Pyridobenzoxazepine and Pyridobenzothiazepine Derivatives as Potential Central Nervous System Agents: Synthesis and Neurochemical Study

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In order to characterize the pharmacological profile of the different chemical classes of pyridobenzazepine derivatives, a series of N-methylpiperazinopyrido $[1,4]$ - and $-[1,5]$ - benzoxaand benzothiazepine derivatives were prepared. The affinities for $D_2, D_1, 5-HT_2$, and cholinergic (M) receptors were measured. In comparison to dibenzazepine reference compounds, a strong decrease of the affinities was observed, less pronounced, however, for the substituted analogues. Oxazepine and thiazepine analogues like clozapine (except 8-chloro-6-(4-methylpiperazin-l-yl) pyrido[2,3-6][l,4]benzoxazepine (9) and 8-chloro-6-(4-methylpiperazin-l-yl)pyrido[2,3-6][l,4] benzothiazepine (11)) were found to be inactive against apomorphine stereotypies. In the openfield test in rats, different molecules showed a high disinhibitory activity as observed with anxiolytic drugs. Moreover, 8-chloro-5-(4-methylpiperazin-l-yl)pyrido[2,3-6][l,5]benzoxazepine (14) presented a clozapine-like profile that was confirmed in the behavioral model in dogs and showed most of the behavioral characteristics described for antipsychotic drugs. Its neurochemical profile, in particular the 5-HT₂/D₂ ratio, was also compatible with atypical antipsychotic activity.

Introduction

Dibenzazepine derivatives have represented and still represent today an interesting and exciting field of investigation in medicinal chemistry.1-16 A slight modification of the tricyclic structure leads to a profound alteration of the activity profile. Indeed, clothiapine (1) is a typical neuroleptic¹⁷ with a strong propensity to induce extrapyramidal symptoms (EPS), while clozapine (2) and fluperlapine (3) show an atypical antipsychotic profile18-21 and produce only minimal EPS. Clozapine, although

possessing clear antipsychotic activity, has also revealed some disinhibitory or anxiolytic effects not only in animals²²⁻²³ but also in humans.²⁴⁻²⁵ The mixed $5.\mathrm{HT}_2$ / D_2 antagonism of clozapine could be implicated in these bivalent activities since either the $5-HT_2$ antagonism or a combination of the $5-HT_2$ and D_2 affinities seems particularly important to improve the negative symptoms of schizophrenia.²⁶⁻²⁸ Moreover, it is well known that the $5-\text{HT}_2$ antagonism is also characteristic of some anxiolytics like ritanserin. The $5-HT_2/D_2$ concept has been widely developed and has contributed to the development of many works.29-33

Although clozapine represents a great advance in the treatment of psychiatric disorders, the induction of severe toxic effects (granulocytopenia, agranulocytosis,34-36 and seizures,³⁷ etc.) has stimulated the search for an alternative to clozapine. This challenge is currently the focus of the most active efforts in contemporary neuropharmacology.^{21,38} This context led us to synthesize and evaluate a series of N -methylpiperazinopyrido[1,4]- and -pyrido[1,5]benzazepine derivatives. The study of N -methylpiperazinopyridobenzodiazepines has already demonstrated¹⁶ that diarylazepine structural modifications could lead to new promising clozapine-like drugs. Parallel to this study, an investigation of an analogous series of pyridobenzoxazepines and pyridobenzothiazepines was undertaken, the results of which are reported below. Biological studies were conducted through *in vitro* or *in vivo* models.

The ability to interact with dopaminergic $(D_2 \text{ and } D_1)$, serotonergic $(5-HT_2)$, and muscarinic (M) receptors, frequently invoked as potential targets of clozapine and other atypical neuroleptics,^{21,26,38} was evaluated in comparison with different tricyclic reference compounds. Compounds were studied *in vivo* using the antagonism of an apomorphine-induced stereotypies test to evaluate their antidopaminergic potential. An open-field test in rats and a complex operant-conditioning schedule in dogs, successfully employed in previous studies to discriminate between acutely²³ or chronically⁴³ administered typical and atypical antipsychotic drugs, were used to evaluate the newly synthesized and previously reported pyridobenzazepine analogues.⁴⁴

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s C₄H₄O₄ represents fumaric acid. b C₄H₄O₄ represents maleic acid. c 0.5C₂H₅OH.

