Pyrimidinones. 3. N-Substituted 6-Phenylpyrimidinones and Pyrimidinediones with Diuretic/Hypotensive and Antiinflammatory Activity

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In an extensive analysis of the antiviral and interferon-induction structure–activity relationship of 6-arylpyrimidinones we found that modifications at positions 1-4 of the pyrimidine ring resulted in a loss of activity.¹⁻³ However, we uncovered interesting hypotensive and antiinflammatory activity with a series of N-substituted analogues, the results of which we report herein.

Over the past 30 years numerous pyrimidines have been prepared and their pharmacology evaluated⁴ Various 2-amino-4-substituted-5-alkylpyrimidines were reported to have diuretic activity.^{5–8} In addition, numerous 2aminopyrimidines have exhibited bacteriostatic, fungicidal, and antiviral activity.^{8–10} We therefore expanded the analogue scope of our antiviral 5-halo-6-phenylpyrimidines to search for cardioregulatory as well as antiinflammatory activity. These were exploited in a series of N-substituted analogues.

Chemistry. There are two obvious problems in attempting to synthesize N-substituted pyrimidines: regioselective introduction of the desired N-alkyl substituent and unequivocal determination of the position of the substituent. The latter problem was solved through the use of ${}^{1}\text{H}{-}{}^{13}\text{C}$ long-range gated decoupling NMR. This we found generally applicable to a variety of N-alkyl substituents. However, selectivity in introducing the N-substituent appeared to be a function of other substituents on the pyrimidine. Therefore, the results we described should not be generalized beyond the 6-aryl series.

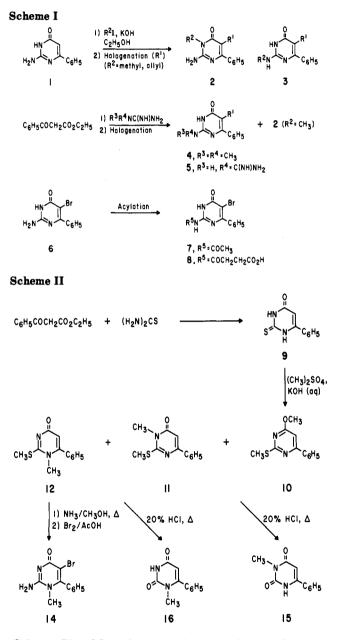
Methylation (CH₃I, KOH, EtOH) of 2-amino-6phenyl-4-pyrimidinone yields the N_3 -methyl product in 75% yield⁸ (Scheme I). Halogenation readily affords the desired 5-halogen derivatives.² Condensation of ethyl benzoylacetate with methylguanidine affords a mixture of N_2 -methyl and N_3 -methyl products with the latter greatly predominating. This is in contrast to the report of N_1 - and N_2 -methyl products in the condensation with ethyl acetoacetate.⁸

The N_2 -methyl is readily recognized in the ¹H NMR spectrum, since the methyl appears as a doublet which collapses to a singlet either with deuterium exchange or N-H irradiation. The position of the C₂, C₄, C₅, and C₆ can be readily assigned in the ¹³C NMR spectrum by ¹H-¹³C coupling (gated and off-resonance decoupling; see Experimental Section).^{12,13} The C₅ exhibits 1,2-coupling (doublet); the C₄ exhibits a small 1,3-coupling; the C₆ exhibits multiple 1,3-coupling; and the C₂ exhibits none. Therefore, the N₃-methyl is readily assigned based on 1,4-coupling with both C₂ and C₄, and of course, the N₂-methyl exhibits a multiplet for C₂ only. No N₁-methyl or alkyl analogues were isolated in these approaches.

Additional N_2 -substituted analogues were prepared via direct condensation, such as the N_2 -dimethyl and the N_2 -amidino, from dimethylguanidine and biguanide, respectively. Acetylation and succinylation of pyrimidinone 6 yielded the 2-acetyl and 2-succinyl derivatives, 7 and 8.

Since we were unable to prepare any N_1 -derivatives via either direct alkylation of the pyrimidinone or condensation of the appropriate guanidine, we explored an alternative route. Reasoning that the temporary removal of the N_2 -amino group might allow N_1 -alkylation to occur, we prepared the 2-(methylthio)-6-phenyl-4-pyrimidinone

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(Scheme II). Methylation with excess dimethyl sulfate and 2 equiv of base afforded 10, 11, and 12 in approxi-

