# N<sup>6</sup>-Substituted Adenosine Receptor Agonists. Synthesis and Pharmacological Activity as Potent Antinociceptive Agents<sup>1</sup>

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Received December 30, 1993<sup>⊗</sup>

Novel  $N^6$ -(indol-3-yl)alkyl derivatives of adenosine were synthesized. The adenosine receptor affinity and the antinociceptive activity of these compounds were assessed in binding studies and the phenylbenzoquinone-induced writhing test. Most of these analogues exhibited a potent analgesic activity without side effects. Among them, compound 3c (UP 202-32) bound to  $A_1$  ( $K_1 = 110$  nM) and  $A_2$  ( $K_1 = 350$  nM) adenosine receptors in a specific manner since it did not interact with many other receptors, especially opioid binding sites. The antinociceptive activity in the phenylbenzoquinone assay (ED<sub>50</sub> = 3.3 mg/kg po) was antagonized by 8-cyclopentyl-theophylline, suggesting that an adenosinergic mechanism underlies the analgesic activity observed with this compound. The data obtained with these new N<sup>6</sup>-substituted adenosine receptor agonists emphasize the interest of such compounds in the treatment of pain.

Since the recognition of the hypotensive, sedative, antispasmodic, and vasodilatory actions of adenosine,<sup>2</sup> a number of adenosine analogues have been synthesized and tested. From these studies, compounds such as  $N^6$ cyclohexyladenosine (CHA),  $N^6$ -(R)-( $\overline{1}$ -phenyl-2-propyl)adenosine (R-PIA), and N-ethyladenosine-5'-uronamide (NECA) were issued (Figure 1). More recently, on the basis of observations that morphine enhances adenosine release from the rat cerebral cortex,<sup>3-5</sup> a potential role for adenosine in analgesia has been postulated.6 The antinociceptive potency of various adenosine analogues administered intrathecally in rats and mice has been found to correlate with the affinity for  $A_1$  receptors,  $^{7,8}$ although a delayed phase associated with A2 receptors has also been reported.9 However a number of side effects were observed with the tested analogues (i.e., CHA, R-PIA) at the analgesic doses.<sup>9,10</sup>

The possibility of inducing analgesia without major side effects with such compounds might be of interest in the development of future drugs for pain treatment in humans. Our strategy to prepare new adenosinerelated compounds was then based on the following observations: The structure-activity relationships already published for N<sup>6</sup>-substituted adenosines<sup>11</sup> showed that the interesting area for modulation of the A<sub>1</sub> binding properties is the N<sup>6</sup> region while the activity at the  $A_2$  receptor is largely due to the  $N^6$ -(2-phenylethyl)amino moiety. A heteroarylethyl group such as 1-substituted-3-ethylindole known for its CNS penetration properties<sup>12</sup> would be an interesting pharmacophore to study. We consequently prepared new  $N^6$ -[(indol-3yl)ethyl]adenosine compounds<sup>13</sup> of formula 3a-at (Figure 1) whose binding results to  $A_1$  and  $A_2$  receptors and antinociceptive activities are described in this paper.

## Chemistry

The commercially available inosine was transformed to 6-chloroadenosine according to the known literature

Figure 1.

methods<sup>14,15</sup> involving acetylation of the free alcohols, treatment with POCl<sub>3</sub>, and deacetylation in the presence of ammonia at low temperature. The known 2′,3′-isopropylideneadenosine-5-carboxylic acid<sup>16–20</sup> was converted to the corresponding acid chloride by reacting it with thionyl chloride. Reaction of the latter with

 $R_A = CH_2OH$ 

**NECA** 

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<sup>\*</sup> Abstract published in Advance ACS Abstracts, November 1, 1994.

#### Scheme 1

## Scheme 2

HOCH<sub>2</sub> OH OH 7a-c 
$$R_1$$
  $R_2$   $R_1$   $R_2$   $R_3$   $R_1$   $R_2$   $R_1$   $R_2$   $R_3$   $R_3$   $R_4$   $R_3$   $R_4$   $R_3$   $R_4$   $R_4$   $R_4$   $R_4$   $R_5$   $R_5$ 

# Scheme 3

different amines gave the desired amides 1.13 The title compounds 3a-at were obtained by the reaction of 6-chloro ribofuranuronamide 1 with different substituted (aminoethyl)indoles (method A) and then by the removal of the isopropylidene group in acidic medium, 1 N HCl (method B) or HCOOH (50%) (method C), as outlined in Scheme 1. To obtain compounds 5a-c, 1-substituted-3-(2-aminoethyl)indoles 7a-c were reacted directly on the 6-chloroadenosine 4 according to Scheme 2 (method D). Most of the 1-substituted-3-(2aminoethyl)indoles involved in structures 3a-at and 5a-c were prepared in one step by a new method<sup>21</sup> (method E) starting from the corresponding tryptamines 6 as shown in Scheme 3. The classical reaction scheme involving the intermediate formylindoles was also used to obtain 1-substituted-3-(2-aminoethyl)indoles as outlined in Scheme 4 (methods F-H), especially for compound 7e which failed to give the attempted product

by method E. The physical—chemical properties of the unknown compounds **7a**—**r** are presented in Table 4. The physical—chemical data of intermediates **9a**—**k** and **10a**,**b** obtained by methods F and G are presented in Tables 1 and 2. Compounds **6a**—**e** were prepared according to the literature methods<sup>22,23</sup> as shown in Scheme 5 (method I). The physical—chemical data of compounds **6a**—**e** as well as of the intermediates of method H, **13a**,**b**, **14a**—**e**, and **15a**—**e**, are presented in Table 3. All physical—chemical data and preparation methods for the final products **3a**—**at** and **5a**—**c** are summarized in Table 5. The intermediary compounds **2** were amorphous solids and only identified by their <sup>1</sup>H NMR spectra.

# Structure-Activity Relationships

Pharmacological data of the final products 3a-at are given in Table 6. Each of the compounds was evaluated for  $A_1$  and  $A_2$  adenosine receptor binding affinity and then in the antinociceptive assay.

Analysis of the  $A_1/A_2$  binding results concerning the sugar moiety showed no difference between alcohols and amides (5a,b vs 3a,ao). Among the different amide functions prepared keeping the N<sup>6</sup>-substituent constant (compounds 3a,l,t in the one hand, 3g,aj,am,as,at in the other hand), the highest activity was observed for R = ethyl or cyclopropyl group. Further lengthening of the R group with the introduction of a heteroatom into the side chain (compounds 3t,am) was unpromising and not further exemplified. Concerning the N<sup>6</sup>-substituent region, its importance in the receptor binding studies was also mentioned before. 10,11,32 We studied three levels of modulation on the indole moiety including the substitution on the indole cycle, the modulation on the ethylamino side chain, and the substitution on the nitrogen of the indole.

When one considers compounds 3g,s,w,x,al, which are 5-substituted indoles, with the same N-substituent (5-substituents respectively H, Cl, CH<sub>3</sub>, OCH<sub>3</sub>, SCH<sub>3</sub>), no significant difference was observed. However we noticed that the 2-position substituent's size may influence the activity on both  $A_1$  and  $A_2$  receptors (compounds 3g,y,ah). In this region, hydrogen was the most effective substituent for  $A_1/A_2$  binding affinities. For the 2-methyl group a slight decrease was observed, while for the 2-phenyl substituent the binding affinity decreased dramatically. One can assume that the steric hindrance caused by a phenyl group placed the indolyl moiety in an unfavorable position for binding to the receptor.

A methyl group in both  $\alpha$  and  $\beta$  positions of the ethylamino side chain was tolerated. However, while the  $A_1$  binding affinity was kept at the same level, the  $A_2$  affinity was reduced, thereby improving the  $A_1$  selectivity. For compound 3g, the  $A_1/A_2$  ratio was 4.8, while for compounds 3ae, ag, the same ratio was 12 and 12.5, respectively.

An important variation was the substitution of the indole's nitrogen. This substitution seems to be required for high affinity, and among various alkyl groups, the most efficient was the  $CH_2CH_2N(CH_3)_2$  group (compounds 3h-j,ak). Selectivity was always in favor of  $A_1$  with a  $K_1$  value of 8.3 nM for compound 3j. When the N-substituent contains an aromatic ring (compounds 3o-q,ai,ao), the magnitude was phenyl  $\sim 2$ -pyridyl  $\sim$ 

#### Scheme 4

R<sub>2</sub> CHO
$$\frac{DMF/K_2CO_3}{reflux} + R_1Cl$$

$$8$$

$$9a-k$$

$$R_2 CHO$$

$$\frac{R_3}{CH_2NO_2}$$

$$\frac{AcONH_4}{Method G}$$

$$R_1$$

$$R_2$$

$$R_3$$

$$R_1$$

$$R_3$$

$$R_3$$

$$R_4$$

$$R_3$$

$$R_4$$

$$R_4$$

$$R_3$$

$$R_4$$

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$$R_5$$

$$R_4$$

$$R_4$$

$$R_4$$

$$R_4$$

$$R_5$$

$$R_4$$

$$R_5$$

$$R_6$$

$$R_7$$

$$R_8$$

$$R_9$$

$$R$$

Table 1. Physical Data of Intermediates 9a-k Obtained by Method F

compd	$R_1$	$R_2$	$^{\%}_{ ext{yield}^a}$	$\operatorname{mp}_{(^{\circ}\mathrm{C})^b}$	recryst solvent
9a	CH <sub>2</sub> -(4-chlorophenyl)	Н	78	122	EtOH
9b	CH <sub>2</sub> -phenyl	H	81	111	EtOH
9c	CH <sub>2</sub> -(2,6-dichlorophenyl)	H	85	160	methoxy-
					ethanol
9d	CH <sub>2</sub> CH <sub>2</sub> OCH <sub>3</sub>	H	98	oil	$\boldsymbol{c}$
<b>9</b> e	cyclopentyl	H	60	oil	c
9f	isopropyl	H	90	$\mathrm{oil}^d$	$\boldsymbol{c}$
9g	CH <sub>2</sub> -(4-methylphenyl)	H	88	118	$O-(iPr)_2^e$
9h	CH <sub>2</sub> -(3,4-dimethylphenyl)	H	98	oil	c
<b>9</b> i	CH <sub>2</sub> -(2,5-dimethylphenyl)	$2-CH_3$	57	155	$\boldsymbol{c}$
9j	cyclopentyl	2-CH <sub>3</sub>	25	oil	f
9k	H .	2-phenyl	99	253	H <sub>2</sub> O <sup>e</sup>

<sup>a</sup> Yields are not optimized. <sup>b</sup> Melting points are uncorrected. <sup>c</sup> Purified by column chromatography (SiO<sub>2</sub>, 10% methanol/chloroform). <sup>d</sup> Crystallized on standing, mp ≤50 °C. <sup>e</sup> Crystallization solvent. <sup>f</sup> Purified by column chromatography (SiO<sub>2</sub>, 5% methanol/chloroform).

