was resuspended in 3 times the original volume of buffer and incubated at 37 °C for 15 min, centrifuged again, and resuspended in the original volume of buffer. A binding assay using [³H]-bremazocine (0.5 nM) was carried out as described above.

Acknowledgment. We greatly appreciate receiving samples of benzomorphan standards 1, its enantiomer, 2, (2'R)- and (2''S)-3, and the enantiomer of (2''S)-3 from Dr. Herbert Merz, Boehringer Ingelheim KG, Germany. We acknowledge the support of this work by the National Institute on Drug Abuse through research grants DA-03933 and DA-06675. **Registry** No. 4, 134133-69-0; 5, 134233-45-7; 6, 134233-46-8; 7, 134233-47-9; 8, 134133-70-3; 8-HCl, 134233-67-3; 9, 134233-48-0; 9-HCl, 134308-16-0; (\pm) -10, 52079-30-8; (5R)-11, 58879-35-9; (5S)-11, 58879-36-0; 12, 134133-71-4; 13, 134233-49-1; 14, 134133-72-5; 15, 134233-50-4; 16 (isomer 1), 134133-73-6; 16 (isomer 2), 134233-51-5; 18 (isomer 1), 134133-74-7; 18 (isomer 2), 134233-52-6; 20, 134133-75-8; 21, 134233-53-7; 22, 134233-54-8; 23, 134233-55-9; 24, 134233-56-0; 25, 134233-57-1; 30, 134233-58-2; 31, 134233-50-3; 32 (isomer 1), 134133-76-9; 32 (isomer 2), 134233-60-6; 33 (isomer 1), 134133-77-0; 33 (isomer 2), 134233-61-7; 33 (isomer 1).HCl, 134233-62-8; 33 (isomer 2).HCl, 134308-15-9; 34 (isomer 1), 134233-63-9; 34 (isomer 2), 134233-64-0; 35 (isomer 1), 134233-65-1; 35 (isomer 2), 134233-66-2.

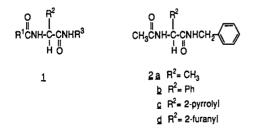
Preparation and Anticonvulsant Activity of a Series of Functionalized α -Heteroatom-Substituted Amino Acids

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Potent anticonvulsant activity has been reported for (R,S)-2-acetamido-N-benzyl-2-methylacetamide (2a). Select α -heteroatom substituted derivatives of 2a have been prepared (26 examples) in which the α -methyl group has been replaced by nitrogen (3a-q), oxygen (3r-u), and sulfur (3v-z) containing moieties. The functionalized amino acid derivatives were evaluated in the maximal electroshock seizure (MES) and horizontal screen (tox) tests in mice. The most active compounds were (R,S)-2-acetamido-N-benzyl-2-(methoxyamino)acetamide (3l), and (R,S)-2-acetamido-N-benzyl-2-(methoxyamino)acetamide (3l), and (R,S)-2-acetamido-N-benzyl-2-(methoxyamino)acetamide (3l), and (R,S)-2-acetamido-N-benzyl-2-(methoxyamino)acetamide (3l). After ip administration, the MES ED₅₀ values for 3l (6.2 mg/kg) and 3n (6.7 mg/kg) compared favorably with phenytoin (9.50 mg/kg).

Nonnaturally occurring amino acids have become increasingly important in the design of pharmacologically active peptides and peptidomimetics.¹ Recently we reported the excellent anticonvulsant activity of certain functionalized amino acid derivatives 1.2-6 Potent protection against maximal electroshock seizures (MES) in mice was observed for functionalized amino acid racemates containing both an N-benzylamide moiety and an acetylated amino group. Systematic variation of the α -substituent revealed that stringent steric and electronic requirements must be met for optimal activity. The median effective dose (ED₅₀) for the α -methyl (2a) (76.5 mg/kg) and α -phenyl (2b) (20.3 mg/kg) derivatives⁴ compared favorably with that observed for phenobarbital⁷ (21.8)mg/kg), while those of the α -pyrrolyl (2c) (16.1 mg/kg) and α -furanyl (2d) (10.3 mg/kg) adducts⁶ rivaled that reported for phenytoin⁷ (9.50 mg/kg). Furthermore, comparison of the two individual enantiomers of 2a,b,d revealed that in each case the anticonvulsant activity resided primarily in the R stereoisomer.^{2,5,6}



In the present study, the synthesis and anticonvulsant properties of a novel series of α -heteroatom-substituted amino acid derivatives (26 examples) are presented. Included in this survey are selected oxygen, nitrogen, and sulfur-functionalized amino acids. Analysis of the composite data set disclosed trends that further define the structure-activity relationships for this class of amino acid derived anticonvulsant agents.

Selection of Compounds

(R,S)-2-Acetamido-N-benzyl-2-methylacetamide³ (2a) represented the parent compound in this study wherein the α -methyl group was replaced by select functionalized nitrogen, oxygen, and sulfur substituents (Table I). In all cases, the racemates were prepared and tested. No attempts were made at this stage to resolve the enantiomeric mixtures. The α -nitrogen-substituted adducts consisted of the parent amino 3a, the monoalkylamino 3b,c, the dialkylamino 3d,e, and the trialkylammonium 3f derivatives, as well as the corresponding monoaryl analogues 3g and 3h. Included in our α -nitrogen subset of compounds were three classes of functionalized amino derivatives. These were the monoacyl derivatives 3i and 3j, the N-hydroxyamino adducts 3k-o, and the Nhydrazino compounds 3p and 3q. The second set of structurally modified amino acid derivatives were the

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- (3) Cortes, S.; Liao, Z.-K.; Watson, D.; Kohn, H. J. Med. Chem. 1985, 28, 601.
- (4) Conley, J. D.; Kohn, H. J. Med. Chem. 1987, 30, 567.
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^{*}Author to whom correspondence should be addressed to at the University of Houston.

[†]University of Houston.

[‡]Lilly Research Laboratories.

For leading references, see: (a) Shuman, R. T.; Ornstein, P. L.; Paschal, J. W.; Gesellchen, P. D. J. Org. Chem. 1990, 55, 738.
 (b) Ojima, I.; Komata, T.; Qiu, X. J. Am. Chem. Soc. 1990, 112, 770 and references therein.

Table I. Selected Physical and Pharmacological Data in Mice for *α*-Heteroatom-Substituted Functionalized Amino Acid Derivatives 3^a X

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no. X mp ⁵ ED ₈₀ 3a NH2 131-133 65.1 [0.5] e 3b NHCH3 115-117 44.5 [0.5] e 3c NHCH2CH3 123-125 42.4 [0.5] e 3d N(CH3)2 104-106 45.3 [1] e 3d N(CH3)2 104-106 45.3 [1] e 3e	3					
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	no.	Х	mp ^b	MES ^c ED ₅₀	tox ^d TD ₅₀	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	3a					
3c NHCH ₂ CH ₃ 123-125 $(37, 2-47, 3)$ e 3d N(CH ₃) ₂ 104-106 45.3 [1] e 3e $-\sqrt{0}$ 171-172 >30, <100 [1]	- •	_		(56.2-75.3)		
3c NHCH ₂ CH ₃ 123-125 42.4 (0.5) e 3d N(CH ₃) ₂ 104-106 45.3 [1] e 3e $-N_{OO}$ 171-172 >30, <100 [1]	3b	NHCH ₃	115-117		е	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	30	NHCH.CH.	193-195		p	
3d N(CH_3)2 104-106 45.3 [1] e 3e $-N_{OO}$ 171-172 >30, <100 [1] e 3f N ⁺ (CH_3)3 BF, ⁻ 171-173 dec >100 e 3g NHPh 183-185 >300 e 3h $-N_{N-N}$ 135-137 ~100 [1] e 3i NHCOCH3 265-267 dec >100, <300 [1] e 3i NHCOCH3 228-230 >300 e 3k NHOH 144-146 dec ~100 [1] e 3h NHOH 144-146 dec ~100 [1] e 3m NICH3OH 159-161 ~30 [1] e 3m N(CH3OCH3 165-167 6.7 (0.5] 50 3m N(CH3OCH3 165-167 6.7 (0.5] e 3a N(CH3OCH3 165-167 6.6 (0.5] e 3a N(CH3OCH3 165-167 6.7 (0.5] e 3a N(CH3OCH3 159-154 556 (0.5] e 3a OH 136-138 80.1 [1] e <		111101120113	120 120		C	
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	3d	$N(CH_3)_2$	104-106		е	
3f N ⁺ (CH _g) ₃ BF ₄ 171-173 dec >100 e 3g NHPh 183-185 >300 e 3h H 135-137 ~100 [1] e 3i NHCOCH ₃ 265-267 dec >100, <300 [1]	3e		171-172	>30, <100 [1]	е	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$						
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		$N^+(CH_3)_3 BF_4^-$				
$\begin{array}{c c c c c c c c c c c c c c c c c c c $						
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	3h		135-137	~100 [1]	е	
3iNHCOCH3265-267 dec> 100, <300 [1]e3jNHCOCF3228-230> 300e3kNHOH144-146 dec~ 100 [1]e3lNHOH144-146 dec~ 100 [1]e3mN(CH3)OH95-97(54-7.2)(383mN(CH3)OCH3165-1676.7 [0.5]50.3o $0 \rightarrow 0$ 149-15131.4 [0.5]e3o $0 \rightarrow 0$ 149-15131.4 [0.5]e3gNHNHPh132-134~ 100 [0.5]e3qNHNHCO2CH2Ph152-15455.6 [0.5]e3rOH136-13880.1 [1]e3t/OCH3145-14698.3 [0.5]> 13t/OCH4155-15750.0(51.1-78.4)3uOPh125-128> 1006e3x'SCH2CH3135-137> 100e3x'SCH2CH3135-137> 100e3x'SCH2CH3135-137> 100e3z/SO(OCH2CH3135-137> 100e3z/SCH2CH3135-137> 100e3x'SCH2CH3135-137> 100e3z/SO(O2CH2CH3135-137> 100e3z/SO(O2CH2CH3135-137> 100e3z/SO(O2CH2CH3135-137> 100e3z/SO(O2CH2CH3135-137> 100e3z/SO(O2CH2CH3135-137> 100e3z/SO(02CH2CH3 <td></td> <td>44</td> <td></td> <td></td> <td></td>		44				
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		NH" V				
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	3i	NHCOCH.	265-267 dec	>100, <300 [1]	е	
31 NH(OCH ₃) 95-97 6.2 [0.5] 46. (5.4-7.2) (38 3m N(CH ₃)OCH ₃ 159-161 ~30 [1] e 3n N(CH ₃)OCH ₃ 165-167 6.7 [0.5] 50. (5.7-7.7) (40 3o \mathcal{O}_{N} 149-151 31.4 [0.5] e e 3p NHNHPh 132-134 ~100 [0.5] e e 3q NHNHCO ₂ CH ₂ Ph 152-154 55.6 [0.5] e 3r OH 136-138 80.1 [1] e 3r' OH 136-138 80.1 [1] e 3t/ OCH ₃ 145-146 98.3 [0.5] >1 3u OPh 125-128 1000' e 3v SCH ₂ CH ₃ 155-157 100 e 3v' SCH ₃ 135-137 >100 e 3v-1 S(O)CH ₂ CH ₃ 135-137 >100 e 3v-1 S(O)CH ₂ CH ₃ 135-137 >100 e 3v-1 S(O)CH ₂ CH ₃ 135-137 >100 e 3y-1 S(O)CH ₂						
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$						
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	31	NH(OCH ₃)	95–97		46.0 [0.5]	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	_				(38.0–56.0)	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$						
300149-15131.4 [0.5] (26.7-37.8)e3pNHNHPh132-134~100 [0.5]e3qNHNHCO2CH2Ph152-15455.6 [0.5]e3rOH136-13880.1 [1]e3e'OCH3145-14698.3 [0.5]>13t'OCH2CH3153-15562.0 [1]>13uOPh125-128>1006e3wSCH3155-157>100e3wSCH2CH3140-142>30, <100 [0.5]	3n	N(CH ₃)OCH ₃	165-167		50.5 [0.5]	
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	-	0			(40.4–59.9)	
3pNHNHPh NHNHCO2CH2Ph132-134 152-154~100 [0.5] 55.6 [0.5] (49.3-63.9)3rOH136-138 (49.3-63.9)80.1 [1] (49.3-63.9)3rOH136-138 (49.3-63.9)80.1 [1] (49.3-63.9)3e'OCH3145-146 (84.4-114.0)98.3 [0.5] 	30	<u>, 7</u>	149-151		е	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				(20.7-37.6)		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$					е	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	3q	NHNHCO ₂ CH ₂ Ph	152 - 154		е	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	9	011	100 100		_	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	3 r	OH	136-138		e	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	3s [/]	OCH.	145-146		>100, <300 [0.5	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		00113	110 110		1 200, 4000 [0.0	
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	3t [/]	OCH ₂ CH ₃	153-155		>112	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				(51.1-78.4)		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		OPh				
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		SCH ₃			е	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$						
3y-mix S(O)CH ₂ CH ₃ 135-137 >100 e 3z S(O ₂)CH ₂ CH ₃ 161-163 >100 e 2a ^h CH ₃ 138-139 76.5 453 2b ^j Ph 202-203 20.3 964 (16.9-24.5) (75 16.1 >3 2c ^k 2-pyrrolyl 174-175 16.1 >3 2d ^k 2-furanyl 178-179 10.3 ~4						
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		$S(O)CH_2CH_3$				
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		$S(O)CH_2CH_3$			е	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		$S(O_2)CH_2CH_3$				
2b ^j Ph 202-203 20.3 96. (16.9-24.5) 2c ^k 2-pyrrolyl 174-175 16.1 >3 (13.2-19.9) 2d ^k 2-furanyl 178-179 10.3 ~4	2a [^]	CH_3	138-139		453.9 ⁱ	
2c ^k 2-pyrrolyl 174-175 (16.9-24.5) (79 2d ^k 2-furanyl 174-175 16.1 >3 10.3 ~4	2b [/]	Ph	909_909		(416.6–501.0) 96.9 [;]	
2c* 2-pyrrolyl 174-175 16.1 >3 2d* 2-furanyl 178-179 10.3 ~4	40	r n	202-203		(79.8–118.4)	
2d^k 2-furanyl 178–179 (13.2–19.9)	2c ^k	2-pyrrolyl	174-175		>30, <100	
				(13.2–19.9)		
(91–116)	2d*	2-furanyl	178-179		~40	
	-haustain!			(9.1-11.6)		
	huenyou.				65.5^{i} (52.5–72.1)	
phenobarbital ⁱ 21.8 69.	phenobarbital ⁱ				(52.5-72.1) 69.0 ⁱ	
	L				(62.8-72.9)	

