hydrous suspension of la (112.6 mg, 0.5 mmol) in dimethylformamide (3 mL). After being stirred with exclusion of moisture for 45 min, the resulting solution was treated with 90% chloroacetaldehyde in ethyl ether (50 *nL,* ca. 0.6 mmol). Stirring was continued at room temperature for 45 min, then at 45° C for 1 h. The reaction mixture was made basic by addition of concentrated aqueous ammonia (2 mL), and after 1 h the obtained solution was evaporated to dryness. The residue after evaporation was dissolved in 50% aqueous ethanol and adsorbed on a portion of silica gel (ca. 3 g, 70-230 mesh) by repeated coevaporation with ethanol. The dried gel was applied onto a silica gel short column $(2.8 \times 9 \text{ cm})$ and the product was eluted with chloroform-methanol (6:1). Fractions 34-47 (at 6 mL) containing the TLC-homogeneous 3a were evaporated to give 22.4 mg (18%) of a white solid. The α were evaporated to give $\sum_{i=1}^{\infty} \sum_{j=1}^{\infty} \sum_{j=1}^{\infty$ scribed in method B.

Method B. Acetal derivative 4a (400 mg, 1.17 mmol) was gently refluxed in 40% acetic acid (25 mL) with a slow distillation of solvent in order to remove ethanol formed in the reaction. During the reaction time ca. 10 mL of a distillate (bp 92-94 ⁰C) were collected. The rest of solvent was then evaporated under diminished pressure and a resulting white residue was coevaporated twice with 2-propanol, which gave the chromatographically pure 3a as a white powder (yield 286 mg, 98%). An analytical sample was crystallized from water: mp 248.5 °C dec; ¹H NMR δ 3.50 (s, 4,4'H and 5'H), 4.67 (t, 1, OH), 5.50 (s, 2, l'H), 7.45 and 7.63 (2 d, *J* - 2.6 Hz, 2, HC=CH), 8.04 (s, 1, 8H), 12.48 (br s, 1, NH). Anal. $(C_{10}H_{11}N_5O_3)$ C, H, N.

3,9-Dinydro-3-[(l,3-dihydroxy-2-propoxy)methyl]-9-oxo- $5H$ -imidazo[1,2-a]purine (3b). Acetal derivative 4b (100 mg, 0.269 mmol) was hydrolyzed in 1.2 N hydrochloric acid (6 mL) at room temperature for 18 h. At this time TLC in solvent C showed that substrate 4b had completely disappeared, and the reaction mixture was made alkaline by addition of concentrated aqueous ammonia (5 mL). An excess of ammonia was gently evaporated after 1 h, and a resulting colorless solution was adsorbed on a portion of silica gel (ca. 3 g, 70-230 mesh) by evaporation. The dried gel was applied onto a silica gel short column

 $(2.5 \times 7 \text{ cm})$ and the product was eluted with chloroform-methanol (6:1). Fractions 35-45 (at 6 mL) contained chromatographically pure 3b and were evaporated to a crystalline solid (yield 58.4 mg, 78%). An analytical sample was recrystallized from 90% aqueous ethanol: mp 250-251 ⁰C dec; ¹H NMR *d* 3.41 (m, 4,3H and 5'H), 3.64 (p, 1, 4'H), 4.60 (t, 2, OH), 5.59 (s, 2, l'H), 7.42 and 7.61 (2 d, $J = 2.6$ Hz, 2, HC=CH), 8.03 (s, 1, 8H), 12.46 (br s, 1, NH). Anal. $(C_{11}H_{13}N_5O_4)$ C, H, N.

Antiviral Activity and Cytotoxicity Evaluation. The methods for measuring virus-induced cytopathogenicity,¹¹ viral plaque formation,¹² host cell DNA synthesis,¹¹ and cell growth,¹² as well as the sources of the different virus strains,^{11,13,14} have been described previously.

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5-(5-Bromothien-2-yl)-2'-deoxyuridine and 5-(5-Chlorothien-2-yl)-2/ -deoxyuridine Are Equipotent to (E) **-5-(2-Bromovinyl)-2'-deoxyuridine in the Inhibition of Herpes Simplex Virus Type I Replication**

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2'-Deoxyuridines with a five-membered heterocyclic substituent in the 5-position were synthesized by palladiumcatalyzed coupling reactions of 5-iodo-2'-deoxyuridine with the activated heteroaromatics. Further modification of the compound with the 5-thien-2-yl substituent gave 5-(5-bromothien-2-yl)-2'-deoxyuridine and 5-(5-chlorothienyl-2-yl)-2'-deoxyuridine. Both compounds show potent and selective activity against herpes simplex virus type 1 and varicella-zoster virus.

Introduction

Several 5-substituted pyrimidine nucleosides show potent anti herpes virus activity.¹ Among them, 5-iodo-2'-deoxyuridine (IdUrd) and 5-(trifluoromethyl)-2' deoxyuridine have been in clinical use for years. The most active congeners among the 5-substituted 2'-deoxyuridine derivatives are (E) -5- $(2$ -halogenovinyl)-2'-deoxyuridines.² These compounds are particularly active against herpes simplex virus type 1 (HSV-I) and varicella-zoster virus. Their antiviral selectivity is primarily due to a preferential phosphorylation by the virus-encoded thymidine kinase and a greater inhibitory effect of the 5'-triphosphates on the viral DNA polymerase than the cellular DNA polymerases.¹

From a structure-activity relationship study of some 30 5-substituted 2'-deoxyuridine analogues, the following features for the ideal 5-substitutent were proposed:³ the substituent should be unsaturated, in conjugation with the pyrimidine base, not more than four carbon atoms long, it should possess *E* stereochemistry, and include an electronegative hydrophobic function.

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Table I. Yields of Coupling Reactions between the Zinc Chloride Salt of the Heterocycles and Silylated IdUrd"

° See Scheme I for structures.

(E)-5-(2-Bromovinyl)-2'-deoxyuridine (BrVdUrd) is a compound that meets these requirements. In vitro, BrVdUrd is a potent inhibitor of HSV-I [minimum inhibitory concentration (MIC) = $0.02 \mu g/mL$ and VZV (MIC = $0.003 \mu g/mL$), but a rather weak inhibitor of HSV-2 (MIC = $1 \mu g/mL$).⁴

In a previous study,⁵ we have reported the synthesis and antiviral activity of 5-isoxazole-, 5-thiazole-, and 5 pyrrole-substituted 2'-deoxyuridines. From this series of compounds 5-(3-bromoisoxazol-5-yl)-2'-deoxyuridine emerged as a selective anti-HSV-1 and anti-VZV agent. These results prompted us to broaden this structure-activity relationship study to other heterocyclic substituents.

Here we report on the synthesis and antiviral activity of 5-thien-2-yl- and of 5-furan-2-yl-substituted 2'-deoxyuridines, which were synthesized through Pd(0)-catalyzed C-C bond-formation reaction.

