cl_2 . The CH_2Cl_2 extract was washed with 5% aqueous NaHCO3 and H2O, dried over MgSO₄, and concentrated to an oil. The crude oil was chromatographed (eluant 10% EtOAc-CH₂Cl₂) to yield 0.305 g (96%) of pure 9, recrystalized (MeOH) to white crystalline plates: mp 240–241 °C; $[\alpha]_{D}^{22}$ –17.86° (c 1.40, pyridine); UV λ_{max} (MeOH) 218.5 (log ε 4.16); IR (KBr) 3300 (OH), 2940 (CH), 1730 (α,βunsaturated C==O), 1760 (formyl, C==O), 1615 (C==C) cm⁻¹; ¹H NMR (CDCl₃) δ 7.9 (1 H, s, CHO), 5.95 (1 H, s, C₂₂-H), 5.6 (1 H, dd, C_{16} -H), 4.9 (2 H, dd, $J_{21,22}$ = 16 Hz, C_{21} -H), 3.45 (1 H, d, C_{17} -H), 2.84 (1 H, dd, J = 16, 9 Hz, C₁₅-H), 2.03 (3 H, s, acetate), 0.99 (6 H, s, C_{18} - and C_{19} - CH_3), ¹H NMR (pyridine- d_5) δ 8.15 (1 H, s, C_{16} -OCH), 6.28 (1 H, s, C_{22} -H), 5.76 (1 H, m, C_3 -H), 5.33 (2 H, m, C₂₁-H), 3.41 (1 H, d, C₁₇-H), 2.84 (1 H, dd, J = 16, 9 Hz, C₁₅-H), 2.08 (3 H, s, OCOCH₃), 0.90 (3 H, s, C₁₈-H), 1.09 (3 H, s, C19-H). Anal. (C₂₆H₃₀O₇) C, H.

Method B. To a stirred mixture of 4 (0.15 g, 0.35 mmol), DMAP (0.1 g), and Et_3N (0.8 mL) in dioxane (10 mL) was added formic acid (3 mL) and acetic anhydride (2 mL) at 22 °C. The mixture was then refluxed at 90 °C for 45 min, cooled, added to ice-H₂O, and extracted with CH₂Cl₂. The combined extract was washed with 5% aqueous NaHCO₃, 1 N HCl and H₂O, dried over MgSO₄, and concentrated to an oil, which was chromatographed (eluant 5% EtOAc-CH₂Cl₂) to yield 0.075 g (49%) of pure 9 along with diacetyl derivative 3 (50%).

Method C. To a solution of 300 mg of 3-acetylgitoxigenin (4) in dry pyridine (5 mL) was added acetic formic anhydride (2 mL). It was left at 0–5 °C for 2 h and then poured into ice water and extracted with CHCl₃. The chloroform layer was washed successively with 1 N HCl, H_2O , 5% NaHCO₃, and H_2O , dried, and evaporated. The residue was recyrstallized (MeOH) to give 275 mg (86%) of **9** as colorless needles, mp 241–242 °C.

 3β , 5β , 14β , 16β , 17β) - 16-Acetoxy-3-(formyloxy)-14hydroxycard-20(22)-enolide (10) (Gitoxigenin 3*β*-Formate 16 β -Acetate). To a solution of 580 mg of 16-acetylgitoxigenin (5) in pyridine (10 mL) was added acetic formic anhydride (4 mL) at 0-5 °C. The resultant solution was stirred at 0-5 °C for 2 h, added to ice-H₂O, and extracted with $CHCl_3$ (2 × 100 mL). The CHCl₃ extract was washed with 1 N HCl, H₂O, 5% NaHCO₃, and H_2O , dried over MgSO₄, and evaporated to an oil. The crude oil was purified via flash chromatography (CH₂Cl₂, 5% EtOAc- $CH_2Cl_2,\,10\%$ EtOAc-CH_2Cl_2) to yield 521 mg (84.4%) of 10 as colorless needles: mp 224–227 °C; $[\alpha]^{22}_{D}$ –39.77° (c 0.88, pyridine); IR (KBr) 3525 (OH), 1785, 1745, 1710 (C=O), 1630 (C=C) cm⁻¹; UV λ_{max} (MeOH) 218 nm (ϵ 14 241); ¹H NMR (CDCl₃) δ 8.03 (1 H, s, $\overline{C_3}$ -OCHO), 5.90 (1 H, s, C_{22} -H), 5.45 (1 H, ddd, J = 9, 9, 2 Hz, C_{16} -H), 5.20 (1 H, m, C_{3} -H), 5.03, 4.78 (2 H, d, J = 16 Hz, C_{21} -H), 3.18 (1 H, d, J = 9 Hz, C_{17} -H), 2.70 (1 H, dd, J = 16, 10 Hz, C₁₅-H), 1.95 (3 H, s, OCOCH₃), 0.95, 0.93 (6 H, s, C₁₈-H and C_{19} -H). Anal. ($C_{25}H_{34}O_7$) C, H.