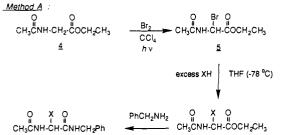
^a The compounds were administered intraperitoneally. ED_{50} and TD_{50} values are in milligrams per kilogram. Number in parentheses are 95% confidence intervals. Time of peak effects in hours as determined in the Experimental Section is denoted in brackets. ^b Melting points (°C) are uncorrected. ^cMES = maximal electroshock seizure test. Compound was suspended in 30% PEG unless otherwise noted. ^d Tox = neurologic toxicity determined from horizontal screen unless otherwise noted. ^eNot determined. ^fReference 8. ^fCompound 3u was suspended in acacia. ^hReference 3. ^jNeurologic toxicity determined using the rotorod test. ^jReference 4. ^kReference 6. ^fReference 7.

 α -oxygen-substituted compounds 3r-u. This group was comprised of the α -hydroxy adduct 3r, the two α -alkoxy derivatives 3s and 3t,⁸ and the α -phenoxy compound 3u.

A similar battery of α -sulfur-substituted compounds (i.e., $3v-x^8$) was selected for evaluation. Attempts to synthesize the parent α -thiol derivative in the series were unsuccessful, however. In addition to 3v-x, both the sulfoxide 3y and the sulfone 3z derivatives of the ethylthio adduct 3w were prepared.

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Scheme I. Preparation of α -Heteroatom-Substituted Functionalized Amino Acid Derivatives 3



[NaCN] CH₃OH

Method B

3

$$3! \xrightarrow{BBr_3} CH_3CNH-CH-CNHCH_2Ph} \begin{array}{c} excess XH (X M^*) \\ THF(-78 °C) \\ 0 \\ H \\ CH_2CI_2 \end{array} \xrightarrow{Or} CH_3CNH-CH-CNHCH_2Ph \xrightarrow{Or} 3 \\ 0 \\ Z \\ 2 \\ XH \\ THF (-78 °C) \end{array}$$

Method C.

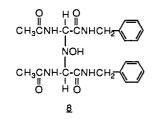
$$\begin{array}{c} (CH_3)_3O^* & \bigcap_{i=1}^{H-4} \\ BF_4 & \bigcap_{i=1}^{H-4} \\ 3d & \begin{array}{c} BF_4 & \\ BF_4 & \\ CH_3NO_2 & O \\ CH_3NO_2 & O \\ 0 \end{array} \end{array} \xrightarrow{} CH_3CNH-CH_2CH_2Pn \\ O \\ 0 \\ 0 \\ 0 \\ 0 \end{array}$$

Method D

Chemistry

Four different synthetic approaches (Scheme I, methods A-D) were employed for the preparation of most of the α -heteroatom-substituted, functionalized amino acid derivatives 3. In the first route (method A), the α -bromo ester 5 was prepared in near quantitative yield using the protocol of Kober and Steglich⁹ from commercially available¹⁰ ethyl acetamidoacetate (4). Treatment of 5 with an excess of the nucleophilic heteroatom species furnished the corresponding ethyl 2-substituted-2-acetamidoacetate 6 (61-92% yields). Formation of 6 is presumed to proceed through the intermediacy of ethyl 2-(acetylimino)acetate. In most cases, 6 was isolated as a thick oil and used directly in the subsequent step without extensive purification. Treatment of 6 with benzylamine and a catalytic amount of NaCN¹¹ gave the desired product in moderate yields (29-74%). This method was used to synthesize compounds **3a-c**, **3e**, **3g**, and **3h**. Attempts to utilize this protocol to prepare 3d, 3k, 3m, and 3n were unsatisfactory. In these cases the final benzylamine-mediated step did not proceed cleanly.

The difficulty encountered in converting several α functionalized esters 6 to the corresponding benzylamides 3 led to the development of the second procedure (method B) depicted in Scheme I. In this route the benzylamide moiety was incorporated within the framework of the amino acid derivative prior to the introduction of the α heteroatom substituent. Treatment of 3t with BBr₃ in CH₂Cl₂ led to the formation of the presumed α -bromo derivative 7. This adduct could not be fully purified or characterized due to the sensitivity of 7 to moisture and its poor solubility in nonhydroxylic solvents. Accordingly, either the addition of an excess of the heteroatom species or the sequential addition of triethylamine and the nucleophilic heteroatom-containing reagent to a THF mixture containing 7 furnished 3 in moderate amounts (21-38% yields from 3t). This method was employed for the preparation of compounds 3d, 3k, 3p, and 3u-w. Included in the product mixture for 3k were the diastereomeric adducts 8a and 8b in which 2 equiv of 7 reacted with NH₂OH. Implementation of this procedure for the synthesis of 31-o was not successful due to the difficulty encountered in obtaining solutions of the free hydroxyamines in nonhydroxylic solvents. Use of methanolic solutions of the hydroxyamines furnished only 3s.



Employment of the third protocol (method C) outlined in Scheme I provided a convenient procedure to circumvent this obstacle. Methylation of the dimethylamino adduct 3d with trimethyloxonium tetrafluoroborate¹⁰ in nitromethane furnished the quaternary ammonium derivative 3f in high yields. Subsequent treatment of 3f with a methanolic solution containing the requisite hydroxyamine led to production of 31-o in good yields (42-82%).

Synthesis of the α -hydroxy (3r), α -ethylthio (3w), and α -thiophenoxy⁸ (3x) amino acid derivatives was accomplished using the last technique depicted in Scheme I in which 3t was treated with BF₃:Et₂O in the presence of H₂O, EtSH, and PhSH, respectively (Scheme I, method D). A similar protocol was utilized by us for the preparation of α -heteroaromatic functionalized amino acid derivatives.^{6,8} This procedure proved superior than that of method B for the preparation of 3w.

Two of the remaining compounds listed in Table I, 3i and 3j, were obtained by treatment of 3a with acetic anhydride and trifluoroacetic anhydride, respectively. The final compounds 3y and 3z were prepared directly from the α -ethylthio adduct 3w. Interestingly, treatment of 3w with *m*-chloroperbenzoic acid in CH_2Cl_2 led to the stereoselective production of the α -sulfoxide 3y-1. ¹³C NMR analysis of the initial reaction mixture indicated the presence of only a single diastereomeric (enantiomeric pair) compound. The precise stereochemical identity of this adduct has not been established. Correspondingly, use of stoichiometric amounts of NaIO₄ at room temperature in an aqueous methanolic solution yielded a 2:1 diastereomeric mixture (¹³C NMR analysis) of 3y-1 and 3y-2 in which the major compound present corresponded to the product generated in the *m*-chloroperbenzoic acid reaction (3y-1). Attempts to completely separate these diastereomers by either TLC or recrystallization proved unsuccessful. Accordingly, a 2:3 mixture of 3y-1 and 3y-2 obtained after fractional recrystallization was analyzed for anticonvulsant activity and is identified as 3y-mix. We have tentatively attributed the diastereoselectivity of the m-chloroperbenzoic acid mediated process to the preorganization of the oxidant with 3w in CH₂Cl₂. Employment of excess $NaIO_4$ with 3w at elevated temperatures (50–60) °C) gave the α -sulfone 3z in 32% yield.

Of note, all 17 α , α -diamino acid derivatives (3a-q), including the trimethylammonium adduct 3f, were well-

⁽⁹⁾ Kober, R.; Steglich, W. Liebigs Ann. Chem. 1983, 599.

⁽¹⁰⁾ Aldrich Chemical Co.

⁽¹¹⁾ Hogberg, T.; Strom, P.; Ebner, M.; Ramsby, S. J. Org. Chem. 1987, 52, 2033.

defined, stable compounds.¹² Only 3f, 3i, and 3k melted with decomposition.