Table 2. Physical Data of Intermediates 10a-k Obtained by Method G

compd	R <sub>1</sub>	$R_2$	$R_3$	% yield <sup>a</sup>	mp (°C) <sup>b</sup>	recryst solvent
10a	CH <sub>2</sub> -(4-chlorophenyl)	H	H	86	178	CH <sub>3</sub> NO <sub>2</sub> c
10 <b>b</b>	CH <sub>2</sub> -phenyl	H	Н	71	130	$\boldsymbol{c}$
10c	CH <sub>2</sub> -(2,6-dichlorophenyl)	H	H	86	170	$\boldsymbol{c}$
10 <b>d</b>	CH <sub>2</sub> CH <sub>2</sub> OCH <sub>3</sub>	H	H	81	132	$\boldsymbol{c}$
10e	cyclopentyl	H	H	80	oil	d
10 <b>f</b>	isopropyl	H	Н	98	oil	d
10g	CH <sub>2</sub> -(4-methylphenyl)	H	Н	91	172	c
10 <b>h</b>	CH <sub>2</sub> -(3,4-dimethylphenyl)	H	Н	64	135	c
10i	CH <sub>2</sub> -(2,5-dimethylphenyl)	2-CH <sub>3</sub>	CH <sub>3</sub>	85	160	iPrOH <sup>e</sup>
10j	cyclopentyl	$2-CH_3$	CH <sub>3</sub>	69	oil	d
10k	H _	2-phenyl	Н	75	220	iPrOH <sup>e</sup>

 $^{a,b}$  See corresponding footnotes in Table 1.  $^c$  Crystallization in the reaction medium (CH<sub>3</sub>NO<sub>2</sub>).  $^d$  Purified by column chromatography (SiO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>).  $^e$  Crystallization solvent.

3-pyridyl > naphthyl. Different substituents on the aromatic ring were tested, especially in the case of a phenyl ring. No obvious structure—activity relationship

was observed suggesting that this substitution was not critical for the binding affinity (compounds 3a,e-g,n,z,ac,ad,an,ao,aq,ar).

Analysis of the antinociceptive activity concerning the sugar moiety showed that the amide function was better than the alcohols (5a,b vs 3a,ao). The analgesic activity dropped with a ramified or bulky amide function (see 3a,l,t on the one hand and 3g,aj,am,as,at on the other hand). The best function for the antinociceptive effect was an ethyl- or cyclopropyl-substituted amide.

Concerning the  $N^6$ -substituent region, among the substituents on position 5 of the indole (keeping the same N-substitution), the best results were obtained with  $CH_3$  and H (compounds  $3\mathbf{w}$ , $\mathbf{g}$ ). Compounds with  $\mathbf{5}$ -Cl and  $\mathbf{5}$ -OCH $_3$  (compounds  $\mathbf{3s}$ , $\mathbf{x}$ ) were less active. The potency order of the analgesic activity of compounds substituted on the 2-position of the indole was  $H > CH_3 > phenyl$  as observed in the binding studies.

The methyl group in both  $\alpha$  and  $\beta$  positions of the ethylamino side chain has no considerable influence on the analgesic activity which remains very efficient (compounds 3g, ae, ag). The slight modification of the  $A_1/A_2$  ratio, noticed below, seemed not to interfere with the antinociceptive effect.

The substituent nature of the indole's nitrogen plays an important role in the antinociception. The ED<sub>50</sub> value of the unsubstituted compound 3m was 35 mg/ kg, while the methylated compound 3ap was 10-fold more active with an ED<sub>50</sub> value of 3 mg/kg. Compounds substituted with different alkyl groups showed potent analgesic activity (ED<sub>50</sub> < 2 mg/kg), except the cyclopropylmethyl-substituted compound 3ak. Products substituted with unsaturated bond-containing groups, such as allyl or propargyl (3u,v), were slightly less active. while the aromatic ring-containing compounds showed high activities in the major cases. It was however noticed that some compounds (i.e., 3p) had high analgesic activity despite their low binding activity. Concerning the benzyl group, substitutions on the 2-position were most favorable for the analgesic activity (3ao, ar, an, ac) irrespective of the nature of the 5'substituent. From these data, compounds such as 3c,d,g,j,aa,ac emerged as the most active products. The compound 3c (UP 202-32), as an example among them, was further investigated in pharmacological studies, and the first results are given hereunder.

### Scheme 5. Method I

R<sub>3</sub> COOEt 
$$\frac{H_2/Ni \text{ Raney}}{EiOH \, 80^{\circ}C}$$
  $\frac{K_3}{H}$   $\frac{COOEt}{NH_2O}$   $\frac{KOH/H_2O}{H}$   $\frac{NH_2}{HCI/NaNO_2}$   $\frac{NH_2}{H}$   $\frac{13a-e}{HCI \, 10 \, \%}$   $\frac{R_3}{R_2}$   $\frac{R_3}{HCI \, 10 \, \%}$   $\frac{R_3}{R_2}$   $\frac{R_3}{HCI \, 10 \, \%}$   $\frac{R_3}{R_2}$   $\frac{R_3}{HCI \, 10 \, \%}$   $\frac{R_3}$ 

Table 3. Physical Data of Intermediates 13a-e, 14a-e, 15a-e, and 6a-e Involved in Method H

compd	$R_2$	$R_3$	$\%$ yield $^a$	$\operatorname{mp}({}^{\circ}\mathrm{C})^{b}$	recryst solvent
13a	H	CH <sub>3</sub>	57	231	H <sub>2</sub> O <sup>c</sup>
13 <b>b</b>	Cl	H	62	248	$H_2O^c$
13c	$CH_3$	H	52	256	$H_2O^c$
13 <b>d</b>	$OCH_3$	H	29	199	$\mathbf{H_2O}^d$
13 <b>e</b>	$SCH_3$	H	62	228	EtOH
14a	H	$CH_3$	52	175	EtOH
14 <b>b</b>	Cl	H	87	218	$H_2O^c$
14 <b>c</b>	$CH_3$	H	96	175	$H_2O^c$
1 <b>4d</b>	$OCH_3$	H	75	265	$H_2O^c$
1 <b>4e</b>	$SCH_3$	H	59	208	$\mathrm{CH_3CN}^d$
15a	H	$CH_3$	≤100	255	$H_2O^c$
1 <b>5</b> b	Cl	H	≤100	$278^{e}$	$\mathbf{H_2^{-}O^{c,d}}$
1 <b>5c</b>	$CH_3$	H	≤100	254	$H_2O^c$
15d	$OCH_3$	H	≤100	259 - 264	$H_2O^c$
1 <b>5e</b>	$SCH_3$	H	≤100	257	EtOH
6a	H	$CH_3$	78	189 <sup>f</sup>	$iPrOH^g$
6b	Cl	H	72	288 <sup>f</sup>	$\mathrm{Et_2O}^g$
6c	$CH_3$	H	64	286 <sup>f</sup>	$\mathrm{Et_{2}O}^{g}$
6d	$OCH_3$	H	72	120	$\mathrm{Et_{2}O}^{g}$
6e	$SCH_3$	H	67	250 <sup>f</sup>	iPrOH <sup>g</sup>

<sup>a,b</sup> See corresponding footnotes in Table 1. <sup>c</sup> Crystallized in the reaction medium (H<sub>2</sub>O). <sup>d</sup> Purified first by column chromatography (SiO<sub>2</sub>, 10% methanol/chloroform) and then recrystallized. <sup>e</sup> Lit. <sup>22</sup> mp 257–258 °C. <sup>f</sup> Melting point of the hydrochloride. <sup>g</sup> Crystallization solvent.

As is shown in Figure 2, adenosinergic mechanism of action of UP 202-32 was evaluated. The compound bound to  $A_1$  and  $A_2$  receptors and was a selective compound since it did not interact ( $K_i > 10~\mu\mathrm{M}$ ) with adrenergic ( $\alpha_1$ ,  $\alpha_2$ ,  $\beta$ ), muscarinic, dopamine ( $D_1$ ,  $D_2$ ), serotonin (5-HT<sub>1A</sub>, 5-HT<sub>1B</sub>, 5-HT<sub>2A</sub>, 5-HT<sub>2C</sub>), opioid ( $\mu$ ,  $\delta$ ,  $\kappa$ ), histamine ( $H_1$ ) benzodiazepine, GABA-A, and NK<sub>1</sub> receptors. Furthermore, the antinociceptive effect of UP 202-32 ( $3\mathbf{c}$ ) observed in the phenylbenzoquinone assay (ED<sub>50</sub> = 3.3 mg/kg po) appeared to be mediated by an adenosinergic mechanism since it was inhibited by 8-cyclopentyltheophylline (see Figure 2). It will be interesting to see if this compound displays effects at the newly discovered  $A_{2b}$  or  $A_3$  adenosine receptors.