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							pre	ood ssure nge ^b		% in	mma hibitio (mg/	on
entry	\mathbb{R}^{1}	R	2 R ⁶	X	analyses/ref	diureticª	4 h	24 h	25	50	100	200
2	Br	CH3		NH ₂	C ₁₁ H ₁₀ BrN ₃ O: C, H, N, Br	5.52	-12	-7	36	64	74	-
3	Br	H		NHCH ₃	$C_{11}H_{10}BrN_{3}O: C, H, N, Br$	-	NT		42	78	82	92
4	Br	Н		$N(CH_3)_2$	$C_{12}H_{12}BrN_{3}O: C, H, N$	-	+1	+2				
5	Br	Н		NHC(NH)NH ₂	$C_{11}H_{10}BrN_5O: C, H, N, Br$	-	-2	-1				
6	Br	Н		NH ₂	ABPP/1	7.41	+0	-22	13	31	44	
7	Br	Н		NHCOCH ₃	C ₁₂ H ₁₀ BrN ₃ O ₂ : C, H, N, Br	-	+4	+1				
8	\mathbf{Br}	Н		NHCOCH ₂ CH ₂ CO ₂ H	$C_{14}H_{12}BrN_3O_4$: C, H, N, Br^d	\mathbf{NT}	\mathbf{NT}				-	
14	Br		CH_3	NH ₂	$C_{11}H_{10}BrN_{3}O: C, H, N^{g}$	\mathbf{NT}	\mathbf{NT}				-	
17	Ι	Н		NH_2	AIPP/1	-	-7	-12	-	3	8	17
18	I	CH_3		NH ₂	$C_{11}H_{10}IN_{3}O: C, H, N$	4.46	-6	-13	18	30	79	-
19	I	Н		$N(CH_3)_2$	C ₁₂ H ₁₂ IN ₃ O: C, H, N, I	-	+1	+2			-	
20	Ι	Н		NHCOCH ₃	$C_{12}H_{10}IN_{3}O_{2}$: C, H, N, I	-	+3	+9			-	
21	I	Н		NHC(NH)NH ₂	C ₁₁ H ₁₀ IN ₅ O: C, H, N, I, C ^e	-	-2	+3			-	
22	I	Н		NHCH ₃	C ₁₁ H ₁₀ IN ₃ O: C, H, N, I, N ^{<i>t</i>}	-	\mathbf{NT}				-	
23	Ι	CH ₂ CH		\mathbf{NH}_2	$C_{13}H_{12}IN_{3}O: C, H, N$	5.09	-6	-10			-	
24	\mathbf{Br}		ОН	NH ₂	ref 17	-	+6	-1		-	-	
1	Н	Н		$\rm NH_2$	ref 1	-					-	
15	Н	CH_3	н	0	$C_{11}H_{10}N_2O_2$: C, H, N	-	0	+6		-	-	
25	Br	CH_3	н	0	$C_{11}H_9BrN_2O_2$: C, H, N, Br, Br^h	-	-1	+6		-	-	
26	I	CH_3	Н	0	$C_{11}H_9IN_2O_2$ C, H, N, I, I ⁱ	-	+1	+4		-	-	
16	Н	Н	CH_3	0	ref 11		+4	+6	27	47	79	95
27	Br	Н	CH_3	0	$C_{11}H_9BrN_2O_2$: C, H, N, Br, Br^j	-	+1	+1	22	43	57	67
28	I	Н	CH_3	0	$C_{11}H_9IN_2O_2$: C, H, N, I	-	+15	-1	33	44	73	99
29	Cl	Н	CH3	0	$C_{11}H_9ClN_2O_2$: C, H, N, Cl	3.49	+2	+2		-		
36	Br	CH_3	CH_3	0	$C_{12}H_{11}BrN_2O_2: C, H, N, Br$	-	\mathbf{NT}		-	36	59	-
31	Br	H	Н	0	ref 11	3.53	+6	+5	-	46	58	-
32	I	Н	н	0	ref 11	5.43	+7	+11	-	41	48	-
33			hydrochlor			7.18	0	+2				
34			furosemide			31.25	+1	-3				
35			guanethidi	ne			-16	9				

^a Numerical values indicate that the test substance was considered active (i.e., T/C for stage $1 \cdot T/C$ for stage 2 > 3.34) with the greater the value the greater the activity; – indicates inactive and NT is not tested; given orally at 40 mg/kg. ^bRefers to the average mmHg change in mean arterial blood pressure observed at 4 and 24 h after oral administration of 50 mg/kg. ^cCompounds were administered po 1 h before antigen sensitization to 24-h fasted rats. Dexamethasone served as the positive control exhibiting 50% inhibition at 5 mg/kg. ^dEquiv w = 389. ^cCalcd, 37.19; found, 37.79. ^fCalcd, 12.84; found, 13.30. ^g1.0 HBr. ^hCalcd, 28.42; found, 26.87. ⁱCalcd, 38.68; found, 40.6. ^jCalcd, 28.42; found, 27.07.

mately 10%, 70%, and 3% yields, respectively. The Oalkylated derivative 10 is readily identified by its ¹H NMR (OCH₃ at 4.01 ppm). Differentiation of 11 and 12 was accomplished by ¹H-¹³C long-range coupling where 11 exhibited collapse to a multiplet with gated decoupling of the readily identified C₄. In the ¹H NMR the N-CH₃ of 11 is further downfield (3.55 ppm) than in 12 (3.23 ppm)

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as expected due to the deshielding of the adjacent carbonyl in 11. In addition, 11 was hydrolyzed to the known N_3 -methyl-6-phenyluracil and 12 to N_1 -methyl-6-phenyluracil.¹¹

Although the yield was poor, sufficient 12 was prepared and heated in a sealed tube with ammonia-saturated methanol to yield the 2-aminopyrimidinone. Bromination afforded 14 in an overall 1% yield from 9. That no rearrangement occurred during the ammonia treatment was evident by the nonidentity of 14 with the previously prepared 2 and 3.

Lastly, the corresponding N_3 -methyl-5-halopyrimidinediones (uracil) were prepared for biological comparisons. These were prepared via conventional fusion procedures to afford the uracil followed by halogenation as outlined previously.¹

Biology. 6-Phenylpyrimidinones 2, 5, 7, and 8 were evaluated for antiviral activity against SFV challenge in mice and found inactive. In addition the N-alkylated-6halogeno-6-phenyluracils were uniformly inactive in this virus model.

Table I depicts the cardiovascular/renal and antiinflammatory activities of both the 2-aminopyrimidinone and the pyrimidinedione series. Compounds 2, 6, 18, and 23 exhibit diuretic activity when administered orally to fasted rats. These same 2-aminopyrimidinones do exhibit hy-