Chemistry

Several 5-aryl-substituted 2'-deoxyuridines were synthesized previously by Mertes et al.⁶ with either a Pd(0) coupling reaction between silylated IdUrd and the [ZnCl]⁺ salt of the aromatic or a Pd(0)-catalyzed reaction between 5-(chloromercuri)-2'-deoxyuridine and a halogenated aromatic. Direct introduction of a heteroaryl substituent in the 5-position of 2'-deoxyuridine was described previously by Pichat et al.⁷ Their procedure was based on the coupling of [ZnCl]⁺ salts of the heteroaromatic with silylated IdUrd and a Pd(II) catalyst. However, the yields of both 5-thien-2-yl-2'-deoxyuridine and 5-thien-3-yl-2'-deoxy- σ then Ξ σ τ are σ σ τ ⁷. No biological or spectroscopic data were provided for these compounds. Variable results were obtained when we tried to adapt the previously described condensation reaction⁶ to the synthesis of $2'$ deoxyuridine analogues with a five-membered heterocyclic substituent in the 5-position. Therefore, silylated IdUrd and palladium(O) tetrakis(triphenylphosphine) were added at once to the freshly prepared $[ZnCl]^+$ salt of the heter-

Table II. Coupling Reaction between Stannylated Heterocycles and IdUrd"

^o Catalyst A, Pd(Ph₃P)₄; B, PdCl₂(Ph₃P)₂. ^b tol = CH₃C₆H₄CO. c See Scheme II for structures.

ocycle in THF (Scheme I). As originally suggested by Bell et al.,⁸ the reaction was carried out in the presence of an excess of ZnCl₂ in order to dissociate intermolecular complexes between Zn^{2+} and the heterocycle. Using this strategy (Table I), we obtained fairly good yields of 5 thiazol-2-yl and 5-(N-methylimidazol-2-yl)-2'-deoxyuridine (48% and 64%, respectively). The method, however, failed with isoxazole and 2-(trimethylsilyl)thiazole as heterocycles and only poor yields were obtained with furan and *N-* [[2-(trimethylsilyl)ethoxy]methyl]imidazole. Apart from the fact that the method is not general applicable, it also suffers from the disadvantage that strictly anhydrous conditinos have to be used and that the elimination of excess of $ZnCl₂$ during the workup procedure can be problematic.

Therefore, we attempted the coupling reaction between 5-iodo-2'-deoxyuridine and the trialkylstannyl derivatives of the heterocycles⁹ (Scheme II). The trimethylstannyl derivatives are much more stable than the corresponding [ZnCl]⁺ salts, and they can be easily prepared. 2-(Trimethylstannyl)thiophene and 2-(trimethylstannyl)furan react in good yields with both protected and unprotected IdUrd. When the reaction was carried out in a coordinating solvent (THF, dioxane) a Pd(II) catalyst is needed, and partial reduction to dUrd was noted. When a noncoordinating solvent (toluene) was used, a Pd(O) catalyst

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Scheme IV

Scheme V

is needed, and no side reaction was observed (Table II). A more straightforward synthesis could start from unprotected IdUrd making 5-thien-2-yl-1- $(\beta$ -D-2-deoxyribofuranos-l-yl)uracil in one step. This conversion can be done in 77% isolated yield by reaction of IdUrd with 2- $(trimethylstanny)$ thiophene in the presence of $PdCl₂/$ Ph_3P/Et_3N in dioxane.

Since trimethylstannylated heterocycles are noxious, we prefer to carry out the reaction with the tributylstannyl derivatives. The yields were comparable. 5-Thien-2-yl- $1-\beta$ -D-arabinofuranos-1-yluracil was obtained in the same way starting from 5-iodo-1- $(2,3,5$ -triacetyl- β -D-arabinofuranos-1-yl)uracil¹⁰ (Scheme III). The protective groups were removed with ammonia in methanol.

During the preparation of this paper, Crisp and Macolino reported on the palladium-catalyzed coupling of protected 5-iodo-2'-deoxyuridine with functionalized and nonfunctionalized arylboronic acids and aryltrimethylstannanes.¹¹ However, coupling reactions with heteroaryltrimethylstannanes were not described.

5-Thien-2-yl-2'-deoxycytidine was synthesized from 5 thien-2-yl-2'-deoxyuridine $(6d)$ by a classical procedure¹² (Scheme IV). Both hydroxyl groups were protected by acetyl esters (10); the uracil base was converted to the cytosine base with $P OCl₃/triazole$ and aqueous ammonia; the deprotection was carried out with ammonia in methanol.

As a recent report¹³ mentioned that α -nucleosides may also show antiviral activity, we also synthesized the *a*analogues of 5-thien-2-yl-2'-deoxyuridine and 5-furan-2 where the contract of the complete means of the contract of the set that the reactions of silylated uracil bases and a crystalline derivative of α -D-2-deoxyribofuranosyl chloride gave predominantly the α -product when they were carried out in the presence of p-nitrophenol (0.32 equiv) and pyridine

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Scheme VI

(0.28 equiv). The role of p-nitrophenol in this reaction is not very clear. When we tried the reaction without pnitrophenol, using 5-iodouracil as the pyrimidine base, the yields were lower (50-70%) (Scheme V). The use of a higher amount of 5-iodouracil (2.5 equiv) gave higher overall yields but lower stereoselectivity.

The α -anomer of 5-IdUrd was isolated by chromatography on silica gel using a $9/1$ mixture of CH_2Cl_2 and $CH₃CN$. Reaction of 14 with the 2-(tributylstannyl)thiophene in the usual way and deprotection of the hydroxyl functions gave 5-thien-2-yl-1- $(\alpha$ -D-2-deoxyribofuranos-l-yl)uracil. The composition of the anomers was determined by HPLC analysis. Although it is not an absolute rule, a distinction between the α - and the β -isomer could be made by NMR spectroscopy. The H-I' signal of the β -anomers appeared as triplets ($J = \pm 6$ Hz); the H-1' signal of the α -anomers all appeared as double doublets $(\tilde{J} = \pm 8 \text{ Hz} \text{ and } J = \pm 2 \text{ Hz})$. Final proof was provided by comparing the obtained compound with the independently synthesized β -anomer starting from IdUrd.

Halogenation of the furan-2-yl (6a) substituent with Br_2 (1 equiv) in CCl₄ and with N-chlorosuccinimide (2 equiv) in pyridine gave mainly the 5-bromofuran-2-yl and 5 chlorofuran-2-yl derivatives (Scheme VI). Chlorination of the thien-2-yl substituent (10), with 1 equiv of *N*chlorosuccinimide, however, gave a mixture of the monoand the dichlorinated derivatives. The mixture was purified by column chromatography and the acetyl groups were removed with ammonia in methanol giving 5-(5 $chlorothien-2-yl)-1-(\beta-D-2-deoxyribofuranos-1-yl)uracil. In$ contrast, bromination with 1 equiv of $Br₂$ gave only the monobrominated product in high yield. The H NMR spectrum of 5-(5-chlorofuran-2-yl)-1-(2-deoxy- β -D-ribofuranos-1-yl)uracil showed two doublets at δ 6.47 (H-4") and 6.86 $(H-3'')$ $(J = 3.1$ Hz). The magnitude of the coupling constant excludes the formation of a 4-chlorofuran-2-yl substituent.