Pharmacological Evaluation

The α -heteroatom substituted amino acid derivatives 3 were tested for anticonvulsant activity by using the procedures described by Krall et al.¹³ All compounds were administered intraperitoneally (ip) to mice. Table I lists the median effective dose (ED₅₀) values required to prevent seizures in the MES test by racemic 3. Included in this table are the median neurotoxic dose (TD₅₀) values determined for select compounds using the horizontal screen test.¹⁴

Evaluation of the results listed in Table I revealed several important observations. First, the α -amino (3a), α -alkylamino (3b-e), and α -trimethylammonium (3f) derivatives all displayed anticonvulsant activities comparable to that observed for the α -methyl analogue 2a.^{3,4} Second. the α -arylamino derivatives 3g and 3h were devoid of activity at doses below 100 mg/kg. A comparable reduction in activity has been observed in proceeding from the α -methyl derivative (2a) to the corresponding α -benzyl adduct⁴ and has been attributed (in part) to the stringent steric requirements that exist for maximal anticonvulsant activity in this class of compounds. Third, conversion of the α -amino derivative 3a to the corresponding α -acylamino adducts 3i and 3j led to a decrease in activity of the test compound. Fourth, incorporation of an α -N-alkoxyamino moiety (i.e., 31, 3n, 3o) within the backbone of the compound led to a pronounced improvement of the potency of the compound in the MES test compared to either 2a or 3a. A corresponding enhancement in activity was not observed for the two N-hydroxyamino adducts 3k and 3m. The anticonvulsant activities of racemic 3l (ED₅₀ = 6.2 mg/kg) and 3n (ED₅₀ = 6.7 mg/kg) were comparable to that of the (R,S)-2-furanyl derivative 2d⁶ (ED₅₀ = 10.3 mg/kg) and phenytoin⁷ (ED₅₀ = 9.5 mg/kg). Importantly, in the most potent analogues (2d, 3l, and 3n), a functionalized oxygen atom existed two atoms removed from the α -carbon atom. This pattern suggests that a substituted β -heteroatom may be necessary for maximal activity. Fifth, the α -hydrazine derivatives **3p** and **3q** did not display significant anticonvulsant activity. Once again this property has been attributed (in part) to the steric size of these substituents. Sixth, the α -hydroxy (3r) and the two α -alkoxy adducts (3s, 3t) displayed activity comparable to that reported for **2a**. The potency of the α -oxygen series was somewhat diminished from that observed for the corresponding α -amino derivatives (3a-c). In agreement with previous findings⁴ the α -phenoxy adduct 3u displayed no activity at doses of 100 mg/kg or less. Seventh, within the α -sulfur series, only the α -ethylthic adduct 3w exhibited anticonvulsant activity at doses less than 100 mg/kg. Eighth, no enhancement of activity was noted for the three sulfur-oxygenated derivatives 3y-1, 3y-mix, and 3z versus 3w. This observation is consistent with the results obtained for the two N-hydroxyamino adducts 3k and 3m versus 3a.

Conclusions

Straightforward procedures have been employed for the preparation of α -heteroatom-substituted amino acid derivatives. Despite the fact that these compounds have geminal heteroatoms α to the carbonyl, they are chemically well-defined and are expected to serve as useful substrates in future chemical and pharmacological studies. The pharmacological data obtained in this investigation provided additional information concerning the structureactivity profile of functionalized amino acid anticonvulsants. The biological activities for 3 reinforced our notions that stringent steric and electronic requirements exist for maximal anticonvulsant activity in this class of compounds. The potencies of 31 and 3n in the MES test were comparable to those of phenytoin and 2d. Additional studies in progress are aimed at investigating the generality of this class of compounds, as well as their mode of action.

Experimental Section

Chemistry. General Methods. Melting points were determined with a Thomas-Hoover melting point apparatus and are uncorrected. Infrared spectra (IR) were run on Perkin-Elmer 1330 and 283 spectrometers and were calibrated against the 1601-cm⁻¹ band of polystyrene. Absorption values are expressed in wavenumbers (cm⁻¹). Proton (¹H NMR) and carbon (¹³C NMR) nuclear magnetic resonance spectra were taken on Nicolet NT-300 and General Electric QE-300 NMR instruments. Chemical shifts (δ) are in parts per million (ppm) relative to Me₄Si, and coupling constants (J values) are in hertz. Low-resolution mass spectra (MS) were recorded at an ionizing voltage of 70 eV with a Varian MAT CH-5 spectrometer at the Lilly Research Laboratories. Microanalyses were provided by the Physical Chemistry Department of the Lilly Research Laboratories. Benzyl carbazate was obtained from Lancaster Synthesis Ltd., Windham, NH. Thin-layer chromatography were run on precoated silica gel GHLF microscope slides $(2.5 \times 10 \text{ cm}; \text{Analtech No. } 21521)$.

Preparation of α -Heteroatom-Substituted Amino Acids (3). Method A. Synthesis of Ethyl 2-Acetamido-2-substituted-acetates. General Procedure. A cooled (-78 °C) solution of 5⁹ (1 equiv) in THF (1 mmol/10 mL) was added slowly to a THF (1 mmol/4 mL) solution of the nitrogen nucleophile (5-10 equiv) at -78 °C. The reaction was stirred at this temperature (0.5 h) and then at room temperature (1 h). The insoluble materials were filtered and washed with THF. The filtrate was concentrated in vacuo, and the residue was purified by flash chromatography on SiO₂ gel (using the indicated solvent as the eluent) to give the desired product.

By use of this procedure, the following compounds were prepared.

Synthesis of Ethyl 2-Acetamido-2-aminoacetate (6a). Compound 5 (2.00 g, 8.93 mmol) and liquid NH₃ (5-6 equiv) yielded a light brown residue, which on purification by flash column chromatography on SiO₂ gel (5% MeOH/CHCl₃) gave 1.00 g (70%) of 6a as a yellow oil: R_f 0.21 (5% MeOH/CHCl₃); ¹H NMR (CDCl₃) δ 1.31 (t, J = 7.1 Hz, 3 H), 2.03 (s, 3 H), 2.61 (br s, 2 H), 4.24 (q, J = 7.1 Hz, 2 H), 5.21 (d, J = 7.1 Hz, 1 H), 7.50 (d, J = 7.1 Hz, 1 H); ¹³C NMR (CDCl₃) 13.72, 22.68, 59.70, 61.73, 170.40, 170.68 ppm.

Synthesis of Ethyl 2-Acetamido-2-(methylamino)acetate (6b). Use of 5 (2.00 g, 8.93 mmol) and MeNH₂ (2.50 g, 80.6 mmol) gave an oily residue (1.50 g). The residue was purified by flash column chromatography on SiO₂ gel (3% MeOH/CHCl₃) to yield 1.00 g (65%) of 6b as an oil: R_f 0.30 (3% MeOH/CHCl₃); ¹H NMR (CDCl₃) δ 1.32 (t, J = 7.1 Hz, 3 H), 2.07 (s, 3 H), 2.36 (s, 3 H), 4.26 (q, J = 7.1 Hz, 2 H), 5.20 (d, J = 7.4 Hz, 1 H), 6.60 (br s, 1 H) (the remaining amino proton was not detected); ¹³C NMR (CDCl₃) 14.02, 23.06, 30.84, 62.04, 65.72, 170.09, 170.40 ppm.

Synthesis of Ethyl 2-Acetamido-2-(ethylamino)acetate (6c). Employing 5 (2.10 g, 9.38 mmol) and EtNH₂ (1.40 g, 31.04 mmol) gave a brown residue. The residue was purified by flash column chromatography on SiO₂ gel (3% MeOH/CHCl₃) to yield 0.90 g (51%) of 6c as a light yellow oil: R_f 0.36 (4% MeOH/CHCl₃); ¹H NMR (CDCl₃) δ 0.93 (t, J = 6.7 Hz, 3 H), 1.12 (t, J = 6.8 Hz, 3 H), 1.87 (s, 3 H), 2.48 (q, J = 6.7 Hz, 2 H), 4.05 (q,

⁽¹²⁾ For recent reports on the preparation of α,α-diamino acid derivatives and related compounds, see: (a) Katritzky, A. R.; Urogdi, L.; Mayence, A. J. Chem. Soc., Chem. Commun. 1989, 337. (b) Katritzky, A. R.; Urogdi, L.; Mayence, A. J. Org. Chem. 1990, 55, 2206. (c) Bock, M. G.; DiPardo, R. M.; Freidinger, R. M. Ibid. 1986, 51, 3718. (d) Fischer, B.; Hassner, A. Ibid. 1990, 55, 5225 and references therein.

⁽¹³⁾ Krall, R. L.; Penry, J. K.; White, B. G.; Kupferberg, H. J.; Swinyard, E. A. *Epilepsia* 1978, 19, 409.

⁽¹⁴⁾ Coughenour, L. L.; McLean, R. R.; Parker, R. B. Pharmacol. Biochem. Behav. 1977, 6, 351.

J = 6.8 Hz, 2 H), 5.05 (d, J = 7.1 Hz, 1 H), 7.09 (d, J = 7.1 Hz, 1 H) (the remaining amino proton was not detected); ¹³C NMR (CDCl₃) 13.64, 14.55, 22.53, 39.06, 61.38, 64.14, 170.09, 170.20 ppm.

Synthesis of Ethyl 2-Acetamido-2-(dimethylamino)acetate (6d). Compound 5 (2.00 g, 8.93 mmol) and Me₂NH (5–6 equiv) gave 6d (1.50 g, 89%) as a yellow oil: ¹H NMR (CDCl₃) δ 1.25 (t, J = 7.1 Hz, 3 H), 2.02 (s, 3 H), 2.23 (s, 6 H), 4.10–4.25 (m, 2 H), 5.24 (d, J = 8.3 Hz, 1 H), 6.59 (d, J = 8.3 Hz, 1 H); ¹³C NMR (CDCl₃) 14.05, 23.00, 40.28 (2 C), 61.84, 69.24, 169.38, 170.57 ppm.

Synthesis of Ethyl 2-Acetamido-2-(4-morpholino)acetate (6e). Use of morpholine (1.71 g, 19.64 mmol) and 5 (2.00 g, 8.93 mmol) gave an oily residue, which was purified by flash column chromatography on SiO₂ gel (2% MeOH/CHCl₃) to give 1.90 g (93%) of 6e as a thick oil: R_f 0.29 (3% MeOH/CHCl₃); ¹H NMR (CDCl₃) δ 1.32 (t, J = 6.8 Hz, 3 H), 2.07 (s, 3 H), 2.43–2.72 (m, 4 H), 3.58–3.78 (m, 4 H), 4.26 (q, J = 6.8 Hz, 2 H), 5.27 (d, J = 7.9 Hz, 1 H), 6.39 (d, J = 7.9 Hz, 1 H); ¹³C NMR (CDCl₃) 14.21, 23.25, 48.47 (2 C), 62.06, 66.71 (2 C), 69.22, 169.00, 170.46 ppm.

Synthesis of Ethyl 2-Acetamido-2-(*N*-anilino)acetate (6g). Use of aniline (1.83 g, 19.6 mmol) and 5 (2.00 g, 8.93 mmol) provided a brown residue, which was purified by flash column chromatography on SiO₂ gel (CHCl₃-2% MeOH/CHCl₃ gradient) to yield 1.80 g (85%) of 6g: mp 87-89 °C (recrystallized from ethyl acetate/petroleum ether); R_f 0.52 (4% MeOH/CHCl₃); IR (KBr) 3340, 1720, 1635, 1590, 1490, 730, 710 cm⁻¹; ¹H NMR (CDCl₃) δ 1.29 (t, J = 7.1 Hz, 3 H), 1.84 (s, 3 H), 4.27 (q, J = 7.1 Hz, 2 H), 5.89 (d, J = 8.2 Hz, 1 H), 6.43 (d, J = 8.2 Hz, 1 H), 6.68-6.71 (m, 2 H), 6.80-6.83 (m, 1 H), 7.17-7.22 (m, 2 H) (the remaining amino proton was not detected); ¹³C NMR (CDCl₃) 13.96, 22.98, 60.19, 62.41, 113.87 (2 C), 119.29, 129.37 (2 C), 144.09, 169.77, 170.14 ppm; mass spectrum (FD) 237 (M⁺ + 1). Anal. (C₁₂H₁₆N₂O₃) C, H, N.