Table II. Effect of ABPP and AIPP on Adjuvant-Induced Polyarthritis

	dose, mg/kg,	body wt	spleen wt,	arthritis (day 15)			arthritis (day 33)					
treatment ^a	b.i.d.	change, gm	g	incidence	score	% I	incidence	score	% I	PIU	% I	
normal control		+57	1.05 ± 0.13	0/10	0					3.6 ± 1.5		
arthritic control		+5	1.93 ± 0.19	25/30	11.3 ± 0.5		13/14	11.4 ± 1.6		196.9 ± 27.1		
cortisol	3.4	-13	1.79 ± 0.37	9/10	7.5 ± 1.5	34	5/5	15.6 ± 0.4	0	152.6 ± 35.4	42	
	8.6	+1	1.12 ± 0.06	4/10	2.4 ± 1.1	79	5/5	6.6 ± 1.7	42	123.4 ± 32.5	37	
	17.8	-6	1.18 ± 0.20	0/10	0	100	5/5	4.2 ± 1.5	63	43.0 ± 17.7	78	
ABPP	106	-8	2.92 ± 0.35	6/10	3.8 ± 1.6	66	4/5	9.4 ± 3.1	18	79.0 ± 28.0	60	
	208	-12	2.89 ± 0.38	3/10	1.3 ± 0.5	88	2/4	1.5 ± 1.0	87	54.6 ± 8.8	72	
	321	-16	3.11 ± 0.61	2/10	1.8 ± 1.6	84	1/4	0.2 ± 0.2	98	80.6 ± 10.6	59	
AIPP	35	-12	1.73 ± 0.14	10/10	10.0 ± 1.5	0	5/5	11.2 ± 2.1	0	162.6 ± 9.4	17	
	106	-11	1.68 ± 0.27	9/10	10.0 ± 1.6	0	5/5	13.4 ± 1.7	0	124.8 ± 12.3	37	
	208	-14	1.75 ± 0.16	9/10	9.6 ± 1.6	0	5/5	11.4 ± 2.5	0	149.8 ± 22.9	24	
	327	-14	1.95 ± 0.15	10/10	10.9 ± 1.2	0	5⁄5	10.8 ± 1.1	0	147.0 ± 5.6	25	

^a Compounds were administered po to 50% of the animals for 15 days. Treatment was stopped in 50% of the animals on day 15, and this part of the assay was terminated on day 33.

Table III. Schedule Dependence	of the Effects of ABPP on Adjuvant-Produced P	'olyarthritis
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	dose, mg/kg,	body wt		arthritis (day 15)						
days of treatment ^a	b.i.d.	change, g	spleen wt, g	incidence	arthr score	% I	PIU	% I		
normal control ^b		114	1.22 ± 0.02	0/10			3.8 ± 1.0			
arthritic control		+34	1.78 ± 0.12	20/20	15.7 ± 0.1		146.7 ± 12.9			
1, 3, 5, 7, 9, 11, 13, 15	279	+23	2.09 ± 0.6	6/10	2.8 ± 0.6	82	62.5 ± 8.7	57		
1, 4, 7, 10, 13	263	+37	1.77 ± 0.12	6/10	3.7 ± 1.4	76	34.9 ± 7.4	76		
1, 5, 9, 14	267	+34	1.73 ± 0.10	9/10	7.6 ± 1.6	51	59.8 ± 9.5	59		
1, 6, 12	281	+37	1.59 ± 0.08	6/10	5.4 ± 1.6	66	69.6 ± 12.7	53		
1-15	292	+21	2.16 ± 0.17	0/10	0	100	21.1 ± 4.0	86		

^a Compounds were administered po for the stipulated time periods. ^b Vehicle was administered to the normal and arthritic control groups from days 1-15.

potensive activity when given orally to normotensive rats at essentially equivalent doses. Since compound 17 exhibits hypotensive activity but no diuretic activity, it cannot be concluded that these two phenomena are interdependent with this series of compounds. A similar conclusion must be drawn from the pyrimidinediones where several compounds (29, 31, and 32) are diuretic but exhibit no hypotensive response. For comparison, diuretic assay values and hypotensive assay values obtained with hydrochlorothiazide (HCTZ) and furosemide (diuretic standards) and guanethidine (hypotensive standard) are also depicted in Table I. Further evaluation of the diuretic activity of pyrimidinones revealed that natriuresis and kaliuresis accompanies the diuresis associated with these agents (Table IV). Results from identical tests with the standards hydrochlorothiazide and furosemide are also shown in Table IV. The most active pyrimidinones were as efficacious as hydrochlorothiazide but less efficacious than furosemide. The Na^+-K^+ excretion profile with these pyrimidinones was nearly identical to that apparent with the thiazide diuretic.

Both series of compounds exhibit antiinflammatory effects as measured in the reverse passive Arthus reaction (RPAR). The pyrimidinediones are the more potent and consistently active compounds exhibiting a dose-related (25-200 mg/kg) suppression of the RPAR.

The summary structure activity relationship (SAR) data indicate that (1) for cardiovascular activity N_1 - or N_2 substituents abolish activity whereas N_3 -substituents are allowed but not required and (2) for antiinflammatory activity the N_1 -substituent of the pyrimidinedione series is compatible but not necessary for activity whereas the N_3 is tolerated only where N_1 is methylated.

The antiarthritic activity of compounds 6 and 17 was further evaluated in adjuvant-induced polyarthritic rats. The data in Table II indicate that compound 6 (ABPP) caused a dose-related inhibition of this chronic arthritic process as reflected in a significant decrease in the incidence and severity of arthritis on day 15. Further, if treatment is terminated on day 15, the activity of ABPP is sustained as illustrated by the suppression of arthritis on day 33. Inhibition of the PIU values relative to controls also demonstrates the antiarthritic activity of this compound. In contrast, compound 17 (AIPP) was essentially inactive in this model of arthritis (Table II).

As an extension of the aforementioned studies, we further evaluated the effects of ABPP on polyarthritis in rats as a function of the dosing regimen with this compound. The results of these studies (Table III) indicate that compound 6 exerted maximum antiarthritic activity when administered daily from day 1 through day 15. However, greater than 80% suppression of the arthritic response was achieved when ABPP was administered on an alternateday basis.