In conclusion, we synthesized new  $N^6$ -(indol-3-yl)alkyl derivatives of adenosine whose adenosine receptor affinity and marked antinociceptive properties without side effects were evidenced. These data suggest a neuromodulator role of adenosine in analgesia and

emphasize the interest of such compounds in the treatment of pain.

## **Experimental Section**

Melting points were determined on an Electrothermal capillary melting point apparatus and are not corrected. The identity of all compounds was confirmed by  $^1\text{H}$  NMR (200 MHz, Bruker AC 200 spectrometer, solvent Me $_2\text{SO-}d_6$  otherwise noted, TMS = 0 ppm), infrared (Perkin-Elmer 298 spectrometer), and microanalytical data (Carlo Erba, elemental analyzer, Model 1106). All reactions were followed by TLC on Merck Silica gel plates (60F-254). Merck silica gel (0.063–0.200 mm) was used for column chromatography. Unless literature references are given, the starting materials were commercially available or prepared according to the literature.  $^{13,21}$  The following methods A–I are described for specific products. However, identical procedures may be applied to analogous compounds.

Method A.  $\beta$ -D-Ribofuranuronamide, 1-[6-[[2-(1-cyclopentylindol-3-yl)ethyl]amino]-9H-purin-9-yl]-N-cyclopropyl-1-deoxy-2,3-O-(1-methylethylidene)- (2) (R = cyclopropyl,  $R_1 = \text{cyclopentyl}$ ,  $R_2 = R_3 = H$ ). A mixture of 20 g (90 mmol) of 1-cyclopentyl-3-(2-aminoethyl)indole (7e) obtained by methods F-H, 21.8 g (58 mmol) of 2',3'-Oisopropylidene-N-cyclopropyl-6-chloropurine-5'uronamide (1) (R = cyclopropyl), 13 27 mL (190 mmol) of triethylamine, and 300 mL of ethanol was refluxed for 6 h under a nitrogen atmosphere. The ethanol was evaporated and 400 mL of dichloromethane added. This solution was washed with brine. Drying (MgSO<sub>4</sub>) the organic layer and evaporation gave a brown oil purified by column chromatography (SiO<sub>2</sub>, 5% methanol/chloroform) to yield 31.9 g (96%) of  $\beta$ -D-ribofuranuronamide, 1-[6-[[2-(1-cyclopentylindol-3-yl)ethyl]amino]-9H-purin-9-yl]-N-cyclopropyl-1-deoxy-2,3-O-(1-methylethylidene)-(R= cyclopropyl;  $R_1$  = cyclopentyl,  $R_2$  =  $R_3$  = H) (2) as an amorphous solid.  $^1H$  NMR (CDCl<sub>3</sub>):  $\delta$  0.26 (m, 2H, CH<sub>2</sub> cyclopropyl), 0.67 (m, 2H, CH<sub>2</sub> cyclopropyl), 1.38 (s, 3H, CH<sub>3</sub> methylidene), 1.63 (s, 3H, CH<sub>3</sub> methylidene), 1.87 (m, 6H, cyclopentyl), 2.2 (m, 2H, cyclopentyl), 2.61 (m, 1H, cyclopropyl), 3.15 (t, 2H, CH<sub>2</sub> ethylamine, J = 7 Hz), 3.99 (m, 2H, CH<sub>2</sub> ethylamine), 4.69 (d, 1H, ribofuranose, J = 1.4 Hz), 4.29 (s. 1H, ribofuranose), 4.57 (s, 1H, ribofuranose), 4.77 (m, 1H, CHcyclopentyl), 5.31 (m, 2H, ribofuranose), 5.97 (d, 1H, ribofuranose, J = 3.1 Hz), 7.15 (m, 4H, 3H indole and 1H purine), 7.39 (d, 1H, indole, J = 8 Hz), 7.56 (m, 1H, NH ethylamine), 7.64 (d, 1H, indole, J = 7.8 Hz), 7.73 (s, 1H, purine), 8.37 (s, 1H, NH amide).

Method B.  $\beta$ -D-Ribofuranuronamide, 1-[6-[[2-(1-cyclopentylindol-3-yl)ethyl]amino]-9H-purin-9-yl]-N-cyclopropyl-1-deoxy- (3c). A mixture of 5.1 g (8.9 mmol) of  $\beta$ -D-ribofuranuronamide, 1-[6-[[2-(1-cyclopentylindol-3-yl)ethyl]-

compd

recryst<sup>c</sup> solvent

formula

 $mp (°C)^b$ 

% yielda

 $R_1$ 

$$R_2 = \begin{cases} 4 & 3 & 8 & R_3 \\ 0 & 1 & 1 & 1 \\ 0$$

$$R_2 = \begin{cases} 4 & 3 & R_3 \\ 7 & N_1 & 9 \end{cases} NH_2$$

prep methods

 $R_2$ 

 $R_3$ 

7a	CH <sub>2</sub> -(4-chlorophenyl)	H	H	FGH	74	212	EtOH	$C_{17}H_{17}ClN_2$ ·HCl
7b	CH <sub>2</sub> -phenyl	H	H	FGH	58	178	iPrOH	C <sub>17</sub> H <sub>18</sub> ClN <sub>2</sub> ·HCl
7c	CH <sub>2</sub> -(2,6-dichlorophenyl)	H	H	FGH	86°	68	c	$\mathrm{C_{17}H_{16}Cl_2N_2}$
7d	CH <sub>2</sub> CH <sub>2</sub> OCH <sub>3</sub>	H	H	FGH	60	oil	c	$C_{13}H_{17}N_2O$
7e	cyclopentyl	H	H	FGH	58	oil	c	$C_{15}H_{20}N_2$
7 <b>f</b>	isopropyl	H	H	FGH	53	oil	c	$C_{13}H_{18}N_2$
7g	CH <sub>2</sub> -(4-methylphenyl)	H	H	FGH	85	oil	c	$C_{18}H_{20}N_2$
$7\bar{\mathbf{h}}$	$CH_{2}$ -(3,4-dimethylphenyl)	H	H	FGH	89	oil	c	$C_{19}H_{22}N_2$
<b>7</b> i	$CH_{2}$ -(2,5-dimethylphenyl)	$2\text{-CH}_3$	H	FGH	61	250	$\mathbf{Et_2O}^d$	$C_{20}H_{24}N_{2}$ ·HCl
<b>7</b> j	cyclopentyl	$2\text{-CH}_3$	H	FGH	90	oil	c	$C_{16}H_{21}N_2$
7k	$CH_{2}$ -(2,5-dimethylphenyl)	H	$9-CH_3$	FGH	47	87	$\mathrm{iPr}_2\mathrm{O}^d$	$C_{16}H_{22}N_2$
71	cyclopentyl	H	9-CH <sub>3</sub>	FGH	85	oil	c	$C_{16}H_{22}N_2$
7m	$CH_{2}$ -(2,5-dimethylphenyl)	H	$8\text{-CH}_3$	$\mathbf{IE}$	61	178	${ m iPrOH}^d$	$C_{20}H_{24}N_{2}$ ·HCl
7 <b>n</b>	CH <sub>2</sub> -(2-methoxyphenyl)	H	H	${f E}$	46	oil	c	$C_{18}H_{20}N_2O$
<b>70</b>	H	2-phenyl	H	GH	73	266	${ m iPrOH}^d$	$C_{16}H_{16}N_2$ ·HCl
7p	CH <sub>3</sub>	H	H	f	53	201	$iPrOH^d$	$C_{11}H_{14}N_2$ ·HCl
$7\bar{\mathbf{q}}$	CH <sub>2</sub> -(2-fluoro-4-bromophenyl)	H	H	E	35	oil	c	$C_{17}H_{16}BrFN_2$
7 <b>r</b>	CH <sub>2</sub> -(2-methylphenyl)	H	H	$\mathbf{E}$	47	175	$i$ PrOH $^d$	$\mathrm{C}_{18}\mathrm{H}_{20}\mathrm{N}_{2}$ $+\mathrm{HCl}$
					1 2 1			

<sup>a</sup> Yields are not optimized and correspond to the recrystallized product in the final step (method H or E). <sup>b</sup> Melting points are uncorrected and correspond to the product for which the formula is given. All compounds were purified by column chromatography on silica gel [elution with 5% isopropylamine/chloroform except for 7d (elution with 10% isopropropylamine/chloroform) and 7n (elution with 10% methanol/dichloromethane)] and used as such if the purity was acceptable. Further purification was achieved on salts (see formula) by recrystallization. d Crystallization solvent. e Reduction according to method H was performed in an ethyl ether (2)/THF (1) mixture instead of THF to prevent dechlorination. f Synthesis according to the literature method (see ref 24).

amino]-9H-purin-9-yl]-N-cyclopropyl-1-deoxy-2,3-O-(1-methylethylidene)- (2) obtained by method A and 110 mL of HCl (1 N) was heated for 3 h at 60 °C. Upon cooling, the mixture was neutralized with a saturated solution of NaHCO3 and then extracted twice with ethyl acetate. The organic layers were washed with water, dried on magnesium sulfate, and concentrated to yield 4.5 g of a brown solid. Purification by chromatography (SiO2, 5% methanol/chloroform) and crystallization in acetonitrile yielded 2.1 g (44%) of  $\beta$ -D-ribofuranuronamide, 1-[6-[[2-(1-cyclopentylindol-3-yl)ethyl]amino]-9Hpurin-9-yl]-N-cyclopropyl-1-deoxy- (3c), mp 141-142 °C. <sup>1</sup>H NMR: δ 0.49 (m, 2H, CH<sub>2</sub> cyclopropyl), 0.72 (m, 2H, CH<sub>2</sub> cyclopropyl), 1.75 (m, 6H, CH2 cyclopentyl), 2.1 (m, 2H, CH2 cyclopentyl), 2.72 (m, 1H, cyclopropyl), 3.04 (m, 2H,  $CH_2$ ethylamine), 3.79 (m, 2H, CH2 ethylamine), 4.14 (m, 1H, ribofuranose), 4.29 (s, 1H, ribofuranose), 4.57 (s, 1H, ribofuranose), 4.82 (m, 1H, CH cyclopentyl), 5.58 (m, 1H, OH ribofuranose), 5.78 (m, 1H, OH ribofuranose), 5.95 (d, 1H, ribofuranose, J = 7.5 Hz), 7.07 (m, 2H, indole), 7.3 (s, 1H, H<sub>2</sub> indole), 7.45 (d, 1H, indole, J = 8 Hz), 7.63 (d, 1H, indole, J =7.2 Hz), 8.15 (m, 1H, NH ethylamine), 8.28 (s, 1H, purine), 8.41 (s, 1H, purine), 8.96 (d, 1H, NH amide, J = 3.7 Hz).