Table 2. Pyridobenzoxazepinones and Pyridobenzothiazepinones

Scheme 1°

^a Key: (a) OPCl₃; (b) N-methylpiperazine, toluene, Δt. A,B: **benzene** or pyridine. $X = 0$, S. NMP: *N*-methylpiperazin-1-yl.

Chemistry

The lateral benzene rings in the reference dibenzazepine compounds were alternatively replaced by a pyridine, thus determining the preparation of two series of derivatives: 2V-methylpiperazinopyrido[l,4]- and -pyrido[l,5]benzazepines. The synthesis of the amidine derivatives 8-16 was conducted according to the same procedure reported in Scheme 1. The lactams (17,18,21...) were transformed into an imino chloride by the action of phosphorus oxychloride (method A). The crude product directly gave amidines 8-16 (Table 1) by reaction with an excess of iV-methylpiperazine (method A). Compounds 10,12, and 16 were isolated as a free base. Compounds 8, 9,11, 14, and 15 and compound 13 were crystallized as fumarate and maleate salts, respectively. The lactams (Table 2)

" Key: (a) $\text{SnCl}_2\text{-}2\text{H}_2\text{O}$, HCl; (b) $o\text{-}DCB$, Δt . 17: $X = N$, $Y = CH$. 18: $X = CH, Y = N$. ^{*b*}Key: (a) NaH, propylene glycol, Δt ; (b) SOCl₂; **(c) NaN3; (d) o-DCB, At; (e) A1C18, At.**

were obtained by different synthetic pathways which are summarized below.

5H-Pyrido[2,3-6] [l,4]benzothiazepin-6-one derivatives were prepared by two procedures. For the nonsubstituted analogues, the appropriate 3-aminochloropyridine directly reacted with thiosalicylic acid to give lactams 17 and 18 (Scheme 2a, method C). 3-Amino-4-chloropyridine was prepared from 4-chloro-3-nitropyridine (method B) according to the method of Delarge and Lapiere.*⁶ The 8-chlorothiazepinone analogue 21 was obtained following the procedure illustrated in Scheme 2b. In the first step,

 $\mathbf{Scheme}\;3^{a,b}$

^{*a*} Key: Δt , 1,2,4-Cl₃C₆H₃. R = H (22a), Cl (22b). ^{*b*}Key: (a) THF, Δt ; (b) EtONa, DMF, Δt . R = H (23a, 24a), Cl (23b, 24b), CH₃ (23c, 24c).

2-chloronicotinic acid reacted with 4-chlorothiophenol to provide 2- [(4-chlorophenyl)thio] nicotinic acid (19) (method D). The corresponding acid chloride was transformed to an acyl azide, 20, by reaction with $NaN₃$ (method E). The anhydrous product, by thermal decomposition, yields the corresponding isocyanate (method F). In the presence of Lewis acids such as aluminum chloride, the isocyanate group, by an electrophilic attack of the adjacent benzene ring, yields lactam 21.

5H-Pyrido[2,3-b][1,4]benzoxazepin-6-one derivatives were prepared by a method previously described.^{46,47} Lactams **22a,b** were obtained by direct condensation of phenyl salicylate or its 5-chloro analogue and 3-amino-2-chloropyridine (Scheme 3a, method G). Yields were poor due to problems of purification related to the presence of a byproduct such as pyridoxazole.⁴⁷

The synthesis of 6H-pyrido[2,3-6] [l,5]benzothiazepin-5-one derivatives was accomplished according to literature procedure.⁴⁸ The corresponding amidine derivative 16 has previously been described⁴⁹ but was prepared in order to examine its neurochemical and behavioral profile.

The preparation of the $6H$ -pyrido $[2,3-b][1,5]$ benzoxazepin-5-one derivatives **24a-c** was achieved following

the procedure of Hoffmann and Faure⁴⁸ (Scheme 3b). The reaction of 2-chloronicotinoyl chloride with an appropriate 2-aminophenol gave amides **23a-c** (method H). Ring closure was achieved with sodium ethoxide in DMF (160- 170 ^CC) (method **I).**

Results and Discussion

Hereby we report the biological evaluation of different N -methylpiperazinopyridobenzoxazepines and -benzothiazepines in comparison with some reference compounds. The binding affinities for $D_1, D_2, 5\text{-}HT_2$, and M receptors and their ratios compared to those of relevant reference compounds (isoclozapine (4), isoclothiapine (5), chlorpromazine (6), and haloperidol (7)) are reported in Table 3 along with the *in vivo* pharmacological data.

