# Design, Synthesis, and Biological Properties of New Bis(acridine-4-carboxamides) as Anticancer Agents 

I ppolito Antonini, ${ }^{*, \dagger}$ Paolo Polucci, ${ }^{\dagger, \perp}$ Amelia Magnano, ${ }^{\dagger}$ Barbara Gatto, ${ }^{\S}$ Manlio Palumbo, ${ }^{\S}$ Ernesto Menta, $\neq$ Nicoletta Pescalli, $\ddagger$ and Sante Martelli ${ }^{\dagger}$<br>Department of Chemical Sciences, University of Camerino, Via S. Agostino 1, 62032 Camerino, Italy, Department of Pharmaceutical Sciences, University of Padova, Via Marzolo 5, 35131 Padova, Italy, and Novuspharma SpA, Via Ariosto 23, 20091 Bresso, Italy

Recei ved February 21, 2003
To enhance the outstanding biological response shown by the corresponding monomers 4 and 5, two classes of bis-acridine-4-carboxamides, 9 , with a linker between the 4,4' positions, and 13, with a linker between the 1,1' positions, have been prepared as DNA-binding and potential antitumor agents. The noncovalent DNA-binding properties of these compounds have been examined using gel-electrophoresis and fluorometric techniques. The results indicate that (i) target compounds intercalate DNA; (ii) the bis derivatives with the optimal linker are considerably more DNA-affinic than corresponding monomers; (iii) overall affinity is sensitive to the nature of the linker, of the chromophores, and of the substituents at 7,7'; (iv) often, the bis derivatives show a marked AT-preferential binding. In vitro cytotoxic potency of these derivatives toward the human col on adenocarcinoma cell line (HT29) is described and compared to that of reference drugs. Structure-activity relationships are discussed. Some highly DNAaffinic and potent cytotoxic compounds, $\mathbf{9 b}, \mathbf{f}$ and $\mathbf{1 3 b}, \mathbf{c}$, have been selected for the National Cancer Institute ( NCl ) screening on 60 human tumor cell lines and identified as new leads in the antitumor strategies.

## Introduction

Connecting two planar intercalating moieties to obtain a bis derivative generally increases the DNA binding affinity and the drug's residence times in the DNA-bound form. This rational has led to the development of an interesting bis-intercalator family possessing noticeable antitumor properties. Some successful examples are the bis(naphthalimide) analogue, DMP 840 (1), reported to be a topoisomerase II poison, ${ }^{1}$ presently undergoing clinical trials, ${ }^{2}$ and the related LU 79553 (2), highly effective against tumor xenografts in vivo. ${ }^{3}$ Moreover, a series of bisimidazoacridones (e.g., WMC$26,3)$ exhibited highly selective cytotoxicity toward human colon carcinoma cells both in culture and in xenografts. ${ }^{4}$ However, for the latter derivatives, bisintercalation appears an unlikely process, and a groove binding mode has been proposed, although the precise nature of this interaction is still to be clarified. ${ }^{4} \mathrm{Re}-$ cently, bis(acridine-4-carboxamides) and dicationic bis-(9-methyl phenazine-1-carboxamides) have been synthesized and studied. ${ }^{5,6}$ I nterestingly, the cytotoxic profile and in vitro activity of these derivatives are consistent with topoisomerase I rather than topoisomerase II poisoning effects. Hence, different modes of DNAbinding and cytotoxic action can be observed, which are still compatible with prominent antitumor efficacy. Altogether, the above results confirm that bis-interca-

[^0]lators can be considered as a promising class of anticancer compounds. Nevertheless, according to the information thus far available, general structure-activity relationships are still far from being firmly established, also given the fact that different cellular targets are possibly recognized by different drugs.

