Mercaptoacyl Amino Acid Inhibitors of Atriopeptidase. 1. Structure-Activity Relationship Studies of Methionine and S-Alkylcysteine Derivatives

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A broad series of N-(3-mercaptoacyl) amino acid derivatives was evaluated for their ability to inhibit atriopeptidase (neutral endopeptidase, EC 3.4.24.11) in vitro and in vivo. Structural parameters studied were (i) the substituent on the 2-position of the 3-mercaptopropionyl moiety, (ii) the amino acid component, (iii) the S-terminal derivative, and (iv) the C-terminal derivative. Optimum activity was observed for derivatives of methionine and S-alkylcysteines. N-[3-Mercapto-2(S)-[(2-methylphenyl)methyl]-1-oxopropyl]-L-methionine was identified as a highly effective inhibitor of atriopeptidase meriting evaluation as a potential cardiovascular therapeutic agent.

Introduction

The atrial natriuretic factors (ANF) are peptide hormones first isolated from cardiac atrial tissue.¹ Related peptidic hormones, brain natriuretic peptide² and urodilatin,³ have been identified. Many pharmacological actions have been demonstrated for ANF, most prominently vasorelaxant and natriuretic activities.⁴ The half-life of these peptides in circulation is quite short,⁵ with proteolytic cleavage occurring in several positions,⁶ but predominantly between Cys-7 and Phe-8 in the disulfide ring (see Figure 1). The cleavage products possess little biological activity.⁷ Consequently, prolonging the action of ANF may result in useful pharmacological effects.

The enzymes participating in the degradation of ANF have been studied extensively.^{6,8} These studies led to the identification of neutral endopeptidase (NEP, EC 3.4.24.11, also termed atriopeptidase) as a critical enzyme in the degradation.⁹ This enzyme¹⁰ is a zinccontaining metallo-protease found in the kidney, brain (where it has been termed enkephalinase), and other tissues. It participates in the degradation of numerous peptidic substrates. NEP cleaves ANF *in vitro*,⁶ and inhibition of NEP *in vivo* has been shown to produce beneficial vasodilatory and natriuretic responses.^{11,12} Thus, development of a practical enzyme inhibitor would permit evaluation of the clinical utility of inhibition of NEP.¹³

Prior to our studies, inhibitors of NEP had been described by several laboratories whose efforts were directed at inhibition of "enkephalinase" to achieve an analgesic effect.¹⁴ These enzyme inhibitors featured a variety of functional groups capable of binding to the active-site zinc atom. Reported enkephalinase inhibitors¹⁵ contained a mercapto,^{14,16} carboxy,¹⁷ phosphonic acid,¹⁸ or hydroxamic acid¹⁹ group as the zinc ligand. We began by evaluating the relative utility of the first three ligand types, and some of our results with carboxyalkyl-dipeptides have been reported.²⁰ This report



Figure 1. Primary sites for metabolic cleavage of ANF.

describes the first phase of our examination of mercapto compounds.

Since the *in vitro* enzyme inhibition assay was wellestablished,²¹ a critical element in developing an effective agent was the establishment of a relevant *in vivo* assay for cardiovascular activity. Initially, activity of test compounds was determined in spontaneously hypertensive rats by measuring potentiation of the hypotensive response to exogenous ANF.²² An additional and more useful assay was found to be the hypotensive response produced by test compounds in DOCA-salt rats.²² This is a volume-dependent hypertensive model in which endogenous plasma ANF is elevated. With these tools in hand, we were equipped to evaluate a series of mercaptoacyl amino acids.

We began our survey of mercapto compounds with a series related to thiorphan (1).¹⁴ While exploring a variety of amino acid components, we found encouraging activity in mercaptoacyl derivatives of S-alkylcysteines and of methionine. Furthermore, we soon discovered that both the underivatized mercapto acids 2 (*ipso*-drugs) and protected forms 3 (prodrugs) were capable



of expressing good *in vivo* activity. Within both the cysteine and methionine classes, broad structureactivity relationships were established by variation of the following parameters: (*i*) the substituent on the 2-position of the 3-mercaptopropionyl moiety, (*ii*) the amino acid component, (*iii*) the S-terminal derivative, and (*iv*) the C-terminal derivative.

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Scheme 1



Δ neat or gl. AcOH; *ii*. H2,10% Pd/C, EtOAc; *iii*. (a) 40% aq. Me2NH, 37% aq. HCHO;
 (b) Δ; *iv*. AcSH, CH₂Cl₂.







Method A-3



/. Na, EtOH; //. n-BuLi, IsoPrzNH, HCHO; ///. AcSH, CH2Cl2; /v. TFA

Method A-4



i. isobutylene, t-BuOH, cat. H_2SO4: // LDA/THF, CH_2O: iii. MsCl, Et_3N, Et_2O; /v. AcSH, K_2CO3, DMF, 100°C; v. TFA/CH_2Cl2.

Chemistry

For variation of the mercaptoacyl moiety, we prepared the appropriate 2-substituted 3-(acetylthio)propionic acids (7) by Michael addition of thiolacetic acid to the respective 2-substituted acrylic acids (6) (see Scheme 1, methods A-1 and A-2). These were in turn available by classical malonic acid Mannich reaction of 5 or 9. The homologous 2-[(acetylthio)methyl]-4-phenylbutyric acid (14) was prepared as shown in Scheme 1, method A-3. For 1-[(acetylthio)methyl]cyclopentanecarboxylic acid (18) (Scheme 1, method A-4), the tert-butyl ester 15 of cyclopentanecarboxylic acid was prepared, and the ester-enolate was treated with formaldehyde to produce the alcohol 16a in high yield. Mesylation was followed by facile thioacetate displacement at the neopentyl center. Final ester deprotection furnished the desired acid 18.



i. SOCI₂, **R**'OH, reflux; ii.(a) NEt₃, EtO₂CCI, THF; (b) NH₄OH; iii. TFA, CH₂CI₂; iv. EDC, **R"R**"'NH, DMF; v. HCI, dioxane; vi. amine, 2 equiv., Δ .



D-Cysteir





The amino acid components were generally commercially available. S-(2-Phenylethyl)-L-cysteine was prepared by alkylation of L-cysteine with 2-phenylethyl bromide. Methyl and ethyl esters 20 of the amino acids were prepared by reaction of the amino acid 19 or N-Boc amino acid 21 with SOCl₂ in the corresponding alcohol (Scheme 2, methods B-1 and B-2). Amino acid amides 23 were prepared from the N-Boc amino acid 21 and appropriate amine by either mixed anhydride- or carbodiimide-mediated coupling (Scheme 2, methods B-3-1 and B-3-2), with subsequent removal of the BOC group. Amino acid amides 23 were also prepared by heating the amino acid ester 20 with excess amine (Scheme 2, method B-4).

The standard method for acylation of an amino acid ester 20 with the acetylthio acid 7 employed carbodiimide coupling and gave generally very good yields. Reaction of a racemic acetylthio acid 7 with a single enantiomer of the amino ester 20 furnished two diastereomeric amido esters 25 and 26 (Scheme 3, method Table 1. Physicochemical Data for N-[3-(Acetylthio)-2-substituted-1-oxopropyl]amino Acid Esters



compd	stereo	AA-OR	$method^a$	mp, °C⁰	yield (%)°	optical rotation $[\alpha]_D$, deg (c. solvent)	NMR, CH δ (solvent) ^d	TLC mobility	formula	anal.
L				P 1	Dhowyly	nothul P U				
95.0	g	S hopey I - Cys-OFt	C-1ef	83-85 NB-1	- nenyn 99	-735(05 M)	A 79 (C)	factor	CH. NO.S.	CHN
20a 96a	פ	S-bongyl L Cys-OEt	C-1/	79-74	16	-70.0(0.0, M)	4.72(C)	alowor	C ₂₇ H ₂₉ NO ₄ S ₂	C H N
20a 95h	n g	S-bongyl-D-Cys-OEt	C-1/	72-73	10	$\pm 15.6(0.65 \text{ M})$	4.70(C)	alower	C ₂₇ H ₂₉ NO ₄ S ₂	C H N
200 26h	P	S-bengyl-D-Cys-OEt	C-1/	84-85	20	+15.0(0.05, M) +75.9(0.55 M)	4.70 (C)	fastor	$C_{24}H_{29}NO_4S_2$	C H N
200	R R	S-(A-Mo-bonzyl)-I-Cys-OMe	C-1#	04 00 nil	19	-762(0.55, M)	4.00(C)	faster	$C_{24}H_{29}NO_4S_2$	CHN
200	P	S-(4-Me-benzyl)-L-Cys-OMe	C-18	oil	10	-49(0.2 (0.0, M))	A A 3 (D)	alowor	C ₂₄ H ₂₉ NO ₄ S ₂	C H N
200 25d	S	S-(4-MeO-benzyl)-L-Cys-OMe	C-1 ^b	oil	18	-665(0.2, M)	4.46(D)	fastor	CatHanNO-Sa	C H N
26d	R	S-(4-MeO-benzyl)-L-Cys-OMe	C_{-1}^{h}	oil	20	+30(01 M)	4.40 (D)	alower	CalHasNOrSet0 5HaO	C H N
25e	S	S-(3 4-Meg-henzyl)-L-Cys-OEt	C-1/	90-92	15	-711(0.55 M)	4.67 (C)	faster	CoeHaoNO4So	CHN
260	R	S-(3 4-Meg-benzyl)-L-Cys-OEt	Č-1/	oil	17	-84(0.75 M)	471 (C)	slower	CocHooNO So	CHN
25f	S	S-(2-nhenvlethyl)-I-Cvs-OEt	C-1*	63-64	28	-512(05 M)	4.67 (C)	faster	CarHanNO4Sa	CHN
26f	R	S-(2-phenylethyl)-L-Cvs-OEt	Č-1*	84-86	24	+5.3(0.5 M)	470 (C)	slower	Cor HanNO4Sa	CHN
27g	R.S	S-trityl-L-Cvs-OMe	$C-1a^{l,m}$	oil	46	+5.9(0.5, M)	4.39 (C)	510 11 01	CarHarNO4Sa	$\tilde{\mathbf{C}}$, \mathbf{H} , \mathbf{N}^n
27h	s'	S-methyl-L-Cys-OEt	Č-1aº	oil	56	-25.9(0.4, M)	1.00 (0)		C18H25NO4S2	C. H. N ^p
25i	\tilde{s}	S-ethyl-L-Cys-OEt	C-1	foam	8		4.67 (C)		- 10202	-,-,-
27k	R	S-ethyl-L-Cys-OEt	C-1a ^q	oil	14	+8.1(0.5, M)			$C_{19}H_{26}NO_4S_2$	C, H, N^r
25l	\boldsymbol{s}	S-tert-butyl-L-Cys-OMe	C-1 ^s	oil	47	-44.9(0.2, M)	4.75 (C)	faster	$C_{20}H_{29}NO_4S_2$	C. H. N
261	R	S-tert-butyl-L-Cys-OMe	C-1 ^s	oil	33	+8.3(0.14, M)	4.49 (C)	slower	$C_{20}H_{29}NO_4S_2$	C, H, N
27m	R,S	L-Met-OMe	$C-1a^{t,u}$	oil	55	-38.9 (0.55, M)	4.38, 4.29 (D)		$C_{18}H_{25}NO_4S_2$	C, H, N ^v
25n	S	L-Met-OMe ^w	$\mathbf{D}^{\mathbf{x},\mathbf{y}}$	78 - 80	82	-67.9(0.5, E)	4.60 (C)		$C_{23}H_{27}NO_4S_2$	C, H, N
26n	R	L-Met-OMe ^w	D ^y	77-79	89	+39.2(0.5, E)	4.62 (C)		$C_{23}H_{27}NO_4S_2$	C, H, N
25 0	\boldsymbol{s}	L-ethionine-OEt	C-1 ^z	foam	12	-60.6 (0.6, M)	4.55 (C)	faster	$C_{20}H_{29}NO_4S_2$	C, H, N ^{aa}
260	R	L-ethionine-OEt	C-1 ^z	foam	11	-0.3 (0.4, M)	4.57 (C)	slower	$C_{20}H_{29}NO_4S_2$	C, H, N
36	\boldsymbol{s}	S-(4-Me-benzyl)-L-Cys-OBu-t	Abb	oil	12	-82.1 (0.48, M)	4.56 (C)	faster	$C_{27}H_{36}NO_4S_2$	C, H, N
37	R	S-(4-Me-benzyl)-L-Cys-OBu-t	\mathbf{A}^{bb}	oil	14	-23.5(0.46, M)	4.60 (C)	slower	$\mathrm{C}_{27}\mathrm{H}_{35}\mathrm{NO}_4\mathrm{S}_2$	C, H, N
				R	a = Met	thyl, $\mathbf{R}_{\mathbf{B}'} = \mathbf{H}$				
25p	\boldsymbol{s}	S-benzyl-L-Cys-OEt	$\mathbf{D}^{cc,x}$	oil	39	-122.4 (0.3, M)	4.75 (C)		$C_{18}H_{25}NO_4S_2$	C, H, N
_				R.	= n-Pr	anv = H				
27q	RS	S-ethyl-L-Cys-OBu-t	\mathbf{A}^{dd}	oil	65	-39.4 (0.25, M)	4.69 (C)		$\mathrm{C_{17}H_{31}NO_4S_2}$	C, H, N ^{ee}
					$R_B + R_i$	$\mathbf{B}' = (\mathbf{CH}_2)_4$			a	a
27s	-	L-Met-OEt	C-1a ^y	40-42	76	-36.6(0.50, E)			$C_{16}H_{27}NO_4S_2$	C, H, N
				$R_B =$	2-Phen	ylethyl, $R_{B'} = H$				
25t	\boldsymbol{s}	L-Met-OEt	C-1 ^y	82 - 5	17	-53.8 (0.50, E)	4.72 (C)		$C_{20}H_{29}NO_4S_2$	C, H, N
26t	R	L-Met-OEt	C-1 ^y	100 - 102	2 10	-4.2(0.50, E)	4.70 (C)		$C_{20}H_{29}NO_4S_2$	C, H, N
				$R_{P} = 2-1$	Nanhth	vlmethvl $\mathbf{R}_{\mathbf{v}} = \mathbf{I}$	н			
25u	S	S-(4-Me-benzyl)-L-Cys-OEt	C-1#	foam	25	-61.0(0.5, M)	4.65 (C)	faster	C ₂₀ H ₂₂ NO ₄ S ₂	C. H. N ^{gg}
26u	\tilde{R}	S-(4-Me-benzyl)-L-Cys-OEt	Č-1#	foam	23	-20.7(0.4, M)	4.65 (C)	slower	C29H33NO4S2	C. H. N ^{hh}
		- (D - 1	 NT1-41				- 20002	-,,-
05	a	C (4 Mo howevel) & Cura OFt	C 1 <i>i</i> i	- π _B = 1	Naphu oo	-40.6(0.5 M)		fastor	C H NO S	СИМ
20V	S P	S-(4-Me-benzyl)-L-Cys-OEt	C-1"	foam	20	-40.0(0.0, M)	4.07 (C)	laster	C_{29} C_{33} NO_4 S_2	C, H, N
200	n	S-(4-Me-Delizy)-L-Cys-OLt	0-1-	Ioani	20	-36.6 (0.6, MI)	4.01 (C)	slower	0291133140462	0, 11, 1₩
]	$R_B = (4-C)$	loroph	enyl)methyl, R _{B′} :	= H	_		
25w	\boldsymbol{s}	S-benzyl-L-Cys-OEt	C-1 ^y	89-90	17			faster		
26w	R	S-benzyl-L-Cys-OEt	C-1 ^y	103-4	22	-17.9 (0.50, E)	4.69 (C)	slower	$C_{24}H_{28}ClNO_4S_2$	C, H, N
			H	$R_{\rm B} = (2 - M_{\rm C})^2$	ethylph	enyl)methyl, R _R	= H			
25y	\boldsymbol{s}	L-Met-OEt	C-1 ^y	91.5-94	<u>19</u>	-62.9 (1.0, M)	4.53 (C)	faster	$C_{20}H_{29}NO_4S_2$	C, H, N
26y	R	L-Met-OEt	C-1 ^y	foam	14	-11.7 (0.6, M)	4.53 (C)	slower	$C_{20}H_{29}NO_4S_2$	C, H, N
	<u> </u>									

