

Articles

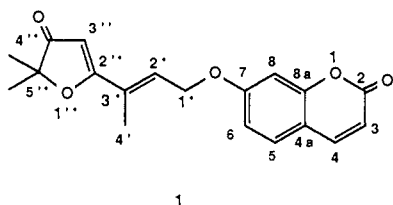
Synthesis and Cytostatic Activity of Geiparvarin Analogues

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In an attempt to determine some of the structural features of geiparvarin (1) that account for its cytostatic activity in vitro, a series of geiparvarin analogues (4a-g) modified in the 3(2*H*)-furanone moiety have been designed and synthesized. The preparation of 4a-g was achieved through a new approach to the 3(2*H*)-furanone ring based on the elaboration of isoxazole derivatives. Among these synthetic analogues, 4b, the 5-methyl-5-ethyl derivative, proved as active as 1 in inhibiting the proliferation of murine and human tumor cell lines in vitro. As a rule, substitutions at the C₅ atom of the 3(2*H*)-furanone moiety of 1 slightly decreased the cytostatic activity of geiparvarin. Several geiparvarin analogues described in this study (i.e. the 5-methyl-5-ethyl derivative 4b, 3(2*H*)-furanimine 4c, 5-methyl derivative 4f, and 5-ethyl derivative 4g) showed such activity in vitro and deserve further investigation for their antitumor potentials in vivo.

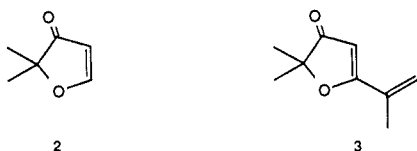
Natural products isolated from extracts of plants and herbs hold much promise in providing valuable and novel antitumor agents.¹ This is the case of geiparvarin (1), a



naturally occurring product that has been the subject of several synthetic investigations since its isolation from the leaves of *Geijera parviflora* Lindl and characterization in 1967.^{2,3} It is important to note that the central structural feature of the molecule, namely the 3(2*H*)-furanone ring, is common to an increasing number of antitumor agents, including such diverse substances as jatrophone,⁴ eremantolides A, B, and C,⁵ and chilenone A.⁶

Furthermore, during studies directed to define the reactivity of the 3(2*H*)-furanone system toward potential bionucleophiles, Smith⁷ has shown that simple 3(2*H*)-furanones and alkenylfuranones, such as 2 and 3, respectively, undergo 1,4 and 1,6 conjugate addition of propanethiol under acidic and basic conditions.

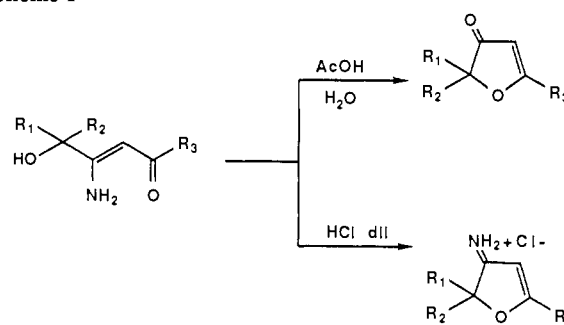
Taking into account that geiparvarin itself as well as jatrophone and eremantolides reacted with propanethiol via 1,6 conjugate addition to afford Michael adducts, the same authors⁷ suggested that this reactivity of the alkenyl-3(2*H*)-furanone system can be considered as the possible mode of biological action of this class of antitumor agents. For instance, geiparvarin and propenylfuranone 3 displayed activity against P-388 lymphocytic leukemia with T/C values of 130 at 400 mg/kg and 128 at 100 mg/kg, respectively.⁷



