

modifications of the side chain at position 21 have little effect on binding to FKBP (e.g., FK-506 vs ascromycin),¹⁷ suggesting that this portion of the molecule may not be involved in binding to the protein. This is consistent with the lack of NOEs between the ethyl side chain of ascromycin and FKBP. It is interesting to note that those portions of ascromycin that were found to be in close proximity to FKBP from the NOE data are the same parts of the molecule in common with rapamycin, a potent immunosuppressant which also binds tightly to FKBP. Indeed, as noted previously,¹⁴ the piperidine ring of rapamycin and FK-506 is involved in binding to FKBP and is likely to be oriented in a similar manner in the two ligand/receptor complexes.

Conclusions. From an analysis of NOEs observed in heteronuclear 3D NOE spectra of the [U - ^{13}C]ascromycin/FKBP complex, the conformation of ascromycin when bound to its putative target protein, FKBP, was determined. The conformation of ascromycin was found to be very different from the X-ray structure of FK-506 determined in the absence of FKBP, but more closely resembles that of uncomplexed rapamycin. In addition, from the observation of ascromycin/FKBP NOEs in the 3D spectra, those portions of ascromycin that are in close proximity to FKBP were identified, and they include parts of the piperidine, pyranose, and cyclohexyl rings. By using the structural information obtained from this study it may be possible to design ascromycin analogues that are structurally dissimilar from ascromycin but which nonetheless maintain the necessary functionality in the proper orientation for binding to FKBP. It is hoped that analogues could be designed with improved physicochemical properties and potentially lower toxicity, thus making them superior immunosuppressive agents.

During the preparation of this manuscript, a report on the X-ray crystal structure of the FK-506/FKBP complex appeared.²³ The conformation of ascromycin when bound to FKBP as determined by NMR is very similar to the conformation of FK-506 when bound to FKBP as determined by X-ray crystallography. In both structures of the bound ligands, the 9,10 amide bonds are trans and the relative orientation of the pyranose and cyclohexyl rings are the same. In the X-ray structure roughly 50% of the ligand surface is buried at the protein-ligand interface with the region around the allyl and cyclohexyl groups being exposed to solvent. We observed no NOEs from the ethyl group of ascromycin (analogous to the allyl group in FK-506) to the protein, which is consistent with this portion of the molecule being exposed to solvent. However, we do observe several NOEs from the cyclohexyl ring (Figure 5) to the protein, which clearly indicates that a portion of this ring is in close proximity to the protein in solution.

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Supplementary Material Available: A table of 1H and ^{13}C chemical shifts of [U - ^{13}C]ascromycin bound to FKBP (1 page). Ordering information is given on any current masthead page. The coordinates of ascromycin when bound to FKBP will be deposited

in the Brookhaven Protein Data Bank.

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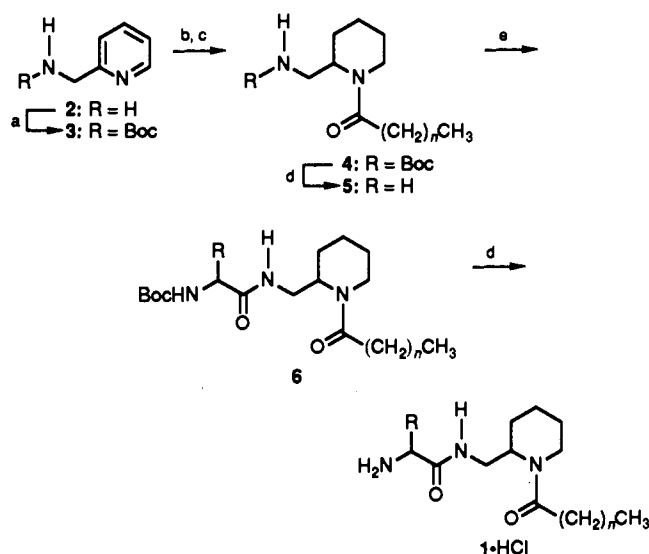
Substituted 2-(Aminomethyl)piperidines: A Novel Class of Selective Protein Kinase C Inhibitors