Method C.  $\beta$ -D-Ribofuranuronamide, N-cyclopropyl- $1\text{-}deoxy\text{-}1\text{-}[6\text{-}[[2\text{-}[1\text{-}(2,5\text{-}dimethylbenzyl)indol-}3\text{-}yl]ethyl]\text{-}$ amino]-9H-purin-9-yl]- (3g). A mixture of 19.6 g (31 mmol) of  $\beta$ -D-ribofuranuronamide, N-cyclopropyl-1-deoxy-1-[6-[[2-[1-(2,5-dimethylbenzyl)indol-3-yl]ethyl]amino]-9H-purin-9-yl]-2,3-O-(1-methylethylidene)- obtained by method A and 590 mL of formic acid (50%) was heated for 1.25 h at 70 °C. The excess of formic acid was removed, water was added, and the mixture was concentrated in vacuo. Methanol was then added and the mixture concentrated. The white solid obtained upon triturating in water was filtered and purified by column chromatography (SiO<sub>2</sub>, 10% methanol/chloroform). Crystallization in a mixture of methanol/ethyl ether yielded 12.8 g (69%) of the expected  $\beta$ -D-ribofuranuronamide, N-cyclopropyl-1-deoxy-1-[6-[[2-[1-(2,5-dimethylbenzyl)indol-3-yl]ethyl]amino]-9H-purin-9yl]- (3g) as a monohydrate, mp 130–132 °C. <sup>1</sup>H NMR:  $\delta$  0.51 (m, 2H, CH<sub>2</sub> cyclopropyl), 0.72 (m, 2H, CH<sub>2</sub> cyclopropyl), 2.11 (s, 3H, CH<sub>3</sub>), 2.17 (s, 3H, CH<sub>3</sub>), 2.72 (m, 1H, CH cyclopropyl), 3.05 (t, 2H, CH<sub>2</sub> ethylamine, J = 6.5 Hz), 3.79 (m, 2H, CH<sub>2</sub> ethylamine), 4.15 (m, 1H, ribofuranose), 4.30 (s, 1H, ribofuranose), 4.60 (m, 1H, ribofuranose), 5.28 (s, 2H, CH<sub>2</sub> benzyl), 5.58 (d, 1H, OH ribofuranose, J = 5.9 Hz), 5.79 (d, 1H, OH ribofuranose, J = 3.7 Hz), 5.97 (d, 1H, ribofuranose, J = 7.5Hz), 6.58 (s, 1H, benzyl), 7.1 (m, 5H, indole and benzyl), 7.36 (d, 1H, indole, J = 7.6 Hz), 7.68 (d, 1H, indole, J = 7 Hz), 8.09(m, 1H, NH ethylamine), 8.27 (s, 1H, purine), 8.41 (s, 1H, purine), 8.97 (d, 1H, NH amide, J = 3.6 Hz).

Method D.  $N^8$ -[2-(1-Benzylindol-3-yl)ethyl]adenosine (5b). A mixture of 8 g (28 mmol) of 1-benzyl-3-(2-aminoethyl)indole prepared according to methods F-H, 200 mL of ethanol, 4.2 g (42 mmol) of triethylamine, and 4 g (14 mmol) of 6-chloropurine riboside was refluxed for 6 h. After standing for one night at room temperature, the crystalline product was filtered off and washed with ethanol. Purification by column chromatography (SiO<sub>2</sub>, 10% ethanol/dichloromethane) followed by recrystallization in ethanol gave 4.4 g (63%) of the expected N<sup>6</sup>-[2-(1-benzylindol-3-yl)ethyl]adenosine (5b), mp 158 °C. <sup>1</sup>H NMR:  $\delta 3.05$  (t, 2H, CH<sub>2</sub>, J = 8.1 Hz), 3.7 (m, 2H, CH<sub>2</sub>), 4 (m, 1H, ribofuranose), 4.16 (m, 1H, ribofuranose), 4.6 (q, 1H, ribofuranose, J = 6.2 and 11.4 Hz), 5.22 (d, 1H, OH, J = 5.6Hz), 5.35 (s, 2H, CH<sub>2</sub> benzyl), 5.5 (m, 2H, 2OH), 5.9 (d, 1H, ribofuranose, J = 6.2 Hz), 7.2 (m, 9H), 7.6 (d, 1H), 8.02 (m, 1H, NH), 8.26 (s, 1H, purine), 8.37 (s, 1H, purine).

Method E. 1-(2,5-Dimethylbenzyl)-3-(2-aminoethyl) $indole^{13,21}$  (7) ( $R_1=dimethylbenzyl, R_2=R_3=H$ ). To a solution of tryptamine (32.6 g, 200 mmol) in 300 mL of dry DMF was added 9.2 g (230 mmol) of NaH (60%) under a nitrogen atmosphere. The mixture was stirred at room temperature for 30 min, and 34 mL (230 mmol) of 2,5dimethylbenzyl chloride was added dropwise. The reaction was slightly exothermic. The temperature was then maintained at 50-55 °C for 3 h. Upon cooling, the solution was filtered off and the solid washed with DMF. The organic layer was concentrated; the residue was taken up with 300 mL of chloroform, washed twice with water, dried, and concentrated. A brown oil was obtained. When purified by column chromatography (SiO2, 5% isopropylamine/chloroform) followed by treatment with a 2-propanol/HCl (6 N) solution, it gave 30.4 g (48%) of the expected 1-(2,5-dimethylbenzyl)-3-(2-aminoethyl)indole as a hydrochloride, mp 184 °C.  $^{1}H$  NMR:  $\delta$  2.13 (s, 3H, CH<sub>3</sub>), 2.23 (s, 3H, CH<sub>3</sub>), 3.04 (s, 4H, CH<sub>2</sub>CH<sub>2</sub>), 5.3 (s,