^a Chromatographic column separations employ Baker silica gel, 60–200 mesh or flash silica gel, 32–64 mesh, with the indicated eluants. ^b White solid unless otherwise stated. ^c Theoretical yield of each diastereomer is 50%. ^d Chemical shift values are for amino acid α-proton. Chloroform-d (C); DMSO-d (D). ^e See the Experimental Section for this example of method C-1. ^f CH₂Cl₂-EtOAc (49:1). ^g EtOAc-hexane (4:21). ^h EtOAc-hexane (1:4). ⁱ C: calcd, 59.48; found, 59.04. ^j EtOAc-hexane (25:170); MeOH-hexane (25:170). ^k CH₂Cl₂; CH₂Cl₂-EtOAc (49:1). ^l See the Experimental Section for this example of C-1a. ^m CH₂Cl₂; CH₂Cl₂-EtOAc (1.5:98.5); (1:49). ⁿ C: calcd, 70.35; found, 69.73. ^o EtOAc-hexane (3:17). ^p C: calcd, 56.37; found, 55.75. ^q EtOAc-hexane (1:3). ^r C: calcd, 57.40; found, 56.99. ^s EtOAc-hexane (3:17). ^t See the Experimental Section for this example of method C-1a. ^u CH₂Cl₂; CH₂Cl₂-EtOAc (49:1). ^v C: calcd, 56.37; found, 55.75. ^q EtOAc-hexane (1:3). ^r C: calcd, 57.40; found, 56.69. ^s EtOAc-hexane (3:17). ^t See the Experimental Section for this example of method C-1a. ^u CH₂Cl₂; CH₂Cl₂-EtOAc (49:1). ^v C: calcd, 56.37; found, 55.47. ^w Benzoylthio in place of acetylthio. ^x See the Experimental Section for this example of method D. ^y Et₂O-hexane (1:1). ^z EtOAc-hexane (2.5:17.5). ^{aa} C: calcd, 58.37; found, 58.6b Prep 500 silica gel cartridges CH₂Cl₂; CH₂Cl₂-EtOAc (100:1; 200:3; 50:1). ^c CH₂Cl₂-EtOAc (99:1; 49:1; 9:1). ^{dd} CH₂Cl₂; CH₂Cl₂-EtOAc (49:1). ^e C: calcd, 54.08; found, 53.59. ^f Silica gel, EtOAc-hexane (1:9; 1.1:8.9). ^{gg} C: calcd, 66.51; found, 66.95. ^{hh} C: calcd, 66.51; found, 66.99. ⁱⁱ EtOAc-hexane (1:9). ^{ji} C: calcd, 66.51; found, 66.06.

C-1) (Table 1). In the majority of examples, these diastereomers could be separated by column chromatography. The ester and thioester functions of 25 and 26 were saponified²³ with 1 N NaOH in aqueous methanol to give the mercapto acids 28 and 29, respectively (Scheme 3, method C-2) (Table 2). Problems arose in the S-alkylcysteine series during preparation of the mercapto acids 28 (n = 1) from the precursor ethyl esters 25 (n = 1). In these cases, saponification with NaOH in aqueous ethanol resulted in racemization of

the cysteine chiral center to the extent of approximately 15%, yielding diastereomer **33** in addition to **28**. The amount of diastereomer could be determined readily in the NMR spectrum, particularly the signal of the amino acid α -proton. The increased susceptibility of S-alky-lated cysteine and O-alkylated serine derivatives to racemization is known,²⁴ and we observed this problem also with O-alkylserine examples. Once the ester function has been converted to a carboxylate, no further racemization occurs, since **28** is not converted to **33**

Table 2. Physicochemical Data for N-[3-Mercapto-2-substituted-1-oxopropyl]amino Acids



						optical				
compd	stereo	AA	method	mp. °C ^a	(%)	deg(c, solvent)	(solvent) ^b	mobility	formula	anal
					Dh -	wellwethel P	<u>ц</u>			
<u>00</u> -	a	Channel + Care	C 01	- il	= rne	$my_{1} methy_{1}, n_{B'} =$		6t		O II M
208	<u>о</u>	S-benzyl-L-Cys	0-2-	011	14	-2.1(1.0, M)	4.70(0)	laster	$C_{20}\Pi_{23}NO_{3}S_{2}U.1C\Pi_{2}CI_{2}$	C, H, N°
29 8	ĸ	S-benzyl-L-Cys	0-2	011	84	-46.7(1.1, M)	4.62 (C)	slower	$C_{20}H_{23}NO_3S_2U.2CH_2CI_2$	C, H, N
28b	S	S-benzyl-L-Cys	C-2	oil	95	+38.2(0.6, M)	4.61 (C)	slower	$C_{20}H_{23}NO_3S_2 \cdot 0.75H_2O$	C, H, N
29b	R	S-benzyl-D-Cys	C-2	oil	85	+13.5 (0.5, M)	4.68 (C)	faster	$C_{20}H_{23}NO_3S_20.5H_2O$	C, H, N
28c	\boldsymbol{s}	S-(4-Me-benzyl)-L-Cys	E-4	oil	86	-32.6 (0.4, M)	4.68 (C)	faster	$C_{21}H_{25}NO_3S_20.5H_2O$	C, H, №
			C-2		78					
29 c	R	S-(4-Me-benzyl)-L-Cys	E-4 ^s C-2	oil	84 58	-60.9 (0.5, M)	4.61 (C)	slower	$C_{21}H_{25}NO_3S_2H_2O$	C, H, N^h
28d	\boldsymbol{S}	S-(4-MeO-benzvl)-L-Cvs	C-2	foam	46	-19.3 (0.4, M)	4.69 (C)	faster	C21H25NO4S20.5H2O	C. H. N
29d	R	S-(4-MeO-benzyl)-L-Cys	C-2	oil	65	-44.2(0.1, M)	4.41 (D)	slower	CatHasNO4Sa0.5HaO	C. H. N
28e	S	S-(3 4-Mearbenzyl)-I-Cvs	Č-2	oil	99	-180(0.3 M)	4 70 (C)	faster	ConHarNOaSa	CHW
200	P	S-(3 A-Mes-benzyl)-L-Cys	Č.2	oil	60	-565(0.2 M)	4 61 (C)	slower	CasHarNO-S	C H N/
10C	ŝ	S (2 phopulothul) I Cus	C.2	oil	00	$\pm 4.9 (0.75 \text{ M})$	4.01(0)	footor	C. H. NO.S. 0 5H.O	C H N
401	מ	G (0 mb analatharl) r Cara	0-2	-1	94 05	$\pm 4.0 (0.70, M)$	4.71(0)	aster	$C_{21} H_{25} NO_3 S_2 0.5 H_2 O$	C, Π, N
201	n	S-(2-phenylethyl)-L-Cys	0.2	on c.	00	-39.7(0.75, M)	4.00(0)	slower	$C_{21}H_{25}NO_3S_2U.75H_2U$	C, H, N^{*}
30g	R,S	S-trityi-L-Cys	C-2a	Ioam	81	+10.5(0.5, M)	4.21 (C)		$C_{32}H_{31}NO_{3}S_{2}O.5H_{2}O$	C, H, N
SOP	R, S	S-methyl-L-Cys	C-2a	oil	83	-32.1(0.4, M)	4.74, 4.68 (C)	-	$C_{14}H_{19}NO_3S_2O.5H_2O$	C, H, N ^m
2 8j	S	S-ethyl-L-Cys	C-2	oil	89	-18.6(0.3, M)	4.73 (C)	faster	$C_{15}H_{21}NO_{3}S_{2}H_{2}O$	C, H, N
30k	S, R	S-ethyl-L-Cys	C-2a	oil	99	-21.9 (0.8, M)	4.73/4.70	-	$C_{15}H_{21}NO_3S_2$	C, H, N^n
281	\boldsymbol{s}	S-tert-butyl-L-Cys	C-2	oil	91	+0.4(0.56, M)	4.80 (C)	faster	$C_{17}H_{26}NO_3S_20.5H_2O$	C, H, №
291	R	S-tert-butyl-L-Cys	C-2	foam	97	-32.2(0.47, M)	4.74 (C)	slower	$C_{17}H_{26}NO_3S_20.75H_2O$	C, H, N^p
30m	R, S	L-Met	C-2a	132-135	82	-34.6(0.86, M)	4.34, 4.27 (D)		C15H21NO3S20.5H2O	C. H. N
28 n	Ś	L-Met	C-2	oil	99	+4.7(0.50, E)	4.64 (C)		C15H91NO3S90.125H9O	C. H. N ^q
29n	R	I-Met	E-3r	foam	92	-38.3 (0.50, E)	4.62 (C)		CieHaiNOsSer0 125HaO	CH N ^s
280	S	Lethionine	C -2	oil	79	-41.8(0.2 M)	4 65 (C)	faster	Cue Han NOs Sert 5HaO	CHN
200	R	Lethionine	Č-2	oil	00	-660(0.5, M)	4.60 (C)	slower	CH. NO.S.	CHN
200		Decimonine	0-2	UII	50	00.0 (0.0, 11)	4.00 (0)	310461	0161129110302	0, 11, 11
					$R_B =$	Methyl, $R_{B'} = H$				
2 8p	\boldsymbol{s}	S-Benzyl-L-Cys	C-2	oil	98	-50.1(0.7, M)	4.80 (C)	-	$C_{14}H_{19}NO_3S_2H_2O$	C, H, N
					D	Duranal D				
00-	חמ	Q Ettherit + Corre	T 0/	- 11	ng 1	a -rropyi, $n_{B'} - n$	400 404(0)			O II N
auq	<i>п</i> , 5	S-Ethyl-L-Cys	E-3.	on	92	-34.1(0.55, M)	4.80, 4.84 (C)		$C_{11}H_{21}NO_3S_2$	C, H, N
30r	R, S	L-Met	C-2a	011	89	-37.1(0.9, M)	4.77 (C)		$C_{11}H_{21}NO_3S_2O.7H_2O$	С, Н, N
					R.	$+ \mathbf{R}_{\mathbf{R}'} = (\mathbf{C}\mathbf{H}_{\mathbf{a}})_{\mathbf{a}}$				
30e		1-Met	C-2	85-87	87	-15.7(0.50 E)	-	-	CueHerNO.S.	CHN
005		B-MICO	01	00 01	0,	10.1 (0.00, 1)			012112110382	0, 11, 11
				R	3 = 2-P	henylethyl, $R_{B'} =$	= H			
28t	\boldsymbol{s}	L-Met	C-2	oil	96	-12.5(0.5, E)	4.80 (C)		$C_{15}H_{23}NO_3S_2$	C, H, N^u
29t	R	L-Met	C-2	oil	94	-22.9(0.5, E)	4.69 (C)		$C_{15}H_{23}NO_3S_20.75H_2O$	C, H, N
				ъ	0 M	- Labordon addred D	- 11			
~~	~			$R_B =$	= Z-INA]	phinyimetnyi, R _B		. .	a	a
28u	S	S-(4-Me-benzyl)-L-Cys	C-2	Ioam	91	+9.9(0.4, M)	4.65 (C)	faster	$C_{25}H_{27}NO_{3}S_{2}$	С, Н, N
29u	R	S-(4-Me-benzyl)-L-Cys	C-2	foam	62	-50.1(0.25, M)	4.55 (C)	slower	$C_{25}H_{27}NO_3S_2$	C, H, N ^v
				Rn =	= 1.Nai	hthvlmethvl R.	/ = H			
2817	ç	S-(A-Me-bonzyl)-I-Cys	C-2	foom	- 1-11aj 61	$\pm 183(0.7 \text{ M})$	A B A (C)	fostor	C.,H.,NO.S.,0 25H.O	СИМ
20V	5	S (4 Mo hopsyl) L Cys	C 2	foam	50	+10.0(0.7, M)	4.54(0)	aster	C H NO S	C H N
27V	л	S-(4-Me-DelizyI)-L-Cys	C-2	Ioam	99	-102(0.4, M)	4.04 (C)	slower	C26H27INO3B2	С, п, м
				$R_{B} = (4$	-Chlor	ophenyl)methvl.	$R_{B'} = H$			
28w	\boldsymbol{S}	S-(4-Me-benzvl)-L-Cvs	C-2	film	91	-3.0 (0.50, E)	4.71 (C)	faster	CanHaaClNOaSa	C. H. N
29w	\tilde{R}	S-(4-Me-benzyl)-L-Cvs	Č-2	film	99	-48.7 (0.50 E)	4.63 (C)	slower	ConHoo CINO So	C. H Nw
20.11			~ -			1011 (0.00, 11)	1.00 (0)	510 001	-2022220110302	0, 11, 11
				$R_{B} = [1, 1]$	l']-Bipl	henyl]-4-ylmethy	$l, R_{B'} = H$			
28x	\boldsymbol{S}	S-(4-Me-benzyl)-L-Cys	E-4	foam ^x	71	-2.2(0.6, M)	4.49 (D)	faster	$C_{27}H_{29}NO_3S_2$	C, H, N
29x	R	S-(4-Me-benzyl)-L-Cys	E-4	foam [*]	27	-62.2(0.3, M)	4.45 (D)	slower	$C_{27}H_{29}NO_3S_2$	C, H, N
				D (9	Mahl		D - U		. –	
00	a	• M-4	~ •	$R_{\rm B} = (2$	-wetny	yipnenyi)metnyi,	$\pi_{B'} = H$			0.17.37
28y	5	T-1A161	U-2	114 - 120	93	+8.0(0.6, M)			$U_{15}H_{23}NU_{3}S_{2}$	U, H, N

^a White solid unless otherwise stated. ^b Chemical shift values are for amino acid α -proton. Chloroform-d (C); DMSO-d (D). ^c TLC in (CH₂Cl₂-MeOH-AcOH) system, typically 97.5:2.5:0.25. ^d See the Experimental Section for this example of method C-2. ^e N: calcd, 3.51; found, 3.10. ^f C: calcd, 61.14; found, 61.86. ^g H: calcd, 6.46; found, 5.69. ^h See the Experimental Section for this example of method E-4. ⁱ C: calcd, 63.28; found, 62.82. ^j C: calcd, 63.28; found, 62.55. ^k H: calcd, 6.40; found, 5.84. ^l See the Experimental Section for an example of method C-2a. ^m C: calcd, 52.15; found, 51.65. H: calcd, 6.26; found, 5.54. ⁿ C: calcd, 55.02; found, 54.20. ^o C: calcd, 56.01; found, 56.45. ^p H: calcd, 7.24; found, 5.94. ^q H: calcd, 6.53; found, 5.87. ^r See the Experimental Section for this example of method E-3. ^s N: calcd, 4.25; found, 4.71. ^t See the Experimental Section for this example of method E-3. ^s C: calcd, 56.66; found, 55.38. ^s Flash silica gel, CH₂Cl₂-CH₃OH-glacial AcOH (97.5:2.5:0.25).

under the reaction conditions. Changes in reaction conditions did not eliminate the racemization problem, and alternative acid hydrolysis was not a clean process. In general, the diastereomeric contaminant in the product acid **28** was considered tolerable for biological assay. Methionine derivatives were not susceptible to this racemization.