Since its discovery in 1977, the ubiquitous enzyme protein kinase C (PKC)¹ has received extensive pharmacological investigation and has emerged as a pivotal mediator in cellular regulation, signal transduction, and neoplastic promotion. Physiological activity of the enzyme is regulated by the allosteric modulators Ca^{2+} , diacylglycerol (DAG), and phosphatidylserine (PS).^{1,2} These modulators interact with the enzyme's regulatory domain while the catalytic domain possesses the site of ATP and substrate binding.³ Phorbol esters have been found to substitute for diacylglycerol as a potent enzyme stimulator,⁴ and in animals, phorbol ester activation of PKC causes intense inflammation. Unlike diacylglycerol, phorbol esters are not rapidly metabolized and thus can effect prolonged enzyme stimulation often leading to neoplastic events.⁵

Given the biological responses induced by activators of protein kinase C, the development of inhibitors of this enzyme may lead to therapeutic agents useful in the treatment of chronic inflammatory and proliferative diseases. Several natural and synthetic agents have been identified as PKC antagonists. These include (1) nonselective phospholipid competing agents such as triphenylethylenes,⁶ chlorpromazine,⁷ and trifluoperazine,⁸ (2)

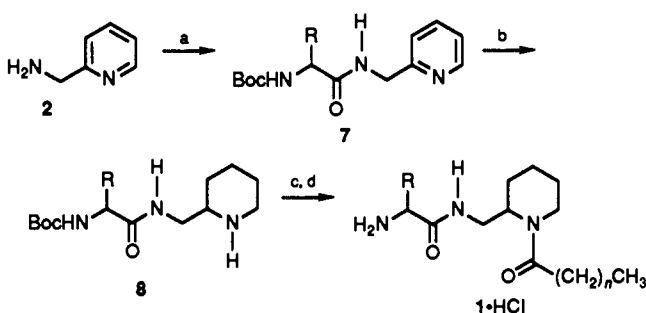
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Scheme I^a

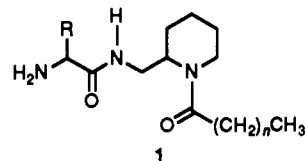
peptide analogues⁹ which lack utility due to in vivo instability, (3) agents interacting with the ATP binding site of the enzymes catalytic domain including staurosporine¹⁰ and isoquinolinesulfonamides¹¹ which also interact with the highly homologous domains of cGMP- and cAMP-dependent kinase, thus limiting their usefulness as specific PKC inhibitors, and (4) inhibitors interacting with the enzymes regulatory domain such as lipoidal amines,¹² aminoacridines,¹³ and calphostin C,¹⁴ the most potent, selective PKC inhibitor known to date.

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Scheme II^a

^a (a) L- or D-BocHNCH(R)CO₂H, EDC, HOBT·H₂O, NMM, CH₂Cl₂, 0 °C to room temperature; (b) H₂, PtO₂, HOAc, EtOH, 60 psi; (c) CH₃(CH₂)_nCOCl, pyr, DMAP, CH₂Cl₂, 0 °C to room temperature or CH₃(CH₂)_nCO₂H, EDC, DMAP, CH₂Cl₂, room temperature; (d) anhydrous HCl, dioxane.

During our course of investigations relating to the development of novel antiinflammatory agents, piperidinylamides of the general structure 1 were discovered to possess selective PKC inhibitory activity. In this communication, we report the synthesis and preliminary pharmacological evaluation of a series of analogues based on structure 1.

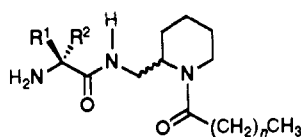


Scheme I (method A)¹⁵ depicts the synthesis of our general series. Catalytic hydrogenation of *N*-Boc¹⁶ protected 2-(aminomethyl)pyridine (3) over PtO₂ followed by acid chloride treatment or carbodiimide coupling¹⁷ with a carboxylic acid efficiently produced racemic amides 4. Acidic removal of the *N*-Boc protecting group yielded amino amides 5. This sequence provides common intermediates for derivatization and is amenable to large-scale preparation. Coupling of enantiomerically pure *N*-Boc-protected amino acids with racemic intermediate 5 employing EDC/HOBT¹⁸ and subsequent amine deprotection with HCl in dioxane under anhydrous conditions provided the diastereomeric targets 1 as hydrochloride salts.