Binding Profile of Oxa- and Thiazepines. In each series, either $X = 0$ or $X = S$, the introduction of a pyridine ring in the tricyclic structure reduced the affinities for each receptor site. Nonsubstituted derivatives (8,10,13, and 16) were less potent than the substituted analogues in either the oxazepine or thiazepine series. A halogen substituent was more favorable as shown for 9 and 14 than amethylgroup (15). Indeed, the halogen group in different antipsychotic diarylazepine analogues has been considered as an important structural element for the recognition

Table 3. Neurochemical Data: Binding Affinities and Ratio Values and Pharmacological Data of N-Methylpiperazinopyridobenzothiazepines, -Pyridobenzoxazepines, and Reference Compounds

	\mathbf{D}_{2}^{a}	D_1^{α}	5. H T ₂	Ma	5-HT ₂ / D_2^b	D_2/M^b	D_2/D_1^b	apomorphine antagonism ^e	
8	4.96	5.04	4.65	6.07	0.94	0.82	0.98	0	
9	6.44	5.97	7.35	5.86	1.14	1.10	1.08	100	
10	5.18	5.09	5.94	6.13	1.15	0.84	1.02	0	
11	6.58	6.17	7.35	5.75	1.12	1.14	1.06	100	
12	5.64	4.81	6.55	6.68	1.16	0.84	1.17		
13	4.83	5.62	5.85	5.47	1.21	0.88	0.86		
14	5.92	6.52	7.19	5.65	1.21	1.05	0.91		
15	5.91	6.12	7.01	5.42	1.19	1.09	0.97		
16	5.39	5.79	6.29	5.61	1.17	0.96	0.93		
clozapine (2)	7.34	6.94	8.41	7.60	1.15	0.97	1.06	0	I*
isoclozapine (4)	7.90	7.55	8.75	7.68	1.11	1.03	1.05	NT	$1.7*$
clothiapine (1)	8.35	7.85	9.23	6.87	1.10	1.21	1.06	100	$0.72*$
isoclothiapine (5)	7.52	6.62	8.38	8.25	1.11	0.90	1.14	NT	I^*
fluperlapine (3)	6.88	6.85	8.41	7.60	1.22	0.96	1.00	NT	
chlorpromazine (6)	8.70	7.45	8.48	7.22	0.97	1.20	1.17	NT	$2.6*$
haloperidol (7)	9.06	7.12	7.63	5.13	0.84	1.77	1.27	100	$0.14*$

^a-log K_i.^b From -log K_i.^c Percent inhibition, by 20 mg/kg (sc) of drug, 1 h after apomorphine administration (2.5 mg/kg, sc). NT = not tested. * = from ref 1, ED_{50} in mg/kg, sc. 1 = inactive.

part of the receptor interaction.⁵⁰ Its favorable influence might be related not only to an electronic effect but also to the increased lipophilicity. The serotonergic affinity was consistently retained throughout these different modifications. The compounds evaluated in this study showed an evolution of the binding affinities similar to that of the pyridobenzodiazepine derivatives previously described.¹⁶ Thus, for the isomers 10 and 16, we observed two different evolutions of the binding profile. The isomer -[1,5]- (16) was more dopaminergic and serotonergic but less muscarinic than its isomer -[1,4]- (10). A similar evolution was noted for the oxazepine analogues 8 and 13 except for the D_2 affinity. Very few differences were noted between 8-chloro thiazepine 11 and 8-chloro oxazepine 9. Compounds 9 and 11 showed a higher dopaminergic potential. Compound 12, analogue of isoclothiapihe, where the pyridine nitrogen replaced the C-Cl group, presented reduced dopaminergic $(D_2 \text{ and } D_1)$ affinities while serotonergic and muscarinic affinities were less modified. As previously reported,²⁰ this part of the molecule has a great influence on the cholinergic activity. Substituted compounds like 14 and 15 showed reduced dopaminergic and muscarinic potentials while serotonergic activity was retained. As previously mentioned, the pyridobenzazepine compounds were less potent than the dibenzazepine analogues14,15 and thus constituted a new opportunity, in the group of diarylazepines, for the preparation of atypical antipsychotics.¹⁶ This therapeutic class possessed the ability to reduce preferentially the activity of the mesolimbic dopaminergic system.^{51,52} Moreover, interactions with nondopaminergic systems, such as 5-hydroxytryptamine, could also contribute to their activity.53-56 A functional D_2 receptor blockade may be unnecessary for some atypical drugs to produce their antipsychotic effect⁵⁷ or to block some of the effects of DA agonists. It has been shown that at clinically effective doses, clozapine produced a lower occupancy at striatal D_2 receptors than classical antipsychotic drugs.⁵⁸ Moreover, clozapine is known to interact with other binding sites.39-42 However, the implication of other receptors such as D_4 or 5-HT_{1C} in the atypical neuroleptic profiles remains to be confirmed.