Antiviral Activity

Substitution at the C-5 position of 2'-deoxyuridine with a thiazol-2-yl group, as in 3a, led to some activity against HSV-I, whereas a N-methylimidazol-2-yl substitution gave a totally inactive product (3b) (Table III).

Introduction of a furan-2-yl or thien-2-yl group at the C-5 position afforded compounds (6c and 6d, respectively) with marked activity against HSV-I. To a lesser extent, these compounds were also inhibitory to VZV, and 6c was also active against vaccinia virus (VV). However, neither 6c nor 6d proved active against HSV-2 or thymidine kinase deficient (TK⁻) mutants of HSV-1 or VZV.

A thiophene ring at the C-5 position of $1-\beta$ -D-arabinofuranos-1-yluracil (9b) and 1- $(\beta$ -D-2-deoxyribofuranos-1yl)cytosine **(12b)** led to compounds that were as active as 6c against TK⁺ strains of HSV-1 and VZV. Again, these

^a Concentration required to reduced virus-induced cytopathogenicity (HSV, VZV, VSV, VV) or plaque formation (CMV) or cell growth by 50%; for morphologic alteration, it corresponded to the minimum concentration required to cause a microscopically detectable alteration of normal cell morphology. *^b* Not determined.

compounds were inactive against TK⁻ mutants of HSV-1 and VZV.

The four molecules containing an additional bromine or chlorine atom at the 5-position of the thiophene **(16b** and **16d)** or furan moiety **(16a** and **16c)** showed a clear increase in antiviral activity (i.e. 10-fold for HSV-I and at least 600-fold for VZV) when compounds **16b** and 16d were compared to the parent molecule compound **6d.** Yet, this halogenation of the thiophene or furan ring did not make the compounds **16a-d** active against either HSV-2 or TK- mutants of HSV-I or VZV.

Our data indicate that introduction of an halogen (bromine or chlorine) in the heterocycle substituted at C-5 of the pyrimidine ring of $1-(\beta-D-2-deoxyribofuranos-1$ yl)uracil imparts strong activity against HSV-I and VZV. The 5-bromothien-2-yl and 5-chlorothien-2-yl derivatives **(16b,d)** conferred a similar potency against HSV-I (MIC $\approx 0.02 \ \mu g/mL$) as BrVdUrd, which is one of the most potent in vitro anti-HSV-1 agents that has ever been described.^{1,2}

The α -anomeric forms of the furan-2-yl (15c) and the thien-2-yl **(15d)** derivatives were also evaluated for antiviral activity. They exhibited little, if any, activity against any of the viruses examined. Likewise, 5-iodo-1- $(\alpha$ -D-2deoxyribofuranos-l-yl)uracil failed to show any antiviral activity.

Except for 5-thien-2-yl-1- $(\beta$ -D-2-deoxyribofuranos-1yl)cytosine (12b), which showed slight activity against CMV (IC₅₀ = 50 μ g/mL), none of the test compounds proved active against CMV. None of the compounds showed activity against VSV, caused a microscopically detectable alteration of cell morphology, or inhibited cell growth at the highest concentrations tested. None of the compounds showed any activity against human immunodeficiency virus type 1 (HIV-I) or type 2 (HIV-2) (data not shown). This is in contradiction with the abstract of a recent patent by Johansson et al.,¹³ who claim strong activity (0.05-10 μ M) for the α -anomer of 5-thien-2-yl-2'-deoxyuridine **(16d).**

The mechanism of antiviral action of 5-(5-bromothien-2-yl)-2'-deoxyuridine and 5-(5-chlorothien-2-yl)-2'-deoxyuridine has not been assessed in the present study. According to their spectrum of antiviral activity (and in analogy with the activity spectrum of the (E) -5-(2-halovinyl)-2'-deoxyuridines (i.e. BrVdUrd), for which the mode of action has been established, $15-18$ it can be postulated that 5-(5-bromothien-2-yl)-2'-deoxyuridine and 5-(5-chlorothien-2-yl)-2'-deoxyuridine are preferentially phosphorylated by the HSV-I- and VZV-encoded thymidine kinases and, once converted to their 5'-triphosphate form, they interact with viral DNA synthesis. Whether the 5-(5 halothien-2-yl) derivatives are eventually incorporated into (viral) DNA is an interesting question that should be addressed in further studies.

Experimental Section

Melting points were determined in capillary tubes with a Buchi-Tottoli apparatus and are uncorrected. Ultraviolet spectra were recorded with a Philips PU 8700 UV/vis spectrophotometer. The ¹H NMR and ¹³C NMR spectra were determined with a JEOL FX 90Q spectrometer with tetramethylsilane as internal standard for the ${}^{1}H$ NMR spectra and DMSO- d_6 (39.6 ppm) for the ${}^{13}C$ NMR spectra ($s = singlet, d = doublet, t = triplet, br s = broad$ signal, $m =$ multiplet. Precoated Merck silica gel F254 plates were used for TLC, and the spots were examined with UV light and sulfuric acid-anisaldehyde spray. Column chromatography was performed on Merck silica gel (0.063-0.200 mm). Anhydrous solvents were obtained as follows: tetrahydrofuran and dioxane were refluxed overnight on lithium aluminum hydride and distilled; dichloromethane was stored for 1 week on anhydrous calcium chloride, filtered, and distilled; water was removed from N , N -dimethylformamide by distillation with benzene followed by distillation in vacuo; toluene was refluxed overnight on sodium and distilled. Pyridine was dried by distillation after it had been refluxed for 24 h on potassium hydroxide.

5-Thiazol-2-yl-1- $(\beta$ -D-2-deoxyribofuranos-1-yl)uracil (3a). With the same procedure as described for 3b, thiazole was coupled to the 5-position of IdUrd in 48% yield. The final purification consisted of chromatography on an XAD column (instead of benzoylation, extraction, and deprotection). This product was
crystallized from 2-propanol. Mp: 217–219 °C.⁶ UV (MeOH): $\lambda_{\text{max}} = 261 \text{ nm}$ ($\epsilon = 2850$), $\lambda_{\text{max}} = 320 \text{ nm}$ ($\epsilon = 14100$). Anal. $(\overline{C_{13}}H_{13}N_3O_5S)$: C, H, N.