 $(3\beta,5\beta,14\beta,16\beta,17\beta)$ -14,16-Dihydroxy-3-[(methoxycarbonyl)oxy]card-20(22)-enolide (11) (Gitoxigenin 3-Methoxycarbonate) and $(3\beta,5\beta,14\beta,16\beta,17\beta)$ -14-Hydroxy-3,16-bis(methoxycarbonyl)oxy]card-20(22)-enolide (12) (Gitoxigenin 3,16-Dimethoxycarbonate). To a solution of 2.00

⁽¹⁸⁾ Krimen, L. I. Org. Synth. 1970, 51, 1.

⁽¹⁹⁾ Haack, E.; Kaiser, F.; Spingler, H. Chem. Ber. 1956, 89, 1353.

g of gitoxigenin (2) in pyridine (30 mL) was added ClCOOCH₃ (6 mL) at 0–5 °C. It was stirred at 0–5 °C for 6 h, poured into ice–H₂O, and extracted with CHCl₃. The CHCl₃ layer was washed successively with 1 N HCl, H₂O, 5% NaHCO₃, and H₂O, dried over MgSO₄, and evaporated to an oil (2.05 g). The crude oil was purified via flash chromatography on silica gel (20% EtOAc-CH₂Cl₂). The fractions with 12 (R_f 0.55) were recrystallized (MeOH) to give 0.30 g (11%) of 12: mp 213–214.5 °C; $[\alpha]^{22}_{\rm D}$ –15.29° (c 0.85, pyridine); IR (KBr) 3550 (OH), 1780, 1745, 1735 (C=O), 1650, 1635 (C=C); UV $\lambda_{\rm max}$ (MeOH) 218.5 nm (ϵ 14685); ¹H NMR (CDCl₃) δ 5.94 (1 H, s, C₂₂-H), 5.35 (1 H, dd, J = 9, 9, 2 Hz, C₁₆-H), 4.95 (2 H, m, C₂₁-H), 4.95 (1 H, m, C₃-H), 3.76 (3 H, s, C₃-OCOOCH₃), 3.69 (3 H, s, C₁₆-OCOOCH₃), 3.13 (1 H, d, J = 9 Hz, C₁₇-H), 2.73 (1 H, dd, J = 16, 9 Hz, C₁₅-H), 0.95 (6 H, s, C₁₈-H and C₁₉-H). Anal. (C₂₇H₃₈O₉) C, H.

The fractions with $R_f 0.25$ were recrystallized (MeOH) to give 0.83 g (33.0%) of 11: mp 213-214.5 °C; $[\alpha]^{22}_D + 2.78^{\circ}$ (c 1.08, pyridine); IR (KBr) 3520, 3425 (OH), 1790, 1750, 1730 (C=O), 1620 (C=C) cm⁻¹; UV λ_{max} (MeOH) 220.5 nm (ϵ 15018); ¹H NMR (pyridine- d_6) δ 6.19 (1 H, s, C₂₁-H), 5.58 (2 H, m, C₂₂-H), 5.06 (1 H, m, C₃-H), 4.95 (1 H, m, C₁₆-H), 3.75 (3 H, s, C₃-OCOOCH₃), 3.25 (1 H, d, J = 8 Hz, C₁₇-H), 2.70 (1 H, dd, J = 15, 8 Hz, C₁₅-H), 1.11 (3 H, s, C₁₉-H), 0.88 (3 H, s, C₁₈-H). Anal. (C₂₇H₃₈O₈) C, H. In addition, 0.54 g (27%) of gitoxigenin (2) was also recovered.