Synthesis of Ethyl 2-Acetamido-2-(3-pyrazolylamino)acetate (6h). Use of 5 (2.00 g, 8.93 mmol) and 3-aminopyrazole (1.85 g, 22.32 mmol) and purification of the reaction product by chromatography on SiO₂ gel (2% MeOH/CHCl₃) gave 1.80 g (89%) of 6h as a yellow oil: R_f 0.35 (8% MeOH/CHCl₃); ¹H NMR (CDCl₃) δ 1.21 (t, J = 7.1 Hz, 3 H), 1.89 (s, 3 H), 4.20 (q, J = 7.1 Hz, 2 H), 5.64 (d, J = 1.8 Hz, 1 H), 5.71 (br s, 1 H), 5.73 (d, J= 7.1 Hz, 1 H), 7.29 (d, J = 1.8 Hz, 1 H), 7.98 (d, J = 7.1 Hz, 1 H) (the remaining amino proton was not detected); ¹³C NMR (CDCl₃) 13.73, 22.49, 61.41, 62.02, 91.79, 130.53, 153.02, 169.96, 170.93 ppm.

Synthesis of Ethyl 2-Acetamido-2-(hydroxyamino)acetate (6k). Use of 5 (2.10 g, 9.38 mmol) and anhydrous NH₂OH¹⁵ (0.93 g, 28.00 mmol) gave an oily residue. The residue was purified by flash column chromatography on SiO₂ gel (5% MeOH/CHCl₃) to give 1.00 g (61%) of 6k. The product was recrystallized from EtOH to give a white flaky solid: mp 119–121 °C; R_f 0.24 (5% MeOH/CHCl₃); IR (KBr) 3300, 1750, 1660, 1540, 1390, 610 cm⁻¹; ¹H NMR (DMSO- d_8) δ 1.19 (t, J = 6.9 Hz, 3 H), 1.87 (s, 3 H), 4.10 (q, J = 6.9 Hz, 2 H), 5.09 (dd, J = 4.0, 80 Hz, 1 H), 6.06 (br s, 1 H), 7.63 (s, 1 H), 8.50 (d, J = 8.0 Hz, 1 H); ¹³C NMR (DMSO- d_8) 14.05, 22.46, 60.82, 67.37, 169.19, 169.48 ppm; mass spectrum (FD) 177 (M⁺ + 1). Anal. (C₆H₁₂N₂O₄) C, H, N.

Synthesis of Ethyl 2-Acetamido-2-(methylhydroxyamino)acetate (6m). MeNHOH (17.39 mmol) (prepared from MeNHOH·HCl (2.00 g, 23.95 mmol) and NaOMe (0.94 g, 17.39 mmol)) and 5 (1.00 g, 4.46 mmol) gave an oily residue. The residue was triturated with EtOAc (5 mL) and the solid (0.70 g, 82%) that remained was filtered and recrystallized from EtOH to give 6m as a white solid: mp 148–150 °C; R_f 0.34 (5% MeOH/CHCl₃); IR (KBr) 3320, 3200 (br), 1760, 1660, 1530, 1470, 720, 640 cm⁻¹; ¹H NMR (DMSO- d_6) δ 1.17 (t, J = 7.0 Hz, 3 H), 1.89 (s, 3 H), 2.37 (s, 3 H), 4.00–4.20 (m, 2 H), 5.04 (d, J = 9.2 Hz, 1 H), 8.17 (s, 1 H), 8.43 (d, J = 9.2 Hz, 1 H); ¹³C NMR (DMSO- d_6) 14.04, 22.28, 43.78, 60.79, 71.46, 168.29, 170.23 ppm; mass spectrum (FD) 192 (M⁺ + 1). Anal. (C₇H₁₄N₂O₄·0.25H₂O) C, H, N.

Synthesis of Ethyl 2-Acetamido-2-(methoxymethylamino)acetate (6n). MeNHOMe (17.40 mmol) (prepared from MeNHOMe HCl (2.18 g, 22.32 mmol) and NaOMe (0.94 g, 17.40 mmol)) and 5 (1.00 g, 4.46 mmol) gave a residue, which was purified by flash column chromatography on SiO₂ gel (1% MeOH/CHCl₃) to give 0.60 g (66%) of **6n** as an oil: R_f 0.53 (2% MeOH/CHCl₃); ¹H NMR (CDCl₃) δ 1.35 (t, J = 7.0 Hz, 3 H), 2.12 (s, 3 H), 2.62 (s, 3 H), 3.46 (s, 3 H), 4.30 (q, J = 7.0 Hz, 2 H), 5.36 (d, J = 8.9 Hz, 1 H), 6.66 (d, J = 8.9 Hz, 1 H); ¹³C NMR (CDCl₃) 14.06, 22.89, 40.30, 60.01, 61.89, 70.16, 168.14, 170.53 ppm.

Synthesis of 2-Acetamido-N-benzyl-2-substituted-acetamides (3). General Procedure. A mixture of 6 (1 equiv), benzylamine (1.2 equiv), and NaCN (0.1 equiv) in MeOH (1 mmol/25 mL) was stirred at 45-50 °C (18 h). The solvent was removed in vacuo, and the residue was purified with either trituration with EtOAc or flash column chromatography on SiO₂ gel (using the indicated solvent as the eluent).

By use of this procedure, the following compounds were prepared.

Synthesis of 2-Acetamido-N-benzyl-2-aminoacetamide (3a). Compound 6a (1.00 g, 6.25 mmol), benzylamine (0.80 g, 7.5 mmol), and NaCN (0.03 g, 0.61 mmol) gave a residue that solidified on standing (18 h). The reaction mixture was triturated with EtOAc (20 mL). The white solid (1.00 g, 72%) that remained was filtered and then further purified by recrystallization from EtOAc: mp 131-133 °C dec; R_f 0.21 (5% MeOH/CHCl₃); IR (KBr) 3300, 1650 (br), 1530 (br), 1450, 740 cm⁻¹; ¹H NMR (DMSO- d_6) δ 1.83 (s, 3 H), 2.35 (br s, 2 H), 4.28 (d, J = 4.4 Hz, 2 H), 4.91 (d, J = 7.0 Hz, 1 H), 7.20–7.32 (m, 5 H), 8.31 (br s, 1 H); ¹³C NMR (DMSO- d_6) 22.66, 42.05, 60.29, 126.67, 127.10 (2 C), 128.18 (2 C), 139.23, 169.24, 170.67 ppm; mass spectrum, m/e (relative intensity) 222 (M⁺ + 1, 100), 221 (M⁺, 29), 133 (8). Anal. (C₁₁H₁₅N₃O₂) C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2-(methylamino)acetamide (3b). Compound 6b (1.50 g, 8.63 mmol), benzylamine (1.11 g, 10.35 mmol), and NaCN (0.04 g, 0.82 mmol) gave a brown residue that was purified by flash column chromatography on SiO₂ gel (2% MeOH/CHCl₃) to yield 1.00 g (49%) of 3b: mp 115–117 °C (recrystallized from ethyl acetate/petroleum ether); R_f 0.33 (3% MeOH/CHCl₃); IR (KBr) 3240, 1610 (br), 1500 (br), 1430, 725, 670 cm⁻¹; ¹H NMR (DMSO- d_{θ}) δ 1.87 (s, 3 H), 2.18 (s, 3 H), 4.20–4.29 (m, 2 H), 4.87 (d, J = 7.9 Hz, 1 H), 7.24–7.35 (m, 5 H), 8.14 (d, J = 7.9 Hz, 1 H), 8.55 (br s, 1 H) (the remaining amino proton was not detected); ¹³C NMR (DMSO- d_{θ}) 22.52, 31.37, 42.04, 65.99, 126.68, 127.12 (2 C), 128.18 (2 C), 139.28, 169.51, 169.83 ppm. Anal. (C₁₂H₁₇N₃O₂) C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2-(ethylamino)acetamide (3c). Use of 6c (0.90 g, 4.79 mmol), benzylamine (0.62 g, 5.75 mmol), and NaCN (0.03 g, 0.51 mmol) gave an oily residue that was purified by flash column chromatography on SiO₂ gel (3% MeOH/CHCl₃) to give 0.35 g (29%) of 3c as a white solid: mp 123-125 °C (recrystallized from ethyl acetate/hexane); R_f 0.34 (4% MeOH/CHCl₃); IR (KBr) 3250, 1620 (br), 1510 (br), 1450 (br), 740, 680 cm⁻¹; ¹H NMR (DMSO- d_6) δ 0.93 (t, J = 6.8 Hz, 3 H), 1.81 (s, 3 H), 2.08 (br s, 1 H), 2.40-2.48 (m, 2 H), 4.22 (d, J = 5.5 Hz, 2 H), 4.90 (d, J = 7.8 Hz, 1 H), 7.20-7.27 (m, 5 H), 8.08 (d, J = 7.8 Hz, 1 H), 8.48 (t, J = 5.5 Hz, 1 H); ¹³C NMR (CDCl₃) 15.14, 22.97, 37.65, 43.53, 65.68, 127.44 (2 C), 127.50, 128.64 (2 C), 137.73, 169.75, 171.20 ppm. Anal. (C₁₃H₁₉N₃O₂) C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2-(4-morpholino)acetamide (3e). Use of 6e (0.59 g, 2.56 mmol), benzylamine (0.28 g, 2.82 mmol), and NaCN (0.01 g, 0.26 mmol) gave a thick oily residue. The residue was triturated with EtOAc (5 mL), and the white solid (0.35 g) that remained was collected by filtration to give 3e. The filtrate was concentrated, and the residue was purified by flash column chromatography on SiO₂ gel (2% MeOH/CHCl₃). The initial fractions furnished a trace amount (0.09 g) of 2-acetamido-N-benzyl-2-(benzylamino)acetamide. Continued elution gave additional amounts (0.20 g) of 3e.

2-Acetamido-N-benzyl-2-(benzylamino)acetamide: yield 0.09 g (11%); mp 135–138 °C; R_f 0.52 (4% MeOH/CHCl₃); IR (KBr) 3250 (br), 1630 (br), 1500 (br), 1425, 750, 700 cm⁻¹; ¹H NMR (DMSO- d_6) δ 1.83 (s, 3 H), 3.52 (br s, 1 H), 3.56 (d, J = 13.6 Hz, 1 H), 3.66 (d, J = 13.6 Hz, 1 H), 4.23 (d, J = 5.4 Hz, 2 H), 4.89 (d, J = 8.0 Hz, 1 H), 7.05–7.38 (m, 10 H), 8.20 (d, J = 8.0 Hz, 1 H), 7.05–7.38 (m, 10 H), 8.20 (d, J = 8.0 Hz, 1 H), 4.87, 64.41, 126.70, 127.13 (2 C), 128.00 (2 C), 128.13 (2 C), 128.22 (2 C), 128.30, 139.29, 140.12, 169.61, 169.90 ppm; mass spectrum, m/e (relative intensity) 312 (M⁺ + 1, 12), 311 (M⁺, 7), 178 (11), 177 (100). Anal. (C₁₆H₂₁N₃O₂·0.25H₂O) C, H, N.