It has been reported that selective defects in T-cell function and immune interferon regulation may contribute to the etiology of rheumatoid arthritis.¹⁹ In addition there are preliminary reports that interferon treatment of rheumatoid arthritis results in a partial amelioration of some of the symptoms.²⁰ Since (1) ABPP is better tolerated in the clinic than interferon.^{21,22} (2) ABPP exhibits

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Table IV. Dose-Response Relationships for Effects of Selected Pyrimidinones on Urine Volume and Na^+ and K^+ Excretion in Conscious Rats^a

		dose, mg/kg						
entry	urinary excretion	0.3	1.0	3.0	10	30	100	
2	UV ^b	1.0	0.9	1.4	2.3	2.5	2.8	
	Na ⁺	1.0	1.2	1.3	$\overline{2.1}$	$\overline{2.5}$	2.8	
	K+	1.1	1.0	1.3	$\overline{1.7}$	2.2	2.2	
6	UV	1.1	1.2	$\frac{\frac{1.4}{1.3}}{\frac{1.3}{1.3}}$	2.5	$\overline{2.6}$	$\overline{2.7}$	
	Na ⁺	1.1	1.3	1.5	2.3	2.9	2.8	
	K+	1.1	$\frac{1.3}{1.0}$	$\frac{1.5}{1.1}$	1.3	1.5	1.3	
18	UV		1.4	1.6	2.2	$\overline{2.7}$	2.0	
	Na^+	$\frac{1.3}{1.1}$	$\frac{1.4}{1.2}$	$\frac{\underline{1.6}}{\underline{1.6}}$	$ \begin{array}{r} 2.3 \\ 2.1 \\ 1.7 \\ 2.5 \\ 2.3 \\ 1.3 \\ 2.2 \\ 1.9 \\ 1.4 \\ \end{array} $	2.4	2.7	
	K ⁺	1.1	1.1	$\overline{1.2}$	1.4	$\frac{1}{2.1}$	1.6	
29	UV	0.9	0.9	0.8	$\overline{1.1}$	1.6	1.6	
	Na ⁺	1.1	1.1	0.9	1.1	1.6	1.6	
	K+	0.9	1.2	1.1	1.3	$\frac{1}{1.7}$	1.9	
31	UV	0.9	1.0	1.1	1.2	1.5	$\overline{2.1}$	
	Na ⁺	1.2	1.2	1.0	1.2	1.4	1.9	
	K+	1.0	1.0	1.0	0.9	1.2	1.8	
HCTZ ^c	UV	1.9	2.3			2.6	$\overline{2.7}$	
	Na ⁺	$\frac{1.9}{1.7}$ 1.2	2.0	$\overline{2.1}$	$\overline{2.4}$	2.4	2.5	
	K ⁺	1.2	1.6	1.6	1.5	1.5	1.7	
furosemide	ŪV	1.1	$\frac{\frac{2.3}{2.0}}{\frac{1.6}{1.2}}$	$\frac{\frac{2.2}{2.1}}{\frac{1.6}{1.3}}$	$ \frac{\frac{2.4}{2.4}}{\frac{1.5}{2.2}} \\ \overline{1.5} $	$\begin{array}{c} \underline{2.5}\\ \underline{2.5}\\ \underline{2.2}\\ \underline{2.6}\\ \underline{2.9}\\ \underline{1.5}\\ \underline{2.7}\\ \underline{2.4}\\ \underline{2.1}\\ \underline{1.6}\\ \underline{1.7}\\ \underline{1.5}\\ \underline{1.4}\\ \underline{2.6}\\ \underline{2.4}\\ \underline{1.5}\\ \underline{4.7}\\ \underline{3.1}\\ \underline{2.2}\\ \end{array}$	$\begin{array}{c} 2.8\\ 2.8\\ 2.2\\ 2.7\\ 2.8\\ 1.3\\ 2.0\\ 2.7\\ 1.6\\ 1.6\\ 1.9\\ 2.1\\ 1.9\\ 2.1\\ 1.9\\ 2.5\\ 1.7\\ 2.5\\ 1.7\\ 2.5\\ 1.7\\ 2.5\\ 3.2\\ \end{array}$	
	Na ⁺	1.1	1.1	$\overline{1.1}$	$\frac{1.5}{1.5}$	3.1	4.5	
	K ⁺	1.0	1.0	1.3	1.6	$\frac{1}{22}$	32	

^a Values are urinary excretion in treated rats/urinary excretion in control rats. Underscoring indicates values significantly greater than 1.0 (P < 0.05); i.e., excretion in treated was significantly greater than in the corresponding control. Mean urine volume of all control rats was 4.2 ± 0.1 (SE) mL/5 h; control Na⁺ excretion was 0.68 ± 0.02 mequiv/5 h; and K⁺ excretion was 0.24 ± 0.01 mequiv/5 h. ^bUrine volume. ^c Hydrochlorothiazide.

antiarthritic activity as noted above, and (3) there appears to be a correlation with antiarthritic activity and interferon induction (note that AIPP is inactive and a poor interferon inducer), it appears justifiable to evaluate ABPP and other interferon inducers in patients with rheumatoid arthritis.

Experimental Section

Melting points were determined on a Thomas-Hoover unimelt and are uncorrected. ¹H NMR spectra were run on a Varian T60-A, FT-80 or EM-390. ¹³C NMR spectra were recorded on a Varian CFT-20. All data are relative to internal Me₄Si. All solvents and reagents employed were reagent grade and used as received.

 $^{1}\text{H}-^{13}\text{C}$ coupling spectra (gated decoupling) were performed by employing a filter bandwidth of 4000, sweep width of 2304, acquisition time of 1.77 s, pulse width of 9, pulse delay of 3.00 s, and the decoupler turned off.

Halogenation procedures are as previously described.²

6-Phenyl-2,4-pyrimidinedione. To 24 g (0.4 mol) of urea was added 200 mL of ethyl benzoylacetate, and the reaction mixture was allowed to stir at 135 °C for 72 h. The mixture was cooled, allowed to stand at room temperature for 18 h, and filtered. The resulting solids were washed well with ether and dried at 60 °C to yield 16.0 g. The solids were triturated with water, cooled, and filtered to yield 14.0 g (18.6%) of 6-phenyl-2,4-pyrimidinedione: ¹H NMR (Me₂SO-d₆) 7.86-7.46 (m, 5 H, Ph), 5.81 (s, 1 H, vinyl).