 $(3\beta,5\beta,14\beta,16\beta,17\beta)$ -3-Acetoxy-14-hydroxy-16-[(methoxy-

carbonyl)oxy]card-20(22)-enolide (13) (Gitoxigenin 3β -Acetate 16β -Methoxycarbonate). To a solution of 500 mg of 3-acetylgitoxigenin (4) in pyridine (5 mL) was added ClCOOCH₃ (2 mL) at 0-5 °C for 5 h. The reaction mixture was poured into ice-H₂O and extracted with CHCl₃. The CHCl₃ was washed successively with 1 N HCl and H₂O, dried, and evaporated, and the residue (0.52 g) was dissolved in 30 mL of CH₂Cl₂. To the solution was added 4-(dimethylamino)pyridine ($D ilde{M}A ilde{P}$) (0.1 g), triethylamine (0.6 mL), and ClCOOCH₃ (2.0 mL) at 0-5 °C. The solution was stirred at room temperature for 5 h, poured into ice water, and extracted with CH₂Cl₂. The CH₂Cl₂ was washed with H₂O, dried, and evaporated to give an oil. The oil was chromatographed on silica gel (40% EtOAc-CH2Cl2) and recrystallized from EtOAc-Et₂O to give 328 mg (57%) of 13: mp 220-222 °C; $[\alpha]^{22}$ _D -20.20° (c 0.99, pyridine); UV λ_{max} (MeOH) 218 nm (ϵ 14446); IR (KBr) 3525 (OH), 1780, 1750, 1720 (C-O), 1635 (C-C) cm^{-1} ; ¹H NMR (CDCl₃) δ 5.95 (1 H, s, C₂₂-H), 5.35 (1 H, ddd, J = 10, 10, 2 Hz, C₁₆-H), 5.03 (1 H, m, C₃-H), 4.93 (2 H, m, C₂₁-H), 3.68 (3 H, s, C_{16} -OCOOCH₃), 3.20 (1 H, d, J = 9 Hz, C_{17} -H), 2.70 $(1 \text{ H}, \text{dd}, J = 16, 10 \text{ Hz}, \text{C}_{15}\text{-}\text{H}), 2.03 (3 \text{ H}, \text{s}, \text{OCOCH}_3), 0.93, 0.95$ (6 H, s, C₁₈-H and C₁₉-H). Anal. (C₂₇H₃₈O₈) C, H.

(3β,5β,14β,17β)-3-Acetoxy-14-hydroxy-16-ketocard-20-(22)-enolide (14) and (3β,5β,14β,17α)-3-Acetoxy-14-hydroxy-16-ketocard-20(22)-enolide (15). To 1.00 g of 3-acetylgitoxigenin (4) in 70 mL of acetone was added 0.8 mL of Jones reagent (CrO₃-H₂SO₄) at -10 to 0 °C, stirred at -10 to 0 °C for 3 min and then added to ice-H₂O. The quenched reaction was extracted with CHCl₃, and the extract was washed with 200 mL of H₂O, dried over MgSO₄, and concentrated to crude crystals (1.00 g), which were recrystallized (acetone-Et₂O) to give 525 mg (52%) of 14: mp 209-211 °C (lit.²⁰ mp 195-210 °C); [a]²²_D+73.0° (c 0.97), acetone); IR (KBr) 3550 (OH), 1785, 1748, 1710 (C==O), 1635 (C==C); UV λ_{max} (MeOH) 216 nm (ε 10010); ¹H NMR (CDCl₃) δ 5.94 (1 H, m, W_{h/2} = 2 Hz, C₂₂-H), 5.05 (1 H, m, C₃-H), 4.61, 4.98 (2 H, dd, J = 18 and 2 Hz, C₂₁-H), 3.08 (1 H, s, C₁₄-OH), 2.05 (3 H, s, OCOCH₃), 1.00 (3 H, s, C₁₈-H), 1.05 (3 H, s, C₁₈-H). Anal. (C₂₂H₃₄O₆) C, H.