⁽¹⁵⁾ Hurd, C. D. Inorg. Synth. 1939, 1, 87.

Functionalized a-Heteroatom-Substituted Amino Acids

2-Acetamido-N-benzyl-2-(4-morpholino)acetamide (3e): yield 0.55 g (71%); mp 171-172 °C (recrystallized from EtOAc); $R_{\rm f}$ 0.35 (4% MeOH/CHCl₃); IR (KBr) 3250, 1620, 1515, 1440, 740, 690 cm⁻¹; ¹H NMR (DMSO- d_6) δ 1.87 (s, 3 H), 2.29-2.45 (m, 4 H), 3.51 (t, J = 4.1 Hz, 4 H), 4.23 (dd, J = 6.1, 15.1 Hz, 1 H), 4.28 (dd, J = 6.1, 15.1 Hz, 1 H), 5.08 (d, J = 9.1 Hz, 1 H), 7.12-7.30 (m, 5 H), 8.24 (d, J = 9.1 Hz, 1 H), 8.58 (t, J = 6.1 Hz, 1 H); ¹³C (DMSO- d_6) 22.39, 42.20, 48.43 (2 C), 66.03, 69.24 (2 C), 126.76, 127.13 (2 C), 128.21 (2 C), 139.42, 168.02, 170.20 ppm; mass spectrum, m/e (relative intensity) 292 (M⁺ + 1, 1), 233 (8), 158 (19), 157 (100), 116 (26), 115 (100), 106 (29), 91 (72). Anal. (C₁₅H₂₁N₃O₃) C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2-(N-anilino)acetamide (3g). 6g (2.00 g, 8.47 mmol), benzylamine (1.09 g, 10.00 mmol), and NaCN (0.04 g, 0.84 mmol) gave a white solid that separated during the course of the reaction. The precipitate was filtered and purified by recrystallization from absolute EtOH to give 1.10 g (44%) of 3g: mp 183-185 °C; IR (KBr) 3270, 1630, 1520, 1490, 1430, 740, 690 cm⁻¹; ¹H NMR (DMSO-d₆) δ 1.84 (s, 3 H), 4.31 (d, J = 5.8 Hz, 2 H), 5.67 (t, J = 8.1 Hz, 1 H), 6.69 (d, J = 8.1 Hz, 1 H), 6.59-6.64 (m, 1 H), 6.70-6.72 (m, 2 H), 7.06-7.11 (m, 2 H), 7.20-7.33 (m, 5 H), 8.41 (d, J = 8.1 Hz, 1 H), 8.72 (t, J = 5.8 Hz, 1 H); ¹³C NMR (DMSO-d₆) 22.46, 42.25, 60.42, 113.21 (2 C), 117.22, 126.72, 127.16 (2 C), 128.18 (2 C), 128.77 (2 C), 138.99, 145.88, 168.65, 169.70 ppm; mass spectrum, m/e (relative intensity) 297 (M⁺, 2), 239 (7), 164 (28), 163 (100), 122 (20), 121 (100), 106 (47), 104 (65), 93 (63), 91 (77). Anal. (C₁₇H₁₉N₃O₂) C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2-(3-pyrazolylamino)acetamide (3h). A solution of 6h (1.60 g, 7.1 mmol) in MeOH (40 mL) containing benzylamine (0.83 g, 7.8 mmol) and NaCN (50 mg, 1 mmol) was stirred at 45-55 °C (18 h). TLC analysis (8% MeOH/CHCl₃) of the reaction mixture indicated the presence of only a minor amount of product. A second lot of NaCN (50 mg, 1 mmol) was then added, and the reaction was allowed to proceed at 45-55 °C (6 h) and then at room temperature (48 h). The solvent was removed in vacuo, and the residue was triturated with EtOAc (15 mL). The insoluble solid that remained was filtered and purified by flash column chromatography on SiO_2 gel (7% MeOH/CHCl₃) to give 0.90 g (44%) of 3h: mp 135-137 °C; R, 0.35 (8% MeOH/CHCl₃); IR (KBr) 3230 (br), 1620 (br), 1500 (br), 1430, 730, 690 cm⁻¹; ¹H NMR (DMSO- d_{6}) δ 1.82 (s, 3 H), 4.29 (d, J = 5.9 Hz, 2 H), 5.51–5.55 (m, 3 H), 7.18–7.40 (m, 6 H), 8.36 (br s, 1 H), 8.53 (br s, 1 H), 11.66 (br s, 1 H); ¹³C NMR (DMSO-d₆) 22.59, 42.29, 61.79, 90.68, 126.67, 127.07 (2 C), 128.17 (2 C), 129.10, 139.41, 153.53, 169.19, 169.67 ppm; mass spectrum, m/e (relative intensity) 288 (M⁺ + 1, 64), 287 (M⁺, 2), 230 (28), 229 (100), 153 (46). Anal. (C₁₄H₁₇N₅O₂.0.5H₂O) C, H, N.

Preparation of Functionalized α -Heteroatom-Substituted Amino Acids (3). Method B. General Procedure. A BBr₃ solution (1 M in CH₂Cl₂, 1.1 equiv) was added to a solution of $3t^{6,8}$ (1 equiv) in CH₂Cl₂ (10 mmol/125 mL). The mixture was stirred at room temperature (5 h) and then concentrated to dryness in vacuo to give 7 as a pale yellow crystalline material. The bromo adduct 7 was then dissolved in THF (10 mmol/250 mL), cooled (-78 °C), and then added over a 15-min interval to a cooled (-78 °C) solution of the heteroatom nucleophile in THF (1 mmol/1 mL). The reaction mixture was stirred at this temperature (30 min) and then at room temperature (90 min). The insoluble salts were filtered, and the filtrate was concentrated in vacuo. The residue was then purified by flash column chromatography on SiO₂ gel (using the indicated solvent as the eluent).

By using this procedure, the following compounds were prepared.

Synthesis of 2-Acetamido-N-benzyl-2-(dimethylamino)acetamide (3d). By making use of 3t (3.00 g, 12.0 mmol), BBr₃ (1 M in CH₂Cl₂, 13.2 mL, 13.2 mmol), and Me₂NH (5-6 equiv), a brown residue was obtained that was purified by flash column chromatography on SiO₂ gel (2.5% MeOH/CHCl₃) to give 1.20 g (40%) of 3d. The product was recrystallized from ethyl acetate/hexane to give light yellow cubic crystals: mp 104-106 °C; R_f 0.39 (5% MeOH/CHCl₃); IR (KBr) 3280, 1670 (br), 1500 (br), 1460, 760, 700 cm⁻¹; ¹H NMR (DMSO- $d_{\rm g}$) δ 1.91 (s, 3 H), 2.11 (s, 6 H), 4.22 (dd, J = 5.2, 14.7 Hz, 1 H), 4.34 (dd, J = 6.1, 14.7 Hz, 1 H), 5.11 (d, J = 8.3 Hz, 1 H), 7.23-7.31 (m, 5 H), 8.18 (d, J = 8.3 Hz, 1 H), 8.55 (br s, 1 H); ¹³C NMR (DMSO- $d_{\rm g}$) 22.43, 40.33 (2 C), 42.28, 69.42, 126.73, 127.27 (2 C), 128.21 (2 C), 139.49, 168.49, 170.31 ppm; mass spectrum (FD) 250 (M⁺ + 1). Anal. (C₁₃-H₁₉N₃O₂) C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2-(hydroxyamino)acetamide (3k). Use of 3t (2.00 g, 8.0 mmol), BBr₃ (1 M in CH_2Cl_2 , 8.8 mL, 8.8 mmol), and anhydrous NH_2OH^{15} (5–6 equiv) gave an oily residue. The residue was separated into three components by flash chromatography on SiO₂ gel (7.5% MeOH/ CHCl₃).

2-Acetamido-N-benzyl-2-(hydroxyamino)acetamide (3k): yield 0.14 g (7%); mp 144–146 °C dec (recrystallized from EtOH); R_f 0.30 (8% MeOH/CHCl₃); IR (KBr) 3320 (br), 1660 (br), 1540 (br), 1460, 750, 700 cm⁻¹; ¹H NMR (DMSO-d₆) δ 1.88 (s, 3 H), 4.31 (d, J = 5.7 Hz, 2 H), 5.08 (dd, J = 4.4, 8.1 Hz, 1 H), 5.94 (dd, J = 2.8, 4.4 Hz, 1 H), 7.19–7.35 (m, 5 H), 7.52 (d, J = 2.8 Hz, 1 H), 8.26 (d, J = 8.1 Hz, 1 H), 8.42 (t, J = 5.7 Hz, 1 H); ¹³C NMR (DMSO-d₆) 22.69, 42.25, 67.86, 126.69, 127.14 (2 C), 128.18 (2 C), 139.08, 168.53, 169.67 ppm; mass spectrum (FD) 238 (M⁺ + 1). Anal. (C₁₁H₁₅N₃O₃) C, H, N.

Compound 8a: yield 0.05 g (3%); mp 177-179 °C (recrystallized from EtOH); R_f 0.27 (8% MeOH/CHCl₃); IR (KBr) 3240 (br), 1640 (br), 1510 (br), 1450, 690 cm⁻¹; ¹H NMR (DMSO- d_6) δ 1.82 (s, 6 H), 4.25-4.34 (m, 4 H), 5.21 (d, J = 9.3 Hz, 2 H), 7.20-7.33 (m, 10 H), 8.16 (d, J = 9.3 Hz, 2 H), 8.26 (t, J = 5.8 Hz, 2 H), 8.51 (s, 1 H); ¹³C NMR (DMSO- d_6) 22.54 (2 C), 42.30 (2 C), 67.55 (2 C), 126.63 (2 C), 127.13 (4 C), 128.11 (4 C), 139.02 (2 C), 168.24 (2 C), 169.33 (2 C) ppm; mass spectrum (FD) 442 (M⁺ + 1). Anal. (C₂₂H₂₇N₅O₅) C, H, N.

Compound 8b: yield 0.10 g (6%); mp 184–186 °C (recrystallized from MeOH); R_f 0.18 (8% MeOH/CHCl₃); IR (KBr) 3300 (br), 1660 (br), 1530 (br), 1450, 740, 700 cm⁻¹; ¹H NMR (DMSO- d_6) δ 1.87 (6 H), 4.20 (dd, J = 5.3, 15.3 Hz, 2 H), 4.44 (dd, J = 6.2, 15.3 Hz, 2 H), 5.28 (d, J = 9.0 Hz, 2 H), 7.15–7.31 (m, 10 H), 8.00 (d, J = 9.0 Hz, 2 H), 8.39 (dd, J = 5.3, 6.2 Hz, 2 H), 8.51 (s, 1 H); ¹³C NMR (DMSO- d_6) 22.50 (2 C), 42.58 (2 C), 69.98 (2 C), 126.73 (2 C), 127.23 (4 C), 128.22 (4 C), 139.08 (2 C), 167.60 (2 C), 169.57 (2 C) ppm; mass spectrum (FD) 442 (M⁺ + 1). Anal. (C₂₂H₂₇N₅O₅) C, H, N.