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Scheme I

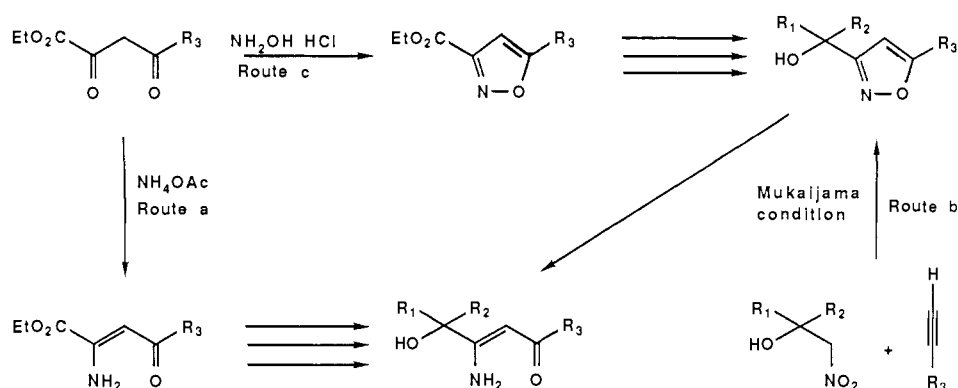


Our recent studies on 3(2*H*)-furanone rings functionalized at the 5-position led us to devise a simple and efficient route to 1 utilizing a new approach based on isoxazole chemistry.⁸

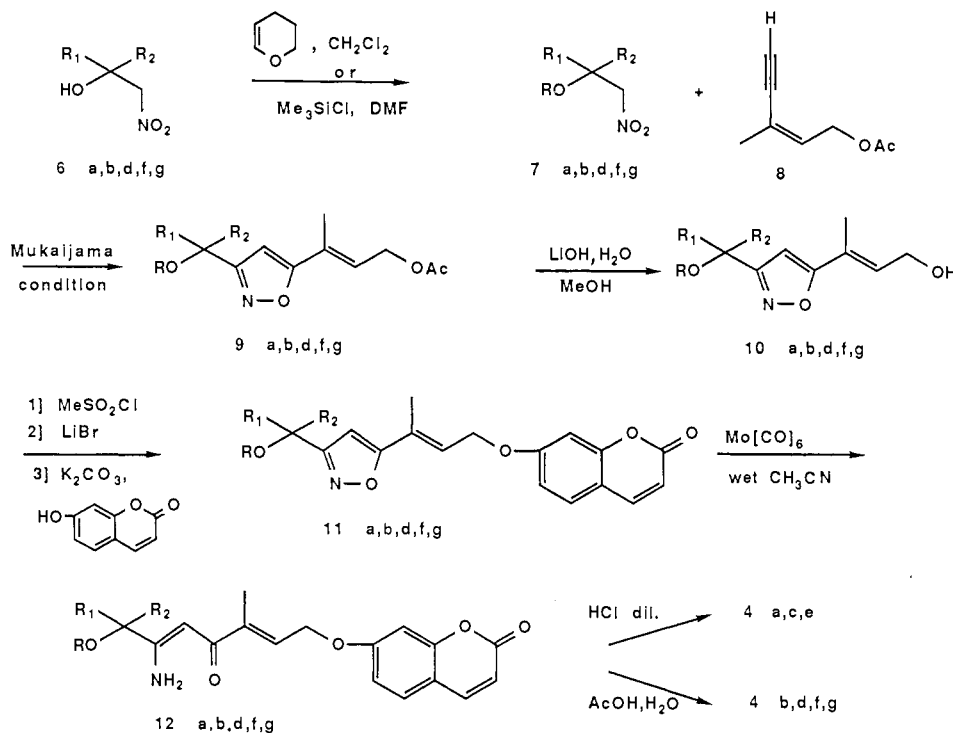
In order to investigate structure-activity relationships of 1, we have undertaken a program aimed at synthesizing new geiparvarin analogues modified on the 3(2*H*)-furanone ring, assuming that the coumarin moiety of the molecule plays a part only in conferring the right degree of lipophilicity. Recently, a series of geiparvarin analogues in which the coumarin ring was replaced by 4*H*-1-benzopyran-4-one and xanthone have been reported.⁹

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Scheme II



Scheme III



7,9 a,b,d R = Me₃Si- ; f,g R = THP

10,11,12 a,b,d R = H ; f,g R = THP

In this paper, we report the synthesis and the cytostatic activity of a series of new geiparvarin analogues (4a-g) as well as the cytostatic activity of a group of simple 3-(2H)-furanone and 3-(2H)-furanimines (5a-i), some of which have previously been reported¹⁰ and others that were newly synthesized.

Chemistry

The analogues 4a-g as well as the newly synthesized 3-(2H)-furanones and 3-(2H)-furanimines (5a-i) were prepared according to the routes depicted in Schemes I-III.