Table 5. Physical Data and Preparation Methods of Compounds 3a-at and 5a-c

compd	$R_1$	$R_2$	$R_3$	R <sub>4</sub>	$rac{ ext{prep}}{ ext{methods}^a}$	$^{\%}$ yield $^{b}$	$(^{\circ}\mathbf{C})^{\circ}$	recryst <sup>d</sup> solvent	formula <sup>e</sup>
<b>5</b> a	CH <sub>2</sub> -(4-chlorophenyl)	Н	Н	CH <sub>2</sub> OH	FGH D	67	181	EtOH	$C_{27}H_{27}ClN_6O_4$
5b	CH <sub>2</sub> -phenyl	H	H	CH <sub>2</sub> OH	FGH D	63	158	$EtOH^f$	$C_{27}H_{28}N_6O_4$
5c	CH <sub>2</sub> -(2,6-dichlorophenyl)	H	H	CH <sub>2</sub> OH	FGH D	58	192	EtOH <sup>f</sup>	$C_{27}H_{26}Cl_2N_6O_4$
<b>3</b> a	CH <sub>2</sub> -(4-chlorophenyl)	H	H	CONH cyclopropyl	FGH AB	57	225	g	$C_{30}H_{30}ClN_7O_4$
3b	CH <sub>2</sub> CH <sub>2</sub> OCH <sub>3</sub>	H	H	CONH cyclopropyl	FGH AB	53	132	CH <sub>3</sub> CN/iPr <sub>2</sub> Og,h	$C_{26}H_{31}N_7O_5$
3c	cyclopentyl	H	H	CONH cyclopropyl	FGH AB	44	141 - 142	$\mathrm{CH_3CN}^{g,h}$	$C_{28}H_{33}N_7O_4$
3d	isopropyl	H	H	CONH cyclopropyl	FGH AB	55	135	$CH_3CN^h$	$C_{26}H_{31}N_7O_4$
<b>3</b> e	CH <sub>2</sub> -(4-methylphenyl)	H	H	CONH cyclopropyl	FGH AB	45	144	$Et_2O$	$C_{31}H_{33}N_7O_4-H_2O$
3f	$CH_2$ -(3,4-dimethylphenyl)	H	H	CONH cyclopropyl	FGH AB	56	134	$\mathrm{Et}_2\mathrm{O}^{g,h}$	$C_{32}H_{35}N_7O_4\cdot 0.9H_2O$
3g	$CH_2$ -(2,5-dimethylphenyl)	H	H	CONH cyclopropyl	EAC	69	130-132	$MeOH/Et_2O^h$	$C_{32}H_{35}N_7O_4 \cdot H_2O$
3h	CH <sub>2</sub> CH <sub>2</sub> -morpholino	H	H	CONH cyclopropyl	EAC	79 [52]	142	iPrOH	C29H36N8O5C6H8O7
<b>3</b> i	$CH_2CH_2N(CH_3)_2$	H	H	CONH cyclopropyl	EAC	64 [66]	130-131	iPrOH	C <sub>27</sub> H <sub>34</sub> N <sub>8</sub> O <sub>4</sub> ·C <sub>6</sub> H <sub>8</sub> O <sub>7</sub>
3 <b>j</b>	CH <sub>2</sub> CH <sub>2</sub> -piperidino	H	H	CONH cyclopropyl	EAC	83 [73]	138	$\mathrm{EtOH}^{h,i}$	C <sub>30</sub> H <sub>38</sub> N <sub>8</sub> O <sub>4</sub> •C <sub>6</sub> H <sub>8</sub> O <sub>7</sub>
3k	CH <sub>2</sub> CH <sub>2</sub> -pyrrolidino	H	H	CONH cyclopropyl	EAC	74 [67]	125	$\mathrm{EtOH}^{h,i}$	C <sub>29</sub> H <sub>38</sub> N <sub>8</sub> O <sub>4</sub> ·C <sub>6</sub> H <sub>8</sub> O <sub>7</sub>
31	CH <sub>2</sub> -(4-chlorophenyl)	H	H	CONHC(CH <sub>3</sub> ) <sub>2</sub> CH <sub>2</sub> OH	FGH AC	31	189	j	$C_{31}H_{34}ClN_7O_5$
$3\mathbf{m}$	H	H	H	CONH cyclopropyl	AC	45	130-131	CH <sub>3</sub> CN <sup>g,h</sup>	C <sub>23</sub> H <sub>25</sub> N <sub>7</sub> O <sub>4</sub> •0.2H <sub>2</sub> O
3n	CH <sub>2</sub> -(3,4-dichlorophenyl)	H	H	CONH cyclopropyl	EAB	38	141	MeOH/Et <sub>2</sub> Og,h	C <sub>30</sub> H <sub>29</sub> Cl <sub>2</sub> N <sub>7</sub> O <sub>4</sub> ·0.8H <sub>2</sub> O
3о	CH <sub>2</sub> -(3-pyridyl)	H	Н	CONH cyclopropyl	EAC	61	239	MeO(CH <sub>2</sub> ) <sub>2</sub> OH	$C_{29}H_{30}N_8O_4\cdot \\ 0.5MeO(CH_2)_2OH$
3p	CH <sub>2</sub> -(1-naphthyl)	H	H	CONH cyclopropyl	EAC	35	146	iPrOH <sup>g</sup>	$C_{34}H_{33}N_7O_4$
3q	CH <sub>2</sub> -(2-pyridyl)	H	H	CONH cyclopropyl	EAC	30	122	$\mathrm{Et}_2\mathrm{O}^{h,k}$	$C_{29}H_{30}N_8O_4$
3r	CH <sub>2</sub> -(4-chlorophenyl)	5-C1	H	CONH cyclopropyl	IE AC	77	154	$\mathrm{H}_2\mathrm{O}^h$	C <sub>30</sub> H <sub>29</sub> Cl <sub>2</sub> N <sub>7</sub> O <sub>4</sub> ·1.1H <sub>2</sub> O
3 <b>s</b>	$CH_2$ -(2,5-dimethylphenyl)	5-Cl	H	CONH cyclopropyl	IE AC	49	139	EtOH	C <sub>32</sub> H <sub>34</sub> ClN <sub>7</sub> O <sub>4</sub> ·1.1H <sub>2</sub> O
3t	CH <sub>2</sub> -(4-chlorophenyl)	H	H	CONHCH <sub>2</sub> CH <sub>2</sub> OCH <sub>3</sub>	IE AC	57	193	$\mathrm{EtOH}^h$	C <sub>30</sub> H <sub>32</sub> ClN <sub>7</sub> O <sub>5</sub>
3u	CH <sub>2</sub> CH=CH <sub>2</sub>	H	H	CONH cyclopropyl	EAC	44	117	m	$C_{26}H_{29}N_7O_4\cdot 0.9H_2O$
3v	$CH_2C=CH_2$	H	H	CONH cyclopropyl	EAC	58	123	m	C <sub>26</sub> H <sub>27</sub> N <sub>7</sub> O <sub>4</sub> ·H <sub>2</sub> O
3w	$CH_2$ -(2,5-dimethylphenyl)	5-CH <sub>3</sub>	H	CONH cyclopropyl	IE AC	52	129	m	C <sub>33</sub> H <sub>37</sub> N <sub>7</sub> O <sub>4</sub> ·0.8H <sub>2</sub> O
3x	CH <sub>2</sub> -(2,5-dimethylphenyl)	5-OCH <sub>3</sub>	H	CONH cyclopropyl	IE AC	53	182	m	C <sub>33</sub> H <sub>35</sub> N <sub>7</sub> O <sub>5</sub> ·0.1H <sub>2</sub> O
3y	$CH_2$ -(2,5-dimethylphenyl)	2-CH <sub>3</sub>	H	CONH cyclopropyl	FGH AC	66	144	pentane <sup>h</sup>	C <sub>33</sub> H <sub>37</sub> N <sub>7</sub> O <sub>4</sub> ·0.7H <sub>2</sub> O
3z	CH <sub>2</sub> -(4-methoxyphenyl)	H	H	CONH cyclopropyl	EAC	72	134	$MeOH^h$	C <sub>31</sub> H <sub>33</sub> N <sub>7</sub> O <sub>5</sub> ·0.8H <sub>2</sub> O
3aa	cyclopentyl	2-CH <sub>3</sub>	H	CONH cyclopropyl	FGH AC	54	140	Et <sub>2</sub> O <sup>h</sup>	C <sub>29</sub> H <sub>35</sub> N <sub>7</sub> O <sub>4</sub>
3ab	н	2-phenyl	H	CONH cyclopropyl	G <sup>n</sup> H AC	42	180	iPrOH	C <sub>29</sub> H <sub>29</sub> N <sub>7</sub> O <sub>4</sub> ·0.4H <sub>2</sub> O
3ac	CH <sub>2</sub> -[2-(N,N-dimethylamino)phenyl]	Н	Н	CONH cyclopropyl	EAC	81	128-129	$\mathrm{Et_2O^h}$	$C_{32}H_{38}N_8O_4$
3ad	CH <sub>2</sub> -(3-nitrophenyl)	H	H	CONH cyclopropyl	EAC	36	129	$\mathrm{Et}_2\mathrm{O}^{h,l}$	C <sub>30</sub> H <sub>30</sub> N <sub>8</sub> O <sub>6</sub> •0.3H <sub>2</sub> O
3ae	CH <sub>2</sub> -(2,5-dimethylphenyl)	H	9-CH <sub>3</sub>	CONH cyclopropyl	FGH AC	46	135	m	$C_{33}H_{37}N_7O_4$
3af	cyclopentyl	Н		CONH cyclopropyl	FGHJ AC	49	130	$\mathrm{Et}_2\mathrm{O}^h$	$C_{29}H_{35}N_7O_4$
3ag	CH <sub>2</sub> -(2,5-dimethylphenyl)	H		CONH cyclopropyl	IºE AC	56	137	Et <sub>2</sub> Og,h	C <sub>33</sub> H <sub>37</sub> N <sub>7</sub> O <sub>4</sub>
	$CH_2$ -(2,5-dimethylphenyl)			CONH cyclopropyl	GHE AC <sup>n</sup>		136	pentane <sup>h</sup>	C <sub>38</sub> H <sub>39</sub> N <sub>7</sub> O <sub>4</sub> ·0.75H <sub>2</sub> O
3ai	CH <sub>2</sub> -(5-chloro-2-thienyl)	H	H	CONH cyclopropyl	EAC	70	137	$\mathrm{Et_2O}^h$	C <sub>28</sub> H <sub>28</sub> ClN <sub>7</sub> O <sub>4</sub> S·H <sub>2</sub> O
3ai	$CH_2$ -(2,5-dimethylphenyl)		H	CONHCH <sub>2</sub> CH <sub>3</sub>	EAC	46	125	CH <sub>3</sub> CN	$C_{31}H_{35}N_7O_4\cdot 0.45H_2O$
	CH <sub>2</sub> -cyclopropyl	Ĥ	H	CONH cyclopropyl	EAC	38	134	MeOH <sup>h</sup>	$C_{27}H_{31}N_7O_4\cdot 0.5H_2O$
3al	$CH_2$ -(2,5-dimethylphenyl)		H	CONH cyclopropyl	IE AC	38	137	MeOH <sup>h</sup>	C <sub>33</sub> H <sub>37</sub> N <sub>7</sub> O <sub>4</sub> S·0.55H <sub>2</sub> O
	$CH_2$ -(2,5-dimethylphenyl)		H	CONHCH <sub>2</sub> CH <sub>2</sub> - morpholino	EAC	69	114	Et <sub>2</sub> O <sup>h</sup>	C <sub>35</sub> H <sub>42</sub> N <sub>8</sub> O <sub>5</sub>
3an	CH <sub>2</sub> -(2-methoxyphenyl)	H	H	CONH cyclopropyl	EAC	61	117	$\mathrm{Et}_2\mathrm{O}^h$	$C_{31}H_{33}N_7O_5$
3ao	CH <sub>2</sub> -phenyl	H	H	CONH cyclopropyl	EAC	47	138	MeOH <sup>h</sup>	C <sub>30</sub> H <sub>31</sub> N <sub>7</sub> O <sub>4</sub> ·H <sub>2</sub> O
3ap	CH <sub>3</sub>	H	H	CONH cyclopropyl	p AC	44	168	CH <sub>3</sub> CN	$C_{24}H_{27}N_7O_4$
3aq	CH <sub>2</sub> -(2-fluoro-4- bromophenyl)	H	Н	CONH cyclopropyl	EAC	57	136	MeOH <sup>h</sup>	C <sub>30</sub> H <sub>29</sub> BrFN <sub>7</sub> O <sub>4</sub> ·H <sub>2</sub> O
3ar	CH <sub>2</sub> -(2-methylphenyl)	H	H	CONH cyclopropyl	EAC	68	177	$CH_3CN^h$	$C_{31}H_{33}N_7O_4$
3as		H	H	CONH <sub>2</sub>	EAC	48	187	$\text{Et}_2\text{O}^h$	C <sub>29</sub> H <sub>31</sub> N <sub>7</sub> O <sub>4</sub>
3at	CH <sub>2</sub> -(2,5-dimethylphenyl)		H	CONHCH <sub>3</sub>	EAC	48	192	CH <sub>3</sub> CN	C <sub>30</sub> H <sub>33</sub> N <sub>7</sub> O <sub>4</sub>