Our aim is to exploit the potential of new acridinebased bis-intercalators as anticancer drugs. The novel compounds are derived from two interesting series of acridine-4-carboxamides we have previously examined. ${ }^{7,8}$ In these derivatives the chromophore moiety is either 9 -acridone (e.g., 4) ${ }^{7}$ or acridine (e.g., 5). ${ }^{8}$ Given the outstanding biological response shown by the above monomers, we synthesized and investigated the biochemical and pharmacological properties of a series of bis anal ogues derived from these monomers.

## Chemistry

Schemes 1 and 2 show the synthetic pathways leading to target derivatives $\mathbf{9}$ and 13. According to the first scheme, the reaction of either 1-chl oro-9-oxo-9,10-dihy-droacridine-4-carboxylic acid (6) ${ }^{9}$ or 1-chloro-7-methoxy-9-oxo-9,10-dihydroacridine-4-carboxylic acid (7) ${ }^{7}$ with the suitable diamine, by the "mixed anhydride" method, ${ }^{10}$ afforded the intermediate bis(acridine-4-carboxamides) 8a-f. All the diamines were commercially available, except the $\mathrm{N}^{1}, \mathrm{~N}^{2}$-bis(2-aminoethyl)- $\mathrm{N}^{1}, \mathrm{~N}^{2}$-dimethyl-1,2ethanediamine, needed for $\mathbf{8 d}$, prepared according to the literature. ${ }^{11}$ Nucleophilic substitution of $\mathbf{8 a}-\mathbf{f}$ with $\mathrm{N}^{1}, \mathrm{~N}^{1}$-dimethyl-1,2-ethanediamine yielded the target bis(acridine-4-carboxamides) 9a-f. Cleavage with aqueous HBr of the methoxy derivatives $9 \mathrm{e}, \mathrm{f}$ gave the hydroxy derivatives $\mathbf{9 g}, \mathbf{h}$, respectively. Finally, the bis-


Figure 1. Structures of DMP 840 (1), LU 79553 (2), WMC-26 (3), bis-functionalized 9-acridones (4), and bis-functionalized acridine (5).

## Scheme $1^{\text {a }}$


 for $\mathbf{a}-\mathbf{d} ; \mathrm{X}=\mathrm{OMe}$ for $\mathbf{e}, \mathbf{f} ; \mathrm{Y}=\left(\mathrm{CH}_{2}\right)_{2} \mathrm{~N}(\mathrm{Me})\left(\mathrm{CH}_{2}\right)_{2}$ for $\mathbf{a}, \mathbf{e}, \mathbf{g}$, and $\mathbf{i} ; \mathrm{Y}=\left(\mathrm{CH}_{2}\right)_{3} \mathrm{~N}(\mathrm{Me})\left(\mathrm{CH}_{2}\right)_{3}$ for $\mathbf{b}, \mathbf{f}, \mathbf{h}$, and $\mathbf{j} ; \mathrm{Y}=\left(\mathrm{CH}_{2}\right)_{3}$ for $\mathbf{c}$; $\mathrm{Y}=\left(\mathrm{CH}_{2}\right)_{2} \mathrm{~N}(\mathrm{Me})\left(\mathrm{CH}_{2}\right)_{2} \mathrm{~N}(\mathrm{Me})\left(\mathrm{CH}_{2}\right)_{2}$ for $\mathbf{d}$.
acridines $9 \mathbf{i}, \mathbf{j}$ were obtained by reduction with aluminum amal gam of bis-9-acridones $\mathbf{9 a}, \mathbf{b}$, respectively.

As shown in Scheme 2, the target compounds 13a-e were prepared by reaction of the appropriate 9 -acridone-4-carboxamide ( $\mathbf{1 0}-\mathbf{1 2})^{7}$ with either $\mathrm{N}^{1}$-(2-aminoethyl)-$\mathrm{N}^{1}$-methyl-1,2-ethanediamine or $\mathrm{N}^{1}$-( 3 -aminopropyl) - $\mathrm{N}^{1-}$ methyl-1,3-propanediamine in 2-ethoxyethanol at 120 ${ }^{\circ} \mathrm{C}$. The hydroxy derivatives $\mathbf{1 3 f}, \mathbf{g}$ were obtained by refluxing the corresponding methoxy derivatives 13d,e in aqueous HBr , while the reduction with aluminum amalgam of the 9 -acridone derivatives $\mathbf{1 3 a}, \mathbf{b}$ afforded the acridine derivatives $\mathbf{1 3 h}, \mathbf{i}$.