In previous synthetic studies,^{8,10,11} we have reported the controlled reaction conditions to obtain 3-(2H)-furanones

and 3-(2H)-furanimines. This reaction is based on acid-promoted cyclocondensation of the gamma-hydroxy-beta-enaminone, namely treatment with acetic acid in THF/water at room temperature or with hydrochloric acid in THF at room temperature (Scheme I).

The gamma-hydroxy-beta-enaminone was obtained both through regiospecific reaction of ammonium acetate in the presence of acetic acid on ethyl 2,4-dioxoalkanoate followed by elaboration of the ester function and through reductive ring opening of the appropriate isoxazole. The latter in turn was obtained both through [3 + 2] cycloaddition of a nitrile oxide generated from the suitable nitro derivative under Mukaijama conditions on an acetylenic compound and through regiospecific cyclocondensation of hydroxylamine hydrochloride on ethyl 2,4-dioxoalkanoate followed by elaboration of the ester function (Scheme II).

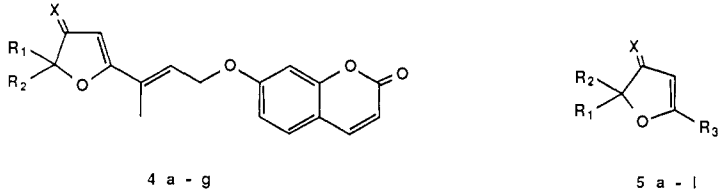
The choice of the synthetic route to be used in the preparation of the gamma-hydroxy-beta-enaminone depends on the availability of the starting material: in the case of simple 3-(2H)-furanones and 3-(2H)-furanimines (5a-i) (R₃ = alkyl,

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Table I. Properties of Geiparvarin Analogues (4a-g) and 3(2H)-Furanones and Iminofuranones 5a-i



no.	R ₁	R ₂	R ₃	X	mp, °C	salt	solvent	yield, %	formula	anal. calcd (found)				
										C	H	N	Cl	
1	Me	Me		O	ref 8									
4a	Me	Me		NH	270-275	chloride	MeOH/Et ₂ O	92	C ₁₉ H ₂₀ NO ₄ Cl	63.05 (63.09)	5.53 (5.53)	3.87 (3.88)	9.80 (9.72)	
4b	Me	Et		O	118		Et ₂ O	94	C ₂₀ H ₂₀ O ₅	70.57 (70.63)	5.92 (5.90)			
4c	Me	Et		NH	178-180	chloride	Me ₂ CO	90	C ₂₀ H ₂₂ NO ₄ Cl	63.92 (63.98)	5.86 (5.85)	3.72 (3.70)	9.44 (9.35)	
4d	-(CH ₂) ₅ -			O	157-160		Me ₂ CO	93	C ₂₂ H ₂₂ O ₅	72.11 (72.20)	6.05 (6.09)			
4e	-(CH ₂) ₅ -			NH	145-150	chloride	MeOH/Et ₂ O	88	C ₂₂ H ₂₄ NO ₄ Cl	68.13 (68.22)	6.19 (6.20)	3.61 (3.63)	9.14 (8.95)	
4f	Me	H		O	165-166		MeCN	90	C ₁₈ H ₁₆ O ₅	69.22 (69.18)	5.16 (5.20)			
4g	Et	H		O	130-131		AcOEt	92	C ₁₉ H ₁₈ O ₅	69.92 (69.94)	5.56 (5.56)			
5a	Me	Me	^a	O	ref 10									
5b	Ph	Ph	Me	NH	200-dec.	chloride	CHCl ₃ /Et ₂ O	86	C ₁₇ H ₁₆ ClNO	71.45 (71.44)	5.06 (5.06)	4.90 (4.90)	12.41 (12.40)	
5c	H	H	<i>o</i> -MeO-Ph	O	ref 10									
5d	H	H	Ph	O	ref 10									
5e	Me	Me	<i>o</i> -MeO-Ph	O	ref 10									
5f	Me	Me	<i>i</i> -C ₄ H ₉	NH	ref 10									
5g	Me	Me	Ph	NH	ref 10									
5h	Me	Me	<i>n</i> -C ₅ H ₁₁	NH	ref 10									
5i	Me	Me	<i>i</i> -C ₄ H ₉	NH	139-140	oxalate	CHCl ₃ /Et ₂ O	94	C ₁₂ H ₁₉ NO ₅	56.02 (56.15)	7.44 (7.50)	5.44 (5.43)		

^a CH₂C(CH₃)=CHCH₂OH.

aryl), routes a and c are the best ones, whereas in the case of complex 3(2H)-furanones route b is more suitable, allowing the assembly of highly functionalized partners. The synthesis of analogues 4a-g was performed according to route b as shown in Scheme III.