Alternatively, a more direct route as outlined in Scheme II (method B)¹⁵ can be employed. Conventional carbodiimide coupling¹⁸ of enantiomerically pure *N*-Boc-protected amino acids with 2-(aminomethyl)pyridine (2) gave the desired chiral amides 7, which readily undergo catalytic hydrogenation to provide the piperidinylamides 8 as a mixture of diastereomers. Amidation of the piperidine nitrogen as previously described and removal of the *N*-Boc protecting group with anhydrous HCl in dioxane yielded the hydrochloride salts of the desired diastereomeric targets 1.

- (15) All new compounds reported herein exhibit ¹H NMR, IR, and elemental analysis (within 0.4% of the calculated value) in agreement with the assigned structures.
- (16) Abbreviations used are as follows: Boc, *tert*-butoxycarbonyl; DMAP, 4-(*N,N*-dimethylamino)pyridine; EDC, 1-[3-(dimethylamino)propyl]-3-carbodiimide hydrochloride; HOBT, 1-hydroxybenzotriazole; NMM, 4-methylmorpholine.
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Table I. Substituted 2-(Aminomethyl)piperidines and Their Inhibition of PKC versus PKA



compound	R ¹	R ₂	n	method ^a	PKC inhibition: IC ₅₀ , μM ^b	PKA inhibition: % at 100 μM
staurosporine					0.0027	IC ₅₀ = 27 nM
sphingosine					57 ± 5	IC ₅₀ = 98 μM
5			12	A	36 ± 2	46 ± 24
9	CH ₃	H	11	A	52 ± 6	0
10	CH ₂ CH(CH ₃) ₂	H	11	A	154 ^c	0
11	CH ₂ CH ₂ SCH ₃	H	11	A	159 ^c	0
12	CH ₂ Ph	H	11	A	agonist	
13	H	H	11	A	51 ± 4	0
14	CH ₂ OH	H	11	A	45 ± 5	0
15	CH ₂ CH ₂ C(O)NH ₂	H	11	A	72 ± 13	0
16	CH ₂ CH ₂ CO ₂ Bn	H	11	A	92 ± 12	0
17	CH ₂ CH ₂ CO ₂ H	H	11	A	inactive	0
18	CH ₂ Im	H	11	A	60 ± 6	0
19	(CH ₂) ₃ NH ₂	H	11	A	23 ± 7	0
20	(CH ₂) ₄ NH ₂	H	11	A	20 ± 3	0
21	(CH ₂) ₃ NHC(NH)NH ₂	H	11	A	72 ± 9	0
22	H	(CH ₂) ₄ NH ₂	11	A	34 ± 12	0
23	(CH ₂) ₄ NH ₂	H	5	B	inactive	0
24	(CH ₂) ₄ NH ₂	H	7	B	inactive	0
25	(CH ₂) ₄ NH ₂	H	9	B	299 ^c	0
26	(CH ₂) ₄ NH ₂	H	10	B	102 ^c	0
27	(CH ₂) ₄ NH ₂	H	12	B	48 ± 19	0
28	(CH ₂) ₄ NH ₂	H	13	B	38 ± 5	0
29	(CH ₂) ₄ NH ₂	H	14	B	45 ± 1	0
30	(CH ₂) ₄ NH ₂	H	15	B	46 ± 1	0
31	(CH ₂) ₄ NH ₂	H	16	B	36 ± 6	38 ± 3

^a All new compounds exhibit ¹H and IR data in agreement with assigned structures. Elemental analysis for final compounds were within ±0.4% of theoretical values for C, H, N. ^b Concentration-inhibition curves were determined at inhibitor concentrations up to 100 μM. ^c Represents an extrapolated IC₅₀ value.

The ability of synthesized compounds 5 and 9–31 to suppress PKC activity was tested *in vitro*. The inhibition of ³²P incorporation into histones by partially purified rat brain PKC in the presence of phorbol myristate acetate (PMA), Ca²⁺, and phosphatidylserine was assayed.¹⁹ In addition, specificity was assessed by measuring the inhibition of cAMP-dependent kinase (PKA).²⁰ These results are summarized in Table I.