The mother liquor was concentrated to dryness (470 mg) and then recrystallized from acetone-Et₂O, giving 107 mg (10.8%) of 15: mp 232-235 °C (lit.²⁰ mp 232-235 °C); $[\alpha]^{22}_{D} + 131.5^{\circ}$ (c 0.20, acetone); IR (KBr) 3490 (OH), 1788, 1755 (lactone C=O), 1740 (OCCH₃), 1705 (ketone), 1633 (C=C); UV λ_{max} (MeOH) 216 nm (ϵ 9620); ¹H NMR (CDCl₃) δ 5.91 (1 H, m, $W_{h/2} = 2$ Hz, C₂₂-H), 5.03 (1 H, m, C₃-H), 4.64, 4.99 (1 H, dd, J = 18 Hz, C₁₅-H), 2.28, 2.68 (2 H, d, J = 18 Hz, C₁₅-H), 2.39 (1 H, s, C₁₇-H), 2.28, 2.68 (2 H, d, J = 18 Hz, C₁₅-H), 2.39 (1 H, s, C₁₄-OH), 2.04 (3 H, s, OCOCH₃), 1.25 (3 H, s, C₁₈-H), 1.00 (3 H, s, C₁₉-H). Anal. (C₂₅H₃₄O₆) C, H.

(20) Satoh, D.; Morita, J. Yakugaku Zashi 1962, 82, 156.

 $(3\beta,5\beta,14\beta,16\beta,17\beta)-3-[(O-2,6-Dideoxy-\beta-D-ribo-hexo$ $pyranosyl-(1-4)-O-2,6-dideoxy-\beta-D-ribo-hexopyranosyl)$ oxy]-14,16-dihydroxycard-20(22)-enolide (18) was preparedby using a modification of the Satoh and Aoyama¹⁴ method usedfor stepwise degradation of digitoxin and gitoxin.

(i) Oxidation of Gitoxin (1) with NaIO₄. To a solution of gitoxin (1) (5.0 g, 6.40 mmol) in MeOH (1500 mL) was added a solution of sodium metaperiodate (5 g) in H₂O (50 mL). The resultant mixture was stirred at 22 °C for 2 h and precipitated NaIO₃ removed by filtration. The filtrate was concentrated under vacuum to about one-third its original volume and extracted with CH₂Cl₂, and the CH₂Cl₂ extract was washed with H₂O, dried over MgSO₄, and concentrated to give crude dialdehyde 16, 4.397 g (88%), as a white amorphous powder. It was then recrystallized (acetone/*n*-hexane): mp 183–185 °C; IR (KBr) 3400 (OH), 1725 (α,β -unsaturated C=O), 1740 (CHO), 1620 (C=C) cm⁻¹.

(ii) **Reduction of 16 with NaBH**₄. To a solution of dialdehyde 16 (5.0 g, 6.42 mmol) in MeOH (500 mL) was added NaBH₄ (0.59 g) in small portions, and the mixture was stirred for 2 h and then quenched with 5% HOAc to pH 6.5. The mixture was concentrated at 40 °C under vacuum and extracted with CH₂Cl₂, and the CH₂Cl₂ extract was washed with H₂O, dried over MgSO₄, and evaporated in vacuo to give crude dimethylol 17, 4.503 g (81%), as a white powder: mp 185 °C (acetone/*n*-hexane); IR (KBr) 3450 (OH), 1730 (α , β -unsaturated C=O, 1630 (C=C) cm⁻¹.