 $5-(N-Methylimidazol-2-yl)-1-(\beta-D-2-deoxyribofuranos-1-d)$ yl)uracil (3b). A solution of 5 -iodo-1- $(\beta$ -D-2-deoxyribofuranos-l-yl)uracil (280 mg, 0.76 mmol) and hexamethyldisilazane (1.15 g, 7.1 mmol) in 5 mL of anhydrous pyridine was stirred at

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room temperature overnight. The solution was evaporated in vacuo at 30 °C. The residue and tetrakis(triphenylphosphine)palladium(O) (450 mg, 0.39 mmol) were mixed in dry tetrahydrofuran (20 mL), and the suspension was stirred vigorously at room temperature in the dark. A clear yellow solution was obtained after 20 min. This solution was transferred to another flask containing 10 mmol of the zinc chloride complex of *N*methylimidazole. The latter was prepared by treating a solution of Af-methylimidazole (0.74 mL, 9.5 mmol) in dry tetrahydrofuran (20 mL) at -70 °C with *n*-butyllithium $(6.3 \text{ mL}, 10.1 \text{ mmol of a})$ 1.6 M solution in hexane) followed by a solution of freshly fused zinc chloride (1.33 g, 9.5 mmol) in THF (4 mL). A heavy white precipitate was formed which was warmed slowly to room temperature over 0.5 h. Then the palladium(O) complex of silylated IdUrd was added. After 5 min 2 equiv of $ZnCl₂$ (2.66 g, 20 mmol) in THF (8 mL) was added. This mixture was refluxed for 4 h. Chromatographic purification of the reaction mixture could not remove ZnCl₂ completely. Therefore, the product-containing fractions were evaporated and treated with an excess of benzoyl chloride (1.5 mL, 1.3 mmol) in dry pyridine (30 mL) at room temperature for 6 h. This mixture was concentrated, diluted with 100 mL of CHCl₃, and extracted four times with 100 mL of water to remove the excess of zinc salts. The organic layer was dried, evaporated, and taken up in 30 mL of methanol saturated with ammonia. Finally a second chromatographic purification yielded 150 mg (64%) of 5-(N-methylimidazol-2-yl)-1-(β -D-2-deoxyribofuranos-l-yl)uracil. The product was crystallized from MeOH/Et_sO. Mp: 209-211 °C. UV (MeOH): $\lambda_{\text{max}} = 228 \text{ nm}$ $\alpha_{\text{max}} = 10400$. $\lambda_{\text{max}} = 285$ nm ($\epsilon = 8600$). ¹H NMR (DMSO-d_e): δ 2.23 (m, 2 H, $\overline{H\text{-}2}$), 3.51 (br s, 5 H, CH₃ and H-5'), 3.84 (m, 1 H, H-4'), 4.26 (m, 1 H, H-3'), 5.02 (br s, 1 H, 5'-OH), 5.31 (br s, 1 H, 3'-0H), 6.20 (t, 1 H, H-I'), 6.89 (d, 1 H, H-5"), 7.20 (d, 1 H, H_1 , 3 -OH), 0.20 (t, 1 H, H-1), 0.33 (d, 1 H, H-3), 1.20 (d, 1 H,
H₋₄''), 8.08 (s, 1 H, H-6), 11.39 (s, 1 H, NH) ppm. ¹³C, NMR (DMSO- d_6): δ 33.7 (CH₃), 40.2 (C-2'), 61.3 (C-5'), 70.6 (C-3'), 85.0 (C-I'), 87.8 (C-4'), 105.8 (C-5), 123.1 (C-5"), 127.5 (C-4"), 141.9 (C-2"), 142.3 (C-6), 150.2 (C-2), 161.7 (C-4) ppm. Anal. (C-2⁻), 142.5 (C-0), 150.2 (C-
(C.,**H, N.O..¹/ H,O)**: C **H** N

5-Furan-2-yI-l-(0-D-2-deoxyribofuranos-l-yl)uracil (6c). Method A. To a solution of 5-iodo-1-(3,5-di-O-toluoyl- β -D-2deoxyribofuranos-1-yl)uracil (4a) (150 mg, 0.26 mmol) in 30 mL of THF was added 2-(trimethylstannyl)furan (160 mg, 2 mmol) and a catalytic amount of dichlorobis(triphenylphosphine)palladium (20 mg). The mixture was refluxed for 2 h, cooled, and evaporated to give 103 mg (73%) of 5-furan-2-yl-l-(3,5-di-0 toluyl-/3-D-2-deoxyribofuranos-l-yl)uracil as an oil.

Method B. A solution of 5 -iodo-1-(3,5-di-O-toluoyl- β -D-2deoxyribofuranos-l-yl)uracil (150 mg, 0.26 mmol), 2-(trimethylstannyl)furan (460 mg, 2 mmol), and tetrakis(triphenylphosphine)palladium(O) (20 mg) in toluene (30 mL) was refluxed for 7 h. The volatiles were removed by evaporation and the mixture was purified chromatographically (CHCl₃/MeOH, 99/1). Yield: 130 mg (95%).

The combined fraction of A and B (232 mg, 0.44 mmol) were treated for 2 h at room temperature with 20 mL of a 0.1 M sodium methoxide solution in MeOH. The solution was neutralized with acetic acid and evaporated to dryness. Chromatographic purification yielded 110 mg (85%) of pure 6c as a foam.

Method C. To a suspension of IdUrd (270 mg, 0.76 mmol) in dioxane (20 mL) was added 2-(trimethylstannyl)furan (1 g, 3.4 mmol) and dichlorobis(triphenylphosphine)palladium (20 mg) catalyst. The mixture was heated for 1 h at 90 °C, cooled, and evaporated. Column chromatography $[(1)$ CHCl₃, (2) CHCl₃/ MeOH, 92/8] yielded 230 mg (89%) of pure 6c as a foam.

Method D. By starting with 280 mg (0.76 mmol) and following the same method as that described for 3b, only 23 mg (10% yield) of 6c was obtained. The product was crystallized from acetone. Mp: 206-208 °C. UV (MeOH): $\lambda_{max} = 249$ nm (ε = 13850), λ_{max} $= 316$ nm ($\epsilon = 12000$). ¹H NMR (DMSO-d₆): δ 2.18 (dd, 2 H, H-2'), 3.61 (t, 2 H, H-5'), 3.85 (m, 1 H, H-4'), 4.29 (m, 1 H, H-3'), 5.06 (t, 1 H, 5'-OH), 5.25 (d, 1 H, 3'-OH), 6.22 (t, 1 H, H-I'), 6.52 (dd, 1 H, H-4"), 6.96 (dd, 1 H, H-3"), 7.61 (dd, 1 H, H-5"), 8.33 $(s, 1 H, H-6), 11.60 (s, 1 H, NH)$ ppm. ¹³C NMR (DMSO- d_6): δ 40.2 (C-2'), 61.3 (C-5'), 70.6 (C-3'), 85.1 (C-I'), 87.8 (C-4'), 105.9 (C-5), 108.2 (C-3"), 111.8 (C-4"), 135.0 (C-6), 141.8 (C-5"), 146.6 $(C-2'')$, 149.6 $(C-2)$, 160.3 $(C-4)$ ppm. Anal. $(C_{13}H_{14}N_2O_6)$: C, H, N.

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5-Thien-2-yl-1-(β-D-2-deoxyribofuranos-1-yl)uracil (6d). **Method A.** To a solution of 5-iodo-1- $(3,5$ -di-O-toluyl- β -D-2deoxyribofuranos-l-yl)uracil (300 mg, 0.52 mmol) in THF (20 mL) was added 2-(trimethylstannyl)thiophene (440 mg, 1.79 mmol) and dichlorobis(triphenylphosphine)palladium(II) (20 mg, 0.04 mmol). The mixture was refluxed overnight, cooled, and evaporated. The product was purified by column chromatography (241 mg, 87%) and then treated with a 0.1 M solution of sodium methoxide in methanol (20 mL) for 3 h at room temperature. The solution was neutralized with HOAc and evaporated. The resulting oil was purified by chromatography on a silica gel column $[(1)$ CHCl₃/MeOH, 95/5, (2) CHCl₃/MeOH, 93/7]; 130 mg (93%) of 6d was obtained as a foam.