The isoxazole scaffolds (9a,b,d,f,g) destined to become the 3(2H)-furanone moiety of the geiparvarin analogues (4a-g) could be assembled through 1,3-dipolar cycloaddition between two readily available precursors, namely the primary nitro derivatives (7a,b,d,f,g), easily accessible by protection of the hydroxyl function as the trimethylsilyl ether or tetrahydropyranyl ether of the Henry adducts (6a,b,d,f,g) between the appropriate carbonyl compound and nitromethane, and the acetate (8) of the commercially available enyne. Thus, the nitrile oxide generated from 7a,b,d,f,g) under Mukaiyama conditions cycloadded regioselectively and chemoselectively to the acetyl derivative 8 to produce the 3,5-substituted isoxazoles (9a,b,d,f,g) in 80-90% yield. Removal of the acetyl group from the intermediates 9a,b,d) by treatment at 20 °C with lithium hydroxide solution (MeOH/H₂O 9/1) followed by acid workup furnished diols 10a,b,d). In entries f and g the tetrahydropyranyl group was maintained in order to avoid ether formation on both hydroxyl functions in the next step. Compounds 10f,g) were obtained from 9f,g) through basic hydrolysis with lithium hydroxide followed by neutral workup.

The primary hydroxyl group of 10a,b,d,f,g) reacted, selectively in the case of entries a, b, and d, with methanesulfonyl chloride in the presence of triethylamine to afford the corresponding crude mesylates, which on treatment with 7-hydroxycoumarin in the presence of lithium bromide gave rise to a 70-80% yield of the key intermediates

11a,b,d,f,g). At this point the stage was set for the unmasking of the isoxazole nucleus to reveal the latent 1,3-diketone precursor of the 3(2H)-furanone moiety. In this case, reductive ring opening of 11a,b,d,f,g) cannot be performed with catalytic hydrogenation owing to the presence of a C-C double bond in the side chain.

Among the nonhydrogenolytic methods for the ring opening of the isoxazole, we chose the protocol introduced by Nitta, involving molybdenum hexacarbonyl in wet acetonitrile. This procedure generated the expected enaminones 12a,b,d,f,g) in 80-95% yield. On exposure of 12b,d,f,g) to acetic acid at 20 °C for 24 h, an easy cyclodehydration took place, affording 4b,d,f,g) in 90-95% yield. On exposure of 12a,b,d) to hydrochloric acid at room temperature for 3 h, cyclodehydration took place, affording 3(2H)-furanimines 4a,c,e) in 86-92% yield.

Biological Evaluation

Geiparvarin (1), analogues 4a-g) with modifications in the 3(2H)-furanone moiety, and compounds with modified 3(2H)-furanone (5a, 5c, 8d, 5e) and 3(2H)-furanimine (5b, 5f, 5g, 5h, 5i) moieties were evaluated for their inhibitory effects on the proliferation of murine (L1210, FM3A) and human (Raji, Molt/4F, CEM, H9 and MT4) tumor cells (Table II).

Among the geiparvarin analogues modified in the 3(2H)-furanone moiety (4a-g), compound 4b) proved as active as the parent compound in inhibiting the proliferation of murine and human tumor cell lines in vitro. Its 50% inhibitory dose (ID₅₀) was 2-3 μg/mL for all cell lines except for MT4. Interestingly, both 1 and 4b) proved 10-fold more inhibitory to the growth of the HTLV-I (human T-cell leukemia virus type 1) transformed T4 lymphocyte