(iii) Hydrolysis of 17. To a solution of dimethylol 17 (4.5 g, 5.6 mmol) in MeOH (500 mL) was added with stirring at 22 °C 0.05 N HCl (75 mL) and the reaction mixture stirred for 24 h. The reaction mixture was neutralized with 5% aqueous NaHCO₃ to pH 6.5, concentrated in vacuo to one-third its original volume, and then extracted with CH₂Cl₂. The CH₂Cl₂ extract was washed with H_2O , dried over MgSO₄, and evaporated at 20 °C in vacuo to give crude 18. It was chromatographed (silica gel, eluant 9% EtOH-CH₂Cl₂) and recrystallized from CH_2Cl_2 -n-hexane to give pure 18, yield 3.847 g (85%), as white granular powder: mp 198–201 °C; UV λ_{max} (MeOH) 218 nm; $[\alpha]^{22}_{D}$ +10.13° (c 0.27 MeOH); IR (KBr) 3440 (OH), 1725 (α,β-unsaturated C=O), 1620 (C=C) cm⁻¹; ¹H NMR (pyridine- d_5) δ 6.18 (1 H, s, C₂₂-H), 5.53 $(2 \text{ H}, \text{d}, J = 16 \text{ Hz}, \text{C}_{21}\text{-}\text{H}), 5.37 (1 \text{ H}, \text{m}, \text{C}_{16}\text{-}\text{H}), 4.60 (2 \text{ H}, \text{m}, \text{m})$ C_3 -H and $C_{3'}$ -H), 4.35 (1 H, m, $C_{4'}$ and $C_{4'}$ -H), 4.19 (1 H, m, C_3 -H), 3.48 (2 H, m, $C_{5'}$ -H and $C_{5''}$ -H), 3.23 (1 H, d, J = 8 Hz, C_{17} -H), 2.68 (2 H, dd, J = 9, 16 Hz, C₁₅-H), 1.39 and 1.44 (6 H, d, J =6 Hz, C_{6} -H and $C_{6'}$ -H), 1.06 (3 H, s, C_{19} -H), 0.83 (3 H, s, C_{18} -H). Anal. $(C_{35}H_{54}O_{11})$ C, H.

(3β,5β,14β,16β,17β)-3-[(2,6-Dideoxy-β-D-ribo-hexopyranosyl)oxy]-14,16-dihydroxycard-20(22)-enolide (19) was prepared from 18 following a similar procedure as used for producing 18 from gitoxin (1). Gitoxigenin bisdigitoxoside (18) was oxidized by NaIO₄ to the corresponding dialdehyde in 75% yield, which on reduction with NaBH_4 and subsequent acid hydrolysis with 0.05 N HCl gave crude gitoxigenin monodigitoxoside 19 in 70% overall yield. The crude 19 was purified by column chromatography, eluant 20% MeOH-CH2Cl2 (increasing in MeOH), and recrystallized from EtOH-Et₂O: mp 214-216 °C; $[\alpha]^{22}$ _D +10.13 (c 0.27, MeOH); IR (KBr) 3500 (OH), 1730 (α,β-unsaturated C==O), 1615 (C==C) cm⁻¹; ¹H NMR (pyridine-d₅) δ 6.19 (1 H, s, C₂₂-H), 5.54 (2 H, d, J = 16 Hz, C₂₁-H), 4.40 (1 H, m, C₃-H), 4.24 (1[°]H, m, C₃-H), 4.18 (1 H, m, C₄-H), 3.57 (1 H, m, C₅-H), 3.24 (1 H, d, J = 8 Hz, C₁₇-H), 2.68 (1 H, dd, J = 16, 10 Hz, C₁₅-H), 1.56 (3 H, d, J = 6 Hz, $C_{6'}$ -CH₃), 1.08 (3 H, s, C_{19} -H), 0.88 (3 H, s, C₁₈-H). Anal. (C₂₉H₄₄O₈) C, H.

(3β,5β,14β,16β,17β)-3-[(2,6-Dideoxy-3,4-O-(1-methylethylidene)-β-D-*ribo*-hexopyranosyl)oxy]-14,16-dihydroxycard-20(22)-enolide (20). To a solution of 19 (2.5 g, 4.8 mmol) in anhydrous acetone (60 mL) was added p-TsOH (25 mg) at 22 °C with stirring. The reaction mixture was stirred for 5 h at 22 °C; by this time the reaction was almost complete as evident from TLC (10% MeOH/CH₂Cl₂). The mixture was concentrated to half the volume in vacuo at 30 °C and to the residue was added H₂O (5 mL). The precipitated solid was extracted with CH₂Cl₂, and the CH₂Cl₂ extract was washed with H₂O and dried over MgSO₄. The product was purified by column chromatrography (eluant 10% EtOAc-CH₂Cl₂) to give pure 20: yield 2.2 g (72%); mp 214-216 °C; (α]²²_D +13.12 (c 0.15 MeOH); UV λ_{max} (MeOH) 218 nm; IR (KBr) 3490 (OH), 1730 (α ,β-unsaturated C=O), 1615 (C=C) cm⁻¹; ¹H NMR (CDCl₃) δ 5.91 (1 H, s, C₂₂-H), 4.98 (2 H, d, J = 16 Hz, C_{21} -H), 4.40 (1 H, m, C_{16} -H), 4.0 (1 H, m, C_{8} -H), 4.75 (1 H, dd, J = 8, 3 Hz, C_{1} -H), 4.40 (1 H, m, C_{3} -H), 3.53 (1 H, dd, J = 9, 5 Hz, C_{4} -H), 3.48 (1 H, m, C_{5} -H), 2.90 (1 H, m, C_{17} -H), 2.40 (1 H, dd, J = 7, 6 Hz, C_{15} -H), 1.35 and 1.48 (6 H, s, acetonide CH₃), 1.24 (3 H, d, J = 6 Hz, C_{6} -H), 0.94 (6 H, s, C_{18} -H and C_{19} -H). Anal. (C_{32} H₄₈O₈) C, H.