Two approaches were utilized in the preparation of individual diastereomers of the mercaptoacyl amino acids (e.g., **28n** and **29n**). One method (Scheme 4a, method D) employed the resolved S-benzoylthio acid 31 reported by Bindra.²⁵ For methionine, the methyl ester 20n was acylated with either enantiomer of acid 31, each furnishing a single amide diastereomer (25n and 26n) of known configuration at the mercaptopropionyl chiral center. These were hydrolyzed respectively to the acids 28n and 29n. For S-alkylcysteine derivatives, because of the susceptibility to racemization during ester saponification, alternative approaches were employed. In one method (Scheme 4b, method E-1), the

Scheme 4



(acetylthio)propionic acid **7f** was converted into the corresponding chloride **35**, and this was used to acylate the S-alkylcysteine acid **19** rather than an ester. Following separation of diastereomers **38a** and **39a**, removal of the S-acetyl provided enantiomerically pure acids **28c** and **29c** (Table 3). In a second method

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(Scheme 4c, method E-2), the amino acid *tert*-butyl ester **20p** was converted to amides **36** and **37**, which were de-esterified to **38a** and **39a** and then deacetylated to **28c** and **29c**, respectively. This approach was limited, since in some cases the diastereomeric *tert*-butyl esters could not be separated by chromatography.

S-Benzoylthio and S-acetylthio amides were prepared via carbodiimide coupling with the carboxylic acid (31 or 7f) or by use of the acid chloride (32 or 35) (Schemes 5 and 6) (Table 4). Hydrolysis of the S-protected amides gave the mercapto amides (Table 7).

Assignment of chirality at the mercaptopropionyl center was required in each case where diastereomers were separated. A rigorous method employed correlation with materials derived from the benzoylthio acid enantiomers 31-R or 31-S, as shown in Scheme 5. Thus, L-methioninamide 23g was converted by 31-R or 31-S to mercaptans 47a and 46a, respectively. Mercaptan 46a was identical to the mercaptan obtained from the less polar diastereomer (48a) of the S-acetyl compounds. This correlation was applied similarly with L-methionine, as described in Scheme 7. A mixture of S-acetyl diastereomeric acids (38b and 39b) was prepared from 35 and L-methionine, and the more polar diastereomer 39b was hydrolyzed to 29n. Mercapto acid 29n was identical with the material prepared from S-benzoyl acid **31-R.** This rigorous correlation method was laborious and not applicable to mercaptoacids other than 2-(mercaptomethyl)-3-phenylpropionic acid.

A more general, although less rigorous, method for assigning chirality employed the assignments determined by the rigorous method above and established correlations between stereochemistry and (a) TLC relative polarity, (b) optical rotation, and (c) NMR chemical shift values. These correlations, illustrated in Figure 2, proved quite reliable. The first correlation observed was between the relative TLC polarity of (acetylthio)acyl L-amino acid ethyl esters (53/54) and specific rotations. The (S)-diastereomers 53 were less polar and possessed strongly negative rotations, while the (R)diastereomers 54 were more polar and possessed small negative rotations. In general, the difference in specific rotation ranged between 50° and 70°. Hydrolysis to the corresponding mercapto acids gave reversal of these relative rotations, as seen in Table 2. In all pairs of diastereomers separated, the TLC mobilities and relative rotations followed this pattern.

For derivatives of L-methionine, stereochemistry was correlated with NMR chemical shift values. The products with (S)-configuration (on the basis of polarity and rotation) at the mercaptopropionyl center showed the SCH₃ singlet at δ 2.10, while those with (R)-configuration showed this singlet at δ 2.00. Ultimately, a further verification of stereochemical assignments was secured via the X-ray structure of Sch 42495²² (**25y**), which agreed with the initial assignment based on the above arguments.

Pharmacology

The *in vitro* assay for NEP inhibition is wellestablished,²¹ as is the assay for ACE inhibition.²⁶ Initially, potentiation of the hypotensive response to exogenous ANF was employed as the key *in vivo* assay. In this assay, spontaneously hypertensive rats (SHR) are challenged with intravenous ANF, and the acute





compd	stereo (*)	AA-OH	method ^a	mp, °C ^b	yield (%)	optical rotation $[\alpha]_D$, deg (c, solvent)	NMR, CH ð (solvent) ^c	TLC mobility	formula	anal.
					$R_B = F$	henvlmethyl				
38 a	S	S-(4-Me-benzyl)-L-Cys-OH	E-1 ^{d,∉} E-2 ^g	oil	6 56	-23.0 (1.0, M)	4.65 (C)	faster	$C_{23}H_{27}NO_4S_20.2CH_2Cl_2$	C, H, N ^f
39 a	R	S-(4-Me-benzyl)-L-Cys-OH	E-1 ^h E-2	oil	10 59	-1.3 (1.0, M)	4.58 (C)	slower	$C_{23}H_{27}NO_4S_20.25$ AcOH	C, H, N
38b	\boldsymbol{S}	L-Met-OH	Ē-1	95-98	21	-36.6(0.5, E)	4.58 (C)	-	$\mathrm{C_{17}H_{23}NO_4S_2}$	C, H, N
				R в =	[1.1'-B	iphenvl]-4-vlmet	hvl			
38c	\boldsymbol{s}	S-(4-Me-benzvl)-L-Cvs-OH	$E-1^i$	123 - 125	17	-45.5 (0.6, M)	4.59 (C)	faster	C₂9H31NO₄S₂	C. H. N
39c	R	S-(4-Me-benzyl)-L-Cys-OH	$E-1^i$	131-135	17	-7.1(0.3, M)	4.61 (C)	slower	$C_{29}H_{31}NO_4S_2$	C, H, N
					Rв	= n-Propyl				
40d	R,S	S-ethyl-L-Cys-OH	E-2	oil	76	-26.9 (0.5, M)	4.79 (C)		C13H23NO4S20.15 CF3CO2H	C, H, N

^a Chromatographic column separations employ Baker silica gel, 60–200 mesh or flash silica gel, 32–64 mesh, with the indicated eluants. ^b White solid unless otherwise stated. ^c Chemical shift values are for amino acid α -proton. Chloroform-d (C); DMSO-d (D). ^d See the Experimental Section for this example of method E-1. ^e Silica gel, CH₂Cl₂-CH₃OH-glacial AcOH (100:1:0.1), faster moving component. ^f H: calcd, 5.97; found, 5.55. ^g See the Experimental Section for this example of method E-2. ^h CH₂Cl₂-CH₃OH-glacial AcOH (100:1:0.1). ⁱ CH₂Cl₂-CH₃OH-glacial AcOH (97.5:2.5:0.25).



reduction in blood pressure is measured. The rats are treated with the test compound, the challenge is repeated, and the increase in hypotensive response is determined. This assay is directly linked to the desired mechanism of action. While this assay was guite useful. an alternative assay employing DOCA-salt rats proved even more relevant. This is a volume-dependent model of hypertension²⁷ (in contrast to the SHR, which is a multifaceted genetic model) with elevated levels of plasma ANF,²⁸ and we have found NEP inhibitors highly effective in reducing blood pressure in this model. This assay can be highly variable, but regular use under the conditions described in the Experimental Section provided reliable results without requiring large groups of animals. The DOCA-rat assay was employed as the primary indicator of compounds with good in vivo

Scheme 6



activity. As further evidence of the mechanism of action, in vivo inhibition of ANF degradation has been demonstrated³⁴ for compounds active in the DOCA-rat assay.

Results and Discussion

The S-benzyl-L-cysteine derivative **28a** ((S,R)-stereochemistry, as shown), the first member of the series,²⁹ showed potent inhibitory activity against NEP (IC₅₀ = 9 nM) and ACE (IC₅₀ = 24 nM) (Table 5). Good *in vivo* activity was observed in both the ANF potentiation and DOCA-rat assays at 30 mg/kg sc. The (R,R)-diastereomer **29a** (also derived from L-cysteine) showed a similar profile, but with greatly diminished ACE inhibitory potency (IC₅₀ > 1000 nM). The known preference³⁰ of ACE for (S)-chirality at the mercaptoacyl center is a further correlation for our stereochemical assignments. The stereochemical tolerance at the 2-position of the

 Table 4. Physicochemical Data for N-[2-[(Acetylthio)methyl]-; 2-[(Benzylthio)methyl]-; 2-(Mercaptomethyl)-3-phenyl-1-oxopropyl]amino Acid Amides



compd	Q	stereo (*)	AA-NR"R"	method ^a	mp, °C ^b	yield (%)	optical rotation $[\alpha]_D$, deg (c, solvent)	$\begin{array}{c} \text{NMR,} \\ \text{CH } \delta \\ (\text{solvent})^c \end{array}$	TLC mobility	formula	anal.
46a	н	S	L-Met-NH ₂	F-3 ^d	133-137	79	-8.4(0.5, E)	4.60 (C)		C ₁₅ H ₂₂ N ₂ O ₂ S ₂ 0.25H ₂ O	C, H, N
47a	н	R	L-Met-NH ₂	F-3	135-136	83	-66.3 (1.0, C)	4.19 (D)		$C_{22}H_{26}N_2O_3S_2$	C, H, N
48 a	Ac	\boldsymbol{S}	L-Met-NH ₂	He	149 - 151	17	-87.5 (0.5, C)	4.53 (C)		$C_{17}H_{24}N_2O_3S_2$	C, H, N
49 c	Ac	R	L-Met-NH ₂	н	119-121	15	+5.0(0.5, C)	4.48 (C)		$C_{17}H_{24}N_2O_3S_2$	C, H, N
50c	Ac	R, S	$L-Met-N(CH_2)_4$	\mathbf{H}^{f}	63-5	30	-49.4(0.5, E)			$C_{21}H_{30}N_2O_3S_2$	C, H, N
50d	Ac	R, S	$L-Met-N(CH_2)_5$	Hg	red oil	59				$C_{22}H_{32}N_2O_3S_2H_2O_3A_2O_3S_2H_2O_3S_2H_2O_3S_2H_2O_3S_2H_2O_3S_2H$	C, H, N
43 a	Βz	S	$L-Met-NH_2$	$F-1^h$	176 - 180	36	-97.6(0.5, C)			$C_{22}H_{26}N_2O_3S_2$	C, H, N
4 4a	Βz	R	$L-Met-NH_2$	F-2 ^j	145-149	86	+32.9(0.5, C)	4.21 (D)		$C_{22}H_{26}N_2O_3S_2$	C, H, N
50e	Ac	R, S	L-Met-NHCH ₂ CH ₂ OH	\mathbf{H}^{j}	115 - 118	61	-31.1(0.5, E)	4.42; 4.48 (C)		$C_{19}H_{28}N_2O_4S_2$	C, H, N
50f	Ac	R, S	$L-Met-N(CH_2)_2O(CH_2)_2$	\mathbf{H}^{k}	oil	75	-30.0(0.5, E)			$C_{21}H_{30}N_2O_4S_20.25H_2O$	C, H, N
50g	Ac	R, S	L-Met-NHCH ₂ CO ₂ Et	\mathbf{H}^{l}	100 - 103	61	-35.6(0.5, E)	4.50; 4.58 (C)		$C_{21}H_{30}N_2O_5S_2 \cdot 0.25H_2O$	C, H, N
$48\bar{h}$	Ac	S	S-(4-Me-benzyl)-L-Cys-NH ₂	\mathbf{G}^{ρ}	foam	7.6	-38.2(0.6, M)	4.44 (D)	faster	$C_{23}H_{28}N_2O_3S_2$	C, H, N
49h	Ac	R	S-(4-Me-benzyl)-L-Cys-NH ₂	$\mathbf{G}^{m.n}$	foam	25	-1.6(0.5, M)	4.45 (D)	slower	$C_{23}H_{28}N_2O_3S_2$	C, H, N
51h	н	\boldsymbol{S}	S-(4-Me-benzyl)-L-Cys-NH ₂	$\mathbf{J}^{p,q}$	130 - 132	75	-4.2(0.14, M)	4.53 (D)	faster	$C_{21}H_{26}N_2O_2S_2$	C, H, N
52h	н	R	$S-(4-Me-benzyl)-L-Cys-NH_2$	\mathbf{J}^q	foam	65	-29.0 (0.33, M)	4.44 (D)	slower	$C_{21}H_{26}N_2O_2S_20.5H_2O$	C, H, N

^a Chromatographic column separations employ Baker silica gel, 60–200 mesh or flash silica gel, 32–64 mesh, with the indicated eluants. ^b White solid unless otherwise stated. ^c Chemical shift values are for amino acid α -proton. Chloroform-*d* (C); DMSO-*d* (D). ^d See the Experimental Section for this example of method H. ^f Silica gel column, Et₂O. ^g Silica gel column, MeOH-CH₂Cl₂, 1:49. ^h See the Experimental Section for this example of method F-1. ⁱ See the Experimental Section for this example of method F-2. ^j MeOH-CH₂Cl₂, 1:19. ^h MeOH-CH₂Cl₂, 3:97. ^l Et₂O-hexane, 1:1. ^m See the Experimental Section for this example of method G. ⁿ Silica gel (Prep 500, 2 cartridges), CH₂Cl₂-EtOAc (4:1) faster moving component. ^o (Prep 500, 2 cartridges), CH₂Cl₂-EtOAc (4:1). ^p See the Experimental Section for this example of method G. (97.5:2.5:0.25) and (97.5:2.5:0.25).