1,3-Dimethyl-5-bromo-6-phenyl-2,4-pyrimidinedione (36). To 2.67 g (10 mmol) of 5-bromo-6-phenyl-2,4-pyrimidinedione was added 50.0 mL of water and 1.0 g (41 mM) of NaOH. When all was in solution, 2.24 mL (\approx 2.97 g, \approx 23.5 mmol) of dimethyl sulfate was added and the reaction mixture was heated at reflux for 18 h. The reaction mixture was cooled and poured into a separatory funnel containing 250 mL of CH₂Cl₂. The aqueous layer was washed twice with methylene chloride, and the combined organic layers were dried over Na₂SO₄ (anhydrous), filtered, and evaporated to dryness under vacuum to yield 2.40 g of an amorphous foam. The crude material was dissolved in 50 mL of acetone, 20 g of silica gel added, and the slurry evaporated. The powder was placed on top of 300 g of silica gel and the column eluted with 1:1 ethyl acetate/hexane (R_f 0.3 in 1:2 EtOAc/hexane) to yield 1.50 g (50.8%): ¹H NMR (CDCl₃) 7.50 (m, 3 H, Ph), 7.15 (m, 2 H, Ph), 3.45 (s, 3 H, CONCH₃CO), 3.10 (s, 3 H, NCH₃). Anal. $(C_{12}H_{11}BrN_2O_2)$ C, H, N, Br.

3-Methyl-6-phenyl-2,4-pyrimidinedione (15). To 2.4 g (26.6 mmol) of N-methylurea was added 25.0 mL of ethyl benzoylacetate, and the mixture was allowed to stir at 135 °C for 18 h. After cooling to room temperature, the precipitate formed was filtered and washed well with ether to yield 1.70 g (31.6%). Recrystallization from acetone/ethanol/hexane gave 1.40 g (26%). This material was identical (spectroscopically and chromatographically) with the pyrimidinone obtained via hydrolysis at the 2-SCH₃ analogue.

Hydrolysis of (Methylthio)pyrimidinones (15, 16). To 12.7 g of 2-(methylthio)-3-methyl-6-phenyl-4-pyrimidinone (54.7 mmol) was added 150 mL of a 20% aqueous HCl solution. The reaction mixture was heated at reflux, with stirring, for 2 1/2 h. After cooling, the resulting solids were filtered, washed with water, and dried at 60 °C to yield 10.5 g. Recrystallization from ethanol/DMF yielded 7.90 g (71.5%): ¹H NMR (Me₂SO-d₆) 7.76 (m, 2 H, Ph), 7.56 (m, 3 H, Ph), 5.96 (s, 1 H, vinyl), 3.16 (s, 3 H, NCH₃). Anal. (C₁₁H₁₀N₂O₂) C, H, N.

In similar fashion 1.0 mmol of 2-(methylthio)-1-methyl-6phenyl-4(3*H*)-pyrimidinone was hydrolyzed to the N_1 -methyl isomer in 65% yield: ¹H NMR (Me₂SO- d_6) 7.44 (s, 5 H), 5.40 (s, H), 2.96 (s, 3 H). Anal. (C₁₁H₁₀N₂O₂) C, H, N.

The assignments are consistent with the ¹H NMR spectra wherein the enforced orthogonal phenyl group of the N_1 isomer appears as a singlet (2) and the N_3 -methyl group is further downfield due to additional deshielding effects of both carbonyl groups.

2-Amino-1-methyl-6-phenyl-4-pyrimidinone. To 400 mg (1.72 mM) of 1-methyl-2-(methylthio)-6-phenyl-4-pyrimidinone was added, in a glass bomb, 12 mL of a saturated solution of NH_3/CH_3OH . The glass bomb was sealed and heated at 170 °C for 18 h. The reaction was cooled, the bomb opened, and the solution evaporated to a yellow oil under vacuum. The desired material had R_{f} 0.3 in 15% MeOH/CHCl₃ and was contaminated with 1-methyl-6-phenyl-2,4-pyrimidinedione as the major byproduct. The crude material was dissolved in ethanol, 8 g of silica gel was added, and the slurry was evaporated to a powder under vaccuum. The powder was placed on top of 25 g of silica gel and eluted with 7% MeOH/CHCl₃, collecting 15.0-mL fractions until desired material was obtained. Evaporation of solvent to dryness gave 180 mg (52%). Recrystallization from methanol/ether gave 60 mg (17%): ¹H NMR (Me₂SO- d_6) 7.50 (s, 5 H, Ph), 5.5 (s, 1 H, vinyl), 3.3 (s, 3 H, NCH₃). MS: calcd, 201.0902; found, 201.0893.

⁽²²⁾ Rios, A. A.; Stringfellow, D. A.; Fitzpatrick, F. A.; Reele, S. B.; Gutknecht, G. D.; Hersh, E. M. J. Biol. Response Modif., in press.

2-Thioxo-6-phenyl-4-pyrimidinone (9). To 106 g (1.4 mol) of thiourea was added 1 L of ethanol and 400 mL of 25% sodium methoxide/methanol and 180 g (0.93 mol) of ethyl benzoylacetate. The reaction solution was heated at reflux for 18 h and cooled to 5 °C. A precipitate formed that was filtered and washed with methanol, dissolved in water (1 L), and neutralized by the addition of concentrated HCl. The resulting solids were filtered, washed well with water followed by acetone, and dried at 60 °C overnight to yield 119.7 g: mp 263-5 °C.¹⁸

Methylation of 2-Thioxo-6-phenyl-2,4-pyrimidinedione (10-12). To 119.7 g (0.586 mol) of 2-thio-6-phenyl-2,4-pyrimidinedione was added 250 mL of water and 51.2 g (1.28 mol) of sodium hydroxide. The reaction mixture was stirred at ambient temperature until solution was complete. Dimethylsulfate (122 mL, \approx 166.9 g) was added dropwise at 50-60 °C. When the addition was complete, the pH was slightly acidic and a precipitate formed. The solids were filtered and washed well with water and dried at 40 °C under vacuum. The aqueous solution was extracted twice with toluene and chloroform. The organic layers were combined, dried (Na₂SO₄ anhydrous), and evaporated to dryness to yield 0.9 g of 1-methyl-2-(methylthio)-6-phenyl-4-pyrimidinone: ¹H NMR (Me₂SO-d₆) 7.46 (s, 5 H, Ph), 5.70 (s, 1 H, vinyl), 3.23 (s, 3 H, NCH₃), 2.46 (s, 3 H, SCH₃). Anal. (C₁₂H₁₂N₂OS) C, H, N, S.