Atypical neuroleptics seem to possess a higher $5-HT_2/$ D2 binding ratio than classical neuroleptics.²⁶ This concept is corroborated by different works.^{29–31,33} In our series, all molecules, except 8 and 11, presented a ratio superior to 1.12, considered as the critical value.²⁶ Nevertheless, as mentioned by Meltzer et al.,²⁶ some molecules of the diarylazepine class, like clothiapine, were not properly differentiated and remained just borderline. In our study, 11, which, incidentally, behaved as an antidopaminergic in the apomorphine test, had a value near 1.12. Compound 1. The aponex parameters, made when their their compound on the possessed a value superior to 1.12, but all other compounds, inactive in the apomorphine-antagonism test, had a value superior to 1.12. As reported by Meltzer et al.²⁶D₂/D₁ ratios were presented as being less discriminant than $5\text{-}HT_2/D_2$ ratio values. Moreover, in our study, they also appeared more markedly differentiated by the 5-HT₂/D₂ and D₂/M ratio values. For a long time, a correct balance of the antidopaminergic and anticholinergic activities appeared important to reveal an atypical neuroleptic profile.^{1-3,8} If we examine the D_2/M binding ratio, it is interesting to note that among the reference compounds, a value superior to 1.0 was observed for the typical neuroleptic drugs (clothiapine, haloperidol, and chlorpromazine) while the atypical antipsychotics

(clozapine and fluperlapine) presented a value inferior to 1.0. Some azepines such as 9 and 11 showed values more related to those of typical antipsychotic compounds. These compounds presented an activity in the apomorphineantagonism test. The anticholinergic activity prevents the inhibition of nigrostriatal cell firing linked to the dopamine-mediated effects, but it is also known that $5-HT_2$ antagonism attenuates the effects of the dopamine blockade. 55

In **ViVo Pharmacological Evaluation.** The study of atypical neuroleptics presents many problems related to the characterization of the antipsychotic effect. The biggest problem is to find analogy and/or homology between animal and human behaviors. The newly developed model by Bruhwyler et al.²³ has allowed the detection of potential antipsychotics and the discrimination between typical and atypical antipsychotic properties.^{23,43} In this study, two supplementary models were used, the antagonism of apomorphine stereotypies and the open-field test in rats. With the exception of chlorosubstituted analogues, pyridobenzoxa- and thiazepines did not inhibit apomorphine-mediated stereotypies in rats (Table 3). Compounds 9 and 11 decreased stereotypies in a similar manner to that of haloperidol and clothiapine.

In behavioral models such as the open-field test in rats or the temporal-conditioning procedure in dogs, 14 showed an antipsychotic potential similar to clozapine.⁴⁴ Results obtained in the open-field test with other derivatives reveal mainly disinhibitory properties characterized by a high increase in the total ambulation score and a decrease in the defecation score (unpublished data). As generally observed for this chemical class, the clinical activity is borderline between antidepressant and anxiolytic. In the temporal-conditioning schedule in dogs, different molecules appeared clearly as disinhibitors and, using a cluster analysis, different thiazepines (10 and 12) were classified among antidepressant compounds (unpublished data). Such activities should be confirmed using appropriate experiments like Porsolt's test. The most interesting compound seems to be 14, which, like clozapine (up to 24 mg/kg, ip), did not significantly modify the total ambulation score⁴⁴ and did not induce catalepsy nor stereotypy/ hyperkinesia but reduced all the motor effects. The conservation of the clozapine behavioral profile with a reduction of motor side effects and the complete disappearance of sialorrhea ⁵⁹ as we showed for some pyridopearance or siatorriea, as we showed for some pyrido-
benzodiazepines.¹⁶ could represent a significant progress in psychosis treatment. Compound 14 has been selected for further investigation.