Scheme 7



mercaptopropionyl chain for NEP inhibition is a general feature of the series reported here (with several apparent exceptions). Such a relationship has been reported



Figure 2. Correlation of stereochemistry with TLC mobility, relative optical rotation, and chemical shift.

by others both for inhibitors of NEP^{16b,31} and inhibitors of thermolysin.³² For S-benzylcysteine, the L-amino acid confers superior activity relative to the D-amino acid, as seen by the activity of **28b** and **29b**.³³ In general, we have observed greatly diminished NEP and ACE inhibitory activity for mercaptoacyl derivatives of the D-amino acids.³⁴



In the S-alkylcysteine series (Table 5), the influence of the substituent on the sulfur atom was examined. For

Table 5. In Vitro and in Vivo Biological Activity of N-[3-Mercapto-2-substituted-1-oxopropy]amino Acids



			IC50	, nM	ΔBP, mmHg ^a		
compd	stereo (*)	AA	NEP	ACE	DOCA	ANF	
-		$R_B = Phenyly$	methyl Re' = H				
2 8a	S	S-benzyl-L-Cys	9 9	24	17 (10 sc) 30 (30 sc)	35 (30 sc)	
29 a	R	S-benzyl-L-Cys	32	>1000	21 (10 sc) 55 (30 sc)	45 (90 sc)	
28b	\boldsymbol{S}	S-benzyl-D-Cys	19	115		11 (30 sc)	
29b	R	S-benzyl-D-Cys	14	1,250	18 (30 po)	19 (30 po)	
28c	\boldsymbol{S}	S-(4-Me-benzyl)-L-Cys	11	19	17 (30 sc)	48 (30 sc)	
29 c	R	S-(4-Me-benzyl)-L-Cys	20	1,200	33 (30 sc) 28 (30 po)	48 (30 sc)	
28d	\underline{s}	S-(4-MeO-benzyl)-L-Cys	2	95	28 (30 po)	16(30 sc)	
29d	R	S-(4-MeO-benzyl)-L-Cys	17	290	0 (30 po)	23 (30 sc)	
28e	S	S-(3,4-Me ₂ -benzyl)-L-Cys	370	460		23 (30 sc)	
29e	R	S-(3,4-Me ₂ -benzyl)-L-Cys	541	>1000	a (aa	0(30 sc)	
28f	S	S-(2-phenylethyl)-L-Cys	110	80	0 (30 po)	37 (30 sc)	
29f	R	S-(2-phenylethyl)-L-Cys	133	>1000	38 (30 po)	23 (30 sc)	
30g	R, S	S-trityl-L-Cys	274	≫1000	a (aa)	0(30 sc)	
30h	S	S-Me-L-Cys	1.5	77	0 (30 po)	30 (30 sc)	
28]	8	S-etnyl-L-Cys	Z.4	28	21 (0.1 sc) 22 (1 sc) 61 (10 po)	19 (30 sc)	
30k	R, S	S-ethyl-L-Cys	6.5	45		25 (30 sc)	
281	S	S-t-Bu-L-Cvs	49	170	17 (30 po)	25 (30 sc)	
29 1	R	S-t-Bu-L-Cys	231	640	_ 、	0 (30 sc)	
3 0m	<i>R</i> , <i>S</i>	L-Met	4	120	42 (30 sc) 18 (30 po)	25 (30 sc)	
2 8n	\boldsymbol{S}	L-Met	6.5	35	31 (10 po)	23 (30 sc)	
29n	R	L-Met	3.5	>1000	30 (10 po)	50 (30 sc)	
280	\underline{s}	L-ethionine	5	215		20 (30 sc)	
29 0	R	L-ethionine	31	>1000	0 (30 po)	30 (30 sc)	
		$R_B = Met$	thyl, $\mathbf{R}\mathbf{B'} = \mathbf{H}$				
28p	\boldsymbol{S}	S-benzyl-L-Cys	46	58	0 (30 sc)	19 (30 sc)	
		RB = n - Pr	opyl, $\mathbf{R}\mathbf{B}' = \mathbf{H}$				
3 0q	R, S	S-ethyl-L-Cys	>300	1000			
30r	R, S	L-Met	2.5	>1000			
		$\mathbf{R}\mathbf{B}=\mathbf{R}$	$\mathbf{B'} = (\mathbf{CH}_2)_4$				
3 0s	-	L-Met	3.5	700	24 (30 sc)		
		$\mathbf{P}_{\mathbf{D}} = \mathbf{D}_{\mathbf{b} \circ \mathbf{m}}$	othul Pn' - U		•		
98+	q	I.Mot	e_{LII} , $\mathbf{RB} - \mathbf{R}$	150	18(10m)		
29t	R	I-Met	> 300	>1000	6(3 po)		
200	10		1 1 1 1 1-(. 1000	0(0 p0)		
	~	RB = 2-Naphth	yImethyl, $RB' =$	H		a (aa)	
28u	S	S-(4-Me-benzyl)-L-Cys	>300	460		0 (30 sc)	
29u	R	S-(4-Me-benzyl)-L-Cys		3100		0 (30 sc)	
		RB = 1-Naphth	ylmethyl, RB' =	Н			
28v	S	S-(4-Me-benzyl)-L-Cys	2	>1000	47 (1 sc) 26 (3 sc) 8 (10 po) 33 (30 po)	0 (30 sc)	
29v	R	S-(4-Me-benzyl)-L-Cys	39	90	00 (00 p0)	0 (30 sc)	
201	11	RB = (4-Chloroph)	enyl)methyl, RB'	' = H		0 (00 BC)	
28w	S	S-benzyl-L-Cys	3	44	20 (30 po)	19 (30 sc)	
29w	R	S-benzyl-L-Cys RB = $\begin{bmatrix} 1 & 1' \end{bmatrix}$ -Binheny	147 /l-4-vlmethyl R	>1000 B' = H		0 (30 sc)	
28x	\boldsymbol{s}	S-(4-Me-benzvl)-L-Cvs	180	180	19 (30 po)	0(30 sc)	
29x	R	S-(4-Me-benzyl)-L-Cys	290	>1000	(-• ₽•)	0 (30 sc)	
		$R_B = (2-Methvlph)$	enyl)methyl. RB	′ = H			
28y	S	L-Met	7	100	24 (1 po)	15 (30 po)	
	~				43 (3 sc)	60 / 6 • • •	
25y	8	L-Met-OEt	>300	250	27 (1 po) 48 (3 po) 70 (10 po)	36 (30 sc)	

^a The numbers indicating the change in blood pressure in DOCA rats or ANF potentiation represent average values determined in four or more animals. ^b S-Acetyl.

the compounds with arylalkyl substituents, the 4-methylbenzyl (28c, 29c) and 4-methoxybenzyl (28d, 29d) (derivatives showed good activity, while the 3,4-dimethylbenzyl (**28e**, **29e**), 2-phenylethyl (**28f**, **29f**), and triphenylmethyl (**30g**) derivatives showed reduced potency. For simple alkyl substituents,³⁵ good activity was

Mercaptoacyl Amino Acid Inhibitors of Atriopeptidase

observed for the methyl (**30h**) and ethyl (**28j**) derivatives and somewhat reduced activity for the *tert*-butyl derivatives (**28l**, **29l**.). These results serve to establish steric limits for binding in this region of the enzyme. In the methionine series, both methionine (**28n**, **29n**) and ethionine (**28o**, **29o**) derivatives showed good activity, with particularly good *in vivo* activity for methionine. It is known that NEP shows a moderately broad tolerance for the side-chain on the amino acid moiety of the inhibitor.¹⁰ The *in vivo* activity of this set of compounds parallels most closely the *in vitro* NEP inhibitory activity.

The substituent at the 2-position of the mercaptopropionyl moiety is an important determinant for in vitro activity (Table 5). A broad range of substituents was examined. Activity against both NEP and ACE was substantial for the captopril-like derivative with a methyl group (28p). With an *n*-propyl substituent (30r), NEP inhibitory activity was very good, while ACE inhibitory activity was weak. A similar profile was observed with the α, α -cyclic derivative **30s**. A 2-phenvlethyl substituent (28t, 29t) led to reduced activity against both enzymes. A2-naphthylmethyl substituent (28u, 29u) produced less activity than the isomeric 1-naphthylmethyl (28v, 29v). Substituted benzyl groups were also examined. A 4-chloro substituent (28w, 29w) was well tolerated, as was a 2-methyl substituent (28y), but a 4-phenyl substituent (28x, 29x) less so. We interpret these in vitro results as demonstrating a broad structural tolerance in this region of the molecule. This is consistent with the lack of strong S/R configurational preference at the chiral center and the tolerance for disubstitution. The substituents providing the poorest NEP inhibitory activity all show significant linear extension from the chiral center and indicate a distant spatial intolerance in this region.

Of the compounds in Table 5, six different substituents at the 2-position of the mercaptopropionyl moiety provided derivatives with IC_{50} below 10 nM for inhibition of NEP. In vivo activity varies within this group. The greatest activity, especially upon oral administration, was seen with the 2-methylbenzyl derivative (**28y**). Oral bioavailability thus appears to be structure-dependent. The primary objective of our studies was to identify an NEP inhibitor with sufficient *in vivo* activity to permit evaluation of the therapeutic potential of this mechanism. Compound **28y** showed sufficient activity for evaluation in a battery of mechanism-related assays, and detailed SAR was compiled on a set of close analogs of this structure. These studies are described in the succeeding report.³⁴

The C-terminal and S-terminal groups of these compounds are capable of influencing both *in vitro* and *in vivo* activity (Table 6). Since a free mercapto group is required for inhibition of NEP, an S-terminal group can influence the release of the active species. S-Acetyl mercaptoacyl amino acids based on methionine (**38b**) and S-alkylcysteine (**38a**, **39a**) showed good *in vivo* activity, as did two S-acylmethionine esters (**26n**, **27s**). Thus, both thioester and ester functions are effectively bioactivated.³⁶ These doubly protected prodrugs serve as excellent precursors to the free mercaptoacids, and many of these are very effective upon oral administration.

Since NEP is an endopeptidase, in contrast to the

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Table 6. In Vivo Biological Activity of

N-[3-(Acetylthio)-2-(phenylmethyl)-1-oxopropyl]amino Acids/ Esters



^a The numbers indicating the change in blood pressure in DOCA rats or ANF potentiation represent average values determined in four or more animals. ^b Benzoyl in place of acetyl.

carboxydipeptidase ACE, substrates (or inhibitors) of the enzyme do not require a C-terminal carboxylic acid.¹⁰ Thus, the mercaptoacyl methioninamides (**46a**, **47a**) are moderately active *in vitro* vs NEP, but inactive vs ACE (Table 7). As a group, the S-acetylated and S-benzoylated amides (particularly **48a**, **48h**, **49a**, **50c**, **50d**, **50f**) showed generally good *in vivo* activity, although not superior to that of the mercapto acids. The free mercapto acids and S-acetyl acid alkyl esters were the focus of subsequent studies, yielding many compounds with excellent *in vivo* activity. Ultimately, Sch 42495 (**25y**) was chosen for clinical evaluation.³⁴

Summary

Starting with a hypothesis on the potential utility of inhibitors of neutral endopeptidase in the treatment of cardiovascular diseases, we examined a broad structural range of inhibitors. *In vivo* models relevant to this mechanism of action were employed. In the mercapto class, we developed SAR covering several structural parameters. The results presented here allowed us to focus on specific mercaptoacyl derivatives of methionine, leading ultimately to a potential clinical candidate.

Experimental Section

Melting points were determined with a Thomas-Hoover melting point apparatus and are uncorrected. The ¹H NMR spectra were obtained on Varian instruments FT80 H (80 MHz), CFT-20 (80 MHz), and EM-390 (90 MHz) and are reported as ppm downfield from Me₄Si with multiplicity, number of protons, and coupling constants in Hertz indicated parenthetically. NMR solvents: C, chloroform-d; D, DMSOd. The EI mass spectra were determined with a Finnigan MAT CH-5 spectrometer. The rotations were determined with a Rudolf Autopol at ambient temperature. Rotations: $[\alpha]^{26}_{D}$ (solvent, concentration). Solvents: C, chloroform; E, ethanol; H, water; M, methanol. Microanalyses and the above determinations were performed by the Physical-Analytical Chemistry Department, Schering-Plough Research Institute.