 $(3\beta, 5\beta, 14\beta, 16\beta, 17\beta) - 3 - [(2, 6 - Dideoxy - 3, 4 - O - (1 - methyl - 1))]$ ethylidene)-β-D-ribo-hexopyranosyl)oxy]-14-hydroxy-16-(formyloxy)card-20(22)-enolide (23). Method A. To a stirred mixture of 20 (0.45 g, 0.8 mmol), DMAP (0.2 g), and Et₃N (1.0 mL) in CH₂Cl₂ (10 mL) was slowly added formic acetic anhydride (2.5 mL, 29 mmol) at 10 °C. The mixture was stirred for 30 min, added to ice-H₂O, and extracted with CH₂Cl₂. The combined extract was washed with 5% aqueous $NaHCO_3$ and H_2O , dried over MgSO4, and concentrated to an oil. The crude oil was chromatographed (eluant 10% EtOAc-CH₂Cl₂) to yield 0.3 g (62%) of pure 23; after recrystallization (EtOAc-Et₂O): mp 197–198 °C; $[\alpha]^{22}_{D}$ +13.347 (c 0.29, MeOH); UV λ_{max} (MeOH) 218 (log ϵ 4.21); IR (KBr) 3440 (OH), 1765 (C=O), 1730 (α,β -unsaturated C=O), 1620 (C=C) cm⁻¹; ¹H NMR (CDCl₃) δ 7.9 (1 H, s, CHO), 5.95 (1 H, s, C_{22} -H), 5.60 (1 H, dd, J = 10, 10, 2 Hz, C_{16} -H), 4.94 (2 H, d, J = 16 Hz, C_{21} -H), 4.75 (1 H, dd, J = 8, 3Hz, C₁-H), 4.40 (1 H, m, C₃-H), 4.01 (1 H, m, C₃-H), 3.54 (1 H, dd, J = 9, 5 Hz, C_{4'}-H), 3.53 (1 H, m, C₅-H), 2.78 (1 H, dd, J =10, 16 Hz, C₁₅-H), 1.36 and 1.49 (6 H, s, acetonide CH₃'s), 1.26 $(3 \text{ H}, d, J = 6 \text{ Hz}, C_{6}\text{-H}), 0.98 (3 \text{ H}, \text{ s}, C_{18}\text{-H}), 0.94 (3 \text{ H}, \text{ s}, C_{19}\text{-H}).$ Anal. (C₃₃H₄₈O₉) C, H.

Method B. To a solution of 20 (0.50 g) in dry pyridine (6 mL) was added acetic formic anhydride (5 mL) at 0-5 °C, and the mixture was stirred at 0-5 °C for 2 h and poured into ice water, and crystals were filtered. The crystals were recrystallized (Et-OAc-Et₂O) to yield 420 mg (80%) of 23 as colorless needles, mp 197-199 °C.