Conclusion

In this study, we confirmed our previous results¹⁶ and we showed that the proposed modulations of dibenzazepine structures led to interesting new compounds. Different pyridobenzazepine derivatives, although possessing reduced *in vitro* binding affinities compared to those of their dibenzazepine analogues, could be classified among neuroleptic drugs. Other molecules presented a disinhibitory activity that could be related to an anxiolytic potential.

Experimental Section

Melting points were determined with a Tottoli (Buchi) melting point apparatus in open capillary tubes and are uncorrected. All compounds were characterized by physical methods using IR (Perkin-Elmer Model 297 spectrophotometer) and 'H NMR (Bruker AW 80 spectrometer with HMDS as the internal

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standard). Column chromatography was carried out using silica gel 60, 230-400 mesh (Merck). Microanalyses were performed in house (Carlo Erba CHNS-0 EA1108 elemental analyzer) and were within $\pm 0.4\%$ of the theoretical values.

Method A. General Method for the Preparation of Oxazepine and Thiazepine Derivatives. 10-(4-Methylpiperazin-1-yl)pyrido[4,3-b][1,4]benzothiazepine(12). Lactam 17 (2.28 g, 0.01 mol) was heated to reflux in the presence of an excess of phosphorus oxychloride (20 mL) and N , N -dimethylaniline (0.5 mL) for 20 h. The brown solution was dissolved in anhydrous toluene and evaporated under reduced pressure. The crude product was used without any further purification in the next step.

The crude imino chloride was dissolved in toluene (20 mL), and an excess of N-methylpiperazine (10 mL, 0.09 mol) was added. The mixture was heated to reflux for 2-4 h. The solvent was then evaporated under reduced pressure and the residue dissolved in CHCl₃ (100 mL) and washed with water $(2 \times 50 \text{ mL})$. The organic layer was clarified with Norit, dried over anhydrous MgSO₄, and concentrated under reduced pressure. The residue was purified by liquid/solid chromatography on Kieselgel 60 (Merck) using acetone as eluent. Recrystallization in petroleum ether (100-140 °C) gave pale yellow crystals: yield $35-40\%$; mp 143-145 °C; IR (KBr) 1583,1554,1477,1430,1242,1002,823,777 cm-¹ ; !H NMR (CDC13) *S* 8.28 (s, HC(1)), 7.98 (d, HC(3)), 7.44- 7.13 (m, benzene and $HC(2)$), 3.53 (m, $-CH_2CH_2$ -), 2.40 (m, $-CH_2$ -CH₂-), 2.26 (s, CH₃-). Anal. (C₁₇H₁₈N₄S) C, H, N, S.

Compounds 10 and 16 were prepared and isolated in a similar manner.

Compounds 8,9,11,14, and 15 were isolated as fumarate salts. The organic layer was concentrated to give an oil. The oil was dissolved in hot ethanol (20 mL) and treated with an equivalent of fumaric acid in hot ethanol (10 mL). The product crystallized as a white solid and was recrystallized from methanol.

Compound 13 was isolated as a maleate salt. The organic layer was removed to leave the crude free base which was converted into the maleate salt and crystallized from ethyl methyl acetone-ether-acetone. Filtration gave the salt as a white powder.

Method B. 3-Amino-4-chloropyridine. To a solution of 4-chloro-3-nitropyridine (4.1 g, 0.032 mol) in 12 N HC1 (25 mL) was added dropwise under stirring a mixture of SnCl2-2H₂O (30.2 g, 0.134 mol) and 12 N HCl $(50$ mL). The temperature was maintained between 50 and 60 °C. After addition, the suspension was heated to 90 °C for 5 min and then cooled at 0 °C for 2 h. The precipitate was collected by filtration and dissolved in water (200 mL). The solution was made basic with a 10% NaOH solution and extracted with CHCl₃ $(4 \times 200 \text{ mL})$. The organic layer was dried over anhydrous MgSO₄ and concentrated after addition of petroleum ether (100-140 °C, 100 mL) under reduced pressure until crystallization. The product was collected by filtration, washed with petroleum ether (40-60 °C), and dried under reduced pressure: yield 55 %; mp 58-59 °C; IR (KBr) 1640, $1570, 1558, 1486, 1416, 1328, 1244, 821, 691$ cm⁻¹; ¹H NMR (CDCl₃) 6 8.05 (s, HC(1)), 7.80 (d, HC(6)), 7.07 (d, HC(5)), 4.18 (br s, H_2N). Anal. (C₅H₅ClN₂) C, H, N.