All the target compounds $\mathbf{9}$ and $\mathbf{1 3}$ were examined, as water-sol uble hydrochl oride salts obtained by usual methods, for their DNA-binding properties and their antineoplastic activity.

## Results and Discussion

DNA-Binding Properties. To assess whether the new compounds are able to intercalate into the double helix, we performed DNA unwinding studies using 6 -[2-(dimethylamino)ethyl]-11-methoxy-2-(2-pi peridi noethyl)-2,5,6,7-tetrahydropyrazol o-[3,4,5-mn]pyrimido[5,6,1-de]-acridine-5,7-dione (PPAC), ${ }^{12}$ an acridone derivative that

## Scheme $\mathbf{2 a}^{\text {a }}$





a Reagents: (i) $\mathrm{H}_{2} \mathrm{~N}-\mathrm{Y}-\mathrm{NH}_{2}$; (ii) $\mathrm{HBr} 48 \%$; (iii) $\mathrm{Al} / \mathrm{Hg}$. Structures: $\mathrm{R}=\left(\mathrm{CH}_{2}\right)_{2} \mathrm{~N}\left(\mathrm{CH}_{3}\right)_{2}$ for $\mathbf{a}-\mathbf{i}$; $\mathrm{X}=\mathrm{H}$ for $\mathbf{a}, \mathbf{b} ; \mathrm{X}=\mathrm{NO}_{2}$ for $\mathbf{c}$; $\mathrm{X}=\mathrm{OMe}$ for $\mathbf{d}, \mathbf{e} ; \mathbf{Y}=\left(\mathrm{CH}_{2}\right)_{2} \mathrm{~N}(\mathrm{Me})\left(\mathrm{CH}_{2}\right)_{2}$ for $\mathbf{a}, \mathbf{d}, \mathbf{f}$, and $\mathbf{h} ; \mathbf{Y}=\left(\mathrm{CH}_{2}\right)_{3} \mathrm{~N}(\mathrm{Me})\left(\mathrm{CH}_{2}\right)_{3}$ for $\mathbf{b}, \mathbf{c}, \mathbf{e}, \mathbf{g}$, and $\mathbf{i}$.


Figure 2. The bis-acridine derivative $\mathbf{9 b}$ unwinds plasmid DNA. Increasing concentrations of PPAC or $\mathbf{9 b}$ were incubated with negatively supercoiled pBR322 $(0.25 \mu \mathrm{~g})$ for 1 h at $25^{\circ} \mathrm{C}$ in TE buffer pH 8.0, and then run for 4 h at $5 \mathrm{~V} / \mathrm{cm}$ on a $1 \%$ agarose gel (Tris-borate 50 mM , EDTA 1 mM ). C refers to the negatively supercoiled form of pBR322.
showed the ability to bind to DNA in an intercalative mode, as control. The example reported in Figure 2 shows that derivative $\mathbf{9 b}$ alters the migration of the supercoiled form of the plasmid through the agarose gel more dramatically than our positive control. In fact, at drug concentrations up to $1 \mu \mathrm{M}$, the gel mobility decreases consistently, while further addition of compound causes an increase in the plasmid mobility. At $100 \mu \mathrm{M}$, precipitation of the DNA in the well occurs. This electrophoretic behavior is consistent with unwinding of the DNA duplex so that the number of negative supercoils is progressively reduced, and further untwisting by the drug binding is compensated for by the formation of positive supercoils, characterized by increased rate of migration. Hence, we can conclude that the mechanism of DNA binding of $\mathbf{9 b}$, as well as of the other DNA binding derivatives bel onging to this family (not shown), is intercalation, and the degree of unwinding of the bis derivatives is higher than that of the mono-acridone derivative used as control. It is noteworthy that the drug-DNA complex formed upon brief incubation is stable even upon dilution in the gel wells and during the overnight gel run (the drug was not present in the agarose gel ).

