Pi Aspartate-Based Peptide ot-((2,6-Dichlorobenzoyl)oxy)methyl Ketones as Potent Time-Dependent Inhibitors of Interleukin-18-Converting Enzyme

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Interleukin-1 β -converting enzyme (ICE) is the processing protease responsible for the production of the potent inflammatory mediator interleukin-1 β (IL-1 β).¹ Inhibitors of this intracellular protease are viewed as potential therapeutic agents for the treatment of a variety of chronic inflammatory disease states.² The detection and characterization of ICE as the IL-1 β processing enzyme was independently reported by research groups at Immunex^{1a-d} and Merck.^{1e,f} The human enzyme has subsequently been purified from THP-1 cells, the cDNA cloned, and the amino acid sequence determined.^{1c,d,f} The enzyme is a heterodimer consisting of 20- and 10-kD noncovalently bound subunits and has been characterized as a thiol (cysteine) protease.³ Interestingly, the primary structure of ICE bears no sequence homology with any known member of the cysteine (or serine) protease superfamily.⁴

ICE cleaves the biologically inactive human IL-1 β precursor protein (pro-IL-1 β) between residues Asp¹¹⁶ and Ala¹¹⁷ to release the biologically active mature cytokine.^{1e,f,5} Peptide-based α -((2,6-dichlorobenzoyl)oxy)methyl ketones⁶ 1-3 which contain a P₁L-aspartic acid residue were synthesized and evaluated as time-dependent inhibitors of the enzyme. The selection of the P_2 Val and Ala in inhibitors 2 and 3 as hydrophobic residue replacements for the native P_2 His is based on previously reported peptide substrate data.^{1f,5}

$$
\begin{array}{l}\n... \cdot P4 \cdot P3 \cdot P2 \cdot P1 \cdot P1' \cdot P2' \cdot P3' \cdot P4' \cdot \dots \\
... \cdot \text{Tyr} \cdot \text{Val} \cdot \text{His} \cdot \text{Asp}^{116} \cdot \text{Ala}^{117} \cdot \text{Pro} \cdot \text{Val} \cdot \text{Arg} \cdot \dots \\
+ \text{ICE cleavage site} \\
\text{on pro-IL-1B}\n\end{array}
$$

The synthesis of the requisite peptide α -((arylacyl)oxy)methyl ketones was carried out using methodology described by Krantz⁶ and is exemplified by the synthesis of inhibitor 1 (Scheme 1). N -(Benzyloxycarbonyl)aspartic acid *P-tert-hutyl* ester 4 (Z-Asp(OtBu)-OH; Bachem) was converted to Z -Asp(OtBu)-CH₂Br 5 upon (1) activation of the carboxylate as a mixed anhydride (1.4 equiv of N -methylmorpholine, 1.3 equiv of ethyl chloroformate, THF, -15 °C); (2) reaction of the mixed anhydride with diazomethane (2 equiv of CH_2N_2 in Et₂O, -15 °C \rightarrow 25 °C); and (3) HBr-catalyzed decomposition of the *in situ* generated diazo ketone (excess glacial HOAc-45% HBr $(1:1)$, $0°C$) in 75% yield. Noteworthy is the survival of the tert-butyl (side chain) ester during the acid-catalyzed decomposition of the diazo ketone. Treatment of 5 with