Method C. 11H-Pyrido[4,3-b][1,4]benzothiazepin-10-one (17). Thiosalicylic acid (1.62 g, 0.01 mol) and 3-amino-4 chloropyridine (1.28 g, 0.01 mol) in o-dichlorobenzene (20 mL) were heated to reflux for 2-3 h. After the reaction was complete, the solvent was eliminated and the residue was taken up in water (100 mL). The pH was adjusted to 5-6, and the mixture was extracted with CHCl₃ (4 \times 50 mL) and dried over anhydrous MgS04. The organic layer was evaporated under reduced pressure, and the residue was taken up in a 10% NaHCO₃ solution. After the solution had stirred for 30 min, the product was collected by filtration, washed with water, and dried at room temperature: yield 40-50%; mp 244-246 °C; IR (KBr) 1662,1590,1559,1474, 1398,1383,1286,1188, 836, 743 cm"¹ ; *^lH* NMR (CDCI3/DMSO*de) &* 10.76 (br s, HNCO), 8.41 (s, HC(D), 8.21 (d, HC(3)), 7.65- 7.31 (m, benzene and $HC(4)$). Anal. $(C_{12}H_8N_2OS)$ C, H, N, S.

Method D. 2-[(4-Chlorophenyl)thio]nicotinic Acid (19). To 4-chlorothiophenol (28.8 g, 0.2 mol) and NaH (4.8 g, 0.2 mol) in propylene glycol (50 mL) was added 2-chloronicotinic acid (15.7 g, 0.1 mol). The mixture was heated to reflux until the starting materials were consumed, and the solvent was then evaporated under low pressure. The residue was dissolved in a 10% NaHCO₃ solution. The excess of thiophenol was extracted with CHCl₃ (4×50 mL). The solution was made acidic with 6 N aqueous HC1. The separated product was collected by filtration, washed with water, and dried at room temperature: yield 60%; mp 223-225 °C; IR (KBr) 1685, 1568, 1478, 1436, 1253, 1077, 822, 766, 716 cm⁻¹; ¹H NMR (CDCl₃/DMSO- d_6) δ 9.81 (br s, H(OC=0), 8.32-8.12 (dd, dd, HC(6'), HC(4')), 7.47-7.23 (m, benzene), 7.09–6.93 (dd, HC(5')). Anal. $(C_{12}H_8CINO_2S)$ C, H, N, S.

Method E. 2-[(4-Chlorophenyl)thio]nicotinoyl Azide (20). 19 (5.3 g, 0.02 mol) in $S O Cl₂$ (50 mL) and DMF (0.5 mL) was heated to reflux for 1 h. The excess of $S OCl₂$ was evaporated under reduced pressure, and the residue was dissolved in cold anhydrous acetone (30 mL) with vigorous stirring. The acid chloride was added dropwise to a cooled 30% NaN₃ solution. The solution was then diluted with water (200 mL) and allowed to stand for 2 h at 0 °C. The separated product was collected by filtration, washed with cold water, and dried under reduced pressure. The product was used without any further purification in the next step: yield 90-95 *%.*

Method F. 8-Chloro-5H-pyrido[2,3-b][1,4]benzothiazepin-6-one (21). 20 (1 g, 0.0033 mol) in o-dichlorobenzene (15 mL) was added to a suspension of aluminum chloride (1.5 g, 0.011 mol) in o-dichlorobenzene (50 mL) and heated to 160-170 °C for 15 min. The solution was diluted with $CHCl₃$ (150 mL) and extracted three times with 2 N aqueous HC1. The aqueous acid solution was extracted with CHCI₃ (2×50 mL). The combined organic layers were concentrated under reduced pressure, and the residue was dissolved in acetone (30 mL). The product was collected by filtration, washed with acetone, and resuspended in 1 N aqueous HC1 (30 mL). The precipitate was collected by filtration, washed with water, and dried at room temperature: yield 20-30%; mp 310-314 °C; IR (KBr) 1666,1580,1555,1445, ¹/₁₄₁₂, 1347, 1102, 828, 776 cm⁻¹; ¹H NMR (CDCl₃/DMSO-d₆) δ 8.92 (br s, HNCO), 8.14-8.03 (m, pyridine), 7.35 (m, benzene), 7.17 (dd, HC(3)). Anal. $(C_{12}H_7C_1N_2OS)$ C, H, N, S.