Improved Synthesis of 2-Acetamido-N-benzyl-2-(hydroxyamino)acetamide (3k). Compound 7 (prepared from 3t (3.00 g, 12.0 mmol) and BBr₃ (1 M in CH₂Cl₂, 17.2 mL, 17.2 mmol)) was dissolved in THF (250 mL), cooled (-10 °C), and then added dropwise (30 min) to a suspension of NH₂OH (5-6 equiv) in THF (50 mL) at -10 °C. The reaction mixture was stirred (30 min) at this temperature and then allowed to warm to room temperature (1 h). The insoluble materials were filtered, and the filtrate was concentrated in vacuo. The residue was separated into two components by flash column chromatography on SiO₂ gel (7.5% MeOH/CHCl₃).

2-Acetamido-N-benzyl-2-(hydroxyamino)acetamide (3k): yield 0.66 g (23%); mp 144-146 °C dec (recrystallized from EtOH). Compound 8b: yield 0.10 g (5%); mp 184-186 °C (recrystallized from MeOH).

Compound 8a was not observed under these conditions.

Synthesis of 2-Acetamido-N-benzyl-2-(N^2 -phenyl-hydrazino)acetamide (3p). Use of 3t (2.00 g, 8.0 mmol), BBr₃ (1 M in CH₂Cl₂, 10.0 mL, 10.0 mmol), and phenylhydrazine (2.60 g, 24.0 mmol) gave a pale yellow oily residue, which was purified by flash column chromatography on SiO₂ gel (2% MeOH/CHCl₃) to give 0.75 g (29%) of 3p. The product was recrystallized from chloroform/hexane as a light yellow solid: mp 132-134 °C; $R_{\rm f}$ 0.26 (2% MeOH/CHCl₃); IR (KBr) 3300, 1640 (br), 1610, 1520 (br), 1460, 760, 700 cm⁻¹; ¹H NMR (DMSO-d₆) δ 1.89 (s, 3 H), 4.28 (d, J = 5.8 Hz, 2 H), 4.89 (d, J = 5.2 Hz, 1 H), 5.09 (dd, J = 5.2, 7.4 Hz, 1 H), 6.61 (t, J = 7.4 Hz, 1 H), 6.70–7.28 (m, 10 H), 8.29 (d, J = 7.4 Hz, 1 H), 8.60 (t, J = 5.8 Hz, 1 H); ¹³C NMR (DMSO-d₆) 22.88, 42.22, 66.22, 112.66 (2 C), 117.57, 126.65, 127.08 (2 C), 128.15 (2 C), 128.53 (2 C), 139.12, 149.90, 168.66, 170.04 ppm; mass spectrum (FD) 313 (M⁺ + 1). Anal. (C₁₇H₂₀N₄O₂) C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2- $(N^2-(benzy) xy-carbonyl)$ hydrazino) acetamide (3q). By employing 3t (3.00 g, 12.0 mmol), BBr₃ (1 M in CH₂Cl₂, 15.0 mL, 15.0 mmol), and benzyl carbazate (4.58 g, 27.6 mmol), 0.95 g (21%) of 3q was obtained. The product was recrystallized from chloroform/hexane to give a white amorphous solid: mp 152–154 °C; R_f 0.32 (2% MeOH/CHCl₃); IR (KBr) 3325, 1620 (br), 1500 (br), 1440, 740, 680 cm⁻¹; ¹H NMR (DMSO- d_6) δ 1.85 (s, 3 H), 4.27 (d, J = 4.4

Hz, 2 H), 5.00 (s, 2 H), 5.14 (dd, J = 3.1, 8.0 Hz, 1 H), 5.23 (t, J = 3.1 Hz, 1 H), 7.25–7.35 (m, 10 H), 8.26 (d, J = 8.0 Hz, 1 H), 8.56 (br s, 1 H), 8.66 (br s, 1 H); ¹³C NMR (DMSO- d_6) 22.71, 42.23, 65.56, 65.97, 126.69, 127.16 (2 C), 127.61 (2 C), 127.77, 128.13 (2 C), 128.27 (2 C), 136.74, 138.87, 168.04, 169.95 ppm; mass spectrum (FD) 371 (M⁺ + 1). Anal. (C₁₉H₂₂N₄O₄) C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2-phenoxyacetamide (3u). Use of 3t (3.00 g, 12.0 mmol), BBr₃ (1 M in CH₂Cl₂, 15.0 mL, 15.0 mmol), and NaOPh (4.18 g, 30 mmol) gave a brown oily residue, which was purified by flash column chromatography on SiO₂ gel using first CHCl₃ and then 2% MeOH/CHCl₃ as the eluents to give 0.80 g (22%) of 3u. The compound was recrystallized from chloroform/hexane: mp 125-128 °C (softens at 122 °C); R_1 0.58 (3% MeOH/CHCl₃); IR (KBr) 3300, 1650 (br), 1600, 1530 (br), 1490, 1450, 760, 700 cm⁻¹; ¹H NMR (DMSO-d₆) δ 1.63; (s, 3 H), 4.35 (d, J = 5.7 Hz, 2 H), 6.18 (d, J = 9.4 Hz, 1 H); ¹³C NMR (DMSO-d₆) 22.54, 42.24, 76.44, 116.09 (2 C), 121.78, 126.84, 127.26 (2 C), 128.25 (2 C), 128.44 (2 C), 138.84, 155.97, 166.63, 170.73 ppm; mass spectrum (FD) 299 (M⁺ + 1). Anal. (C₁₇H₁₈N₂O₃·0.5H₂O) C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2-(methylthio)acetamide (3v). A cooled (-78 °C) solution of Et₃N (4.85 g, 48.0 mmol) in THF (20 mL) was added to a cooled (-78 °C) solution of 7 (prepared from 3t (4.00 g, 16.0 mmol) and BBr₃ (1 M in CH₂Cl₂, 20.0 mL, 20.0 mmol)) in THF (275 mL). A cooled (-78 °C) solution of excess MeSH (5-6 equiv) in THF (55 mL) was then added. The reaction mixture was stirred at this temperature (30 min) and then at room temperature (1 h). The insoluble materials were filtered, and the filtrate was evaporated to dryness in vacuo. The oily residue obtained was purified by flash column chromatography on SiO₂ gel (2% MeOH/CHCl₃) to give 1.10 g (27%) of 3v as a yellow orange oil. The product was purified by a second flash column chromatography on SiO₂ gel (2% MeOH/CHCl₃) to give 0.72 g of the pure product as a white solid: mp 155-157 °C; R_f 0.65 (3% MeOH/CHCl₃); IR (KBr) 3320, 1650 (br), 1520 (br), 1460, 750 cm⁻¹; ¹H NMR (CD₃NO₂) δ 1.98 (s, 3 H), 2.08 (s, 3 H), 4.39 (dd, J = 6.1, 15.2 Hz, 1 H), 4.49 (dd, J = 6.1, 15.2 Hz, 1 H), 5.51 (d, J = 7.8 Hz, 1 H), 7.15 (d, J = 7.8 Hz, 1 H), 7.17–7.41 (m, 6 H); ¹³C NMR (CD₃NO₂) 12.28, 22.94, 44.26, 56.03, 128.46, 128.60 (2 C), 129.77 (2 C), 140.17, 169.19, 171.06 ppm; mass spectrum (FD) 253 (M⁺ + 1). Anal. ($C_{12}H_{16}N_2O_2S$) C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2-(ethylthio)acetamide (3w). By using the procedure described for the synthesis of 3v, 3t (2.00 g, 8.0 mmol) and EtSH (0.65 g, 10.4 mmol) were converted to 0.80 g (38%) of 3w. The compound was further purified by recrystallization from chloroform/hexane to give a beige solid: mp 146–148 °C; R_{1} 0.60 (4% MeOH/CHCl₃); IR (KBr) 3240, 1620 (br), 1510 (br), 1415, 680, 640 cm⁻¹; ¹H NMR (DMSO-d₆) δ 1.56 (t, J = 7.4 Hz, 3 H), 1.88 (s, 3 H), 2.49–2.67 (m, 2 H), 4.23 (dd, J = 5.9, 15.2 Hz, 1 H), 4.32 (dd, J = 5.9, 15.2 Hz, 1 H), 5.55 (d, J = 9.1 Hz, 1 H), 7.20–7.35 (m, 5 H), 8.59 (d, J = 9.1 Hz, 1 H), 8.75 (t, J = 5.9 Hz, 1 H); ¹³C NMR (DMSO-d₆) 14.73, 22.43, 23.73, 42.10, 53.70, 126.87, 127.14 (2 C), 128.32 (2 C), 139.01, 167.89, 169.02 ppm; mass spectrum (FD) 267 (M⁺ + 1). Anal. (C₁₃H₁₈N₂O₂S-0.25H₂O) C, H, N.

Preparation of Functionalized α -**Hetero**atom-Substituted Amino Acids (3). Method C. General Procedure. A mixture of 3f (1 equiv), and the nitrogen nucleophile (4-5 equiv) in MeOH (1 mmol/1 mL) was stirred at 55-60 °C (3 h). The solvent was removed in vacuo, and the residue was purified by flash column chromatography on SiO₂ gel by using the indicated solvents as the eluent.

By using this procedure, the following compounds were prepared.

Synthesis of 2-Acetamido-N-benzyl-2-(methoxyamino)acetamide (31). Use of a MeOH solution of MeONH₂ (prepared from MeONH₂·HCl (2.83 g, 33.9 mmol) and NaOMe (1.41 g, 26.1 mmol)) and 3f (2.70 g, 7.67 mmol) gave an oily residue that was purified by flash column chromatography on SiO₂ gel (2% MeOH/CHCl₃) to give 0.80 g (42%) of 31. The product was recrystallized from chloroform/hexane: mp 95–97 °C; R_1 0.23 (2% MeOH/CHCl₃); IR (KBr) 3300, 1650, 1620, 1510 (br), 1440, 750, 680 cm⁻¹; ¹H NMR (DMSO- d_6) δ 1.88 (s, 3 H), 3.38 (s, 3 H), 4.22-4.41 (m, 2 H), 5.18 (dd, J = 4.9, 7.8 Hz, 1 H), 6.78 (d, J =4.9 Hz, 1 H), 7.21-7.32 (m, 5 H), 8.33 (d, J = 7.8 Hz, 1 H), 8.56 (br s, 1 H); ¹³C NMR (DMSO- d_{θ}) 22.64, 42.28, 61.42, 66.25, 126.74, 127.19 (2 C), 128.19 (2 C), 139.11, 167.95, 169.66 ppm; mass spectrum (FD) 252 (M⁺ + 1). Anal. (C₁₂H₁₇N₃O₃) C, H, N. Synthesis of 2-Acetamido-N-benzyl-2-(methylhydroxy-