Chemistry. Starting materials were purchased or prepared by literature methods: S-(Phenylmethyl)-L-cysteine hydroTable 7. In Vitro and in Vivo Biological Activity of N-[3-(Acetylthio)-2-(phenylmethyl)-1-oxopropyl]amino Acid Amides



				IC ₅₀	0, nM	ΔBP, n	nmHgª
compd	Q	stereo (*)	AA-NR"R"	NEP	ACE	DOCA	ANF
46 a	Н	S	L-Met-NH ₂	76	>1000		26 (30 sc)
47 a	н	R	$L-Met-NH_2$	94	>1000		
48a	Ac	S	$L-Met-NH_2$			39 (3 p o)	23 (30 sc)
49 a	Ac	R	$L-Met-NH_2$			•	27 (30 sc)
5 0 c	Ac	R, S	$L-Met-N(CH_2)_4$			24 (10 po)	29 (30 sc)
50d	Ac	R, S	L-Met-N(CH ₂) ₅			3 (3 sc)	38 (30 sc)
						32(10 sc)	,
43 a	Bz	S	L-Met-NH ₂			0 (30 sc)	
44a	Bz	R	L-Met-NH2			0 (30 sc)	19 (3 po)
			-			· · ·	52 (10 po)
50f	Ac	R,S	L-Met-N(CH ₂) ₂ O(CH ₂) ₂	>300		20 (10 po)	24 (30 sc)
5 0e	Ac	R, S	L-Met-NHCH ₂ CH ₂ OH			24 (10 po)	25 (30 sc)
50g	Ac	R, S	L-Met-NHCH ₂ CO ₂ Et	>300		0 (10 po)	
$51\mathbf{\check{h}}$	н	Ś	$S-(4-Me-benzvl)-L-Cvs-NH_2$	>300		• •	0 (30 sc)
52h	н	R	S-(4-Me-benzvl)-L-Cvs-NH ₂	>300		17 (3 sc)	17 (30 sc)
			· · · · · · · · · · · · · · · · · · ·			46 (10 po)	
49h	Ac	R	$S-(4-Me-benzvl)-L-Cvs-NH_2$			0 (10 po)	0 (10 sc)
48h	Ac	S	S-(4-Me-benzyl)-L-Cys-NH ₂			29 (10 po)	12 (30 sc)

^a The numbers indicating the change in blood pressure in DOCA rats or ANF potentiation represent average values determined in four or more animals.

chloride (Sigma), S-(phenylmethyl)-L-cysteine ethyl ester hydrochloride (Sigma), N-[(1,1-dimethylethoxycarbonyl]-S-(phenylmethyl)-D-cysteine (Bachem), S-ethyl-L-cysteine (Chemical Dynamics), N-[(1,1-dimethylethoxy)carbonyl]-S-[(4-methylphenyl)methyl]-L-cysteine (Bachem), N-[(1,1-dimethylethoxy)carbonyl]-S-[(3,4-dimethylphenyl)methyl]-L-cysteine (Bachem), N-[(1,1-dimethylethoxy)carbonyl]-S-[(3,4-dimethylphenyl)methyl]-L-cysteine (Bachem), N-[(1,1-dimethylethoxy)carbonyl]-S-[(4-methoxyhenyl)methyl]-L-cysteine (Bachem), N-[(1,1-dimethylethoxy)carbonyl]-S-(1,1-dimethylethoxy)carbonyl]-S-(1,1-dimethylethyl)-L-cysteine (Bachem), N-[(1,1-dimethylethoxy)carbonyl]-S-(1,1-dimethylethyl)-L-cysteine (Bachem), S-(triphenylmethyl)-L-cysteine (Sigma), S-methyl-L-cysteine (Chemical Dynamics), L-ethionine (Chemical Dynamics), D-(-)-S-acetyl- β -mercaptoisobutyric acid (Chemical Dynamics). 2(R,S)-[(Benzoylthio)methyl]-3-phenyl-propionic acid and the individual enantiomers were prepared by a literature procedure.²⁵

Abbreviations: EDC, 3-ethyl-1-[3-(dimethylamino)propyl]carbodiimide hydrochloride; HOBT, 1-hydroxybenzotriazole hydrate; BOC, (1,1-dimethylethoxy)carbonyl.

Method A-1-1. Preparation of (Arylmethylene)propanedioic Acids. [(4-Chlorophenyl)methylene]propanedioic Acid (4a). A powdered mixture of 4-chlorobenzaldehyde (70.25 g, 500 mmol) and malonic acid (52.1 g, 500 mmol) was heated on a steam bath for 6 h and then cooled to room temperature. The mixture was partitioned between Et₂O (400 mL) and 1 N NaOH (1 L). The aqueous layer was washed with Et₂O (300 mL), cooled, and acidified to pH 1 with concentrated HCl to give a white solid which was filtered, washed with water, and dried to give 4a as a white solid (79.1 g, 69%): mp 207-209 °C (lit.²⁵ mp 215 °C dec).

(2-Naphthylmethylene)propanedioic Acid (4b). 2-Naphthaldehyde (15.6 g, 100 mmol) and malonic acid (10.4 g, 100 mmol) in glacial AcOH (6 mL) were heated under reflux at 100 °C for 4 h. The reaction mixture was cooled and diluted with CH₂Cl₂ to give a white precipitate. Filtration gave a white solid A (14.3 g). The filtrate was concentrated *in vacuo*, malonic acid (1.5 g) added, and the mixture treated as above to give solid B (4.82 g). Solids A and B were combined and suspended in water (200 mL), filtered, and washed with water to give 4b as a white solid (16.6 g, 68%): mp 204-205 °C. Anal. (C₁₄H₁₀O₄) C, H.

[[1,1'-Biphenyl]-4-ylmethylene]propanedioic Acid (4c). In a manner similar to that described for 4b, 4-biphenylcarboxaldehyde (18.0 g, 99 mmol) and malonic acid (10.4 g, 100 mmol) were converted to 4c as a pale yellow solid (9.56 g, 31%): mp 208-209 °C. Anal. ($C_{16}H_{12}O_4$) C, H. [(2-Methylphenyl)methylene]propanedioic Acid (4d). o-Tolualdehyde (100 g, 833 mmol) and malonic acid (86 g, 833 mmol) were heated under reflux at 100 °C for 7 h, and the resulting mixture was treated as described in method A-1-1 to give 4d as a white solid (107.7 g, 62%): mp 198-204 °C dec. Anal. ($C_{11}H_{10}O_4$) C, H.

Method A-1-2. Preparation of (Arylmethyl)propanedioic Acid. [(4-Chlorophenyl)methyl]propanedioic Acid (5a).²⁵ Diacid 4a (79 g, 348 mmol) was hydrogenated at 50 psi in EtOAc (1 L) with 10% Pd/C (3.4 g). After filtration and washing with EtOAc (300 mL), concentration *in vacuo* gave 5a as a white solid (77.9 g, 97.7%): mp 162-164 °C.

(2-Naphthylmethyl)propanedioic Acid (5b). In a manner similar to that described for 5a, diacid 4b (16.5 g, 68 mmol) was converted to 5b as a white solid (16.0 g, 99%): mp 150–152 °C. Anal. ($C_{14}H_{12}O_4$) C, H.

[[1,1'-Biphenyl]-4-ylmethyl]propanedioic Acid (5c). In a manner similar to that described for 5a, diacid 4c (1.44 g, 8.6 mmol) was converted to 5c as a white solid (1.40 g, 96%): mp 180-181 °C. Anal. ($C_{16}H_{14}O_4$) C, H.

[(2-Methylphenyl)methyl]propanedioic Acid (5d). Diacid 4d (107 g, 519 mmol) in absolute EtOH (300 mL) and EtOAc (300 mL) was hydrogenated with 10% Pd/C (4.0 g) as described in method A-1-2 to give 5d as a white solid (97.5 g, 90%): mp 137-140 °C. Anal. ($C_{11}H_{12}O_4$) C: calcd, 63.46; found, 63.97; H: calcd, 5.81; found, 6.50.

Method A-1-3. Preparation of α -Methylenearenepropanoic Acids. 4-Chloro- α -methylenebenzenepropanoic Acid (6a). Diacid 5a (39 g, 171 mmol) in distilled water (50 mL) was cooled to 0-5 °C in a salt-ice bath. To this was added 40% aqueous dimethylamine (43.2 mL) to pH 7.5. Additional 5a (38.8 g, 170 mmol) was added, followed by sufficient distilled water to give a clear solution. A 37% aqueous formaldehde solution (40 mL) was added, and the mixture was allowed to reach room temperature and then stirred for 20 h. The solid was filtered off and washed with water, suspended in water (250 mL), and heated on a steam bath 2 h. The solution was cooled and acidified with concentrated HCl. The solid was filtered, washed with water, and dried to give 6a as a white solid (36.1 g, 54%): mp 93-95 °C (lit, ²⁵ mp 95-96 °C).

 α -Methylene-2-naphthalenepropanoic Acid (6b). In a manner similar to that described for 6a, diacid 5b (16.0 g, 66 mmol) was converted to 6b as a white solid (9.73 g, 70%): mp 83-84 °C. Anal. (C₁₄H₁₂O₄-0.12H₂O) C, H.

 α -Methylene-[1,1'-biphenyl]-4-propanoic Acid (6c). In a manner similar to that described for 6a, diacid 5c (9.10 g, 53.5 mmol) was converted to 6c as a white solid (6.68 g, 83%): mp 168-170 °C. Anal. (C₁₆H₁₄O₄·0.1H₂O) C, H.

2-Methyl- α -methylenebenzenepropanoic Acid (6d). Diacid 5d (50 g, 240 mmol) in water (100 mL) was cooled in an ice bath and treated with 40% dimethylamine solution (52.2 mL) (mechanical stirring used). After 10 min, additional 5d (47 g, 226 mmol) and water (200 mL) were added. After 30 min, 37% formaldehyde solution (100 mL) was added. The resulting mixture was warmed to room temperature, stirred for 18 h, and filtered. The white solid (98.5 g) was suspended in water (500 mL) and heated under reflux for 2.5 h. The reaction mixture was cooled, extracted with diethyl ether, acidified with concentrated HCl to pH 2, and filtered to give 6d as a white solid (36.2 g, 44%): mp 82-85 °C. Anal. (C₁₁H₁₂O₂) C, H.

2-Methylenepentanoic Acid (6e). Diethyl *n*-propylmalonate (40 g, 198 mmol) in EtOH (100 mL) was added dropwise to KOH (26 g, 464 mmol) in EtOH (200 mL) and water (50 mL), and the mixture was heated under reflux 6 h. The reaction mixture was cooled, concentrated *in vacuo*, diluted with water (500 mL), acidified to pH 2 with concentrated hydrochloric acid, and extracted with Et₂O. The dried (MgSO₄) Et₂O was concentrated *in vacuo* to give *n*-propylmalonic acid, a white solid (26.7 g, 92%), mp 92-95 °C. Anal. (C₆H₁₀O₄) C, H. In a manner similar to that described for **6a**, *n*-propylmalonic acid (26.0 g, 178 mmol) was converted to **6e**, a clear oil (5.33 g, 26%) which was used in the next step.

Method A-1-4. Preparation of α -[((R, \hat{S})-acetylthio)methyl]benzenepropanoic Acids. 4-Chloro- α -[((R, \hat{S})acetylthio)methyl]benzenepropanoic Acid (7a). Acid 6a (8.0 g, 40.7 mmol) in CH₂Cl₂ (100 mL) was treated with thiolacetic acid (20.9 g, 276 mmol) in CH₂Cl₂ (40 mL) for 48 h and concentrated *in vacuo* to give 7a, a yellow oil (10.4 g, 94%) which was suitable for further use.

 α -[((**R**,**S**)-Acetylthio)methyl]2-naphthalenepropanoic Acid (7b). In a manner similar to that described for 7a, acid **6b** (9.37g, 44 mmol) was converted to 7b, a light tan solid (5.27 g, 41%; contains 13% starting material by NMR); mp 103-106 °C.

 α -[((*R***,S**)-Acetylthio)methyl][1,1'-biphenyl]-4-propanoic Acid (7c). In a manner similar to that described for 7a, acid 6c (6.60g, 27.7 mmol) was converted to 7c as a white solid (3.44 g, 40%): mp 101-103 °C. Anal. (C₁₈H₁₈O₃S) C, H.

2-Methyl- α -[((R,S)-acetylthio)methyl]benzenepropanoic Acid (7d).³⁷ Acid 6d (36.1 g, 205 mmol) and thiolacetic acid (22.4 mL, 430 mmol) were stirred for 3 days. Toluene (200 mL) was added, and the mixture was concentrated *in* vacuo (3 times). The residue was triturated with hexane, filtered, and dried to give 7d as a white waxy solid (44.2 g, 86%); mp 75-76 °C. Anal. (C₁₃H₁₆O₃S) C, H.

 α -[((*R***,S)-Acetylthio)methyl]pentanoic Acid (7e). In a manner similar to that described for 7a, 6e (5.33 g, 47 mmol) was converted to 7e as an oil (6.45 g, 73%). Anal. (C₈H₁₄O₃S) H; C: calcd, 50.50; found, 49.15.**

 α -[((**R**,**S**)-Acetylthio)methyl]1-naphthalenepropanoic Acid (7g). In a manner similar to that described for 7a, acid **6g** (9.5 g, 46 mmol) was converted to **7g** as a white solid (11.95 g, 92%): mp 94–95 °C. Anal. (C₁₆H₁₆O₃S) H; C: calcd, 66.65; found, 66.09.

 α -[((*R***,S**)-Acetylthio)methyl]benzenepropionic Acid (7f). In a manner similar to that described for **6a**, acid **6f** (105 g, 649 mmol) was converted to **7f**, a white wax (used in next step) (153 g, 99%).

Method A-2-1. Preparation of Diethyl (1-Naphthylmethyl)propanedioate (8g). Sodium metal (11.0 g, 478 mmol) was added with cooling to absolute EtOH (650 mL). To the solution was added diethyl malonate (75.6 g, 472 mmol), followed by 1-(bromomethyl)naphthalene (100.0 g, 452 mmol). After 3 h, the mixture was concentrated *in vacuo*, diluted with water, and extracted with Et₂O. The dried (MgSO₄) Et₂O was concentrated *in vacuo* to give 8g as an oil (133.7 g, 98%) which was used in the next step.

Method A-2-2. Preparation of Monoethyl (Arylmethyl)propanedioic Acids. Monoethyl (1-Naphthylmethyl)- **propanedioic Acid** (9g). To 8g (133.7 g, 446 mmol) in EtOH (400 mL) was added a solution of potassium hydroxide (24.9 g, 445 mmol) in ethanol (400 mL), and the mixture was stirred 18 h, concentrated *in vacuo*, diluted with ice-water, and extracted with Et₂O. The aqueous solution was cooled, acidified to pH 2 with concentrated HCl, and extracted with Et₂O. The dried (MgSO₄) Et₂O was concentrated *in vacuo* to give 9g as an oil (used in next step) (100.0 g, 82%).

Monoethyl (Phenylmethyl)propanedioic Acid (9f). In a manner similar to that described for 9g, diethyl benzylmalonate (300 g, 1.2 mol) was converted to 9f as a yellow oil (266 g, 99%) which was used in the next step.

Method A-2-3. Ethyl α -Methylenearenepropanoates. Ethyl α - Methylene-1-naphthalenepropanoate (10g). To 9g (100.0 g, 367 mmol) and diethylamine (39 mL) was added 37% aqueous formaldehyde solution (38 mL) over 30 min at 0-5 °C. The mixture was stirred for 7 h at room temperature and extracted with Et₂O (3 × 500 mL). The Et₂O extracts were washed with 2 N HCl (2 × 500 mL), saturated NaHCO3 (500 mL), and brine (500 mL). The dried (MgSO₄) Et₂O was concentrated *in vacuo* to give 10g, an oil (58.6 g, 66%) which was used in the next step.

Ethyl α -Methylenebenzenepropanoate (10f). In a manner similar to that described for 10g, 9f (266 g, 1.2 mol) was converted to 10f as an oil (165 g, 72%) which was used in the next step.

Method A-2-4. α -Methylenearenepropanoic Acids via Ethyl α -Methylenearenepropanoates. α -Methylene-1naphthalenepropanoic Acid (6g). Ester 10g (12.0 g, 50 mmol) in dioxane (50 mL) was treated with 1 N NaOH (60 mL) and stirred at room temperature 20 h. The reaction mixture was concentrated *in vacuo*, diluted with water, and extracted with EtOAc. The aqueous solution was cooled to 0-5 °C, and concentrated HCl was added slowly to give a precipitate which was filtered off, washed with water, and dried to give **6g** as a white solid (9.62 g, 90%), mp 115-117 °C. Anal. (C₁₄H₁₂O₂0.1H₂O) C, H.