a Preparation methods involve, in the first part, the methods for the preparation of the 1-substituted-3-(2-aminoethyl) indoles 7 used and, in the second part, the preparation method of the compounds 1a,b. b Yields are not optimized and correspond to the recrystallized product in the final step. Values in brackets [X] indicate the percent yield of the salt formation after recrystallization. c Melting points are uncorrected and correspond to the product whose formula is given. d All compounds were purified first by column chromatography (elution with 10% methanol/chloroform, otherwise noted) and then recrystallized if needed. Analyses for C, H, and N were ±0.4% of the expected values for the formula shown. f Column chromatography elution with 10% ethanol/dichloromethane. Column chromatography elution with 5% methanol/chloroform. Crystallization solvent. Column chromatography elution with 20% methanol/chloroform. Two successive column elutions with 10% methanol/chloroform, 20% isopropylamine/chloroform, and 10% methanol/dichloromethane, respectively. Two successive column elutions with 10% methanol/chloroform and then 10% methanol/dichloromethane. Elution with 10% methanol/chloroform. Synthesis of (2-phenyl)indole was achieved according to the literature method (see ref 25). Synthesis of 2-[1-(2,5-dimethylbenzyl)indol-3-yl]propylamine was done according to the literature (see ref 26).

Table 6. Pharmacological Activity of Compounds 3a-at and 5a-c

		$A_1$		A <sub>2</sub>			
	% displa	cement <sup>a</sup>		% displac	ement <sup>a</sup>	analgesic activity, $^b$	
compd	10 <sup>-5</sup> M	10 <sup>-7</sup> M	$K_{i}(nM)$	10 <sup>-5</sup> M	$10^{-7}  \mathrm{M}$	$\mathbf{K}_{i}\left(\mathbf{n}\mathbf{M}\right)$	ED <sub>50</sub> (mg/kg po)
5a	$100 \pm 1$	$71\pm1$	11	$97 \pm 1$	$59 \pm 1$	140	40.8 (27.7-60.2)
5b	$100 \pm 1$	$67 \pm 1$	50	$95 \pm 1$	$30 \pm 2$	250	$> 30 \text{ NDD}^d$
5e	$92 \pm 1$	$15 \pm 3$	$NC^c$	$91 \pm 1$	$35 \pm 2$	NC	inactive
3a	$98 \pm 1$	$46 \pm 2$	50	$91 \pm 1$	$30 \pm 2$	320	0.9(0.6-1.5)
3b	$98 \pm 1$	$61 \pm 2$	23	$90 \pm 1$	$19 \pm 1$	370	1.2(0.7-2.2)
3c	$96 \pm 1$	21*	110	90*	$28 \pm 2$	350	1.5(1.1-2.1)
3d	$87 \pm 2$	$4\pm8$	120	$91 \pm 2$	$14 \pm 3$	390	1.6(0.8-3.4)
3e	$97 \pm 1$	$37 \pm 10$	130	$89 \pm 4$	$21 \pm 6$	340	<3 NDD
3 <b>f</b>	$96 \pm 1$	$29 \pm 1$	160	$83 \pm 1$	$13 \pm 5$	820	>30 NDD
3g	$93 \pm 1$	$26 \pm 3$	140	$85 \pm 3$	$9 \pm 4$	670	2.4(1.8-3.3)
$3\mathbf{h}^e$	$100 \pm 5$	$52 \pm 4$	NC	$88 \pm 3$	$20 \pm 2$	NC	>30 <sup>f</sup>
$3i^e$	$100 \pm 1$	$83 \pm 4$	NC	$95 \pm 3$	$51 \pm 6$	NC	>60 NDD
$3\mathbf{j}^e$	$99 \pm 1$	$90 \pm 4$	8.3	$96 \pm 1$	$44 \pm 1$	120	$7.3(3.1-17.1^g)$
$3\mathbf{k}^e$	$100 \pm 3$	$96 \pm 2$	7.5	$95 \pm 1$	$53 \pm 5$	58	44.2 (28.8-67.9)
<b>3</b> l	$79 \pm 3$	$5 \pm 2$	NC	$50 \pm 8$	0*	NC	inactive
3m	$100 \pm 1$	$91 \pm 1$	7.9	$68 \pm 2$	$29 \pm 1$	560	34.9 (20.2-60.3)
3 <b>n</b>	$100 \pm 1$	$52 \pm 1$	140	$81 \pm 1$	$24 \pm 5$	1400	>30 NDD
<b>30</b>	97*	$78 \pm 5$	26	$98 \pm 1$	$30 \pm 9$	183	>30 NDD
3 <b>p</b>	$93 \pm 2$	$20 \pm 5$	370	$76 \pm 6$	$8\pm1$	1400	2.4 (1.6-3.6)
3q	$96 \pm 3$	$73 \pm 4$	39	$94 \pm 1$	$28 \pm 3$	390	<3 NDD
3r	$100 \pm 6$	$49 \pm 4$	NC	$79 \pm 2$	$23 \pm 2$	NC	inactive
3s	$99 \pm 3$	$47 \pm 9$	150	$77 \pm 1$	$10 \pm 4$	660	6.3 (4.7-8.5)
3t	$90 \pm 4$	$13 \pm 8$	890	$36 \pm 4$	$0 \pm 2$	>10000	inactive
3u	100*	$56 \pm 5$	39 NG	$80 \pm 3$	$17 \pm 7$	280	5.3 (2.7–10.7)
3v	$95 \pm 1$	$64 \pm 5$	NC 62	$83 \pm 0$	$25\pm7 \ 3\pm7$	NC	<10 NDD
3w	$100 \pm 2$	$\begin{array}{c} 50\pm2 \\ 16\pm2 \end{array}$	NC	$\begin{array}{c} 70\pm1 \\ 91\pm3 \end{array}$	$3\pm7$ $17\pm1$	890 NC	0.9(0.6-1.4)
3x	$100 \pm 2 \\ 56 \pm 8$	$\begin{array}{c} 16 \pm 2 \\ 5 \pm 4 \end{array}$	NC NC	$60 \pm 3$	$7\pm 2$	NC NC	46.3 (30.2-71.1) 1.2 (0.4-4.3)
3y 3z	$98 \pm 3$	$\begin{array}{c} 3\pm 4 \\ 42\pm 2 \end{array}$	NC NC	$100 \pm 3$	$\frac{7 \pm 2}{59 \pm 3}$	NC NC	71.6 (15.2-338)
3aa	$\begin{array}{c} 36 \pm 3 \\ 77 \pm 6 \end{array}$	$\begin{array}{c} 42 \pm 2 \\ 15 \pm 5 \end{array}$	895	$92 \pm 1$	$6\pm 1$	2180	0.5 (0.2-1.2)
3ab	$84\pm1$	$\frac{10 \pm 0}{20 \pm 9}$	NC	$\begin{array}{c} 32 \pm 1 \\ 71 \pm 14 \end{array}$	$10 \pm 5$	NC	27.1 (17.2-42.8)
3ac	$99\pm1$	$33 \pm 6$	170	$\begin{array}{c} 71\pm14\\ 92\pm7 \end{array}$	$5\pm7$	760	1.9(0.8-4.4)
3ad	$100 \pm 2$	$63 \pm 0$	29	$99\pm1$	$46\pm4$	182	<10 NC
3ae	$96 \pm 2$	$21\pm2$	180	$84\pm1$	$9\pm2$	2200	3.6 (2.1-6.3)
3af	$96 \pm 2$	$45 \pm 6$	NC	$85\pm1$	$21\pm7$	NC	2.2 (1.2-4.0)
3ag	$97 \pm 4$	$28\pm2$	200	$61 \pm 5$	$0\pm7$	2500	0.9 (0.4-2.0)
3ah	$31 \pm 4$	$6 \pm 3$	NC	$23 \pm 6$	$0\pm4$	NC	>30 NDD
3ai	$100 \pm 2$	$47 \pm 4$	NC	$82\pm1$	$6\pm 2$	NC	26.2 (9.8-69.9)
3aj	$92\pm4$	$38 \pm 6$	160	$86\pm2$	$12\pm 5$	660	<2 NDD
3ak	$97 \pm 1$	$45\pm3$	67	$87 \pm 2$	$25\pm3$	190	>10 NDD
3al	$82 \pm 6$	$13 \pm 3$	NC	$76 \pm 6$	$8 \pm 6$	NC	>10 NDD
3am	$5\pm6$	$5 \pm 5$	NC	$19 \pm 2$	$0\pm7$	NC	>60
3an	$100 \pm 2$	$42 \pm 4$	170	$89 \pm 1$	$23 \pm 2$	810	2.3(1.4-3.9)
3ao	$99 \pm 2$	$35 \pm 1$	40	$94 \pm 1$	$41 \pm 3$	270	18.1 (10.3-31.7)
Зар	$99 \pm 1$	$76 \pm 3$	14	$91 \pm 2$	$29 \pm 3$	270	3 (0.7-7.1)
3aq	$96 \pm 2$	$40 \pm 3$	NC	$92 \pm 3$	$23\pm4$	NC	inactive
3ar	$98 \pm 1$	$40 \pm 1$	NC	$93 \pm 1$	$27 \pm 3$	NC	<3
3as	$87 \pm 2$	$17 \pm 3$	510	$61 \pm 1$	$0 \pm 1$	2600	>30
3at	$80 \pm 2$	$12 \pm 5$	1400	$68 \pm 2$	$0 \pm 5$	3000	10.2 (5.4 - 15.2)