Table 1. Results of ICE and Cathepsin B Assays with Inhibitors 1-3 and 7

<• ICE was partially purified from THP-1 cells using the DEAE-Sephacel and Sephadex G-75 steps described by Black (ref lb). The ICE assay contained 10 mM HEPES (pH 7.5), 25% glycerol, 1 mM dithiotreitol (DTT) and 10 *iM* Suc-Tyr-Val-Ala-Asp-AMC (Bachem) in a volume of $30 \mu L$ in a polystyrene 96-well microtiter plate. Progress curves were obtained at 37 °C over 30 min. Kinetic data were obtained on a Fluoroskan II fluorescence plate reader under control of an Apple Macintosh computer running the DeltaSoft data aquisition program (BioMetallics, Inc.). Nonlinear progress curves were analyzed as described by Tian and Tsou (ref 13). *^b* Bovine spleen cathepsin B was purchased from Sigma and assayed using Z-Phe-Arg-AMC (Bachem) under the conditions described by Barrett (ref 14). Adapted for use in continuous assays at 37 °C . ϵ DCB = (2,6dichlorobenzoyl)oxy. ^{*d*} No inhibition at 20 μ M inhibitor. • Not determined.

Scheme 1. Synthesis of

 α -((2,6-Dichlorobenzoyl)oxy)methyl Ketone 1 and the Structures of Ketones 2, 3, and 7 (Absolute Stereochemistry Is as Shown)

Z = benzyloxycarbonyl

2,6-dichlorobenzoic acid (1.2 equiv of 2,6-Cl₂PhCO₂H, 2.5 equiv of KF, DMF, 12 h, 25 °C) afforded the corresponding $Z-Asp(OtBu)CH₂OC(O)-2.6-Cl₂Ph 6 in >95% yield.$ Deprotection of the *tert-butyl* ester in 6 to yield 1 occurred in quantitative yield following exposure of 6 to a solution of 25% trifluoroacetic acid (TFA) in CH_2Cl_2 (0.5 M solution of 6 in 25% v/v TFA/CH₂Cl₂). Similarly prepared were 2 and 3 from their respective carboxypeptides Z-Val-Asp- (OtBu)-OH and Z-Val-Ala-Asp(OtBu)-OH. The homolog of 1, Z-Glu-CH₂OC(O)-2,6-Cl₂Ph, 7, was synthesized from Z-Glu-OH in analogous fashion.⁷

Evaluation of inhibitors 1-3 against ICE revealed these compounds to be potent time-dependent inactivators⁸ of the enzyme (Table 1). Inhibitor $1, Z$ -Asp-CH₂OC(O)-2,6- $Cl₂Ph$, which contains only a single amino acid residue, was sufficient for relatively rapid inactivation of enzyme. The $k_{\text{obs}}/[1]$ of 7100 M⁻¹ s⁻¹ for 1 is the greatest value ever reported for a single amino acid-based affinity label against a cysteine protease.⁹ Typically, a dipeptide unit is the smallest recognition sequence needed to achieve inactivation rates >1000 M⁻¹ s⁻¹ for this enzyme class.¹⁰ Inclusion of the P_2 Val and $P_3\text{-}P_2$ Val-Ala residues into 1 enhances inhibitory potency in this series. The tripeptide analog Z-Val-Ala-Asp- $CH₂OC(O)-2.6-Cl₂Ph$, 3, possesses

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a second-order rate constant equal to 406 700 M⁻¹ s⁻¹. This rate is substantially faster than the ca . 16 500 M^{-1} s⁻¹ rate observed for the inactivation of ICE with the tetrapeptide diazo ketone, Ac-Tyr-Val-Ala-Asp-CHN_{2.}1f,11 Furthermore, enzyme specificity for an aspartic acid residue at P_1 in this class of inhibitor was clearly demonstrated as α -((2,6-dichlorobenzoyl)oxy)methyl ketone 7, the side chain homolog of 1, was devoid of enzyme affinity. This specificity for aspartic acid and the high inactivation rate observed for 1 suggests that the discriminating primary binding site for ICE is the S_1 subsite, atypical for a cysteine protease.¹⁰

Inhibitors 1-3 were also examined for their selectivity against cathepsin B (Table 1). ((Arylacyl)oxy)methyl ketone 1 displayed the highest selectivity for ICE versus cathepsin B ($> 700:1$), while the more potent methyl ketones 2 and 3 demonstrated *ca.* 100-180-fold preference for ICE.¹²

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