(3β,5β,14β,16β,17β)-3-[(2,6-Dideoxy-β-D-*ribo*-hexopyranosyl)oxy]-14-hydroxy-16-(formyloxy)card-20(22)-enolide (24). Formate 23 (0.49, 0.68 mmol) was stirred at 22 °C with 60% HOAc (54 mL) for 5 h, monitored with TLC (10% EtOH- CH_2Cl_2). The reaction was poured in ice-cold H_2O and extracted with CHCl₃, and the combined CHCl₃ extract was washed with H_2O , 5% NaHCO₃, and H_2O and dried over MgSO₄. The product was concentrated to a crude oil and flash chromatographed (7.5% EtOH-CH₂Cl₂) to yield pure 24 (0.25 g, 67%) as a white amorphous powder: mp 145–150 °C; $[\alpha]^{28}_D$ –14.07° (c 0.32, MeOH); UV λ_{max} (MeOH) 218 nm (log ϵ 4.22); IR (KBr) 3500 (OH), 1740 (CHO), 1710 (α,β -unsaturated C=O), 1615 (C=C); ¹H NMR (CDCl₃) δ 7.9 (1 H, s, CHO), 5.95 (1 H, s, C_{22} -H), 5.58 (1 H, ddd, J = 10, 10, 2 Hz, C_{16} -H), 4.95 (2 H, d, J = 16 Hz, C_{21} -H), 4.0 (1 H, m, C_{3} -H), 4.88 (1 H, dd, J = 8, 2 Hz, $C_{1'}$ -H), 4.05 (1 H, m, $C_{3'}$ -H), 3.78 (1 H, dd, J = 9, 6 Hz, $C_{4'}$ -H), 3.25 (1 H, m, $C_{5'}$ -H), 3.23 (1 H, d, J= 9 Hz, C_{17} -H), 1.29 (3 H, d, J = 6 Hz, C_{6} -CH₃). Anal. (C_{30} H₄₄O₉) C, H.

 $(3\beta,5\beta,14\beta,16\beta,17\beta)$ -3-[(2,6-Dideoxy-3,4-O-(1-methylethylidene)- β -D-*ribo*-hexopyranosyl)oxy]-16-acetoxy-14-hydroxycard-20(22)-enolide (21). To a solution of 0.50 g of 20 in pyridine (6 mL) was added acetic anhydride (6 mL). The resultant solution was stirred at room temperature for 5 h and then added to ice-H₂O. The crystals were filtered and recrystallized (MeOH-Et₂O) to give 449 mg (96%) of 21 as colorless needles: mp 192-195 °C; $[\alpha]^{19}_D$ -7.09 (c 0.13, MeOH); IR (KBr) 3400 (OH), 1785, 1760, 1750 (sh) (C=O), 1645, 1625 (C=C) cm⁻¹; ¹H NMR (CDCl₃) δ 5.93 (1 H, s, C₂₂-H), 5.45 (1 H, ddd, J = 10, 10, 2 Hz, C₁₆-H), 4.90 (2 H, d, J = 16 Hz, C₂₁-H), 4.73 (1 H, dd, J = 8, 3 Hz, C₁-H), 4.38 (1 H, m, C₃-H), 3.99 (1 H, m, C₃-H), 3.53

(1 H, dd, J = 9, 5 Hz, $C_{4'}$ -H), 3.51 (1 H, m, $C_{5'}$ -H), 3.15 (1 H, d, J = 8 Hz, C_{17} -H), 2.70 (1 H, dd, J = 16, 10 Hz, C_{15} -H), 1.95 (1 H, s, C_{16} -OOCCH₃), 1.34, 1.48 (6 H, s, $>C(CH_3)_2$), 1.24 (3 H, d, J = 6 Hz, $C_{6'}$ -H), 0.95 (6 H, s, C_{18} -H and C_{19} -H). Anal. ($C_{34}H_{50}O_9$) C, H.

(3β,5β,14β,16β,17β)-3-[(2,6-Dideoxy-β-D-ribo-hexopyranosyl)oxy]-16-acetoxy-14-hydroxycard-20(22)-enolide (22). A solution of 21 (200 mg) in 60% AcOH (30 mL) was stirred at room temperature for 5 h, poured into water, and extracted with CHCl₃. The CHCl₃ extract was washed with water, dried over $MgSO_4$, and evaporated in vacuo to give an oil. The crude oil flash chromatographed (20% acetone-CH₂Cl₂) to yield 146 mg (78%) of 22 as amorphous powder and 41 mg (28.5%) of 16-acetylgitoxigenin (5). Data for 22: mp 167-170 °C; $[\alpha]^{19}$ -19.07° (c 0.19, MeOH); IR (KBr) 3460 (OH), 1782, 1745 (C=O) cm⁻¹; ¹H NMR (CDCl₃) δ 5.95 (1 H, s, C₂₂-H), 5.43 (1 H, ddd, J = 10, 10, 2 Hz, C_{16} -H), 4.90 (2 H, d, J = 16 Hz, C_{21} -H), 4.85 (1 H, dd, J = 8, 2 Hz, C_{1} -H), 4.08 (1 H, m, C_{3} -H), 3.70 (1 H, dd, $J = 9, 6 \text{ Hz}, C_{4'}\text{-H}), 3.35 (1 \text{ H}, \text{ m}, C_{5'}\text{-H}), 3.18 (1 \text{ H}, \text{ d}, J = 9 \text{ Hz},$ C₁₇-H), 1.96 (3 H, s, C₁₆-OCOCH₃), 0.95 (6 H, s, C₁₈-H and C₁₉-H). Anal. $(C_{31}H_{46}O_9)$ C, H.