Synthesis of 2-Acetamido-N-benzyl-2-(methylhydroxyamino)acetamide (3m). A MeOH solution (30 mL) of MeNHOH (21.74 mmol) (prepared from MeNHOH·HCl (2.36 g, 28.26 mmol) and NaOMe (1.17 g, 21.74 mmol)) and 3f (2.20 g, 6.25 mmol) gave a residue that was purified by flash column chromatography on SiO₂ gel (6% MeOH/CHCl₃) to give 0.95 g (61%) of 3m as a white solid. The product was then purified by recrystallization from EtOH: mp 159–161 °C; R_f 0.32 (8% MeOH/CHCl₃); IR (KBr) 3440 (br), 3300, 1640, 1530, 1460, 750, 700 cm⁻¹; ¹H NMR (DMSO- d_{θ}) δ 1.95 (s, 3 H), 2.43 (s, 3 H), 4.26 (dd, J = 5.7, 15.1 Hz, 1 H), 4.35 (dd, J = 5.7, 15.1 Hz, 1 H), 5.09 (d, J = 9.1 Hz, 1 H), 7.21–7.29 (m, 5 H), 8.05 (s, 1 H), 8.18 (d, J = 9.1 Hz, 1 H), 8.23 (t, J = 5.7 Hz, 1 H); ¹³C NMR (DMSO- d_{θ}) 22.40, 42.34, 43.92, 71.49, 126.62, 127.12 (2 C), 128.12 (2 C), 139.14, 167.82, 170.28 ppm; mass spectrum (FD) 252 (M⁺ + 1). Anal. (C₁₂H₁₇N₃O₃) C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2-(methoxymethylamino)acetamide (3n). A MeOH solution (20 mL) of MeN-HOMe (17.39 mmol) (prepared from MeNHOMe·HCl (2.20 g, 23.02 mmol) and NaOMe (0.94 g, 17.39 mmol)) and 3f (2.10 g, 5.97 mmol) gave a solid residue. Flash column chromatography of the solid on SiO₂ gel (2% MeOH/CHCl₃) yielded pure 3n (1.30 g, 82%). The product was recrystallized from EtOH: mp 165–167 °C; R_f 0.39 (2% MeOH/CHCl₃); IR (KBr) 3300, 1640 (br), 1540 (br), 1460, 750, 700 cm⁻¹; ¹H NMR (DMSO- d_6) δ 1.93 (s, 3 H), 2.43 (s, 3 H), 3.32 (s, 3 H), 4.25 (dd, J = 5.9, 14.9 Hz, 1 H), 4.37 (dd, J = 5.9, 14.9 Hz, 1 H), 5.19 (d, J = 9.4 Hz, 1 H), 7.21–7.35 (m, 5 H), 8.31 (d, J = 9.4 Hz, 1 H), 8.56 (t, J = 5.9 Hz, 1 H); ¹³C NMR (DMSO- d_6) 22.36, 39.68, 42.34, 59.16, 70.33, 126.74, 127.41 (2 C), 128.21 (2 C), 139.30, 167.38, 170.30 ppm; mass spectrum (FD) 266 (M⁺ + 1). Anal. (C₁₃H₁₉N₃O₃) C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2-(N-isoxazolidino)acetamide (30). 3f (1.60 g, 4.55 mmol), isoxazolidine (prepared from isoxazolidine hydrobromide¹⁶ (2.41 g, 15.65 mmol) and NaOMe (0.70 g, 13.04 mmol)) gave 0.80 g (64%) of 30. Compound 30 was recrystallized from chloroform/hexane to give a white amorphous solid: mp 149-151 °C; R_1 0.29 (4% MeOH/CHCl₃); IR (KBr) 3400 (br), 3300, 1650, 1530, 1470, 740, 700, 610 cm⁻¹; ¹H NMR (DMSO- d_6) δ 1.91 (s, 3 H), 2.05-2.20 (m, 2 H), 2.45-2.89 (m, 1 H), 2.98-3.07 (m, 1 H), 3.74-3.90 (m, 2 H), 4.25 (dd, J =6.1, 15.3 Hz, 1 H), 4.35 (dd, J = 6.1, 15.3 Hz, 1 H), 5.23 (d, J =9.2 Hz, 1 H), 7.15-7.35 (m, 5 H), 8.49 (d, J = 9.2 Hz, 1 H), 8.56 (br s, 1 H); ¹³C NMR (DMSO- d_6) 22.26, 28.26, 42.15, 48.94, 66.19, 68.77, 126.64, 127.02 (2 C), 128.13 (2 C), 139.22, 167.43, 170.27 ppm; mass spectrum (FD) 278 (M⁺ + 1). Anal. (C₁₄H₁₉N₃O₃) C, H, N.

Preparation of Functionalized α -Heteroatom-Substituted Amino Acids (3). Method D. General Procedure. 2-Acetamido-N-benzyl-2-ethoxyacetamide (3t) (1 equiv) was suspended in Et₂O (100 mL/10 mmol), and then BF₃·Et₂O (1.6-2.4 equiv) was rapidly added and the resulting solution was stirred (10 min). The nucleophile (H₂O or EtSH) (1.6-4.0 equiv) was then added, and the reaction mixture was stirred at room temperature (18-48 h). The reaction was then quenched by the addition of an aqueous NaHCO₃ (100 mL/10 mmol)/ice mixture. The experimental workup varied slightly for 3r and 3w and is described in the following sections along with the observed spectral properties.

Synthesis of 2-Acetamido-N-benzyl-2-hydroxyacetamide (3r). Reacting 3t (1.00 g, 4.0 mmol), BF₃·Et₂O (0.91 g, 6.4 mmol), and H₂O (0.12 g, 6.7 mmol) followed by aqueous NaHCO₃ workup gave an aqueous reaction mixture. The solution was then extracted with EtOAc (3×50 mL), and the combined EtOAc extracts were dried (Na₂SO₄) and concentrated in vacuo. The residue was purified by flash column chromatography on SiO₂ gel (3%MeOH/CHCl₃) to give 0.30 g (34%) of 3r as a white solid: mp 136-138 °C; R_f 0.14 (3% MeOH/CHCl₃); IR (KBr) 3300, 1620, 1530 (br), 1430 (br), 730, 690 cm⁻¹; ¹H NMR (DMSO-d₆) δ 1.85 (s, 3 H), 4.29 (d, J = 5.9 Hz, 2 H), 5.48 (dd, J = 5.5, 8.6 Hz, 1 H), 6.47 (d, J = 5.5 Hz, 1 H), 7.21-7.35 (m, 5 H), 8.52 (t, J = 5.9 Hz, 1 H), 8.59 (d, J = 8.6 Hz, 1 H); ¹³C NMR (DMSO-d₆) 22.66, 41.99,

⁽¹⁶⁾ Hine, J.; Evangelista, R. A. J. Am. Chem. Soc. 1980, 102, 1649.

71.42, 126.66, 127.22 (2 C), 128.13 (2 C), 139.20, 169.47, 169.62 ppm; mass spectrum, m/e (relative intensity) 223 (M⁺ + 1, 1), 163 (11), 134 (9), 106 (46), 91 (100), 77 (22), 65 (38). Anal. (C₁₁H₁₄N₂O₃) C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2-(ethylthio)acetamide (3w). Use of 3t (2.00 g, 8.0 mmol), BF₃:Et₂O (2.72 g, 19.2 mmol), and EtSH (2.38 g, 38.4 mmol) gave an aqueous reaction mixture. The solution was extracted with CHCl₃ (3×100 mL). The combined CHCl₃ layers were dried (Na₂SO₄) and then concentrated in vacuo to give 1.90 g (89%) of 3w as a white solid: mp 148-149 °C (mixed melting point with an authentic sample was undepressed); R_f 0.60 (4% MeOH/CHCl₃).

Synthesis of 2,2 Diacetamido-N-benzylacetamide (3i). Ac₂O (1 mL) was added to a solution of 3a (1.10 g, 4.98 mmol) in dry pyridine (10 mL), and then CH₂Cl₂ (20 mL) was added. The mixture was stirred at room temperature (4 h), and then the volatile materials were removed in vacuo. The residue was then treated with a saturated aqueous NaHCO₃ solution (50 mL). The white solid (1.20 g, 92%) that remained was filtered, dried (Na₂SO₄), and recrystallized from MeOH: mp 265-267 °C dec; IR (KBr) 3260, 1530, 1500, 740, 690 cm⁻¹; ¹H NMR (DMSO-d₆) δ 1.84 (s, 6 H), 4.26 (d, J = 5.8 Hz, 2 H), 5.71 (t, J = 7.6 Hz, 1 H), 7.20-7.31 (m, 5 H), 8.44 (d, J = 7.6 Hz, 2 H), 8.48 (t, J = 5.8Hz, 1 H); ¹³C NMR (DMSO-d₆) 22.44 (2 C), 42.26, 56.99, 126.62, 127.02 (2 C), 128.12 (2 C), 139.15, 168.19, 169.39 (2 C) ppm; mass spectrum (FD) 264 (M⁺ + 1). Anal. (C₁₃H₁₇N₃O₃) C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2-(trifluoroacetamido)acetamide (3j). Ice-cold trifluoroacetic anhydride (8 mL) was added in one portion to ice-cold 3a (1.00 g, 4.53 mmol). The reaction was accompanied by the evolution of heat. After stirring (5 mm), the volatile materials were removed in vacuo. The residue was treated with a saturated aqueous NaHCO₃ solution (20 mL), and the solid that remained was filtered and washed with H₂O to give 3j (1.00 g, 70%). The product was recrystallized from EtOH: mp 228-230 °C; R_f 0.34 (8% MeOH/CHCl₂); IR (KBr) 3300, 1720, 1660, 1520, 1380, 760, 700 cm⁻¹; ¹H NMR (DMSO-d₆) δ 1.90 (s, 3 H), 4.30 (d, J = 5.1 Hz, 2 H), 5.85 (d, J = 8.0 Hz, 1 H), 7.21–7.35 (m, 5 H), 8.64 (d, J = 8.0 Hz, 1 H), 8.75 (t, J = 5.1Hz, 1 H), 10.04 (s, 1 H); ¹³C NMR (DMSO-d₆) 22.52, 42.52, 57.42, 117.4 (q, J_{CF} = 288.3 Hz), 126.80, 127.16 (2 C), 128.21 (2 C), 138.93, 156.14 (q, J_{CF} = 35.3 Hz), 166.39, 169.88 ppm; mass spectrum (FD) 318 (M^+ + 1). Anal. ($C_{13}H_{14}N_3O_3F_3$) C, H, N. Synthesis of 2-Acetamido-N-benzyl-2-(trimethyl-

Synthesis of 2-Acetamido-N-benzyl-2-(trimethylammonio)acetamide Tetrafluoroborate (3f). A solution of 3d (1.93 g, 7.76 mmol) in nitromethane (7 mL) was added slowly to an ice-cold solution of trimethyloxonium tetrafluoroborate (1.26 g, 8.54 mmol) in nitromethane (6 mL). The reaction mixture was stirred at this temperature (15 min) and then at room temperature (2 h). Anhydrous Et₂O (~50 mL) was added to the reaction mixture, and the white solid that separated was filtered, washed with Et₂O, and dried in vacuo: yield 1.95 g (72%); mp 171-173 °C dec; IR (KBr) 3300, 1680 (br), 1530, 1490, 710 cm⁻¹; ¹H NMR (CD₃NO₂) δ 2.14 (s, 3 H), 3.18 (s, 9 H), 4.50 (d, J = 5.8 Hz, 2 H), 5.70 (d, J = 9.3 Hz, 1 H), 7.30-7.41 (m, 5 H), 7.57 (d, J = 9.3 Hz, 1 H), 7.70 (br s, 1 H); mass spectrum (FD) 264 (M⁺). Anal. (C₁₄H₂₂N₃O₂BF₄) C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2-(ethylthio)acetamide S-Oxide (3y-1). A solution of m-chloroperbenzoic acid (1.00 g (~65%), 3.76 mmol) in CH_2Cl_2 (10 mL) was added dropwise into a stirred, cooled (-10 to -15 °C) CH₂Cl₂ solution (125 mL) of 3w (1.00 g, 3.76 mmol) under N₂. The reaction was stirred (30 mm) at this temperature, and then the *m*-chlorobenzoic acid was precipitated as its ammonium salt by passing NH₃ gas over the surface of the reaction solution. The excess NH₃ was removed by passing N_2 gas through the solution (20 min) at room temperature. The ammonium salt was filtered, and the filtrate was concentrated in vacuo. The residue was purified by flash chromatography on SiO₂ gel (2% MeOH/CHCl₃) to give 0.55 g (52%) of the desired product. Compound 3y-1 was recrystallized from chloroform/hexane as a white granular solid: mp 135-137 °C; R₁ 0.23 (2% MeOH/CHCl₃); IR (KBr) 3300 (br), 1640 (br), 1510 (br), 1370, 1230, 1100, 1020, 900 cm⁻¹; ¹H NMR (DMSO-d₆) δ 1.15 (t, J = 7.5 Hz, 3 H), 1.99 (s, 3 H), 2.49–2.56 (m, 1 H), 2.65-2.72 (m, 1 H), 4.34 (d, J = 5.7 Hz, 2 H), 5.55 (d, J = 9.5 Hz, 1 H), 7.23–7.34 (m, 5 H), 8.74 (d, J = 9.5 Hz, 1 H), 8.77 (t, J =5.7 Hz, 1 H); ¹³C NMR (DMSO-d₈) 7.03, 22.34, 42.40, 42.47, 67.15,

126.89, 127.27 (2 C), 128.24 (2 C), 138.55, 164.66, 170.18 ppm; mass spectrum (FD) 283 (M⁺ + 1). Anal. $(C_{13}H_{18}N_2O_3S)$ C, H, N.