The dried solids were chromatographed on 1.5 kg of silica gel and eluted with 1:2 EtOAc/hexane until R_f 0.9 (1:1, EtOAc/ hexane) appeared. Evaporation of the material yielded 10.3 g of 2-(methylthio)-4-methoxy-6-phenylpyrimidine. This pyrimidine was recrystallized from *n*-hexane to yield 8.5 g of pure material: ¹H NMR (CDCl₃) 8.15–7.98 (m, 2 H, Ph), 7.55–7.4 (m, 3 H, Ph), 7.8 (s, 1 H, vinyl), 4.01 (s, 3 H, OCH₃), 1.65 (s, 3 H, SCH₃). Anal. (C₁₂H₁₂N₂OS) C, H, N, S.

At this point the eluent was switched to 1:1 EtOAC/hexane and solvent collected until R_f 0.5 (2:1 EtOAc/hexane) appeared. Evaporation of solvents left 71 g of 3-methyl-2-(methylthio)-6phenyl-4-pyrimidinone. Recrystallization gave 58.5 g of pure compound: ¹H NMR (CDCl₃) 8.05–7.85 (m, 2 H, Ph), 7.50–7.35 (m, 3 H, Ph), 6.65 (s, 1 H, vinyl), 3.55 (s, 3 H, NCH₃), 2.70 (s, 3 H, SCH₃). Anal. (C₁₂H₁₂N₂OS) C, H, N, S.

The third spot (R_f 0.3 in 1:1 EtOAc/hexane), which was found to be starting material, was discarded and the eluent switched to 2.5% MeOH/CHCl₃ and the elution continued until spot 4 was 1.90 g of 1-methyl-2-(methylthio)-6-phenyl-4-pyrimidinone, which was recrystallized from water to yield 750 mg.

(3,4-Dihydro-4-oxo-6-phenyl-2-pyrimidinyl)guanidine (5). To 5.0 g (50 mmol) of biguanide in 300 mL of absolute ethanol was added 30 mL of toluene and heated to reflux to azeotrope off 100 mL of solvent. The suspension was cooled, and 9.6 g (50 mmol) of ethyl benzoylacetate was added. This was heated to reflux for 6 h. Then 20 mL of water was added and continued to heat at reflux for an additional 30 min. Upon cooling to room temperature CO₂ chips were added to neutralization. The precipitate was filtered and washed with water, followed by acetone and ether, and dried at 60 °C in vacuo to yield 3.6 g (31.5%) of product: ¹H NMR (Me₂SO-d₆/Me₄Si) 11.9-11.6 (b, 1 H, NH), 8.58-8.08 (b, 4 H, NH), 8.0-7.25 (m, 2 H, Ph), 7.63-7.38 (m, 3 H, Ph), 6.28 (s, 1 H, vinyl). Anal. (C₁₁H₁₁N₅O) C, H, N.

2-(Acetylamino)-5-iodo-6-phenyl-4(3H)-pyrimidinone (20). To 2.0 g (6.38 mmol) of 2-amino-5-iodo-6-phenyl-4(3H)-pyrimidinone was added 70.0 mL of acetic anhydride and heated at reflux for 2 h. The reaction mixture was cooled and evaporated to a volume of 2 mL under vacuum. Addition of 20 mL of ethanol induced a precipitate, which was filtered, washed with ethanol, acetone, and ether, and dried at 60 °C to yield 1.60 g (71%) of product: ¹H NMR (Me₂SO- d_6 /TMS) 12.16–11.66 (b, 1 H, NH), 7.75–7.33 (m, 5 H, Ph), 3.58–3.16 (b, 1 H, NH), 2.13 (s, 3 H, CH₃C). Anal. (C₁₂H₁₀IN₃O) C, H, N, I.

2-(Acetylamino)-5-bromo-6-phenyl-4(3H)-pyrimidinone (7). By employing the same procedure as for compound **20**, 1.90 g (83%) of product was isolated: ¹H NMR (Me₂SO- d_6 /Me₄Si) 12.1-11.8 (b, 1 H, NH), 7.8-7.36 (m, 5 H, Ph), 3.5-3.08 (b, 1 H, NH), 2.16 (s, 3 H, CH₃C). Anal. (C₁₂H₁₀BrN₃O₂) C, H, N, Br.

2-Succinimido-5-bromo-6-phenyl-4(3H)-pyrimidinone (8). To 2.8 g (10.5 mmol) of 2-amino-5-bromo-6-phenyl-4(3H)-pyrimidinone was added 70 mL of dry pyridine and 4.5 g (45 mmol) of succinic anhydride. After heating at reflux for 1 h, the solution was cooled and evaporated to dryness under vacuum. The residue was azeotroped twice with water and once with ethanol, then triturated twice with boiling water and filtered, while hot. Repetition of this procedure gave a uniform solid, which was recrystallized from 100 mL of hot ethanol to yield 400 mg (9.6%) of product. Anal. $(C_{14}H_{12}BrN_{3}O_{4})$ C, H, N, Br.

Preparation of 2-Amino-3-methyl-6-phenyl-4(3H)-pyrimidinone and 2-(Methylamino)-6-phenyl-4(3H)-pyrimidinone (2, 3). Following the procedure employed for compound 5, 75 mmol of methylguanidine sulfate was heated with 18.8 g of ethyl benzoylacetate and 36 mL of 25% sodium methoxide/ methanol in 1.1 L of ethanol for 28 h to afford 21.7 g of solids. Recrystallization twice from 100 mL of hot 95% ethanol yieled 4.2 g of a solid, mp 238-239 °C, identical with that obtained from the methylation of 2-amino-6-phenyl-4(3H)-pyrimidinone and assigned N_3 -methyl isomer (2, vide infra).