<sup>a</sup> Values are the mean ± SEM of three determinations except for values marked with an asterisk where only one determination was performed (see the Experimental Section). <sup>b</sup> ED<sub>50</sub> in phenylbenzoquinone-induced writhing test with 95% fiducial limits (see the Experimental Section). CNC: not calculated. NDD: not dose dependent. The ED50 value was not calculable by the linear regression method, although 50% inhibition was observed at the indicated dose. Affinity values for compounds 3h-k were given on salts. The affinity of the corresponding base is as follows. Compound 3h:  $A_1 (10^{-5} M) 98 \pm 1 (10^{-7} M) 63 \pm 3$ ;  $A_2 (10^{-5} M) 90 \pm 1 (10^{-7} M) 12 \pm 4$ . Compound 3i:  $A_1 (10^{-5} M) 100 \pm 2 (10^{-7} M) 91 \pm 1$ ;  $A_2 (10^{-5} M) 94 \pm 1 (10^{-7} M) 44 \pm 1$ . Compound 3i:  $A_1 (10^{-5} M) 99 \pm 2 (10^{-7} M) 94$  $\pm$  2;  $A_2$  (10<sup>-5</sup> M) 92  $\pm$  1 (10<sup>-7</sup> M) 46  $\pm$  3. Compound 3k:  $A_1$  (10<sup>-5</sup> M) 100  $\pm$  1 (10<sup>-7</sup> M) 82  $\pm$  10;  $A_2$  (10<sup>-5</sup> M) 92  $\pm$  1 (10<sup>-7</sup> M) 69  $\pm$  1 / Value for the base. § Activity for the salt as citrate. The result for an oxalate was  $ED_{50} = 10.2$  (7.1–14.6).

2H, CH<sub>2</sub>N indole), 6.59 (s, 1H), 7.1 (m, 5H), 7.4 (d, 1H, J =7.8 Hz), 7.6 (d, 1H, J = 7.5 Hz), 8.1 (s, 3H, NH<sub>2</sub> and HCl).

Method F. 1-Cyclopentyl-3-formylindole (9e) ( $R_1$  = cyclopentyl,  $R_2 = H$ ). A mixture of 20 g (140 mmol) of 3-formylindole, 15.6 mL (150 mmol) of chlorocyclopentane, 100 mL of DMF, and 20.7 g (150 mmol) of K2CO3 was refluxed for 2 h. Upon cooling, the solid was filtered off and washed with DMF and the DMF phase concentrated. The residue was taken up with chloroform, washed twice with water, dried, and concentrated to give a brown oil which was purified by column chromatography (SiO<sub>2</sub>, 10% methanol/chloroform) to yield 17.8 g (60%) of the expected 1-cyclopentyl-3-formylindole as an oil.  $^{1}$ H NMR (CDCl<sub>3</sub>):  $\delta$  1.8 (m, 6H, CH<sub>2</sub> cyclopentyl), 2.2 (m, 2H,

CH<sub>2</sub> cyclopentyl), 4.8 (m, 1H, CH cyclopentyl), 7.4 (m, 2H, indole), 7.81 (s, 1H, H2 indole), 8.3 (m, 1H, indole), 9.99 (s, 1H,

Method G. 1-Cyclopentyl-3-(2-nitrovinyl)indole (10e)  $(\mathbf{R}_1 = \mathbf{cyclopentyl}, \mathbf{R}_2 = \mathbf{R}_3 = \mathbf{H}).$  1-Cyclopentyl-3-formylindole (9e) (17.8 g, 83 mmol) prepared according to method F was mixed with 85 mL (1.6 mol) of nitromethane and 5 g (65 mmol) of ammonium acetate. The mixture was refluxed for 30 min and excess of nitromethane evaporated. The residue was taken up with dichloromethane, washed with water, dried, concentrated, and then purified by column chromatography (SiO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>) to yield 16.8 g (80%) of 1-cyclopentyl-3-(2-nitrovinyl)indole (10e) as an oil.  $^1$ H NMR (CDCl<sub>3</sub>):  $\delta$  1.9 (m, 6H, CH<sub>2</sub> cyclopentyl), 2.21 (m, 2H, CH<sub>2</sub> cyclopentyl), 4.7 (m,

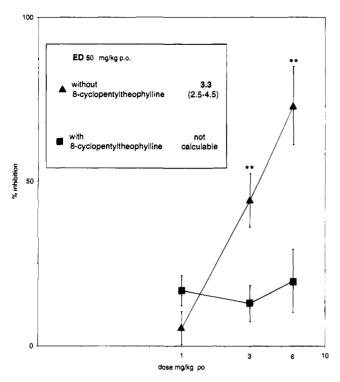


Figure 2. Effect of 8-cyclopentyltheophylline on UP202-32 (3c)-induced antinociception in the phenylbenzoquinone-induced writhing assay. n = 6 (see the Experimental Section). \*\*p < 0.01 as compared to the 8-cyclopentyltheophylline-treated group.

1H, CH cyclopentyl), 7.35 (m, 3H, indole), 7.58 (s, 1H,  $H_2$  indole), 7.67 (m, 2H, vinyl and indole), 8.2 (d, 1H, vinyl, J = 13.3 Hz).

Method H. 1-Cyclopentyl-3-(2-aminoethyl)indole (7e). 1-Cyclopentyl-3-(2-nitrovinyl)indole (10e) (16.8 g, 66 mmol) prepared according to method G and 100 mL of THF were added dropwise to a mixture of 13.7 g (360 mmol) of LiAlH4 and 100 mL of THF. The reaction was exothermic. Upon complete addition, the mixture was refluxed for 1.5 h. A saturated solution of  $H_2O/Na_2SO_4$  was added dropwise at 5 °C to destroy the excess of LiAlH4. The resulting mixture was filtered on Celite and extracted with ethyl acetate. The organic layers were concentrated, and the residue was chromatographed (SiO<sub>2</sub>, 5% isopropylamine/chloroform) to give 8.8 g (58%) of 1-cyclopentyl-3-(2-aminoethyl)indole (7e) as a yellow oil. <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 1.56 (s, 2H, NH<sub>2</sub>), 1.85 (m, 6H, CH<sub>2</sub> cyclopentyl), 2.17 (m, 2H, CH<sub>2</sub> cyclopentyl), 2.95 (m, 4H, CH<sub>2</sub>- $\tilde{CH}_2\tilde{N}$ ), 4.75 (m, 1H, CH cyclopentyl), 7.16 (m, 3H, indole), 7.37 (d, 1H, indole, J = 8.3 Hz), 7.59 (d, 1H, indole, J = 7.7 Hz).

Method I. Ethyl 2-Oxo-5-methylpiperidine-3-carboxylate (12b) ( $\mathbf{R}_3=\mathbf{CH}_3$ ). A mixture of 26.7 g (117 mmol) of ethyl 2-carbethoxy-4-cyanopentanoate (11b), <sup>26</sup> 500 mL of ethanol, and 1 g of Raney nickel was heated for 6 h under H<sub>2</sub> pressure (80 kg) at 80 °C. The catalyst was filtered on Celite and washed with methanol, and the organic layers were concentrated. Crystallization in light petroleum gave 17.7 g (82%) of ethyl 2-oxo-5-methylpiperidine-3-carboxylate (12b) as a white solid, mp 108 °C (lit. <sup>26</sup> mp 100-101 °C). <sup>1</sup>H NMR: δ 0.92 (d, 3H, CH<sub>3</sub>, J=5.8 Hz), 1.18 (t, 3H, CH<sub>3</sub> ester, J=7.05 Hz), 1.6 (m, 1H), 1.9 (m, 2H), 2.78 (m, 1H), 3.13 (m, 1H), 3.3 (m, 1H), 4.09 (q, 2H, CH<sub>2</sub> ester, J=7.05 Hz), 7.73 (s, 1H, NH).

**2,3-Dioxo-5-methylpiperidine 3-Phenylhydrazone (13a)** ( $\mathbf{R_2} = \mathbf{H}, \ \mathbf{R_3} = \mathbf{CH_3}$ ). Ethyl 2-oxo-5-methylpiperidine-3-carboxylate (12b) 36.5 g (200 mmol) prepared as above was dissolved in 500 mL of water containing 14 g (250 mmol) of potassium hydroxide. The solution was stirred at room temperature overnight. This solution cooled in an ice bath was treated with a solution of benzene diazonium chloride (prepared as follows: 24 g (0.26 mol) of aniline was mixed with 400 mL of water and 60 mL of concentrated HCl. The mixture was cooled in an ice bath, and a solution of 20 g (0.29 mol) of

NaNO<sub>2</sub> and 500 mL of water was introduced dropwise and the mixture stirred for 30 min. The pH of the solution was adjusted to 4.5 with a 10% solution of Na<sub>2</sub>CO<sub>3</sub> (250 mL) before mixing with the benzene diazonium chloride solution). The pH of the resulting solution was adjusted to pH 5 by the addition of acetic acid. Stirring was continued for 4 h at 0 °C. The orange solid obtained was filtered and dried to yield 24.9 g (57%) of the expected phenylhydrazone 13a, mp 231 °C (lit.  $^{26}$  mp 238–239 °C).