Several structure-activity relationship observations regarding these analogues can be derived from evaluation of the biological data presented in Table I. First, activity appears to be modulated by the basicity of the amino acid residue as evidenced by comparison of compounds 18–21. The basicity requirement is optimized by inclusion of the amino acid ornithine or lysine to produce the most potent analogues 19 and 20, respectively. Incorporation of the weakly basic histidine or the strongly basic arginine residue results in diminished potency. The presence of an acidic functionality as in derivative 17 abolishes biological response. Esterification, however, produces the inhibitory

derivative 16. Omission of the amino acid fragment produces a nonselective antagonist as shown by compound 5. Second, a long-chain acyl group is essential for activity. Compounds 20 and 23–31 display a direct relationship between PKC inhibitory potency and chain length. Studies with these analogues reveal that alkyl chains shorter than 13 carbons greatly reduce potency and alkyl chains longer than 18 carbons compromise kinase selectivity. Third, the inclusion of bulky nonpolar substituents in the α-chain of the amino acid moiety results in significantly reduced potency as exemplified by analogues 9–12. In fact, the bulky phenylalanine derivative 12 exhibits agonist activity. Those analogues containing a polar amino acid residue tend to be more effective than their nonpolar counterparts as indicated by compounds 13–16. Finally, comparison of 20 and 22 suggests the effect of the stereogenic center of the amino acid fragment upon potency is modest.

To investigate the mechanism of action of these analogues, the inhibitory properties of 20 were studied in more detail. Under conditions which elicit maximal protein kinase activity with purified enzyme, 20 selectively inhibited PKC (IC₅₀ = 19 ± 2 μM)²¹ with no observable inhibition of PKA or calcium/calmodulin-dependent myosin light chain kinase at concentrations up to 100 μM.

Phorbol ester activation of PKC was studied at fixed concentrations of phosphatidylserine (100 μg/mL) and Ca²⁺ (400 μM) with varying concentrations of 20. Lineweaver-Burk and Dixon analyses of the data (Figure 1)

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(20) Protein kinase A activity was determined in the presence of 20 mM Tris buffer, 10 μM histone, 4 ng/mL of the catalytic subunit of protein kinase A, 100 μM dibutyl-cAMP, and 100 μM [³²P]ATP as described in Davis, P. D.; Hill, C. H.; Keech, E.; Lawton, G.; Nixon, J. S.; Sedgwick, A. D.; Wadsworth, J.; Westamcott, D.; Wilkinson, S. E. Potent Selective Inhibitors of Protein Kinase C. *FEBS Lett.* 1989, 259, 61–63.

(21) Reported value represents mean ± SEM of four individual determinations.