 α -Methylenebenzenepropanoic Acid (6f). In a manner similar to that described for 6g, 10f (165 g, 868 mmol) was converted to 6f as a white solid (105 g, 75%), mp 67-68 °C.

Method A-3. Preparation of α -[((*R*,*S*)-Acetylthio)methyl]benzenebutanoic Acid. 1,1-Dimethylethyl 2-(2-Phenylethyl)-3-oxobutanoate (11).³⁸ Sodium (2.3 g, 100 mmol) was added to absolute EtOH (50 mL). The solution was cooled to 0-5 °C, and *tert*-butyl acetoacetate (15.8 g, 100 mmol) was added. After 10 min, 2-(bromoethyl)benzene (18.5 g, 100 mmol) was added and the mixture heated under reflux 4 h. The mixture was cooled and filtered, the filtrate concentrated *in vacuo*, and the residual oil distilled to give 11 as a colorless oil (12.85 g, 49%): bp 100-118 °C (0.1 mm); NMR (C) 3.32 (t,1H), 2.62 (m, 2H), 2.17 (m, 2H). This contains approximately 0.25 equiv of ethyl ester and was used in the next step.

1,1-Dimethylethyl α-Methylenebenzenebutanoate (12).³⁸ At -60 °C, 2.5 M n-butyllithium in hexane (18.75 mL, 47 mmol) was added dropwise to a solution of diisopropylamine (4.75 g, 47 mmol) in dry THF (30 mL). The mixture was warmed to 0 °C and then cooled to -60 °C, and 11 (12.0 g, 45.8 mmol) in THF (20 mL) was added. After 10 min, paraformaldehyde (6.87 g) was added, and the reaction mixture was allowed to warm to room temperature, stirred at that temperature 1 h, heated under reflux 1 h, cooled, and filtered. The filtrate was concentrated in vacuo and the residue partitioned between water and Et₂O. The Et₂O solution was washed sequentially with 1 N HCl, water, and 1 N NaHCO₃. The dried (MgSO₄) Et₂O was concentrated in vacuo to give an oil (10.03 g). Distillation (Kugelrohr at 0.1 mm between 100-150 °C) gave 12 as a colorless oil (3.58 g, 34%): NMR (C) δ 5.42 + 6.07 (C=CH2). This contains approximatey 0.25 equiv of ethyl ester and was used in the next step

1,1-Dimethylethyl α -[((R,S)-Acetylthio)methyl]benzenebutanoate (13). Ester 12 (1.5 g, 6.5 mmol) and thiolacetic acid (0.98 g, 12.9 mmol) were stirred for 3 days and concentrated under high vacuum. The residue was partitioned between Et₂O and 1 N NaHCO₃. The dried (MgSO₄) Et₂O was concentrated *in vacuo* to give an oil (1.91 g) which was chromatographed on silica gel (300 g) (hexane-Et₂O, 9:1) to give **13** as an oil (0.92 g, 47%): NMR (C) δ 3.10 (m, 2H), 2.65 (t, 3H), 2.52 (m, 1H), 1.91 (m, 2H).

 α -[((*R***,S)-Acetylthio)methyl]benzenebutanoic Acid (14).</mark> Ester 13** (0.60 g, 1.9 mmol) and trifluoroacetic acid (10 mL) were stirred for 30 min and then concentrated *in vacuo* to give **14** as an orange oil (0.49 g, 52%): NMR (C) δ 3.17 (m, 2H), 2.71 (t, 2H), 2.70 (m, 1H), 2.01 (m, 2H).

Method A-4. Preparation of 1-[(Acetylthio)methyl]cyclopentanecarboxylic Acid. 1,1-Dimethylethyl Cyclopentanecarboxylate (15). At -20 to -30 °C, isobutylene (100 mL) was condensed in a threaded tube. *tert*-Butyl alcohol (4 mL), cyclopentanecarboxylic acid (25 g, 219 mmol), and concentrated sulfuric acid (1 mL) were added. The tube was sealed and stirred for 4 days at room temperature, cooled, vented, and purged with nitrogen. The mixture was partitioned with Et₂O and 1 N NaHCO₃. The dried (MgSO₄) Et₂O was concentrated *in vacuo* to give a pale yellow oil (25.05 g). Distillation gave 15, a colorless oil (17.7 g, 47%): bp_{0.1} 55-60 °C; NMR (C) 2.63 (quint, 1H), 1.80-1.90 (m, 8H), 1.44 (s, 9H).

1,1-Dimethylethyl 1-(Hydroxymethyl)cyclopentanecarboxylate (16a). *n*-Butyllithium (2.5 M in hexane, 8.3 mL, 21 mmol) was added to a -78 °C solution of diisopropylamine (2.12 g, 21 mmol) in THF (50 mL). This mixture was warmed to 0 °C, stirred for 30 min, cooled to -78 °C, and treated dropwise with 15 (3.00 g, 17.6 mmol) in THF (50 mL) and dipyridyl (0.002 g). Paraformaldehyde (2.58 g) in a separate flask was heated to 160 °C and introduced via a stream of nitrogen into the reaction mixture. After 45 min, the reaction mixture was warmed to room temperature and stirred for 18 h. The mixture was quenched with saturated NH₄Cl solution (100 mL) and then brine (100 mL) and extracted with CH₂Cl₂ (4 × 100 mL). The dried (MgSO₄) CH₂Cl₂ was concentrated *in vacuo* to give **16a** as a yellow oil (2.98 g, 84%): NMR (C) δ 3.53 (s, 2H), 1.92, (m, 2H), 1.50–1.80 (m, 8H), 1.46 (s, 9H).

1,1-Dimethylethyl 1-[[(Methylsulfonyl)oxy]methyl]cyclopentanecarboxylate (16b). Methanesulfonyl chloride (0.98 g, 8.30 mmol) was added to 16a (1.5 g, 7.5 mmol) and triethylamine (0.83 g, 8.8 mmol) in Et₂O (20 mL), and the mixture was stirred at room temperature for 18 h. The reaction mixture was washed with water, and the dried (MgSO₄) Et₂O was concentrated *in vacuo* to give 16b as a yellow oil (2.02 g, 97%): NMR (C) δ 4.24 (s, 2H), 3.02 (t, 3H), 2.07 (m, 2H), 1.50–1.80 (m, 3H),1.46 (s, 9H). This contains 0.15 equiv of methanesulfonyl chloride.

1,1-Dimethylethyl 1-[(Acetylthio)methyl]cyclopentanecarboxylate (17). Ester 16b (1.85 g, 6.65 mmol), thiolacetic acid (1.11 g, 14.6 mmol), and potassium carbonate (0.92 g, 6.6 mmol) in anhydrous DMF (30 mL) were heated under reflux at 100 °C for 4.5 h. The reaction mixture was partitioned between EtOAc and 1 N HCl. The dried (MgSO₄) EtOAc was concentrated *in vacuo* to give a red oil (1.64 g) which was chromatographed on a column of silica gel (300 g) (hexane-Et₂O, 19:1) to give 17 as a red oil (1.25 g, 73%): NMR (C) δ 3.18 (s, 2H), 2.33 (s, 3H), 2.04 (m, 2H), 1.50-1.80 (m, 6H), 1.44 (s, 9H).

1-[(Acetylthio)methyl]cyclopentanecarboxylic Acid (18). Ester 17 (1.25 g, 4.84 mmol) was dissolved in CH_2Cl_2 (40 mL) and TFA (40 mL). After 18 h, the solvent was removed *in* vacuo and the residual solid dried at 50 °C (0.1 mm) to give 18 as a brown solid (0.96 g, 98%): NMR (C) δ 3.23 (s, 2H), 2.35 (s, 3H), 2.12 (m, 2H), 1.60–1.90 (m, 6H).

S-(2-Phenylethyl)-L-cysteine (19a). At 0 °C, nitrogen was bubbled through a mixture of L-cysteine (45.5 g, 376 mmol) in water (180 mL) during the addition of 2 N NaOH (187 mL, 376 mmol), followed by addition of 2-phenylethyl bromide (75.9 g, 410 mmol) in MeOH (800 mL). After 1 h, the mixture was warmed to room temperature and kept for 20 h. The mixture was made acidic to pH 6-7 with glacial AcOH (15 mL) and the white precipitate collected and washed with water to give a white solid (40.8 g), mp 217-219 °C dec. This solid (34 g) was boiled in a mixture of MeOH (180 mL), glacial AcOH (340 mL), and water (700 mL), filtered, concentrated to 500 mL, and chilled to give **19a** as a white solid (22.5 g): mp 224-225 °C dec. Second crop of **19a** as a white solid (5.2 g): mp 221-223 °C.

Method B-1. Preparation of Amino Acid Ester Hydrochlorides from Amino Acids. S-(2-Phenylethyl)-L-cysteine Ethyl Ester Hydrochloride (20a).³⁷ S-(2-Phenylethyl)-L-cysteine (19a) (2.50 g, 1.11 mmol) was added to a solution resulting from the addition of thionyl chloride (2.0 mL) to absolute EtOH (25 mL). The mixture was heated under reflux for 5 h, cooled, and concentrated *in vacuo*. The residue was dissolved in absolute EtOH, treated with Darco, filtered, and concentrated *in vacuo* to give 20a as a white solid (2.90 g, 90%): mp 155-156 °C; $[\alpha]^{26}$ _D -1.5° (0.52, M). Anal. (C₁₃H₁₉-NO₂S+HCl) C, H, N.

S-Methyl-L-cysteine Ethyl Ester (20b). Similarly, S-methyl-L-cysteine was converted to crude 20b as a white solid which was chromatographed on a column of silica gel using CH_2Cl_2 -MeOH-NH₄OH, 170:27:3, as eluant to give 20b as an amber oil (2.20 g, 61%): $[\alpha]^{26}_D$ +25.1° (0.53, M). Anal. (C₆H₁₃NO₂S·H₂O) C, N. H: calcd, 8.28; found, 7.10.

S-Ethyl-L-cysteine Ethyl Ester Hydrochloride (20c). Similarly to 20a, S-ethyl-L-cysteine was converted to 20c as a white solid (99%): mp 130–133 °C (lit.³⁹ mp 137 °C); $[\alpha]^{26}_{D}$ -11.0° (0.50, M). Anal. (C₇H₁₆NO₂S·HCl) C, H, N.

L-Ethionine Ethyl Ester Hydrochloride (20d).³⁸ Similarly, L-ethionine was converted to 20d as a white foam (99%): $[\alpha]^{26}_{D} + 15.5^{\circ} (0.51, M)$ which was used without further purification.

S-(Triphenylmethyl)-L-cysteine Methyl Ester Hydrochloride (20e). Similarly, S-(triphenylmethyl)-L-cysteine in MeOH was converted to 20e as an off-white foam (95%) which was used in the next step.

Method B-2. Preparation of Amino Acid Ester Hydrochlorides from N-[(1,1-Dimethylethoxy)carbonyl]amino Acids. S-(Phenylmethyl)-D-cysteine Ethyl Ester Hydrochloride (20f).³⁸ N-Boc-S-(phenylmethyl)-D-cysteine (2.50 g, 8.0 mmol) was added to a solution resulting from the addition of thionyl chloride (1.2 mL) to absolute EtOH (30 mL) at 0 °C. After 18 h at room temperature, thionyl chloride (0.60 mL) was added. After 3 h, the mixture was concentrated *in vacuo* and the residue washed with Et₂O to give 20f as a white solid (2.05 g, 95%): mp 149-151 °C (lit.⁴⁰ mp 148 °C); [α]²⁶_D -15.5° (0.51, M). Anal. (C₁₂H₁₇NO₂S·HCl) C, H, N.

S-[(4-Methylphenyl)methyl]-L-cysteine Methyl Ester Hydrochloride (20g). Similarly, N-Boc-S-[(4-methylphenyl)methyl]-L-cysteine was converted to 20g as a white foam (94%): $[\alpha]^{26}_{D} -22.9^{\circ}$ (0.41, M). Anal. (C₁₂H₁₇NO₂S·HCl) H, N; C: calcd, 52.26; found, 51.62.

S-[(4-Methylphenyl)methyl]-L-cysteine Ethyl Ester Hydrochloride (20h). Similarly, N-Boc-S-[(4-methylphenyl)-methyl]-L-cysteine was converted to 20h as a white solid (94%): mp 156–160 °C; $[\alpha]^{26}_{D}$ –28.3° (0.87, M). Anal. (C₁₃H₁₉-NO₂S·HCl) H, N; C: calcd, 53.88; found, 52.65.

S-[(3,4-Dimethylphenyl)methyl]-L-cysteine Ethyl Ester Hydrochloride (20j). Similarly, N-Boc-S-[(3,4-dimethylphenyl)methyl]-L-cysteine was converted to 20j as a white foam (98%): $[\alpha]^{26}_{D} - 26.6^{\circ}$ (0.37, M). Anal. (C₁₄H₂₁ NO₂S-HCl) H, N; C: calcd, 55.34; found, 54.16.

S-[(4-Methoxyphenyl)methyl]-L-cysteine Methyl Ester Hydrochloride (20k). Similarly, N-Boc-S-[(4-methoxyphenyl)methyl]-L-cysteine in MeOH was converted to 20k as a white foam (97%): $[\alpha]^{26}_D$ -23.2° (0.82, M). Anal. (C₁₂H₁₇-NO₃S·HCl) H, N; C: calcd, 49.40; found, 48.73.

S-(1,1-Dimethylethyl)-L-cysteine Methyl Ester Hydrochloride (20m). Similarly, N-Boc-S-(1,1-dimethylethyl)-Lcysteine in MeOH was converted to 20m as an oil (75%), which was used in the next step.

Method B-3-1-1. N-[(1,1-Dimethylethoxy)carbonyl]-S-[(4-methylphenyl)methyl]-L-cysteinamide (22a). N-Boc-S-[(4-methylphenyl)methyl]-L-cysteine (6.15 g, 19.2 mmol) in THF (60 mL) was treated with triethylamine (4.44 g, 44 mmol) and cooled to 0-5 °C. Ethyl chloroformate (4.77 g, 44 mmol) in THF (5 mL) was added and the mixture stirred 15 min. Concentrated ammonium hydroxide (2 mL) in THF (5 mL) was added dropwise over 5 min. After 15 min, the reaction mixture was warmed to room temperature, stirred for 18 h, and filtered. The filtrate was concentrated *in vacuo* to give a pale yellow solid (8.0 g) which was dissolved in CH₂Cl₂ and washed with water. The dried (MgSO₄) CH₂Cl₂ solution was concentrated *in vacuo* to give **22a** as a white solid (6.4 g, 99%): mp 140-142 °C; $[\alpha]^{26}_{D} - 7.5$ ° (1.0, M). Anal. (C₁₆H₂₄N₂O₃S) C, H, N.