Table II. Inhibitory Effects of the Geiparvarin Analogues **4a-g** and Their 3(2*H*)-Furanones and Iminofuranones **5a-i** on the Proliferation of Murine or Human Tumor Cells

no.	ID ₅₀ , ^a µg/mL						
	L1210	FM3A	Raji	Molt/4F	MT4	CEM	H9
1	2.07 ± 0.66	3.20 ± 0.39	2.19 ± 0.36	2.84 ± 0.42	0.320 ± 0.077	2.00 ± 0.84	1.97 ± 0.83
4a	18.4 ± 4.6	20.7 ± 0.82	26.0 ± 2.7	25.8 ± 6.5	8.77 ± 6.52	18.8 ± 9.1	21.7 ± 9.9
4b	2.05 ± 0.49	2.96 ± 0.26	2.59 ± 0.56	3.02 ± 0.41	0.530 ± 0.132	2.49 ± 0.73	2.15 ± 0.85
4c	4.71 ± 0.68	14.1 ± 1.55	17.9 ± 4.0	3.17 ± 0.30	2.65 ± 0.23	3.57 ± 1.59	2.58 ± 0.13
4d	1.82 ± 0.58	6.61 ± 2.16	5.43 ± 1.37	12.7 ± 10.3	2.10 ± 0.91	2.72 ± 0.08	2.08 ± 0.93
4e	4.05 ± 1.05	15.8 ± 3.46	23.5 ± 3.6	20.5 ± 6.5	4.93 ± 2.15	4.95 ± 3.51	14.2 ± 12.8
4f	7.05 ± 2.66	15.6 ± 4.76	10.4 ± 2.98	3.32 ± 0.45	2.56 ± 0.68	3.56 ± 0.65	2.91 ± 0.87
4g	9.77 ± 4.71	17.2 ± 2.36	2.43 ± 0.55	2.84 ± 0.19	1.63 ± 0.49	10.27 ± 1.13	4.10 ± 3.47
5a	15.8 ± 4.01	23.3 ± 3.24	16.6 ± 5.7	15.2 ± 7.7	3.05 ± 0.61	3.2 ± 6.8	9.13 ± 4.61
5b	24.2 ± 4.02	25.4 ± 1.6	24.0 ± 5.0	25.3 ± 3.6	26.5 ± 4.5	29.3 ± 2.3	30.0 ± 7.7
5c	>100	>100	>100	>100	44.4 ± 10.6	>100	66.3 ± 31.7
5d	45.8 ± 14.0	31.4 ± 3.6	50.1 ± 2.7	31.2 ± 7.5	31.7 ± 1.1	30.2 ± 3.3	38.6 ± 4.4
5e	40.4 ± 0.26	31.7 ± 4.61	89.1 ± 18.9	53.3 ± 22.9	30.4 ± 6.3	66.1 ± 36.9	63.9 ± 41.7
5f	>100	33.0 ± 3.21	>100	>100	>100	>100	>100
5g	66.4 ± 31.3	24.7 ± 3.7	>100	>100	76.2 ± 25.6	>100	>100
5h	98.3 ± 2.89	24.2 ± 2.14	>100	>100	>100	>100	>100
5i	>100	33.2 ± 6.83	>100	>100	>100	>100	>100

^a 50% inhibitory dose or dose required to inhibit cell proliferation by 50%.

cell line (MT4) than to the growth of the other T4 lymphocyte cell lines. As a rule, substitutions at the C-5 atom of the 3(2*H*)-furanone moiety of geiparvarin slightly decreased the cytostatic activity in vitro. Compound **4a**, the 3(2*H*)-furanimine derivative of **1**, was even 10-fold less cytostatic than geiparvarin. When the 3(2*H*)-furanone moiety of geiparvarin was linked to 4*H*-1-benzopyran-4-one or xanthone rather than to a coumarin ring, the inhibitory effects of these compounds on tumor cell proliferation was substantially decreased.⁹ Compounds **5a**, **5d**, and **5e** were 5–20-fold less cytostatic, whereas **5c** was inactive against most of the tumor cell lines at 100 µg/mL. Replacement of the carbonyl group in 3(2*H*)-furanone by an imine function further decreased the cytostatic activities of these compounds. Only **5b** showed a moderate antiproliferative effect (ID₅₀ = 25 µg/mL).

In conclusion, minor substitutions at C-5 of geiparvarin and deletion of the coumarin part from the furanone or furanimine moiety resulted in a modestly or substantially decreased cytostatic effect of compound **1**. However, several members of the geiparvarin analogues described in this study (i.e. **4b**, **4c**, **4f**, and **4g**) show such activity in vitro that they deserve further investigation for their antitumor potentials in vivo.