Method B. A mixture of IdUrd (270 mg, 0.76 mmol), 2-(trimethylstannyl)thiophene (1 g, 4.1 mmol) and dichlorobis(triphenylphosphine)palladium(II) in dioxane (20 mL) was heated at 95 °C for 4 h. The solution was cooled and evaporated and the mixture was purified chromatographically $[(1)$ CHCl₃/MeOH, 95/5, (2) CHCl3/MeOH, 93/7]; 106 mg (45%) of 6d was obtained as a foam.

Method C. 5-Iodo-1-(β -D-2-deoxyribofuranos-1-yl)uracil (470 mg, 1.33 mmol) and 2-(trimethylstannyl)thiophene (1 g, 4.1 mmol) were added to a hot solution of palladium acetate (15 mg, 0.07 mmol), triphenylphosphine (35 mg, 0.13 mmol), and triethylamine (0.5 mL, 3.5 mmol) in 25 mL of dioxane, which had been refluxed for 10-15 min. The reaction mixture was then stirred further for 4 h at 85 ⁰C, a red solution was formed, and a palladium mirror on the walls of the reaction flask appeared at the end of the reaction. The reaction flask was cooled and evaporated and 320 mg (77%) of 6d was obtained after chromatographic purification $(CHCl₃/MeOH, 93/7)$. The product was crystallized from acetonitrile. Mp: 206-208 °C. UV (MeOH): $\lambda_{\text{max}} = 262 \text{ nm}$ ($\epsilon =$ 10300), $\lambda_{\text{max}} = 319 \text{ nm}$ ($\epsilon = 10000$); ¹H NMR (DMSO-d₆): δ 2.24 $(t, 2 H, H-2)$, 3.65 (br s, 2 H, H-5'), 3.86 (m, 1 H, H-4'), 4.39 (m, 1 H, H-3'), 5.23 (br s, 2 H, 3'-OH and 5'-OH), 6.24 (t, 1 H, H-I'), 7.09 (d, 1 H, H-4"), 7.42 (m, 2 H, H-3" and H-5"), 8.58 (s, 1 H, $H=6$), 11.60 (s, 1 H, NH) ppm. ¹³C NMR (DMSO-de): δ 40.5 (C-2'), 61.0 (C-5'), 70.1 (C-3'), 85.0 (C-I'), 87.7 (C-4'), 108.4 (C-5), 122.7, 125.6 and 126.5 (C-3", C-4", C-5"), 134.1 (C-2"), 135.7 (C-6), 149.4 (C-2), 161.4 (C-4) ppm. Anal. $(C_{13}H_{14}N_2O_5S)$: C, H, N.

 5 -Thien-2-yl-1- β -D-arabinofuranos-1-yluracil (9b). mixture of 590 mg (1.19 mmol) of $1-(2,3,5\text{-}triacetyl-\beta-D-arabino-)$ furanos-l-yl)-5-iodouracil, 500 mg **(1.34** mmol) of 2-(tributylstannyl)thiophene and 30 mg (0.026 mmol) of palladium tetrakis(triphenylphosphine) in 30 mL of toluene was refluxed overnight. After 16 h TLC evaluation (CHCl₃/MeOH, 97/3) showed that only 60% of the starting material had been transformed into a more liphophilic and fluorescent product. Another 500 mg of 2-(tributybtannyl)thiophene and 20 mg of palladium catalyst were added, and the mixture was refluxed for six more hours. After cooling, the black palladium precipitate was filtered off, and the filtrate was evaporated to dryness. The protective groups were removed by treatment of this oil with 20 mL of MeOH saturated with ammonia. After evaporation of the volatiles the mixture was separated by column chromatography; 130 mg (33% yield) of 5 -thienyl-2-yl-1- β -D-arabinofuranos-1-yluracil was isolated. The product was taken up in acetone and precipitated by addition of heptane. Mp: 225-228 °C. UV (MeOH): $\lambda_{\text{max}} = 262 \text{ nm}$ (ϵ $= 10200$), $\lambda_{\text{max}} = 317$ nm ($\epsilon = 9900$). ¹H NMR (DMSO-d₆): δ 3.28 (br s, 2 H, H-5'), 3.72 (br s, 2 H, H-3' and H-4'), 4.08 (m, 1 H, H-2'), 5.29, 5.49, and 5.65 (2 \times d and 1 \times t, D₂O exchangeable, 2'-, 3'- and 5'-OH), 6.10 (d, 1 H, H-I'), 7.08 (m, 1 H, H-4"), 7.39 (m, 2 H, H-3" and H-5"), 8.26 (s, 1 H, H-6), 11.65 (s, 1 H, NH) ppm. ¹³C NMR (DMSO-d_e): δ 60.3 (C-5'), 74.7 (C-3'), 75.5 (C-2'). 84.3 (C-4'), 85.2 (C-I'), 107.2 (C-5), 122.8,125.5, and 126.7 (C-3", C-4", and C-5"), 134.4 (C-2"), 137.5 (C-6), 149.5 (C-2), 161.4 (C-4) ppm. Anal. $(C_{13}H_{14}N_2O_6S)$: C, H, N.

5-Thien-2-yl-1- $(\beta$ -D-2-deoxyribofuranos-1-yl)cytosine (12b). 310 mg (1 mmol) of 5-thien-2-yl-l-(3-D-2-deoxyribofuranos-lyl)uracil was taken up in pyridine (10 mL) and acetic anhydride (2 mL, 21 mmol) was added. The mixture was evaporated and coevaporated twice with toluene (10 mL); TLC evaluation $(CHCl₃/MeOH, 90/10)$ revealed that all of the 6d had been transformed in a more lipophilic product (10). The solution was evaporated to a foam. This foam was dried by coevaporation with pyridine $(2 \times 10 \text{ mL})$ and taken up in pyridine (50 mL) . POCl₃

(0.27 mL, 3 mmol) and crystalline triazole (828 mg, 12 mmol) were added. The solution was stirred at room temperature for 3 h; cooled in an ice bath, and then ammonia $(5 \text{ mL}, 33\% \text{ in H}_2\text{O})$ was added dropwise. After 10 min the mixture was evaporated and purified by flask chromatography $[(1)$ CHCl₃/MeOH, 99/1, (2) $CHCl₃/MeOH$, 95/5)]. The protective groups were removed by treatment with methanol saturated with ammonia. After evaporation the resulting oil was taken up in methanol and crystallized. Yield: 180mg,60%. Mp: 194 ⁰C dec. UV(MeOH): $\lambda_{\text{max}} = 242 \text{ nm}$ ($\epsilon = 14400$), $\lambda_{\text{max}} = 287 \text{ nm}$ ($\epsilon = 8150$). ¹H NMR (DMSO-d6): *S* 2.10 (t, 2 H, H-2'), 3.65 (br s, 2 H, H-5'), 3.79 (m, 1 H, H-4'), 4.20 (m, 1 H, H-3'), 4.96 (t, 1 H, 5'-OH), 5.18 (d, 1 H, $3'$ -OH), 6.16 (t, 1 H, H-1'), 6.90 (br s, 2 H, NH₂), 7.10 (m, 2 H) and 7.55 (t, 1 H) (H-3", H-4", H-5"), 8.00 (s, 1 H, H-6) ppm. ¹³C NMR (DMSO-d_a): δ 41.0 (C-2'), 61.2 (C-5'), 70.3 (C-3'), 85.8 (C-1'), 87.6 (C-4'), 101.1 (C-5), 126.6,127.5 and 128.4 (C-3", C-4", and C-5"), 134.6 (C-2"), 141.1 (C-6), 154.7 (C-2), and 163.4 (C-4) ppm. Anal. $(C_{13}H_{15}O_4N_3S^{1/2}H_2O)$ C, H, N.