Method G. 8-Chloro-5H-pyrido[2,3-b][1,4]benzoxazepin-6-one (22b). Phenyl 5-chlorosalicylate (12.4 g, 0.05 mol) and 3-amino-2-chloropyridine (12.8 g, 0.1 mol) in 1,2,4-trichlorobenzene (10 mL) were heated to reflux under N_2 for 1 h. The liberated phenol was distilled during the experiment, and HC1 was absorbed in a 10% NaOH solution. The hot mixture was then poured into ethanol (100 mL). The separated product was collected by filtration, washed with ethanol, and dried. The product was recrystallized from methanol/DMF $(6/1)$: yield $35-40\%$; mp 313-314 °C; IR (KBr) 1683,1600,1569,1506,1462,1432,1357,1105, 827, 794 cm-¹ ; ^JH NMR (DMSO-d6) *S* 10.68 (br s, HNCO), 7.98 $(dd,HC(2), 7.68-7.50$ (m, 3H), 7.37-7.17 (m, 2H). Anal. (C₁₂H₇-C1N202) C, H, N.

Method H. N-(2-Hydroxy-5-chlorophenyl)-2-chloro-3pyridinecarboxamide (23b). To a solution of 2-amino-4 chlorophenol (15.9 g, 0.1 mol) in THF (150 mL) was added dropwise a solution of 2-chloronicotinoyl chloride (17.5 g, 0.1 mol) in THF (100 mL). The solution was heated to reflux while stirring for 1 h. The mixture was diluted with water (1 L). The separated product was collected by filtration, washed with water, and air-dried; yield 75-80%. A sample was recrystallized from 2-propanol: mp 193-194 °C; IR (KBr) 1657, 1612, 1579, 1583, 1547, 1429, 1394, 1189, 805, 744 cm⁻¹; ¹H NMR (CDCl₃/DMSO d_6) δ 9.79 and 9.46 (br s, HO and HNCO), 8.42 (dd, HC(6')), 8.05-7.82 (m, benzene and HC(4')), 7.38 (dd, HC(5')), 6.86 (d, benzene). Anal. $(C_{12}H_8Cl_2N_2O_2)$ C, H, N.

Method I. 8-Chloro-6H-pyrido[2,3-b][1,5]benzoxazepin-5-one (24b). Compound 23b (26.2 g, 0.1 mol) and NaOEt (6.8 g, 0.1 mol) in DMF (100 mL) were heated to reflux for 3-4 h. The mixture was concentrated under reduced pressure, and the product was collected by filtration, washed with cold methanol, and recrystallized from methanol/DMF $(4/1)$: yield 65%; mp 285-286 °C; IR (KBr) 1686,1596,1573,1498,1423,1384,818,780 cm⁻¹; ¹H NMR (CDCl₃/DMSO-d₆) δ 10.67 (br s, HNCO), 8.42 (dd, HC(2)), 8.22 (dd, HC(4)), 7.37 (dd, HC(3)), 7.27-6.98 (m, benzene). Anal. $(C_{12}H_7C1N_2O_2)$ C, H, N.

Radioligand Binding Study. Experiments on receptor preparations were performed following classical methods previously described, \overline{D}_1 , $^{60, 61}$ D_2 , 62 5-HT₂, 63 and M.⁶⁴ Specific binding was defined as the difference between total and nonspecific

binding (with and without $[3H]$ drug). K_i values were calculated according to the Cheng-Prusoff equation: $65 K_i = I C_{60}/(1 + L/K_i)$ with L the concentration and K_d the apparent dissociation constant of the [³H] ligand obtained from Scatchard analysis of saturation experiments. Each K_i value was determined at least in duplicate with nine concentrations of the drug in triplicate. The mean coefficient of variation for duplicate determinations of the K_i value for the four types of binding sites was $15.3 \pm 3.8\%$ (SD). The pK_i (-log K_i) values are reported in Table 3 and were used in the calculation of binding ratios.

Apomorphine-Antagonism Test in the Rat. Experiments were performed following the classical method previously described.⁶⁶

Open-Field Test in the Rat. Experiments were performed following the classical method previously described.²²

Temporal-Regulation Schedule in the Dog. Experiments were performed following the previously described proce $dure.^{23,43,44}$

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