To the solution from the crystallization was added 50 g of silica gel, and the mixture was evaporated to a powder under vacuum. The powder was placed on 800 g of silica gel (dry), and the column was eluted with 3% methanol/ethyl acetate. The crude solid had two spots on TLC (R_f 0.7 and 0.65, 5% MeOH/EtOAc) with the N_3 -methyl isomer having a R_f 0.7. The fractions containing pure R_f 0.65 were combined and evaporated to dryness to yield 40 mg of the N_2 -methyl isomer (3). ¹H NMR (Me₂SO-d₆/Me₄Si) 8.16-7.91 (m, 2 H, Ph), 7.58-7.36 (m, 3 H, Ph), 2.93-2.86 (d, J = 4.5 Hz, 3 H, NCH₃). Anal. (C₁₁H₁₁N₃O) C, H, N.

Methylation of 2-Amino-6-phenyl-4(3H)-pyrimidinone (2). To 7.0 g (36.4 mmol) of 1 was added 350 mL of absolute ethanol and 3.3 g (57 mmol) of KOH. When all the solid had dissolved, 18.2 g of potassium hydroxide and 7.98 mL (≈0.129 mol) of methyl iodide were added and the reaction mixture was heated at reflux for 5 h. At this point an additional 0.5 g of potassium hydroxide was added and heating was continued for an additional 3 h. The mixture was cooled and evaporated to dryness under vacuum. To the residue was added 150 mL of H₂O, and, at 50 °C, 1 N HCl(aq) was added to dissolve the solid. The solution was cooled to 20 °C and aqueous NaHCO3 solution added until no further precipitation occurred. The solids were filtered, washed well with water, and dried at 50 °C in vacuo to yield 5.4 g (74%) of product. Recrystallization of an analytical sample from methanol/ether gave crystals: mp 240-240.5 °C; ¹H NMR (Me₂SO-d₆) δ 7.88-8.05 (m, 2 H, Ph), 7.38–7.58 (m, 3 H, Ph), 7.05–7.31 (b, 2 H, NH₂), 6.23 $(s, 1 H, vinyl), 3.33 (s, 3 H, NCH_3)$. Anal. $(C_{11}H_{11}N_3O) C, H, N$.

2-(Dimethylamino)-6-phenyl-4(3H)-pyrimidinone (4). Following the procedure employed for compound 5, 0.1 mol of 1-(dimethylamino)guanidine hydrochloride was heated with 14.8 g of lithium carbonate, 6.8 g of sodium ethoxide, and 19.2 g of ethyl benzoylacetate to afford 16.5 g of product (77% yield): ¹H NMR (Me₂SO-d₆/Me₄Si) 8.13-7.91 (m, 2 H, Ph), 7.58-7.33 (m, 3 H, Ph), 6.11 (s, 1 H, vinyl), 3.10 (s, 6 H, N(CH₃)₂). Anal. (C₁₂H₁₃N₃O) C, H, N.

Reversed Passive Arthus Reaction. Male Wistar rats (180-200 g, Charles River, Wilmington, MA) were sensitized with an intravenous injection of 1 mg of bovine serum albumin in 0.2 mL of saline (n = 5-10 rats/group). Sixty minutes following antigen administration rats were challenged with antibody via a subplantar injection of $200 \ \mu \text{g}$ of anti-bovine serum albumin in 0.2 mL of saline in the left hindpaw. The degree of hindpaw swelling (inflammation) was determined by mercury displacement plethysmography. Changes in paw volume over time were determined at 2 h postantibody challenge.

Developing Arthritis Assay. Adjuvant-Induced Polyarthritis. Male rats (200-225 g, Charles River, Wilmington, MA) received a subcutaneous injection of 0.1 mL of complete Freund's adjuvant (4 mg/mL Mycobacterium butyricum in mineral oil) in the tail (n = 10 rats/group) on day 0. An arthritic score, used to assess drug effects on the arthritic process, was determined by assigning a rating of 0-4 to each paw (the maximum severity of arthritis would be 16 if each paw received a rating of 4) on the last day of the assay. The incidence of arthritis was also determined at this time.

Plasma Inflammation Units (PIU). PIU values were determined as described previously.¹⁴

Preparation and Administration of Compounds. Test compounds and reference standards were suspended in water containing 1-2 drops of Tween-80 (pyrogen-free, Accurate

Chemicals, Hicksville, NY) and shaken overnight with large glass beads. All compounds were administered orally, in milligrams per kilogram of body weight, according to the respective dosing regimens described in the table legends. Normal rats and the respective control groups received water plus Tween-80.

Diuretic Assays. A primary assay for diuretic activity was conducted in male rats weighing approximately 160 g. The rats were deprived of food 24 h and water $1^{1}/_{2}$ h before test time. During testing both food and water were withheld. Testing was initiated by simultaneous hydration and oral administration of test agent. This was accomplished by gavage with 25 mL/kg of normal saline (0.9%) containing (carboxymethyl)cellulose (0.5%) and test substance. Rats were placed in metabolism cages and urine collected over the ensuing 5 h. In all instances, the test dose was 40 mg/kg. Criteria for declaring test substances active or inactive were established from a two-stage test as described by Roseberry and Gehan.¹⁵ For each stage, the ratio (denoted T/C) of urine volume in treated animals to urine volume in control rats was determined. When the product of T/C for stage 1 and stage 2 was 3.34 or greater, the test substance was declared active. Secondary testing was carried out on selected pyrimidinones. These agents were subjected to dose-response analyses using a wide range of doses (0.3-100 mg/kg) and tested for effects on urinary Na⁺ and K⁺ excretion. Urinary Na⁺ and K⁺ were measured by flame photometry. In all other respects, i.e., tests animals, route of administration, hydration, and collection periods, secondary testing was identical to primary testing.