Experimental Section

Melting points were determined on a Büchi apparatus and were uncorrected. Reaction courses and product mixtures were routinely monitored by thin-layer chromatography (TLC) on silica gel precoated F₂₅₄ Merck plates. Infrared (IR) spectra were recorded on a Perkin-Elmer 297 spectrometer. Nuclear magnetic resonance (NMR) spectra were obtained with a Bruker 200 spectrometer for solution in CDCl₃, and peak positions are given in parts per million downfield from tetramethylsilane as an internal standard. The various splitting patterns were designated as follow: s, singlet; br s, broad singlet; d, doublet; t, triplet; q, quartet; m, multiplet. The synthesis of **4b** and **4c** are given as examples. The other geiparvarin analogues were prepared according to those procedures from appropriate starting materials. The final compounds (**4a-g**) are listed in Table I with their analytical data. Starting materials, compounds **6a**,^{12a} **6b**,^{12b} **6d**,^{12c} and **6f,g**,^{12d} have been prepared according to known directions. Compounds **7a**, **7b**, **7d**, **7f**, and **7g** were prepared from the corresponding nitro alcohol by standard procedures.^{13a,b}

Preparation of 4b and 4c. 5-[(*E*)-3-(Acetyloxy)-1-methyl-1-propenyl]- α -methyl- α -ethyl-3-isoxazolemethanol Trimethylsilyl Ether (**9b**, R = Me₃Si, R₁ = Me, R₂ = Et). To a solution of **7b** (R = Me₃Si, R₁ = Me, R₂ = Et) (10.7 g, 52 mmol) and **8** (21.5 g, 156 mmol) in dry benzene (50 mL) containing triethylamine (0.5 mL) was added dropwise phenyl isocyanate (14.1 mL, 130 mmol) in benzene (10 mL) at room temperature, and the mixture was allowed to stand overnight. The cooled mixture (5 °C) was filtered, and the filtrate was washed with brine (3 × 50 mL), dried, and concentrated in vacuo. The residue crude oil was flash chromatographed on silica gel (ether/light petroleum 2/8) to give **9a** as an oil (14.7 g, 87%); IR (neat) 1740, 1590 cm⁻¹; ¹H NMR (CDCl₃) δ 0.1 (s, 9 H), 0.87 (t, 3 H, *J* = 7 Hz), 1.6 (s, 3 H), 1.9 (q, 2 H, *J* = 7 Hz), 2.0 (s, 3 H), 4.8 (d, 2 H, *J* = 6 Hz), 6.3 (s, 1 H), 6.4 (br t, 1 H).

5-[(*E*)-3-Hydroxy-1-methyl-1-propenyl]- α -methyl- α -ethyl-3-isoxazolemethanol (**10b**, R = H, R₁ = Me, R₂ = Et). Lithium hydroxide monohydrate (8.4 g, 200 mmol) was added to a stirred solution of **9b** (R = Me₃Si, R₁ = Me, R₂ = Et) (27 g, 83 mmol) in methanol (200 mL) containing 30 mL of water. After 10 min at 0 °C the solution was concentrated in vacuo, and the residue was poured into 2 N hydrochloric acid (30 mL) and extracted with ethyl acetate (3 × 100 mL). The dried organic extracts were evaporated in vacuo, and the residue was flash chromatographed on silica gel (ethyl acetate/light petroleum, 1:1) to give **10a** as an oil (16.5 g, 94%); IR (neat) 3300, 1650, 1560 cm⁻¹; ¹H NMR (CDCl₃) δ 0.89 (t, 3 H, *J* = 7 Hz), 1.6 (s, 3 H), 1.85 (q, 2 H, *J* = 7 Hz), 1.99 (s, 3 H), 2.3 (br s, 1 H), 2.7 (br s, 1 H), 4.4 (d, 1 H, *J* = 6 Hz), 6.2 (s, 1 H), 6.4 (br t, 1 H).