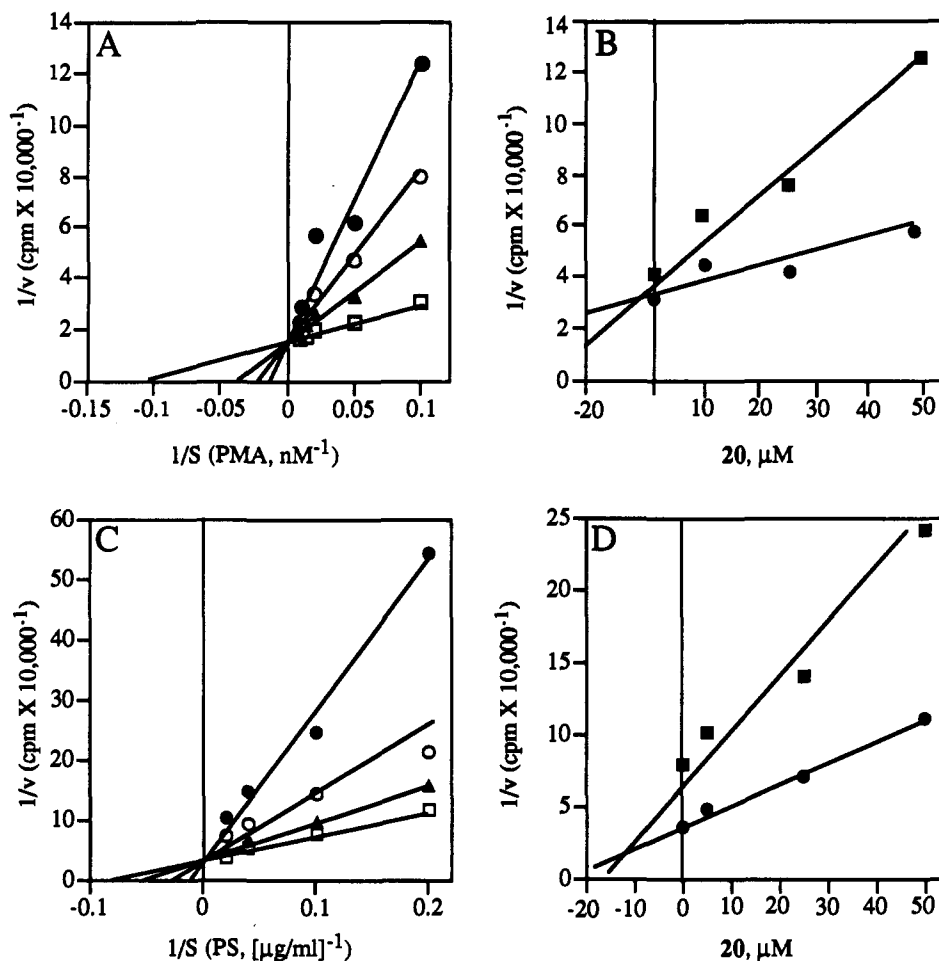


Figure 1. (A) Lineweaver-Burk plot of the effect of PMA on the inhibition of PKC- α by compound 20. PKC activity was determined in the presence of 400 μM "free calcium", 100 $\mu g/mL$ PS, 10 μM ATP, and the indicated concentrations of PMA. Compound 20 was present at 0 μM (■), 10 μM (▲), 25 μM (○), and 50 μM (●). (B) Dixon analysis (1/velocity vs concentration of 20) of the effect of PMA on the inhibition of PKC- α by compound 20. K_i may be determined from the point of intersection of the two lines.²² PMA concentrations used were 10 nM (■) and 50 nM (●). (C) Lineweaver-Burk plot of the effect of PS on the inhibition of PKC- α by 20. PKC activity was determined in the presence of 400 μM "free calcium", 1 μM ATP, and the indicated concentrations of PS. Compound 20 was present at 0 μM (■), 10 μM (▲), 25 μM (○), and 50 μM (●). (D) Dixon analysis (1/velocity vs concentration of 20) of the effect of PS on the inhibition of PKC- α by compound 20. PS concentrations were 5 $\mu g/mL$ (■) and 25 $\mu g/mL$ (●).

indicate this analogue is a competitive inhibitor with respect to phorbol ester activation ($K_i = 5 \pm 3 \mu M$).²¹ Additionally, phosphatidylserine activation of PKC was competitively inhibited by 20 ($K_i = 12 \pm 4 \mu M$, Figure 1).²¹ Compound 20 also inhibited [3H]phorbol dibutyrate binding ($IC_{50} = 23 \pm 4 \mu M$) to the enzyme preparation, further suggesting that inhibition of PKC activity by 20 may be the result of interactions with the enzyme's regulatory domain. Recent studies employing recombinant mutant PKC enzymes support evidence that compound 20 interacts with the regulatory domain of the enzyme.²³ Use of the mutant enzymes localized the effect of compound 20 to the N-terminal region of the C-1 domain known to possess the amino acid sequence that includes the pseudosubstrate region and part of the first of the cysteine-rich repeat sequences.

Synthesis of a novel series of selective PKC inhibitors should provide useful tools for elucidating the physiological role of PKC in various cellular processes. The development of these and related agents for therapeutic evaluation

in cellular and living systems is currently in progress.

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A Selective, Reversible, Competitive Inhibitor of Monoamine Oxidase A Containing No Nitrogen, with Negligible Potentiation of Tyramine-Induced Blood Pressure Rise

Monoamine oxidase (MAO) (EC 1.4.3.4, amine oxidase, flavin containing) consists of two different forms distinguishable by their substrate specificity¹ and their amino acid sequences.² Ratios of MAO A to MAO B of ca. 1 and

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