Competitive displacement ( $\mathrm{C}_{50}$ ) fluorometric assays with DNA-bound ethidium can be used ${ }^{13}$ (a) to deter-
mine, as shown in Table 1, 'apparent' equilibrium constants ( $\mathrm{K}_{\text {app }}$ ) for drug binding, as the $\mathrm{C}_{50}$ value is approximately inversely proportional to the binding constant, ${ }^{14}$ and (b) to establish possible base- or sequencepreferential binding. ${ }^{15}$ In the present study, fluorescence displacement assays were performed at pH 7 to enable comparison under biological conditions.

The $K_{\text {app }}$ values of the new derivatives with CT-DNA, AT, and GC are reported in Table 1. The results indicate that the target compounds are excellent DNA ligands, generally possessing greater affinity than ethidium, but lower than mitoxantrone ( Mx ); two exceptions are 9c with a $K_{\text {app }}$ of $3.3 \times 10^{5}$ and $9 j$ with a $K_{\text {app }}$ of $1.4 \times 10^{10}$; the low affinity of $9 \mathbf{9}$ is probably due to a linker that is too short and devoid of a basic nitrogen $\left(Y=-\left(\mathrm{CH}_{2}\right)_{3^{-}}\right)$ to allow an efficient interaction with DNA. The exceptional affinity of $\mathbf{9 j}$ is difficult to explain, without considering the $K_{\text {app }}$ value ( $1.5 \times 10^{9}$ ) of related monomer 5. ${ }^{8} 9 \mathrm{~d}$ also shows a remarkable $K_{\text {app }}$ value, indi cating that even when $\mathrm{Y}=-\left(\mathrm{CH}_{2}\right)_{2} \mathrm{~N}\left(\mathrm{CH}_{3}\right)\left(\mathrm{CH}_{2}\right)_{2} \mathrm{~N}\left(\mathrm{CH}_{3}\right)-$ $\left(\mathrm{CH}_{2}\right)_{2^{-}}$, an efficient binding can be achieved.

For the derivatives with $Y=-\left(\mathrm{CH}_{2}\right)_{n} N\left(\mathrm{CH}_{3}\right)\left(\mathrm{CH}_{2}\right)_{n}$-, the following should be noted: (a) The respective $\mathrm{K}_{\text {app }}$ values of all the related pairs with $n=2$ or 3 (8a,b, $\mathbf{9 a}, \mathrm{b}, \mathbf{9 e}, \mathrm{f}, \mathbf{9 g}, \mathrm{h}, \mathbf{9 i}, \mathbf{j}, \mathbf{1 3 a}, \mathrm{b}, \mathbf{1 3 d , e}$ 13f,g, and 13h,i) always indicate a binding more (2.3-100 times) efficient for compounds with $n=3$, clearly showing that the best results in terms of DNA affinity are achieved with the longest linker either in the 4,4' or in the 1,1' positions. (b) The $K_{\text {app }}$ values of the intermediates $\mathbf{8 a}, \mathbf{b}$, compared to that of the related target compounds $\mathbf{9 a}, \mathbf{b}$, indicate that $\mathbf{9 a}, \mathbf{b}$ are more DNA affinic ( $4-17$ times) than $\mathbf{8 a}, \mathbf{b}$, pointing out the importance of the basic side chains in positions 1,1' for binding. (c) The substitution at positions 7,7' leads to different results in the subseries 9 and 13: for derivatives $\mathbf{9}$ the affinity rank order is $\mathbf{9 e}, \mathbf{f}$ ( $\mathrm{X}=\mathrm{OMe}$ ) $>\mathbf{9 a , b}(\mathrm{X}=\mathrm{H})>\mathbf{9 g}, \mathbf{h}(\mathrm{X}=\mathrm{OH})$; for 13 the order is 13f,g $(X=O H)>13 a, b(X=H)>13 d, e(X=$ OMe). (d) Generally, the related monomers 4 and 5 possess DNA affinity inferior to that of target derivatives with $n=3$, but superior to that of target derivatives with $\mathrm{n}=2$. A noticeable exception is constituted by the case of $\mathbf{5}$ compared with the pair 13h,i: the monomer is much more DNA affinic than both components of the pair. (e) Effect of the chromophore: com-