7-[[(*E*)-3-[3-(1-Hydroxy-1-methylpropyl)isoxazol-5-yl]-2-butenyl]oxy]-2*H*-1-benzopyran-2-one (**11b**, R = H, R₁ = Me, R₂ = Et). To an ice-cooled solution of **10b** (R = H, R₁ = Me, R₂ = Et) (0.85 g, 4 mmol) in methylene chloride (10 mL) containing triethylamine (0.5 mL, 6 mmol) was added dropwise methanesulfonyl chloride (0.35 mL, 5 mmol) in methylene chloride (5 mL) with stirring. After 30 min at room temperature, the mixture was treated with 2% citric acid (3 mL) and the organic phase separated, dried, and concentrated in vacuo. The crude mesylate was immediately heated at reflux for 0.5 h in acetone solution (20 mL) containing 7-hydroxycoumarin (0.65 g, 4 mmol), potassium carbonate (0.55 g, 4 mmol), and lithium bromide (0.1 g). After removal of the solvent in vacuo, the residue was poured into water (50 mL) and extracted with ethyl acetate (2 × 50 mL). The combined extracts were washed with 5% aqueous ammonia (3 × 20 mL) and brine (30 mL) and dried. The solvent was removed in vacuo to give **11b** (1.33 g, 78%); mp 124–126 °C (AcOEt, light petroleum); IR (CHCl₃) 3300, 1730, 1610, 1570, 1500 cm⁻¹; ¹H NMR (CDCl₃) δ 0.88 (t, 3 H, *J* = 7 Hz), 1.58 (s, 3 H),

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1.85 (q, 2 H, $J = 7$ Hz), 2.1 (s, 3 H), 2.5 (br s, 1 H), 4.85 (d, 2 H, $J = 6$ Hz), 6.2 (s, 1 H), 6.25 (d, 1 H, $J = 9$ Hz), 6.55 (br t, 1 H), 6.8 (m, 2 H), 7.4 (d, 1 H, $J = 8.5$ Hz), 7.65 (d, 1 H, $J = 9$ Hz).

7-[[*(E,E)*-6-Amino-7-hydroxy-3-methyl-7-ethyl-4-oxo-2,5-octadienyl]oxy]-2*H*-1-benzopyran-2-one (**12b**, $R = H$, $R_1 = Me$, $R_2 = Et$). A solution of isoxazole **11b** ($R = H$, $R_2 = Et$) (2.13 g, 6 mmol) in acetonitrile (40 mL) containing water (50 drops) was treated with molybdenum hexacarbonyl (0.81 g, 3 mmol) and heated at reflux for 1.5 h with stirring. Celite (5 g) was added to the cooled solution, and the resulting mixture was evaporated in vacuo. The residue was flash chromatographed on silica gel (ethyl acetate/light petroleum 6/4) to give **12a** (2.0 g, 94%): mp 104–105 °C (methanol); IR (CHCl₃) 3450, 1730, 1615, 1550, 1500 cm⁻¹; ¹H NMR (CDCl₃) δ 0.94 (t, 3 H, $J = 7$ Hz), 1.5 (s, 3 H), 1.5–1.85 (m, 3 H), 1.96 (s, 3 H), 4.8 (d, 2 H, $J = 6$ Hz), 5.3 (s, 1 H), 6.25 (d, 1 H, $J = 9$ Hz), 6.4 (br, 2 H), 6.64 (m, 2 H), 7.76 (d, 1 H, $J = 8.5$ Hz), 7.7 (d, 1 H, $J = 9$ Hz), 10.5 (br s, 1 H).

7-[[3-(4,5-Dihydro-5-methyl-5-ethyl-4-oxo-2-furanyl)-2-butenyl]oxy]-2*H*-1-benzopyran-2-one (**4b**, $R_1 = Me$, $R_2 = Et$, $X = O$). A solution of **12b** ($R = H$, $R_1 = Me$, $R_2 = Et$) (0.92 g, 2.6 mmol) in tetrahydrofuran (20 mL) was stirred at room temperature for 24 h with 75% acetic acid (20 mL). The mixture was poured into brine (50 mL) and extracted with ethyl acetate (3 × 25 mL). The combined organic extracts were washed in turn with saturated aqueous sodium hydrogen carbonate (2 × 10 mL) and brine (2 × 10 mL) and dried. The solvent was removed under reduced pressure and the residue was purified by flash chromatography, eluting with methylene chloride/ethyl acetate 10/1, to give **4b** (0.83 g, 94%): mp 118 °C (ether); IR (CHCl₃) 1730, 1690, 1660, 1610, 1550 cm⁻¹; ¹H NMR (CDCl₃) δ 0.87 (t, 3 H, $J = 7$ Hz), 1.38 (s, 3 H), 1.8 (q, 2 H, $J = 7$ Hz), 2.0 (s, 3 H), 4.85 (d, 2 H, $J = 6$ Hz), 5.65 (s, 1 H), 6.3 (d, 1 H, $J = 9$ Hz), 6.9 (m, 3 H), 7.2 (d, 1 H, $J = 8.5$ Hz), 7.4 (d, 1 H, $J = 9$ Hz).