Method B-3-1-2. S-[(4-Methylphenyl)methyl]-L-cysteinamide (23a). Amide 22a (6.15 g, 19.2 mmol) in CH₂Cl₂ (40 mL) was treated with trifluoroacetic acid (10mL). After 44 h, the reaction mixture was concentrated *in vacuo*, and the residue was diluted with CH₂Cl₂, concentrated *in vacuo*, dissolved in EtOAc and washed with saturated NaHCO₃. The dried (MgSO₄) EtOAc was concentrated *in vacuo* to give 23a as a white solid (1.53 g, 79%): mp 94-95 °C; $[\alpha]^{26}$ -1.3° (0.55, M). Anal. (C₁₁H₁₆N₂OS) C, H, N.

Method B-3-2-1. Preparation of 1-[N-[(1,1-Dimethylethoxy)carbonyl]amino] Acid Amides. 1-[2(S)-[[(1,1-Dimethylethoxy)carbonyl]amino]-4-(methylthio)-1-oxobutyl]piperidine (22b). N-Boc-L-methionine (1.50 g, 6 mmol), piperidine (0.51 g, 6 mmol), HOBT (0.91 g, 6 mmol), and EDC (1.15 g, 6.1 mmol) in anhydrous DMF (40 mL) were stirred for 5 h, concentrated *in vacuo*, and partitioned between EtOAc and water. The EtOAc extract was washed with water, 1 N NaHCO₃, and brine. The dried (MgSO₄) EtOAc was concentrated *in vacuo* to give a colorless oil (1.96 g) which was chromatographed on a column of silica gel (300 g) (CH₂Cl₂-MeOH, 24:1) to give **22b** as a colorless oil (1.84 g, 96%). NMR (C) consistent.

1-[2(S)-[[(1,1-Dimethylethoxy)carbonyl]amino]-4-(methylthio)-1-oxobutyl]morpholine (22c). Similarly, N-Boc-L-methionine (1.20 g, 4.81 mmol) was converted to 22c as an oil (1.27 g, 82%). Et₂O was used as eluant: $R_f 0.5$ (Et₂O); NMR (C) consistent.

2-[[2(S)-[[(1,1-Dimethylethoxy)carbonyl]amino]-4-(methylthio)butanoyl]amino]ethanol (22d). Similarly, N-Boc-L-methionine (1.50 g, 6.0 mmol) was converted to **22d** as an oil (1.43 g, 81%) (CH₂Cl₂-MeOH 19:1): R_f 0.7 (CH₂Cl₂-MeOH, 17:3); NMR (C) consistent.

N-[2(S)-[[(1,1-Dimethylethoxy)carbonyl]amino]-4-(methylthio)-1-oxobutyl]-β-alanine Ethyl Ester (22e). Similarly, N-Boc-L-methionine (1.50 g, 6.0 mmol) was converted to 22e as an oil (1.90 g, 94%) (CH₂Cl₂-MeOH, 19:1): R_f 0.7 (CH₂Cl₂-MeOH, 19:1); NMR (C) consistent.

Method B-3-2-2. Preparation of Amino Acid Amide Hydrochlorides. N-[2(S)-Amino-4-(methylthio)-1-oxobutyl]piperidine Hydrochloride (23b). Amide 22b (1.70 g, 5.37 mmol) was treated with 6 M HCl in dioxane (25 mL). After 30 min, the reaction mixture was concentrated *in vacuo* to give 23b as a foam (1.60 g, crude, suitable for further use).

N-[2(S)-Amino-4-(methylthio)-1-oxobutyl]morpholine Hydrochloride (23c). Similarly, amide 22c (1.25 g, 3.9 mmol) was converted to 23c as a solid (1.20 g, crude, suitable for further use).

2-[[2(S)-Amino-4-(methylthio)butanoyl]amino]ethanol Hydrochloride (23d). Similarly, amide 22d (1.33 g, 4.55 mmol) was converted to 23d as an oil (1.05 g, crude, suitable for further use).

N-[2(S)-Amino-4-(methylthio)-1-oxobutyl]- β -alanine Ethyl Ester Hydrochloride (23e). Using the method described above, amide 22e (1.90 g, 5.68 mmol) was converted to 23e as an oil (1.68 g, crude, suitable for further use).

Method B-4. 1-[2(S)-Amino-4-(methylthio)-1-oxobutyl]pyrrolidine Hydrochloride (23f). L-Methionine methyl ester hydrochloride (4.00 g, 20 mmol) was dissolved in pyrrolidine (10.0 g, 141 mmol) (exotherm). After 18 h, the mixture was concentrated *in vacuo* at 50 °C and partitioned with EtOAc (100 mL), 1 N NaHCO₃ (100 mL), and brine (25 mL). The aqueous solution was treated with NaCl (20 g) and 50% NaOH (10 mL) and extracted with a mixture of EtOAc (100 mL) and EtOH (10 mL). The organic extracts were dried (MgSO₄) and concentrated *in vacuo*. The residue was dissolved in Et₂O, filtered, and concentrated *in vacuo* to give an oil (1.6 g). The oil was taken up in Et₂O and treated with 2 N HCl/ Et₂O (5 mL). The solid was filtered off and recrystallized from MeOH-EtOAc to give **23f** as needles (1.50 g, 31%): mp 124-125 °C. Anal. (C₉H₁₈N₂OS·HCl) C, H, N.

Method C-1. N-[3-(Acetvlthio)-2(S)-(phenvlmethyl)-1oxopropyl]-S-(phenylmethyl)-L-cysteine Ethyl Ester (25a) and N-[3-(Acetylthio)-2(R)-(phenylmethyl)-1-oxopropyl]-S-(phenylmethyl)-L-cysteine Ethyl Ester (26a). EDC (1.05 g, 5.5 mmol) was added to acid 7f (1.19 g, 5.0 mmol), S-(phenylmethyl)-L-cysteine ethyl ester hydrochloride (20f) (1.36 g, 5.0 mmol), HOBT (0.75 g, 5.5 mmol), and N-methylmorpholine (1.1 mL, 10 mmol) in DMF (5 mL). The reaction mixture was stirred for 20 h, concentrated in vacuo, and partitioned between CH₂Cl₂ and water. The dried (MgSO₄) CH_2Cl_2 solution was concentrated in vacuo to give a residue which was chromatographed on flash silica gel (300 mL) (CH2- Cl_2 -EtOAc, 49:1) to give fraction 1: N-[3-(acetylthio)-2(S)-(phenylmethyl)-1-oxopropyl]-S-(phenylmethyl)-L-cysteine ethyl ester **25a** as a white solid (0.49 g, 22%): mp 83-85 °C; $[\alpha]^{26}$ _D -73.5° (0.28, M). Anal. (C₂₇H₂₉NO₄S₂) C, H, N; fraction 2: (0.15 g); fraction 3 (0.50 g); and fraction 4: N-[3-(acetylthio)-2(R)-(phenylmethyl)-1-oxopropyl]-S-(phenylmethyl)-L-cysteine ethyl ester **26a** as a white solid (0.37 g, 16%): $[\alpha]^{26} - 9.4^{\circ}$ (0.51, M). Anal. (C₂₇H₂₉NO₄S₂) C, H, N.

The N-[3-(acetylthio)-2(S)-[(substituted)methyl]-1-oxopropyl]-L-amino acid esters (**25c-f,l,n,o,t-w,y, 36**); N-[3-(-acetylthio)-2(S)-(phenylmethyl)-1-oxopropyl]-S-(phenylmethyl)-D-cysteine ethyl ester (**26b**) (Table 1); N-[3-(acetylthio)-2(R)-[(substituted)methyl]-1-oxopropyl]-L-amino acid esters (**26cf,l,o,t-w,y, 37**); and N-[3-(acetylthio)-2(R)-(phenylmethyl)-1oxopropyl]-S-(phenylmethyl)-D-cysteine ethyl ester (**26b**) (Table 1) were prepared using the procedure described above.

Method C-1a. N-[3-(Acetylthio)-2(R,S)-(phenylmethyl)-1-oxopropyl]-L-methionine Methyl Ester (27m). In a manner similar to that described in method C-1, acid 7f (3.01 g, 12.6 mmol) and L-methionine methyl ester (2.00 g, 12.3 mmol) were converted to N-[3-(acetylthio)-2(R,S)-(phenylmethyl)-1-oxopropyl]-L-methionine methyl ester (27m) (2.61 g, 55%): [α]²⁶_D -38.9° (0.55, M). Anal. (C₁₈H₂₅NO₄S₂) H, N; C: calcd, 56.37; found, 55.47 (Table 1).

The N-[3-(acetylthio)-2(R,S)-[(substituted)methyl]propyl]-Lamino acid esters (**27g,h,k,q,s**) (Table 1) were prepared using the procedure described above.

Method C-2. N-[3-Mercapto-2(S)-(phenylmethyl)-1oxopropyl]-S-(phenylmethyl)-L-cysteine (28a). Under a nitrogen atmosphere, ester 25a (0.48 g, 0.97 mmol) in MeOH^{23.41} (20 mL) was treated with 1 N NaOH (3.1 mL, 3 equiv). After 1 h, 1 N HCl (4 mL) and then water (50 mL) were added. The mixture was extracted with EtOAc (500 mL), and the dried (MgSO₄) EtOAc was concentrated *in vacuo* to give 28a as a viscous oil (0.30 g, 74%): $[\alpha]^{26}_{D} - 2.1^{\circ}$ (1.0, M). Anal. (C₂₀H₂₃-NO₃S₂·0.1CH₂Cl₂) C, H; N: calcd, 3.51; found, 3.10. This material contains 15% D-cysteine diastereomer, as seen in the NMR.

The N-[3-mercapto-2(S)-(arylmethyl)-1-oxopropyl]-L-amino acids (**28c-f,l,n,o,t-w,y**); N-[3-mercapto-2(S)-(phenylmethyl)-1-oxopropyl]-S-(phenylmethyl)-D-cysteine (**28b**) (Table 2); N-[3mercapto-2(R)-(arylmethyl)propyl]-L-amino acids (**29c-f,l,n,o,t-x**); and N-[3-mercapto-2(R)-(phenylmethyl)-1-oxopropyl]-S-(phenylmethyl)-D-cysteine (**29b**) (Table 2) were prepared using the procedure described above.

Method C-2a. N-[3-Mercapto-2(R,S)-(arylmethyl)-1oxopropyl]-L-amino Acids. The N-[3-mercapto-2(R,S)-(arylmethyl)-1-oxopropyl]-L-amino acids (**30g,h,k,m,r,s**) (Table 2) were prepared using method C-2.

Method D. N-[3-(Benzoylthio)-2(S)-(phenylmethyl)-1oxopropyl]-L-methionine Methyl Ester (25n). In a manner similar to that described in method C-1, $\alpha(S)$ -[(benzoylthio)methyl]benzenepropanoic acid (31-S) (1.27 g, 4.0 mmol) and L-methionine methyl ester hydrochloride (0.79 g, 4.0 mmol) were converted to N-[3-(benzoylthio)-2(S)-(phenylmethyl)-1oxopropyl]-L-methionine methyl ester (25n) (1.45 g, 86%): $[\alpha]^{26}_{D}$ -67.9° (0.5, E). Anal. (C₂₃H₂₇NO₄S₂) C, H, N (Table 1).

N-[3-(Benzoylthio)-2(R)-(phenylmethyl)-1-oxopropyl]-L-**methionine methyl ester (26n)** was prepared in a manner similar to that described above.

N-[3-(Acetylthio)-2(S)-methyl-1-oxopropyl]-S-(phenylmethyl)-L-cysteine Ethyl Ester (25p). In a manner similar to that described in method C-1, D-(-)-S-acetyl- β -mercaptoisobutyric acid (1.46 g, 9 mmol) and S-(phenylmethyl)-L- cysteine ethyl ester hydrochloride (2.76 g, 1 mmol) were converted to N-[3-(acetylthio)-2(S)-methyl-1-oxopropyl]-S-(phenylmethyl)-L-cysteine ethyl ester (**25p**) (1.37 g, 39%): $[\alpha]^{26}_{D}$ -122.4° (0.3, M). Anal. (C₁₈H₂₅NO₄S₂) C, H, N (Table 1).

Method E-1. α -[(*R*,*S*)-(Acetylthio)methyl]benzenepropanoyl Chloride (35). Acid 7f (7.10 g, 30 mmol) in toluene (30 mL) was treated with thionyl chloride (3.3 mL). The resulting mixture was heated under reflux for 3 h and concentrated *in vacuo* to give 35, an amber oil (6.94 g) which was used in the next step.

N-[3-(Acetylthio)-2(S)-(phenylmethyl)-1-oxopropyl]-S-[(4-methylphenyl)methyl]-L-cysteine (38a) and N-[3-(Acetylthio)-2(R)-(phenylmethyl)-1-oxopropyl]-S-[(4-methylphenyl)methyl]-L-cysteine (**39a**). S-[(4-Methy]phenyl)methyl]-L-cysteine (6.10 g, 27 mmol) in acetonitrile (50 mL) and water (25 mL) was treated with triethylamine (3.80 mL), followed by dropwise addition of acid chloride 35 (7.0 g, 25 mmol) in acetonitrile (30 mL). The resulting mixture was stirred at room temperature 20 h, filtered, concentrated in vacuo, diluted with water (100 mL), acidified with 1 N hydrochloric acid to pH 2-4, and extracted with EtOAc. The dried (MgSO₄) EtOAc was concentrated in vacuo to give a brown oil (12.82 g). Column chromatography on silica gel (3 L) (CH₂Cl₂-MeOH, 100:1; 7 L); (CH₂Cl₂-MeOH-glacial AcOH, 400:4:0.1; 2.5 L); then (CH₂Cl₂-MeOH-glacial AcOH, 400:5:0.1) gave fraction 1 (1.00 g); fraction 2 (2.38 g); fraction 3 (1.00 g); and fraction 4 (1.56 g). Fraction 1 was dissolved in EtOAc, treated with Darco, filtered, and then chromatographed (with fraction 2-2, see below) on a column of silica gel (2 L) (CH₂Cl₂-MeOH, 100:1, 1.5 L); (CH₂Cl₂-MeOH-glacial AcOH, 400:4:0.1) to give fraction 1-3, N-[3-(acetylthio)-2(S)-(phenylmethyl)-1-oxopropyl]-S-[(4-methylphenyl)methyl]-L-cysteine (**38a**) an oil (0.34 g, 6.3%): $[\alpha]^{26}D - 23.0^{\circ}$ (1.0, **M**). Anal. (C₂₃H₂₇NO₄S₂·0.2CH₂Cl₂) C, N; H: calcd, 5.97; found, 5.55. Fraction 2 was chromatographed on a column of silica gel (2 L), $(CH_2Cl_2-MeOH, 100:1, 2 L)$; $(CH_2Cl_2-MeOH-glacial)$ AcOH, 100:1:0.1) to give fraction 2-1, (0.1 g); fraction 2-2(0.54 g); fraction 2-3 (0.41 g); fraction 2-4 (0.10 g): fraction 2-5 (0.28 g); fraction 2-6 (0.12 g); and fraction 2-7 (0.10 g). Fraction 3 and fractions 2-3 to 2-7 were combined and chromatographed on a column of silica gel (2 L), (CH₂Cl₂-MeOH, 100:1, 2 L); (CH₂Cl₂-MeOH-glacial AcOH, 100:1:0.1) to give fraction 3-4, N-[3-(acetylthio)-2(R)-(phenylmethyl)-1oxopropyl]-S-[(4-methylphenyl)methyl]-L-cysteine (39a) an oil (0.52 g, 9.5%): $[\alpha]^{26} - 1.3^{\circ}(1.0, M)$. Anal. $(C_{23}H_{27}NO_4S_2 - 0.25)$ $CH_3CO_2H)$ C, H, N.