5-Iodo-l-(3,5-di-0-toluoyl-a-D-2-deoxyribofuranos-l-yl) uracil (14a). To a suspension of 5-iodouracil (1.19 g, 5.12 mmol) in 15 mL of dichloroethane was added 4 mL of hexamethyldisilazane and a catalytical amount of ammonium sulfate. The mixture was refluxed for 16 h, cooled, evaporated, and coevaporated three times with 20 mL of m-xylene.

To this oil was added 400 mg (1.024 mmol) of 2,5-di-Otoluyl-a-D-2-deoxyribofuranos-l-yl chloride and 15 mL of dry $CH₂Cl₂$. Finally 0.1 mmol (8 mg) of pyridine was added and the mixture was stirred at room temperature. After 1 h 30 mL of a saturated aqueous sodium bicarbonate solution and 30 mL of dichloromethane were added. The organic phase was dried on Na2SO4, filtrated, and evaporated. Flash chromatography (CHCl₃/MeOH, 95/5) yielded 700 mg (91% yield) of an α/β mixture (1/1).

The pure α -anomer was obtained by a second chromatography on silica gel using a $9/1$ mixture of CH_2Cl_2 and CH_3CN . The structure of the compound was proven after deprotection. UV (MeOH): $\lambda_{\text{max}} = 246$ and 283 nm.

5-Iodo-1- $(\alpha$ -D-2-deoxyribofuranos-1-yl)uracil (14b). The protective groups were removed by treatment of 364 mg (0.61 mmol) of 14a with 15 mL of 0.1 N NaOMe in MeOH for 2 h at room temperature. Chromatographic purification yielded 130 mg (60% yield) of pure 5-iodo-1- $(\alpha$ -D-2-deoxyribofuranos-1-yl)uracil, which was crystallized from acetone. Mp: 204-205 °C. UV (MeOH): $\lambda_{\text{max}} = 287 \text{ nm}$ ($\epsilon = 8200$). ¹H NMR (DMSO-d_e): *6* 1.99 $(m, 1 H, H-2), 2.70$ $(m, 1 H, H-2), 3.52$ (br s, 2H, H-5'), 4.19 $(m,$ 2 H, H-3' and H-4'), 4.83 (br s, 1H, 5'-OH), 5.41 (br s, 1H, 3'-OH), 6.09 (dd, *IH, J =* 7.9 and 1.8 Hz, H-I'), 8.29 (s, 1 H, H-6), 11.70 (s, 1 H, NH) ppm. ¹³C NMR (DMSO-dg): *S* 40.9 (C-2'), 62.0 (C-5'), 68.2 (C-5), 70.8 (C-3'), 86.8 (C-I'), 90.0 (C-4'), 146.4 (C-6), 150.6 (C-2), 161.0 (C-4) ppm. Anal. $(C_9H_{11}N_2O_5I)$: C, H, N.

5-Furan-2-yl-l-(a-D-2-deoxyribofuranos-l-yl)uracil (15c). Following the same method as for 6c but with tributyl instead of (trimethylstannyl)furan 15c was obtained in 45% overall yield. The product was crystallized from ethyl acetate. Mp: 193-195 °C. UV (MeOH): $\lambda_{\text{max}} = 251 \text{ nm}$ ($\epsilon = 12000$), $\lambda_{\text{max}} = 315 \text{ nm}$ (ϵ $= 11\,200$). ¹H NMR (DMSO- d_6): δ 1.99 (m, 1 H, H-2'), 2.73 (m, 1 H, H-2'), 3.45 (d, 2 H, H-5'), 4.24 (m, 2 H, H-3' and H-4'), 5.35 (br s, 2 H, 3' and 5'-OH), 6.19 (dd, 1 H, *J =* 5.8 and 1.8 Hz, H-I'), 6.52 (dd, 1 H, *J* = 3.1 and 0.9 Hz, H-4"), 6.86 (d, 1 H, *J =* 3.1 Hz, H-3"), 7.63 (d, *IH1J =* 0.9 Hz, H-5"), 8.34 (s, 1 H, H-6), 11.32 (s, 1 H, NH) ppm. ¹³C NMR (DMSO-d6): *&* 40.7 (C-2'), 61.9 (C-5'), 70.8 (C-30,86.7 (C-I'), 90.1 (C-4'), 105.2 (C-5), 108.0 (C-3"), 111.8 (C-4"), 135.8 (C-5"), 141.6 (C-6), 146.8 (C-2"), 149.6 (C-2), 160.4 (C-4) ppm. Anal. $(C_{13}H_{14}N_2O_6)$ C, H, N.

5-Thien-2-yl-1- $(\alpha$ -D-2-deoxyribofuranos-1-yl)uracil (15d). Starting from 400 mg (0.66 mmol) of 14a and following the same method as for 9a, 203 mg (55% yield) of 16b was obtained. Deprotection with 0.1 M sodium methoxide gave 16d in 70% yield. The product was crystallized from acetone. Mp: 197-199 °C. UV (MeOH): $\lambda_{\text{max}} = 262 \text{ nm}$ ($\epsilon = 12000$), $\lambda_{\text{max}} = 317 \text{ nm}$ ($\epsilon = 12000$) 11700); ¹H NMR (DMSO-d_e): δ 2.08 (m, 1 H, H-2'), 2.70 (m, 1 H, H-2'), 3.44 (br s, 2 H, H-5'), 4.25 (m, 2 H, H-3' and H-4'), 4.84 (t, 1 H, 5'-OH), 5.47 (d, 1 H, 3'-OH), 6.24 (dd, 1 H, *J =* 7.9 and 1.8 Hz, H-I'), 7.05 (dd, *J =* 4.0 and 4.8 Hz, H-4"), 7.28-7.53 (m, 2 H, H-3' and H-5"), 8.49 (s, 1 H, H-6), 11.60 (s, 1 H, NH) ppm. ¹³C NMR (DMSO-d₆): δ 40.5 (C-2'), 62.1 (C-5'), 71.1 (C-3'), 86.7

(C-I'), 90.2 (C-4'), 108.3 (C-5), 123.3,125.9, and 127.0 (C-3", C-4", and C-5"), 134.5 (C-2"), 137.4 (C-6), 149.9 (C-2), 161.7 (C-4) ppm. Anal. $(C_{13}H_{14}N_2O_5S)$: C, H, N.