4-Methyl-1-oxo-1,2,3,4-tetrahydro-β-carboline (14a) ( $\mathbf{R}_2$  =  $\mathbf{H}$ ,  $\mathbf{R}_3$  =  $\mathbf{CH}_3$ ). The phenylhydrazone 13a prepared as above (24.9 g, 115 mmol) in 500 mL of formic acid (90%) was boiled under reflux for 1 h. The mixture was diluted with 200 mL of water, and hot ethanol was added until the brown oil separated in the solution. Upon cooling, the β-carboline obtained was filtered off and recrystallized from ethanol to yield 11 g (48%) of 4-methyl-1-oxo-1,2,3,4-tetrahydro-β-carboline (14a), mp 175 °C (lit.²6 mp 204–206 °C). ¹H NMR: δ 1.31 (d, 3H, CH<sub>3</sub>, J = 6.5 Hz), 3.24 (m, 2H, CH<sub>2</sub>), 3.58 (m, 1H, CH), 7.05 (t, 1H, J = 7 and 7.7 Hz), 7.21 (t, 1H, J = 7 and 7.7 Hz), 7.4 (d, 1H, J = 8 Hz); 7.58 (s, 1H, NH), 8 (d, 1H, J = 8 Hz), 11.6 (s, 1H, NH indole).

3-(2-Amino-1-methylethyl)indole-2-carboxylic Acid (15a) ( $\mathbf{R_2} = \mathbf{H}, \ \mathbf{R_3} = \mathbf{CH_3}$ ). 4-Methyl-1-oxo-1,2,3,4-tetrahydro- $\beta$ -carboline (14a) (11.9 g, 60 mmol) prepared as above was added portionwise to a mixture of 26.6 mL of ethanol (50%) and 29.3 g (520 mmol) of KOH. The resulting mixture was refluxed for 6 h and kept at room temperature for a night. The solvent was then removed in vacuo and 100 mL of water added. The solution was filtered and acidified with acetic acid. The solid obtained was washed with water, ethanol, and ethyl ether and then dried to yield 13.1 g (100%) of the expected acid, mp 255 °C (lit.  $^{26}$  mp 242—243 °C).  $^{1}$ H NMR:  $\delta$  1.51 (d, 3H, CH<sub>3</sub>, J = 7.3 Hz), 3.1 (m, 2H, CH<sub>2</sub>), 4.03 (m, 1H, CH), 6.93 (t, 1H, J = 7.1 and 7.7 Hz), 7.07 (t, 1H, J = 7.7 and 7.1 Hz), 7.36 (d, 1H, J = 8 Hz), 7.64 (d, 1H, J = 8 Hz), 11.04 (s, 1H, NH indole).

3-(2-Amino-1-methylethyl)indole (6a) ( $\mathbf{R_2}=\mathbf{H}, \mathbf{R_3}=\mathbf{CH_3}$ ). 3-(2-Amino-1-methylethyl)indole-2-carboxylic acid (15a) (13.4 g, 0.06 mol) prepared as above was refluxed with 10% hydrochloric acid (440 mL) for 2 h. Upon cooling, the solution was made alkaline with 30% sodium hydroxide, extracted twice with ether, dried, and concentrated. The brown oil was treated with 9 mL of a 6 N 2-propanol/hydrochloric acid solution to yield 9.9 g (78%) of the expected 3-(2-amino-1-methylethyl)indole (6a) hydrochloride as a white solid, mp 189 °C (lit. 26 mp 224-226 °C for the picrate). 1H NMR:  $\delta$  1.38 (d, 3H, CH<sub>3</sub>, J = 6.8 Hz), 2.98 (m, 1H, CH), 3.37 (m, 1H, CH), 7.09 (m, 2H), 7.24 (s, 2H, H<sub>2</sub> indole), 7.38 (d, 1H, J = 7.9 Hz), 7.62 (d, 1H, J = 7.6 Hz), 8.14 (s, 3H, NH<sub>2</sub> and HCl), 11.09 (s, 1H, NH indole).

**Pharmacology.** The compounds were dissolved in water if available as soluble salts or suspended in an aqueous 1% gum arabic, 0.1% sodium chloride, 0.001% Tween 80 solution when available as free bases. They were administered orally or intraperitoneally under a volume of 0.5 mL/20 g and 0.2 mL/20 g, respectively.

Phenylbenzoquinone-Induced Writhing in Mice. The nociceptive reaction was induced following the method of Siegmund et al.31 One hour after oral administration of the test compound, the mice received intraperitoneally 0.20-0.24 mL of a hydroalcoholic 0.02% phenylbenzoquinone solution. The number of nociceptive reactions (writhings and stretches) were counted from the 5th to the 10th minute. In the protocol of analgesic activity inhibition with 8-cyclopentyltheophylline, used as an A<sub>1</sub> selective adenosine receptor antagonist, an ineffective analgesic dose of 8-cyclopentyltheophylline (10 mg/ kg ip) was administered 30 min after the oral administration of UP 202-32 (3c). Groups of six animals were used for treated and control groups. For each treated group, the inhibition of painful reactions was calculated from unpaired values in comparison with the mean control value. The ED<sub>50</sub> value (dose for which the nociceptive reaction was decreased by 50%) was determined by linear regression on quantitative values.

Binding Studies. A<sub>1</sub> Binding Assays. Preparation of Whole Rat Brain Membranes. Rat brain membranes were obtained according to a previous method.<sup>27</sup> Rats were decapi-

tated and whole brains rapidly removed at 4 °C. They were rapidly washed in ice-cold saline solution, dried, and weighed. They were homogenized in 25 volumes of ice-cold incubation buffer (Tris-HCl, 50 mM, pH 7.4) using an Ultra-Turrax homogenizer and centrifuged at 1090g for 10 min at 4 °C. The supernatant was centrifuged a second time at 48000g for 20 min at 4 °C. The pellet was resuspended in 4 volumes of icecold incubation buffer and homogenized with the Ultra-Turrax homogenizer. The suspension was incubated with adenosine deaminase (ADA) (1 U/mL of suspension) for 30 min at room temperature with gentle stirring and centrifuged as before. The pellet was resuspended in 10 volumes of ice-cold incubation buffer and homogenized with the Ultra-Turrax ice-cold homogenizer. The suspension was stored at 4 °C until required or stored at -20 °C. Receptor binding studies were carried out as previously described by Schwabe and Trost,27 with slight modifications. Incubations (2 mL) were performed at 20 °C for 30 min in poly(styrene) tubes containing incubation buffer, 1 mL of the tissue preparation, 2.5 nM [3H]PIA, and various concentrations of the competing drugs. Nonspecific binding was defined as that remaining in the presence of 10  $\mu$ M PIA. The reaction was terminated by rapid filtration of the solution through Whatman GF/B glass fiber filters, and the latter were washed three times with incubation buffer. The radioactivity was counted in a liquid scintillation  $\beta$  counter with 47% efficiency. Each assay was performed in triplicate.

A2 Binding Assays. Preparation of Rat Striatal Membranes. Rat striatal membranes were obtained according to a previously described method.28 Briefly, the striata were disrupted in 10 volumes of ice-cold homogenization buffer (Tris-HCl, 50 mM, MgCl<sub>2</sub>, 10 mM, pH 7.7) using an Ultra-Turrax homogenizer and centrifuged at 48340g for 10 min at 4 °C. These operations were repeated on the resulting pellet. Before the centrifugation, the suspension was incubated with ADA (1 U/mL of suspension) for 30 min at room temperature, with gentle stirring. The pellet was resuspended in 5 volumes of ice-cold homogenization buffer and homogenized with the Ultra-Turrax homogenizer. This suspension was stored frozen at -80 °C. On the day of use, it was thawed to room temperature and resuspended in 15 more volumes of ice-cold homogenization buffer. It was further homogenized with the Ultra-Turrax homogenizer and then with the glass-Teflon potter. Receptor binding studies were carried out as previously described by Bruns et al.28 with slight modifications. Incubations (2 mL) were performed at 25 °C for 60 min in poly-(styrene) tubes containing incubation buffer (Tris-HCl, 50 mM,  $MgCl_2$ , 10 mM, cyclopentyladenosine, 0.11  $\mu$ M, pH 7.7), 1 mL of the tissue preparation, 50 μL of 4 nM [3H]NECA, and various concentrations of the test compounds. Nonspecific binding was defined as that remaining in the presence of 5 μM NECA. The reaction was terminated by rapid filtration of the solution through Whatman GF/B glass fiber filters, and the latter were washed three times with washing buffer (Tris-HCl, 50 mM, MgCl<sub>2</sub>, 10 mM, pH 7.7). The radioactivity was counted in a liquid scintillation  $\beta$  counter with 47% efficiency. Each assay was performed in triplicate.

Data Analysis. Competition data were analyzed using the nonlinear regression program LIGAND<sup>29</sup> adapted for an IBM-PC<sup>30</sup> and obtained from Elsevier-Biosoft (Cambridge, England). The concentration of unlabeled drug causing 50% displacement of the radioligand from its binding site (IC<sub>50</sub> value) was calculated by log-logit linear regression analysis of data (EBDA). Then, the latter were analyzed by a nonlinear regression program (LIGAND) assuming a model of one binding site. The inhibition constant  $(K_i)$  value was calculated according to the Cheng-Prusoff equation.<sup>33</sup> Each  $K_i$  value was determined from one experiment, each assay being performed in triplicate.

Acknowledgment. The technical assistance of A. Fouquet, J. Goyard, C. Gouedard, A. Gourvil, S. Blanchot, M. Bineau, Q. Kodjo, E. Boisserie, R. Billiras, V. Levenez, C. Delchambre, L. Pentecouteau, C. Antoine, M. L. Prunet, S. Cario, S. O'Sullivan, and R. Dudley and

the secretarial assistance of M. F. De Oliveira are gratefully acknowledged.

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