7-[[3-(4,5-Dihydro-5-methyl-5-ethyl-4-imino-2-furanyl)-2-butenyl]oxy]-2*H*-1-benzopyran-2-one Hydrochloride (**4c**, $R_1 = Me$, $R_2 = Et$, $X = NH$). A solution of enaminone **12b** ($R = H$, $R_1 = Me$, $R_2 = Et$) (1.2 g, 3.3 mmol) in THF (20 mL) containing 5% HCl (20 mL) was stirred at room temperature for 3 h. Evaporation of the solvent in vacuo, followed by extraction with ether (50 mL) to remove some impurities and finally with CHCl₃ (3 × 25 mL). The dried chloroform extracts were evaporated in

vacuo, and the solid residue was crystallized from acetone to give **4c** (1.13 g, 90%): mp 178–180 °C; IR (CHCl₃) 3400–2500, 1720, 1610, 1520 cm⁻¹; ¹H NMR (CDCl₃) δ 0.85 (t, 3 H, $J = 7$ Hz), 1.6 (s, 3 H), 2.0 (m, 5 H), 5.0 (d, 2 H, $J = 6$ Hz), 6.3 (d, 1 H, $J = 9$ Hz), 6.4 (s, 1 H), 7.0 (m, 3 H), 7.65 (d, 1 H, $J = 8.5$ Hz), 8.0 (d, 1 H, $J = 9$ Hz), 11.5 (br s, 2 H).

Cytostatic Assays. The cytostatic effects of the compounds were assessed on the basis of the inhibition of cell proliferation; these assays were carried out with murine leukemia L1210 and mammary carcinoma FM3A cells, human B-lymphoblast Raji and T-lymphoblast Molt/4F cells, and human T4 lymphocyte MT4, CEM, and H9 cells in their exponential growth phase. This assay procedure has been described previously.¹⁴

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Registry No. 1, 36413-91-9; **4a**, 117310-13-1; **4b**, 117310-14-2; **4c**, 117310-15-3; **4d**, 117310-16-4; **4e**, 117310-17-5; **4f**, 117310-18-6; **4g**, 117310-19-7; **5a**, 87064-11-7; **5b**, 117310-20-0; **5c**, 82284-73-9; **5d**, 5198-63-0; **5e**, 117310-21-1; **5f**, 117310-22-2; **5g**, 117310-23-3; **5h**, 117310-24-4; **5i**, 117310-25-5; **6a**, 5447-98-3; **6b**, 22916-76-3; **6d**, 3164-73-6; **6f**, 3156-73-8; **6g**, 3156-74-9; **7a**, 55816-65-4; **7b**, 117310-26-6; **7d**, 117310-27-7; **7f**, 69386-03-4; **7g**, 117310-28-8; **8**, 35272-88-9; **9a**, 117285-95-7; **9b**, 117310-29-9; **9d**, 117310-30-2; **9f**, 117310-31-3; **9g**, 117310-32-4; **10a**, 101417-83-8; **10a** (mesylate), 101417-84-9; **10b**, 117310-33-5; **10b** (mesylate), 117310-37-9; **10d**, 117310-34-6; **10d** (mesylate), 117310-38-0; **10f**, 117310-35-7; **10f** (mesylate), 117310-39-1; **10g**, 117310-36-8; **10g** (mesylate), 117310-40-4; **11a**, 101417-85-0; **11b**, 117310-41-5; **11d**, 117310-42-6; **11f**, 117310-43-7; **11g**, 117310-44-8; **12a**, 101417-86-1; **12b**, 117310-45-9; **12d**, 117310-46-0; **12f**, 117310-47-1; **12g**, 117310-48-2; 7-hydroxycoumarin, 93-35-6.

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