 $5-(5-Bromofuran-2-yl)-1-(\beta-D-2-deoxyribofuranos-1-yl)$ uracil (16a). To a solution of 5-furan-2-yl-l-(3,5-di-0-toluoyl- β -D-2-deoxyribofuranos-1-yl)uracil (250 mg, 0.48 mmol) in 10 mL of CH₂Cl₂ was added dropwise at room temperature a dilution of Br_2 (0.49 mmol) in 5 mL of CCl₄. After 30 min the green solution was extracted with water $(2 \times 20 \text{ mL})$, dried, and evaporated. The resulting oil was treated with 20 mL of sodium methoxide 0.1 M for 2 h at room temperature. Then the solution was neutralized with HOAc and evaporated. Chromatographic purification yielded 120 mg of a white product that was crystallized from EtOAc. A first crystalline fraction contained 57 mg of pure 16a. The filtrate contained a 6/4 mixture (NMR) of mono and dibrominated product. Mp: 109 °C. UV (MeOH): $\lambda_{\text{max}} = 252$ nm ($\epsilon = 12950$), $\lambda_{max} = 319$ nm ($\epsilon = 11800$). ¹H NMR (DMSO-d₈): δ 2.18 (t, 2 H, H-2⁷), 3.61 (br s, 2 H, H-5⁷), 3.84 (m, 1 H, H-4⁷), 4.29 (m, 1 H, H-3'), 5.05 (t, 1 H, 5'-OH), 5.25 (d, 1 H, 3'-OH), 6.19 (t, 1 H, H-1'), 6.83 and 6.60 (2 \times d, 2 \times 1 H, $J = 3.3$ Hz, H-3" (c, 1 11, 11-1), 6.35 and 6.30 (2 \sim d, 2 \sim 1 H, 0 $-$ 3.3 112, 11-3
and H-4"), 8.28 (s, 1 H, H-6), 11.61 (s, 1 H, NH) ppm. ¹³C NMR (DMSO-dg): *S* 40.2 (C-2'), 61.1 (C-5'), 70.2 (C-3'), 85.0 (C-I'), 87.7 (C-4'), 104.6 (C-5), 110.4 and 113.4 (C-3" and C-4"), 120.1 (C-5"), 134.2 (C-6), 148.9 and 149.2 (C-2" and C-2), 159.8 (C-4) ppm. Anal. $(C_{13}H_{13}BrN_2O_6^{-1}/_2H_2O)$: C, H, N.

 $5-(5-Bromothien-2-yl)-1-(\beta-D-2-deoxyribofuranos-1-yl)$ uracil (16b). 5-Thien-2-yl-1- $(\beta$ -D-2-deoxyribofuranos-1-yl)uracil (200 mg, 0.66 mmol) was dissolved in 10 mL of pyridine; 2 mL of acetic anhydride was added. The mixture was evaporated to dryness and coevaporated twice with toluene. TLC analysis $(CHCl₃/MeOH, 90/10)$ showed that the transformation to 5thien-2-yl-l-(3,5-di-0-acetyl-|3-D-2-deoxyribofuranos-l-yl)uracil (10) was complete. The oil was taken up in 25 mL of CHCl3. At 0 °C a dilution of 0.66 mmol Br₂ in 5 mL of CCl₄ was added dropwise (15 min). The red of bromine disappeared immediately, and TLC evaluation revealed that all of the 10 had been transformed into a more lipophilic product. The solution was extracted with water $(2 \times 25 \text{ mL})$, dried, and evaporated to an oil. This oil was taken up in 20 mL of MeOH saturated with ammonia and set aside for 16 h. Chromatographic purification yielded 210 mg (83%) of 5-(5-bromothien-2-yl)-1- $(\beta$ -D-2-deoxyribofuranos-1yl)uracil. The product was precipitated as an amorphous solid from acetone-heptane. UV (MeOH): $\lambda_{max} = 273$ nm ($\epsilon = 8500$), $\lambda_{\text{max}} = 327 \text{ nm}$ ($\epsilon = 12100$); ¹H NMR (DMSO-d₆): δ 2.23 (t, 2 H, H-2'), 3.68 (br s, 2 H, H-5'), 3.84 (m, 1H, H-4'), 4.33 (m, 1H, H-3'), 5.29 (br s, 2 H, D_2O exchangeable, 3' and 5'-OH), 6.20 (t, 1 H, H-I'), 7.20 (m, 2 H, H-3" and H-4"), 8.64 (s, 1 H, H-6), 11.77 (s, 1 H, NH) ppm. ¹³C NMR (DMSO-d6): *6* 40.5 (C-2'), 60.7 (C-5'), 69.8 (C-3'), 85.0 (C-I'), 87.7 (C-4'), 107.5 (C-5'), 111.1 (C-5"), 122.1 and 129.2 (C-3" and C-4"), 135.7 (C-6 and C-2"), 149.1 (C-2), 161.3 (C-4) ppm. Anal. $(C_{13}H_{13}BrN_2O_5S)$ C, H, N.

 $5-(5-Chlorofuran-2-yl)-1-(\beta-D-2-deoxyribofuranos-1-yl)$ uracil (16c). A mixture of 5-furan-2-yl-1- $(3,5$ -di-O-toluoyl- β -D-2-deoxyribofuranos-l-yl)uracil (250 mg, 0.48 mmol) N-chlorosuccimide (NCS) (65 mg, 0.48 mmol) in 10 mL of pyridine was heated for 1 h at 90 °C. TLC evaluation $(CH_2Cl_2/CH_3CN, 9/1)$ revealed a 50% conversion of the starting material into a more lipophilic product. Another 65 mg of NCS was added; after 2 h all of the 6b was transformed. The solution was cooled, evaporated, and taken up in 100 mL of CHCl₃. The solution was extracted with a saturated sodium bicarbonate solution (2×100) mL). The organic layer was dried and evaporated. The protective groups were removed by treatment at room temperature for 2 h with 20 mL of 0.1 M sodium methoxide in methanol. Column chromatographic purification yielded 130 mg (85%) of a yellow chromatographic purification yielded 130 mg (85%) of a yellow
foam which was crystallized from EtOAc. Mp: 175 °C. UV (Main Whith was crystanized from Brock. Mp. 179 C.
(MeOH): $\lambda_{\text{max}} = 252 \text{ nm}$ ($\epsilon = 12400$), $\lambda_{\text{max}} = 318 \text{ nm}$ (e = 11500);
¹H NMR (DMSO-d.): *6* 2.18 (t, 2 H, H-2'), 3.61 (br s, 2 H, H-5'). 3.84 (m, 1 H, H₋₄'), 4.98 (m, 1 H, H-3'), 5.04 (t, 1 H, 5'-OH), 5.95 o.o4 (iii, 1 H, H¹4), 4.20 (iii, 1 H, H¹0), 0.04 (t, 1 H, 0 °OH), 0.20
(d, 1 H, 3'-OH), 6.18 (t, 1 H, H₋9'), 6.47 and 6.86 (9 x d, 9 x 1 *H, J =* 3.1 Hz, H-3" and H-4"), 8.27 Is, 1 H, H-6), 11.65 (s, 1 H, H, J = 3.1 Hz, H-3'' and H-4''), 8.2'' (s, 1 H, H-6), 11.65 (s, 1 H,
NH) ppm^{. 13}C NMR (DMSO-d-); *&* 40.6 (C-2'), 61.2 (C-5'), 70.4 (C-3'), 85.2 (C-I'), 87.8 (C-4'), 104.8 (C-5), 108.7 and 110.4 (C-3" and C-4"), 134.0 (C-5"), 135.5 (C-6), 14(5.9 (C-2"), 149.4 (C-2), 160.0 and C-4"), 154.0 (C-5"), 155.5 (C-6), 146.5 (C-2"), 145.4 (
(C-4) ppm. Anal. (C. H. N.O.Cl.¹/.H.O) C, H, N.