(3β,5β,14β,16β,17β)-3-[(O-2,6-Dideoxy-β-D-ribo-hexopyranosyl- $(1 \rightarrow 4)$ -O-2,6-dideoxy- β -D-ribo-hexopyranosyl- $(1 \rightarrow 4)$ -2,6-dideoxy- β -D-ribo-hexopyranosyl)oxy]-16-acetoxy-14-hvdroxycard-20(22)-enolide (Gitoxin 168-Acetate). To a solution of 2.0 g of gitoxin (1) in 100 mL of pyridine was added dropwise 24 mL of a 10% solution of COCl₂ in toluene for 30 min at 0-5 °C with shaking for 1 h at about 0 °C. Excess of COCl₂ was decomposed by addition of ice water and the reaction product was extracted with CHCl₃. The CHCl₃ extract was washed with 5% HCl, 3% NaHCO₃, and H₂O to neutral, dried over anhydrous MgSO₄, and evaporated in vacuo to give crude crystals, which were chromatographed (10% MeOH-CHCl₃) to yield 1.482 g (71.71%) of gitoxin 3",4"'-cyclocarbonate 25 as colorless crystals: mp 238–240 °C dec; $[\alpha]^{22}_{D}$ +21.6° (c 0.52, CHCl₃); IR (CHCl₃) 3458 (br, OH), 1808 (cyclocarbonyl), 1746 (C=O), 1616 (C=C) cm⁻¹; UV λ_{max} 218 nm (ϵ 14 590). Anal. (C₄₂H₆₂O₁₅) C, H.

To a solution of 1.482 g of 25 in dry pyridine (10 mL) was added acetic anhydride (2 mL) and the mixture was stirred at room temperature for 1 h. The reaction mixture was poured into water and extracted with CHCl₃, and the CHCl₃ extract was washed successively with 1 N HCl, H₂O, 5% NaHCO₃, H₂O, and dried, and evaporated in vacuo to give 1.656 g of crystals. A solution of the crystals in 150 mL of aqueous acetone (acetone $-H_2O$, 3:1, v/v) containing 0.4% KHCO3 was allowed to stand at room temperature for 4 days. The resulting solution was neutralized, concentrated in vacuo, and extracted with CHCl₃. The CHCl₃ extract was washed with H₂O, dried over MgSO₄, and evaporated to give 1.454 g of a crude product. The crude product was chromatographed (50% acetone– CH_2Cl_2) and crystallized from acetone to give 527 mg (33.75%) of 26: mp 238-240 °C; $[\alpha]^{19}_{D}$ +0.83 (c 0.24, MeOH); IR (KBr) 3440 (OH), 1750 (C=O) cm⁻¹ ¹H NMR (CDCl₃) δ 5.94 (1 H, s, C₂₂-H), 5.45 (1 H, dd, J = 9, 9Hz, C_{16} -H), 4.95 (2 H, d, J = 16 Hz, C_{21} -H), 4.23 (1 H, m, 3-H), 1.98 (3 H, s. OCOCH₃), 1.23, 1.30 (9 H, \overline{d} , J = 6 Hz, $C_{6'}$ -H, $C_{6''}$ -H, $C_{6''}$ -H), 0.95 (6 H, s, C_{18} -H, C_{19} -H). Anal. ($C_{43}H_{66}O_{15}$) C, H.

Acknowledgment. This research was supported in part by research Grant HL-21457 awarded by the National Heart, Lung and Blood Institute, DHHS, and by the Medical Research Fund of the Veterans Administration. Analysis of data was expedited using the PROPHET system, a national computer resource supported by the NIH.