Synthesis of 2-Acetamido-N-benzyl-2-(ethylthio)acetamide S-Oxide (3y-1 and 3y-2). A solution of $NaIO_4$ (1.77 g, 8.27 mmol) in H₂O (20 mL) was added dropwise into a stirred solution of 3w (2.00 g, 7.52 mmol) in MeOH (25 mL). A precipitate appeared rapidly. H_2O (~30 mL) was added to the mixture to dissolve most of the suspension, and the reaction mixture was stirred (4 h) at room temperature. The reaction was concentrated in vacuo, and the remaining aqueous mixture was extracted with $CHCl_3$ (3 × 100 mL). The combined $CHCl_3$ extracts were dried (Na_2SO_4) , and the solvent was removed in vacuo. The oily residue (1.95 g, 92%) solidified on drying in vacuo. NMR analysis (DMSO- d_6) of the product showed that it was a 2:1 mixture of the two diastereomers 3y-1 and 3y-2, respectively. The reaction was recrystallized from EtOAc to give nearly pure diastereomer 3y-1 (1.20 g). The EtOAc mother liquor was concentrated, and the remaining residue (0.75 g) was recrystallized from ethyl acetate/hexane to give a diastereomeric mixture (0.41 g) of 3y-1 and 3y-2 in a 2:3 ratio, respectively: mp 135-137 °C (softens at 117 °C); R_f 0.60 (4% MeOH/CHCl₃); IR (KBr) 3300 (br), 1640 (br), 1510 (br), 1370, 1230, 1100, 1020, 900 cm⁻¹; mass spectrum (FD) 283 (M^+ + 1). Anal. ($C_{13}H_{18}N_2O_3S$) C, H, N. The following NMR spectral properties have been assigned to compounds 3y-1 and 3y-2.

Diastereomer 3y-1: ¹H NMR (DMSO- d_6) δ 1.16 (t, J = 7.5 Hz, 3 H), 2.00 (s, 3 H), 2.49–2.72 (m, 2 H), 4.28–4.39 (m, 2 H), 5.56 (d, J = 9.7 Hz, 1 H), 7.21–7.34 (m, 5 H), 8.71–8.77 (m, 2 H); ¹³C NMR (DMSO- d_6) 7.10, 22.43, 42.48, 42.57, 67.23, 126.98, 127.36 (2 C), 128.33 (2 C), 138.63, 164.73, 170.25 ppm.

Diastereomer 3y-2: ¹H NMR (DMSO- d_6) δ 1.13 (t, J = 7.6 Hz, 3 H), 1.94 (s, 3 H), 2.49–2.72 (m, 2 H), 4.28–4.39 (m, 2 H), 5.71 (d, J = 9.9 Hz, 1 H), 7.21–7.34 (m, 5 H), 8.83 (d, J = 9.9 Hz, 1 H), 8.98 (t, J = 5.6 Hz, 1 H); ¹³C NMR (DMSO- d_6) 6.47, 22.43, 41.53, 42.55, 67.90, 126.98, 127.36 (2 C), 128.33 (2 C), 138.39, 164.43, 169.82 ppm.

Synthesis of 2-Acetamido-N-benzyl-2-(ethanesulfonyl)acetamide (3z). An aqueous solution (20 mL) of NaIO₄ (3.00 g, 14.02 mmol) was added to a MeOH solution (20 mL) of 3w (0.95 g, 3.57 mmol). The initial homogeneous solution rapidly became turbid. H_2O (~10 mL) was then added dropwise until the system became homogeneous. The solution was stirred (18 h) at 50-60 °C. MeOH (50 mL) was added to the reaction solution, and the precipitated salt was filtered and washed with MeOH (10 mL). The filtrate was concentrated, and the remaining solution was extracted with CHCl₃ (3 \times 50 mL). The combined CHCl₃ extracts were dried (Na₂SO₄) and concentrated in vacuo. The residue was purified by flash chromatography on SiO₂ gel (1% MeOH/CHCl₃) to give 0.34 g (32%) of the desired product. Compound 3z was further purified by recrystallization from EtOH: mp 161-163 °C; R_{f} 0.34 (3% MeOH/CHCl₂); IR (KBr) 3300, 2940, 1660, 1520, 1310, 1230, 1120, 900 cm⁻¹; ¹H NMR (DMSO- d_{6}) δ 1.22 (t, J = 7.4 Hz, 3 H), 1.99 (s, 3 H), 3.04-3.24 (m, 2 H), 4.31 (dd, J = 5.7, 15.3 Hz, 1 H), 4.41 (dd, J = 5.7, 15.3 Hz, 1 H), 5.93 (d, J = 9.8 Hz, 1 H), 7.22–7.35 (m, 5 H), 9.13 (t, J = 5.7 Hz, 1 H), 9.17 (d, J = 9.8 Hz, 1 H); ¹³C NMR (DMSO-d₆) 5.72, 22.27, 42.63, 45.43, 69.14, 127.02, 127.28 (2 C), 128.33 (2 Č), 138.16, 161.88, 169.83 ppm; mass spectrum (FD) 298 (M⁺). Anal. ($C_{13}H_{18}N_2O_4S$) C, H, N.

Pharmacology. Initial anticonvulsant evaluation of 3 was conducted with at least three dose levels (30, 100, and 300 mg/kg) administered intraperitoneally. All tests were performed with male CF-1 mice from Charles River Breeding Laboratories (Portage, MI). Test solutions of all compounds were dissolved in 30% poly(ethylene glycol) 400. Four mice at each dose level were tested at 0.5, 1, and 4 h after administration to determine if there was protection against MES seizures.

MES seizures were elicited by electrical current (ac 60 cps, 50 mA, 0.2 s) applied via corneal electrodes. A drop of 0.9% saline was instilled on the eye prior to application of the electrodes to ensure electrical contact. Abolition of the hind limb tonic extension component of the seizure was defined as protection in the MES test.

After the time of peak activity and the approximate dose range were determined, a dose-response curve was generated with at least three or four doses and 10–12 mice per dose. The MES ED_{50} is the estimated dose from the dose-response data to protect 50% of the mice in the MES test. Neurological impairment was measured in mice using the horizontal screen (tox) text.¹⁴ Previously trained mice were dosed with the compound and then placed individually on top of a square (13 × 13 cm) wire screen (no. 4 mesh), which was mounted on a vertical rod. The rod was then rotated 180°, and the number of mice that returned to the top of the screen within 1 min was determined. To determine ED₅₀ values, a dose-response curve was determined at the time of peak anticonvulsant activity with at least three to four doses and 12 mice per dose. The TD₅₀ is the estimated dose that impaired 50% of the mice.

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Registry No. 3a, 133873-19-5; 3b, 133873-25-3; 3c, 133873-26-4;

3d, 133873-20-8; 3e, 133873-27-5; 3f, 133873-22-0; 3g, 133873-28-6; 3h, 133873-29-7; 3i, 133873-30-0; 3j, 133873-31-1; 3k, 133873-32-2; 31, 133873-33-3; 3m, 133873-34-4; 3n, 133873-35-5; 3o, 133873-36-6; 3p, 133873-37-7; 3q, 133873-38-8; 3r, 133873-39-9; 3s, 133873-23-1; 3t, 117422-80-7; 3u, 133873-40-2; 3v, 133873-41-3; 3w, 133873-24-2; 3x, 133887-00-0; 3y (diastereomer 1), 133873-42-4; 3y (diastereomer 2), 133886-68-7; 3z, 133873-43-5; 5, 124421-42-7; 6a, 133873-08-2; 6b, 133873-09-3; 6c, 133873-10-6; 6d, 133873-11-7; 6e, 133873-12-8; 6g, 133873-13-9; 6h, 133873-14-0; 6k, 133873-15-1; 6m, 133873-16-2; 6n, 133873-17-3; 7, 133873-18-4; 8 (diastereomer 1), 133873-45-7; 8 (diastereomer 2), 133873-46-8; PhNH₂, 62-53-3; PhCH₂NH₂, 100-46-9; PhNHNH₂, 100-63-0; H₂NNHCOOCH₂Ph, 5331-43-1; NaOPh, 139-02-6; MeSH, 74-93-1; EtSH, 75-08-1; PhSH, 108-98-5; (±)-AcNHCH(NHCH₂Ph)CONHCH₂Ph, 133873-44-6; morpholine, 110-91-8; 3-aminopyrazole, 1820-80-0; isoxazolidine hydrobromide, 111780-15-5.

Arocalciferols: Synthesis and Biological Evaluation of Aromatic Side-Chain Analogues of 1α ,25-Dihydroxyvitamin D_3^{1a}

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Aromatic side-chain analogues (arocalciferols 6-9) of the steroid hormone 1α ,25-dihydroxyvitamin D₃ (1) were synthesized and biologically evaluated. The analogues were prepared by coupling the vitamin D A-ring enyne 14 with the appropriate enol triflate of a modified CD steroid fragment of the type 22. The resulting dienyne 23 was then transformed in three steps to the vitamin D analogues 6-9. Biological evaluation of these analogues have provided information concerning side-chain topographical effects on in vivo and in vitro activity.

Introduction

The steroid hormone 1α ,25-dihydroxyvitamin D₃ [1, 1,25(OH)₂D₃], the physiologically active form of vitamin D (calciferol), is considered to be one of the most potent stimulators of calcitropic effects (intestinal calcium absorption, ICA, and bone calcium mobilization, BCM).² Besides its traditional role as a hormone in calcium homeostasis,³ 1,25(OH)₂D₃ induces differentiation and affects cellular proliferation, indicating its possible use in the treatment of certain cancers and skin disorders.⁴⁻⁶ There

has accordingly been an increased interest in the further development of $1,25(OH)_2D_3$ analogues for these new therapeutic purposes.

The clinical utility of $1,25(OH)_2D_3$ is limited because therapeutically effective doses induce hypercalcemia.⁷ This has led investigators toward the development of an analogue with a more useful therapeutic index, specifically directed toward analogues with high cell differentiating ability and low calcitropic action. A series of remarkably diverse side-chain analogues of $1,25(OH)_2D_3$ exhibiting promising therapeutic indices have in fact been reported

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