5-(5-Chlorothien-2-yl)-1-(β -D-2-deoxyribofuranos-1-yl)uracil (16d). 5-Thien-2-yl-1-(2- β -D-deoxyribofuranos-1-yl)uracil (385 mg) was acetylated as described for 16b. This foam was dissolved in 10 mL of pyridine; 160 mg (1.18 mmol) of NCS was added and the mixture was heated for 4 h at 70 °C. Chromatographic purification $(CH_2Cl_2/CH_3CN, 87/13)$ yielded 190 mg of a foam. The acetyl groups were removed by treatment with 20 mL of methanol saturated with ammonia. After evaporation and purification 120 mg (28% yield) of 13b was obtained. The product was crystallized from acetone. Mp: 220-222 °C. UV (MeOH): $\lambda_{\text{max}} = 274 \text{ nm}$ (ε = 8800), $\lambda_{\text{max}} = 327 \text{ nm}$ (ε = 11500). ¹H NMR (DMSO-d₆): δ 2.23 (t, 2 H, H-2'), 3.68 (br s, 2 H, H-5'), 3.87 (m, 1 H, H-4'), 4.31 (m, 1 H, H-3'), 5.29 (br s, 2 H, 3'-OH and 5'-OH), 6.19 (t, 1 H, H-I'), 7.03 and 7.23 [2 x (d, 1 H, *J* = 3.95 Hz, H-3" and H-4")], 8.64 (s, 1 H, H-6), 11.76 (s, 1 H, NH) ppm. ¹³C NMR (DMSO-dg): *S* 40.5 (C-2'), 60.7 (C-5'), 69.8 (C-30, 85.1 (C-1'), 87.7 (C-4'), 107.5 (C-5), 121.1 and 125.7 (C-3" and C-5"), 127.4 (C-5"), 133.0 (C-2"), 135.7 (C-6), 149.1 (C-2), 161.3 (C-4) ppm. Anal. $(C_{13}H_{13}CIN_2O_5S)$: C, H, N.

Antiviral Activity. The different molecules were evaluated for their antiviral in vitro activity according to well-established procedures.^{17,18} The origin of the viruses [herpes simplex type 1 (HSV-I) (strains KOS, F and Mclntyre), thymidine kinase deficient (TK") HSV-1 strains (B2006 and VMW 1837), herpes simplex virus type 2 (HSV-2) (strains G, 196 and Lyons), varicella-zoster virus (VZV) (strains Oka and YS), TK" VZV (strains 07-1 and YS-R), vaccinia virus (W), vesicular stomatitis virus (VSV), and cytomegalovirus (CMV) (strains AD169 and Davis)] has been described previously.^{18,19} Cytotoxicity measurements

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Registry No. 3a, 133040-36-5; 3b, 134333-64-5; 4a, 31356-86-2; 4b, 54-42-2; 5 (X = O), 51583-40-5; 5 (X = S), 37496-13-2; 6c, 92233-50-6; 6d, 89647-10-9; 7, 84500-33-4; 8, 54663-78-4; 8 (X = O), 118486-94-5; 9b, 134333-65-6; 10,133787-21-0; 11,134333-66-7; 12b, 134333-67-8; 13, 4330-21-6; α -14a, 104267-95-0; β -14a, 31356-86-2; 14b, 60110-67-0; α -15a, 127235-39-6; β -15a, 127235-40-9; α -15b, 134333-68-9; α -15c, 127235-62-5; α -15d, 127235-63-6; /3-15d diacetyl derivative, 133787-21-0; **16a,** 134333-69-0; 16b, 134333-70-3; 16c, 134333-71-4; 16d, 134333-72-5; N-methylimidazole, 616-47-7; 5-iodouracil, 696-07-1; 1,2,4-triazole, 288-88-0; thiazole, 288-47-1; 5-iodo-bis-0-(trimethylsilyl)uracil, 38953-72-9.

Synthesis and Binding of [¹²⁵I2]Philanthotoxin-343, [¹²⁵I2]Philanthotoxin-343-lysine, and [¹²⁵I2]Philanthotoxin-343-arginine to Rat Brain Membranes

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 $125I_2$ -iodinated philanthotoxin-343 (PhTX-343) (10), $[125I_2]$ PhTX-343-arginine (11), and $[125I_2]$ PhTX-343-lysine (12) were synthesized and evaluated as probes for glutamate receptors in rat brain synaptic membranes. It was found that these probes were not specific for the glutamate receptors but may be useful for investigating the polyamine binding site. Filtration assays with Whatman GF/B fiber glass filters were unsuitable because the iodinated PhTX-343 analogues exhibited high nonspecific binding to the filters, thus hindering detection of specific binding to membranes. When binding was measured by a centrifugal assay, $[1^{26}I_2]$ PhTX-343-lysine (12) bound with low affinity ($K_D = 11.4$) \pm 2 μ M) to a large number of sites (37.2 \pm 9.1 nmol/mg of protein). The binding of $[1^{26}I_2]$ PhTX-343-lysine was
sensitive only to the polyamines spermine and spermidine, which displaced $[1^{26}I_2]$ PhTX-343-ly of $(3.77 \pm 1.4) \times 10^{-6}$ M and $(7.51 \pm 0.77) \times 10^{-5}$ M, respectively. The binding was insensitive to glutamate receptor agonists and antagonists. Binding results with $[1^{25}I_2]$ PhTX-343-arginine (11) were similar to PhTX-343-lysine. Considering the high number of toxin binding sites (10000-fold more than glutamate) in these membranes and the insensitivity of the binding to almost all drugs that bind to glutamate receptors, it is evident that most of the binding observed is not to glutamate receptors. On the other hand, PhTX analogues with photoaffinity labels may be useful in the isolation/purification of various glutamate and nicotinic acetylcholine receptors; they could also be useful in structural studies of receptors and their binding sites.

Introduction

Glutamate receptor pharmacology has attracted a great deal of attention in the last few years¹⁻⁴ because of the possible involvement of glutamate receptors in degenerative brain diseases,⁵ mechanism of memory,⁶ and ischemic damage.⁷ Glutamate receptors are classified into three types according to their sensitivity to the exogenous excitatory amino acids quisqualate, kainate, and *N*methyl-D-aspartate (NMDA). A subunit of a kainate type

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