The N-[3-(acetylthio)-2(S)-(arylmethyl)-1-oxopropyl]-L-amino acids **38b** and **38c** (Table 3) and N-[3-(acetylthio)-2(R)-(arylmethyl)-1-oxopropyl]-L-amino acid **39c** (Table 3) were prepared using the procedure described above.

Method E-2. N-[3-(Acetylthio)-2(S)-(phenylmethyl)-1oxopropyl]-S-[(4-methylphenyl)methyl]-L-cysteine (38a). 1,1-Dimethylethyl ester 36 (0.47 g, 9.5 mmol) in CH₂Cl₂ (10 mL) was treated with trifluoroacetic acid (5 mL) and the mixture stirred 18 h. The reaction mixture was concentrated *in vacuo*, the residue was dissolved in CH₂Cl₂ (25 mL) and concentrated *in vacuo* (twice) and then dissolved in Et₂O (100 mL) and concentrated *in vacuo* (twice) to yield 38a as a colorless oil (0.24 g, 56%): $[\alpha]^{26}$ -58.8° (0.58, M). Anal. (C₂₃H₂₇NO₄S₂) C, H, N.

N-[3-(Acetylthio)-2(R)-(phenylmethyl)-1-oxopropyl]-S-[(4-methylphenyl)methyl]-L-cysteine (**39a**) and <math>N-[2(R,S)-[(acetyl-thio)methyl]-1-oxopentyl]-S-ethyl-L-cysteine (**40d**) were prepared by the procedure above (Table 3).

Method E-3. Preparation of N-[3-Mercapto-2-(alkyl or arylmethyl)-1-oxopropyl] Amino Acids. N-[2(R,S)-(Mercaptomethyl)-1-oxopentyl]-S-ethyl-L-cysteine (30q). Under a nitrogen atmosphere at 0 °C, to acid 40d (0.72 g, 2.24 mmol) in MeOH⁴¹ (20 mL) was added 1 N NaOH (7 mL) and the mixture was stirred at that temperature for 20 h. The mixture was partitioned between EtOAc and 1 N HCl. The dried (MgSO₄) EtOAc solution was concentrated *in vacuo* to give 30q as a yellow oil (0.58 g, 92%): $[\alpha]^{26}$ -34.1° (0.54, M). Anal. (C₁₁H₂₁ NO₃S₂) C, H, N.

N-[3-Mercapto-2(R)-(phenylmethyl)-1-oxopropy]-L-methionine (29n). Acetylthio acid 39b (0.50 g, 1.35 mmol) was treated as described above to give **29n** as a white foam: $[\alpha]^{26}_D$ -38.3° (0.50, E). Anal. (C₁₅H₂₁NO₃S₂O.125H₂O) C, H, N.

Method E-4. N-[3-Mercapto-2(S)-(phenylmethyl)-1oxopropyl]-S-[(4-methylphenyl)methyl]-L-cysteine (28c). Under a nitrogen atmosphere, acetylthio acid 38a (0.217 g, 0.48 mmol) was treated with ammonia in absolute MeOH (37% wt/wt, 100 mL). After 1.5 h, 1 N HCl was added to pH 2. Water (200 mL) was added, and the mixture was extracted with EtOAc. The dried (MgSO₄) EtOAc was concentrated *in vacuo* to give 28c as a colorless oil (0.17 g, 86%): $[\alpha]^{26}$ -32.6° (0.4, M). Anal. (C₂₁H₂₅NO₃S₂0.5H₂O) C, H, N.

N-[3-Mercapto-2(R)-(phenylmethyl)-1-oxopropyl]-S-[(4-methylphenyl)methyl]-L-cysteine (29c); N-[3-mercapto-2(S)-[[1,1'-biphenyl]-4-ylmethyl]-1-oxopropyl]-S-[(4-methylphenyl)methyl]-L-cysteine (28x); and N-[3-mercapto-2(R)-[[1,1'-biphenyl]-4-ylmethyl]-1-oxopropyl]-S-[(4-methylphenyl)methyl]-L-cysteine (29x) were prepared using the procedure described in method E-4 (Table 3).

Method F-1. N-[3-(Benzoylthio)-2(S)-(phenylmethyl)-1-oxopropyl]-L-methioninamide (43a). (S)-Benzoylthio acid 31-S (4.50 g, 15.0 mmol) was converted to acid chloride 32-S (4.75g, 99%) using the procedure described in method E-1. This acid chloride (1.90 g, 5.97 mmol) was added dropwise to L-methioninamide (1.00 g, 5.43 mmol) and triethylamine (1.20 g, 11.94 mmol) in acetonitrile (40 mL) and water (20 mL). After 1 h, the reaction mixture was concentrated *in vacuo* and the residue partitioned between 1 N HCl and EtOAc. The dried (MgSO₄) EtOAc solution was concentrated *in vacuo* to give a white solid (2.33 g), mp 130-144 °C. This solid was recrystallized from EtOAc and dried to give **43a** as a white solid (0.84 g, **36%**): mp 176-180 °C; $[\alpha]^{26}$ -97.6° (0.5, C). Anal. (C₂₂H₂₈N₂O₃S₂) C, H, N.

Method F-2. N-[3-(Benzoylthio)-2(R)-(phenylmethyl)-1-oxopropyl]-L-methioninamide (44a). (R)-Benzoylthio acid 31-R (0.51 g, 1.7 mmol), L-methioninamide hydrochloride (0.31 g, 1.7 mmol), EDC (0.32 g, 1.7 mmol), HOBT (0.23 g, 1.7 mmol), and N-methylmorpholime (0.72 g, 1.7 mmol) in DMF (5 mL) were stirred at room temperature for 18 h. Dilution with water gave a precipitate which was collected, washed with water, and dried to give 44a as a white solid, (0.62 g, 86%): mp 145-149 °C; $[\alpha]^{26}_{D}$ +32.9° (1.0, C). Anal. (C₂₂H₂₆N₂O₃S₂) C, H, N.

Method F-3. N-[3-Mercapto-2(S)-(phenylmethyl)-1oxopropyl]-L-methioninamide (46a). Benzylthio amide 43a (0.30 g, 0.69 mmol) in methanol (25 mL) (purged with nitrogen) was treated with 1 N NaOH (3 mL) (purged with nitrogen). After 1 h, the reaction mixture was concentrated *in vacuo* and treated with 1 N HCl (3 mL). The white precipitate was washed with water to give 46a as a white solid (0.19 g, 79%): mp 133-137 °C; $[\alpha]^{26}$ -8.4° (0.13, E). Anal. (C₁₅H₂₂N₂O₂S₂) C, H, N.

 $\begin{array}{l} N-[3-Mercapto-2(R)-(phenylmethyl)-1-oxopropyl]-L-methioninamide (47a). Benzoylthio amide 44a (0.46 g, 1.1 mmol) was converted by the above method to 47a, a white solid (0.30 g, 83%): mp 135-136 °C; [\alpha]^{26}{}_D -66.3^{\circ} (1.0, C). Anal. (C_{15}H_{22}N_2O_2S_2) C, H, N. \end{array}$

Method G. N-[3-(Acetylthio)-2(S)-(phenylmethyl)-1oxopropyl]-S-[(4-methylphenyl)methyl]-L-cysteinamide (48h) and N-[3-(Acetylthio)-2(R)-(phenylmethyl)-1-oxopropyl]-S-[(4-methylphenyl)methyl]-L-cysteinamide (49h). A solution of acid 7f (1.50 g, 6.3 mmol), HOBT (0.92 g, 6.8 mmol), and EDC (1.30 g, 6.8 mmol) in DMF (10 mL) was treated with S-[(4-methylphenyl)methyl]-L-cysteinamide (1.50 g, 6.7 mmol) and N-methylmorpholine (2.20 mL, 20 mmol). The mixture was stirred 20 h and concentrated in vacuo. The residue was partitioned between EtOAc and water. The dried (MgSO₄) EtOAc solution was concentrated in vacuo and chromatographed on Prep 500 silica gel (2 cartridges) $(CH_2Cl_2-EtOAc, 9:1 (4 L); then 4:1)$ to give fraction 1, (0.02) g); fraction 2 (0.76 g); fraction 3 (0.20 g); and fraction 4, N-[3-(acetylthio)-2(R)-(phenylmethyl)-1-oxopropyl]-S-[(4-methylphenyl)methyl]-L-cysteinamide (49h) as an oil (0.76 g, 26%): $[\alpha]^{26}$ $-1.6^{\circ}(0.5, M)$. Anal. $(C_{23}H_{28}N_2O_3S_2)C, H, N$. Fraction 2 was chromatographed on silica gel to give fraction 2-5, N-[3-(acetylthio)-2(S)-(phenylmethyl)-1-oxopropyl]S-[(4-methylphe-

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nyl)methyl]-L-cysteinamide (**48h**) as an oil (0.23 g, 7.6%): $[\alpha]^{26}{}_D$ –4.2° (0.13, M). Anal. $(C_{23}H_{28}N_2O_3S_2)$ C, H, N.

Method H. N-[3-(Acetylthio)-2(S)-(phenylmethyl)-1oxopropyl]-L-methioninamide (48a) and N-[3-(Acetylthio)-2(R)-(phenylmethyl)-1-oxopropyl]-L-methioininamide (49a). Acid chloride 35 (1.53 g, 5.97 mmol) was added dropwise to a solution of L-methioninamide (1.00 g, 5.43 mmol) and triethylamine (1.15 g, 11.40mmol) in acetonitrile (40 mL) and water (20 mL). After 1 h, 1 N hydrochloric acid (13 mL) was added, and the mixture was extracted with Et_2O . The dried (MgSO₄) Et₂O solution was concentrated in vacuo to give a yellow oil (1.98 g). Chromatography on a silica gel column (500 mL) (MeOH-CH₂Cl₂, 1:24) gave N-[3-(acetylthio)-2(S)-(phenylmethyl)-1-oxopropyl]-L-methioninamide (48a) as a white solid: $(0.33 \text{ g}, 17\%) \text{ mp } 149-151 \,^{\circ}\text{C}; [\alpha]^{26}\text{D} - 87.5^{\circ} (0.5, \text{C}).$ Anal. $(C_{17}H_{24}N_2O_3S_2)$ C, H, N] and N-[3-(acetylthio)-2(R)-(phenylmethyl)-1-oxopropyl]-L-methioninamide (49a) as a white solid (0.30 g, 15%): mp 119–121 °C; $[\alpha]^{26}_{D}$ +5.0° (0.5, C). Anal. $(C_{17}H_{24}N_2O_3S_2)$ C, H, N.

The N-[2(R, S)-[(acetylthio)methyl-3-arylpropanoyl]-L-amino acid amides (**50c-g**) (Table 4) were prepared using method H.

Method J. N-[3-Mercapto-2(S)-(phenylmethyl)-1-oxopropyl]-L-methioninamide (51h). Amide 48h (0.23 g, 0.6 mmol) was treated with methanol saturated with ammonia (100 mL) for 1 h. The reaction mixture was purged with nitrogen for 5 min, treated with 0.1 N HCl (200 mL), and extracted with EtOAc. The dried (MgSO₄) EtOAc was concentrated *in vacuo* to give 51h as a white solid (0.18 g, 75%): mp 130-132 °C; $[\alpha]^{26}$ -4.2° (0.13, M). Anal. (C₂₁H₂₆N₂O₂S₂) C, H, N.

N-[3-Mercapto-2(R)-(phenylmethyl)-1-oxopropyl]-L-methioninamide (52h) (Table 4) was prepared similarly.

Biological Assays. All procedures in this study were preformed under protocols approved by the Schering-Plough Research Institute's Animal Care and Use Committee. Animals were treated as recommended by the NIH Guide for the Care and Use of Laboratory Animals and the Animal Welfare Act in a vivarium certified by the American Association of the Accredidation of Laboratory Animal Care.

In Vitro ACE-Inhibitory Activity. The *in vitro* inhibitory activity was determined by the method of Cushman and Cheung.²⁶ The crude ACE was prepared as described.⁴²

In Vitro NEP-Inhibitory Activity. The effects of compounds were evaluated on purified rabbit kidney NEP EC 24.11 kindly provided by Dr. Phillip Crine (University of Montreal) using 10 mM [³H]Leu-enkaphalin as a substrate according to the method decribed by Chipkin et al.⁴³ and Sybertz et al.²¹

Antihypertensive Activity in DOCA Rat. Male Spague– Dawley rats weighing 100–150 g were prepared, and DOCA-Na hypertension was induced by methods described by Sybertz *et al.*^{21,22} Animals were studied 17–21 days after induction of the hypertension, a time at which the hypertension is well established. Animals with mean blood pressure > 150 mmHg were used. Mean blood pressure was recorded after a 90-min equilibration period by the method of Baum.⁴⁴ Compounds for testing were administered subcutaneously in TRIS buffer (2 mL/kg) or orally as a solution or suspension in 0.4% aqueous methylcellulose solution (volume 4 mL/kg). Groups of four animals were employed, and maximum reduction in blood pressure over 4-h period was determined relative to vehicletreated controls. DOCA-Na rats were nonfasted.

Potentiation of ANF Induced Hypotension. Male SHR (270-300 g) were anesthetized with ether, and the caudal artery was cannulated for direct measurement of blood pressure. A jugular vein was cannulated for iv administration of drugs. After a stabilization period of at least 90 min, animals had regained consciousness and were challenged with ANF 103-125 or 99-126, injected iv as a bolus at a dose of 30 mg/kg, both of which are submaximum with respect to lowering blood pressure in the SHR. At 60-90 min after this first challenge with ANF, the animals were dosed sc with drug or vehicle (0.2% aqueous methylcellulose with 20% Tris-HCl) and rechallenge with ANF 15 min later. Responses to this second challenge with ANF were compared in drug-treated and vehicle

control animals according to the methods of Sybertz et al. 21 and references cited therein.

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