Hypotensive Assay—Blood Pressure in Rats. The arterial blood pressures of restrained female Sprague–Dawley rats were measured directly from indwelling aortic cannulas exteriorized at the nape of the neck.¹⁶ The rats were restrained in a towel when observations were made with a Statham transducer (P23G) and Grass polygraph. Mean arterial pressure was obtained by electrical integration of the phasic pressure. Observations were made before, as well as 4 and 24 h after, oral treatment. Compounds were suspended in a (carboxymethyl)cellulose vehicle to provide a dose of 50 mg/kg in a volume of 10 mL/kg. Mean arterial blood pressure values of two animals were averaged at each of the three observation periods. An average change of at least 5 mmHg was required posttreatment to attain statistical significance (P < 0.05).

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Registry No. 1, 56741-94-7; 2 ($\mathbb{R}^1 = \mathbb{H}, \mathbb{R}^2 = \mathbb{C}\mathbb{H}_3$), 92519-10-3; **2** ($\mathbf{R}^1 = \mathbf{Br}, \mathbf{R}^2 = \mathbf{CH}_3$), 92519-08-9; **2** ($\mathbf{R}^1 = \mathbf{I}, \mathbf{R}^2 = \mathbf{H}$), 72943-43-2; 2 ($\mathbb{R}^1 = \mathbb{I}, \mathbb{R}^2 = \mathbb{C}H_3$), 92519-09-0; 2 ($\mathbb{R}^1 = \mathbb{I}, \mathbb{R}^2 = \mathbb{C}H_2\mathbb{C}H\mathbb{C}H_2$), 102649-71-8; 3 ($\mathbb{R}^1 = \mathbb{H}, \mathbb{R}^2 = \mathbb{C}H_3$), 100008-30-8; 3 ($\mathbb{R}^1 = \mathbb{B}r, \mathbb{R}^2$ = CH₃), 102649-63-8; 3 (R¹ = I, R² = COCH₃), 102649-68-3; 3 (R¹ = I, $\mathbf{R}^2 = \mathbf{CH}_3$), 102649-70-7; 4 ($\mathbf{R}^1 = \mathbf{I}$), 102649-67-2; 4 ($\mathbf{R}^1 = \mathbf{H}$), 31937-04-9; 4 (R¹ = Br), 102649-64-9; 5 (R¹ = H), 102649-58-1; 5 ($\mathbb{R}^1 = \mathbb{B}r$), 102649-65-0; 5 ($\mathbb{R}^1 = I$), 102649-69-4; 6, 56741-95-8; 7, 74856-68-1; 8, 102649-59-2; 9, 36822-11-4; 9 (dione), 102649-77-4; 10, 102649-60-5; 11, 74303-68-7; 12, 102649-61-6; 14, 102649-66-1; 15, 61736-36-5; 15 (5-bromo), 102649-72-9; 15 (5-iodo), 102649-73-0; 16, 42542-57-4; 16 (5-bromo), 102649-74-1; 16 (5-iodo), 102649-75-2; 16 (5-chloro), 102649-76-3; H2NCONH2, 57-13-6; C6H5COCH2C-O2CH2CH3, 94-02-0; H3CNHCONH2, 598-50-5; H2NCSNH2, 62-56-6; H₃CNHC(NH)NH₂-1/2H₂SO₄, 598-12-9; (CH₃)2NC(NH)N-H₂·HCl, 1186-46-5; 6-phenyl-2,4-pyrimidinedione, 13345-09-0; 1,3-dimethyl-5-bromo-6-phenyl-2,4-pyrimidinedione, 98854-09-2; 5-bromo-6-phenyl-2,4-pyrimidinedione, 16290-56-5; 2-amino-1methyl-6-phenyl-4-pyrimidinone, 102649-62-7; biguanide, 56-03-1; 1-hydroxy-2-amino-5-bromo-6-phenyl-4-pyrimidinone, 78222-48-7.

Topical Nonsteroidal Antipsoriatic Agents. 1. 1,2,3,4-Tetraoxygenated Naphthalene Derivatives¹

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On the basis of previous observations that both 2,3-dihydro-2,2,3,3-tetrahydroxy-1,4-naphthoquinone (oxoline, 1) and 6-chloroisonaphthazarin (2) had demonstrated antipsoriatic activity in vivo, a series of structural derivatives of 2 were prepared and examined in the Scholtz-Dumas topical psoriasis bioassay. Of these six (5, 6, 9a, 10, 11a, 11b), the most effective compound was found to be 6-chloro-1,4-diacetoxy-2,3-dimethoxynaphthalene (RS-43179, lonapalene, 11a). An extensive series of 1,2,3,4-tetraoxygenated naphthalenes (16-74) incorporating variations of the ester, ether, and aryl substituents were prepared as analogues of 11a to examine the structural requirements for activity and were screened in vivo as inhibitors of arachidonic acid induced mouse ear edema, a topical bioassay capable of detecting 5-lipoxygenase inhibitors. Net lipophilicity, hydrolytic stability, and ring substitution play significant roles in determining the observed in vivo activity. Lonapalene (11a) is currently in clinical development as a topically applied nonsteroidal antipsoriatic agent.

Psoriasis is a chronic, relapsing hyperproliferative and inflammatory disease of the skin characterized by symmetrically distributed silvery red, scaling plaques found on the scalp, the extremities, back, buttocks, and especially the knees and elbows. In severe cases, most of the body surface may be involved. An estimated 6-8 million persons in the United States have psoriasis, contributing to a 2-6% incidence worldwide. While the exact etiology of the disease remains elusive, there now exist a number of treatments which provide temporary relief,² including coal tar and UV irradiation (the Goeckermann regimen), psoralen and UV-A irradiation (PUVA), oral retinoids, corticosteroids, and antimetabolites. Probably the most widespread treatment used, however, is topical anthralin, an irritating and mutagenic compound that leaves a brown

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