Table 1. Melting Points, ${ }^{\text {a }}$ Yields, Formula, ${ }^{\text {b }}$ DNA Binding, ${ }^{\text {c }}$ and Cytotoxic Activity against Human Colon Adenocarcinoma (HT29) of Target Compounds $\mathbf{9 a - j}$ and $\mathbf{1 3 a - i}$, of Intermediate Derivatives 8a,b, and of Related Monomers 4a-d and 5. Reference Drug: Mitoxantrone (Mx).

| compd | mp, ${ }^{\circ} \mathrm{C}$ | yield, \% | formula | $\mathrm{K}_{\text {app }}{ }^{\text {d }} \times 10^{-7} \mathrm{M}^{-1}$ |  |  | binding site preference ${ }^{e}$ | $\frac{\mathrm{IC}_{50}(\mu \mathrm{M})^{\mathrm{f}}}{\mathrm{HT} 29}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  | AT | CT-DNA | GC |  |  |
| 9 a | 116-117(238-240) | 79 | $\mathrm{C}_{41} \mathrm{H}_{49} \mathrm{~N}_{9} \mathrm{O}_{4}$ | 2.9 | 2.6 | 1.5 | A-T (1.9) | 0.48 |
| 9b | 119-120 (226-228) | 85 | $\mathrm{C}_{43} \mathrm{H}_{53} \mathrm{Ng}_{9} \mathrm{O}_{4}$ | 8.2 | 6.3 | 0.79 | A-T (10) | 0.057 |
| 8a | ( $>300$ ) ${ }^{\text {g }}$ | 38 | $\mathrm{C}_{33} \mathrm{H}_{28} \mathrm{Cl}_{3} \mathrm{~N}_{5} \mathrm{O}_{4}$ | 0.09 | 0.15 | 0.0034 | A-T (26) | 5.5 |
| 8b | $(245-247 \mathrm{dec}) 9$ | 46 | $\mathrm{C}_{35} \mathrm{H}_{32} \mathrm{Cl}_{3} \mathrm{~N}_{5} \mathrm{O}_{4}$ | 1.5 | 1.6 | 1.6 | none | 0.34 |
| $4 a^{h}$ | (26-247 dec) |  | $\mathrm{C}_{3}{ }_{3}{ }^{\text {a }}$ | - | 3.7 | - | none | 0.094 |
| 9 c | 220-221 (198-200) | 47 | $\mathrm{C}_{39} \mathrm{H}_{44} \mathrm{~N}_{8} \mathrm{O}_{4}$ | 0.041 | 0.033 | 0.020 | A-T (2.0) | > 10 |
| 9d | (268-270) ${ }^{9}$ | 46 | $\mathrm{C}_{44} \mathrm{H}_{60} \mathrm{Cl}_{4} \mathrm{~N}_{10} \mathrm{O}_{4}$ | 32 | 15 | 40 | none | 0.29 |
| 9 | 217-218 (270-273 dec) | 74 | $\mathrm{C}_{43} \mathrm{H}_{53} \mathrm{Ng}_{9} \mathrm{O}_{6}$ | 5.8 | 4.4 | 4.3 | A-T (1.3) | 0.89 |
| 9 f | 143-144 (250-252 dec) | 66 | $\mathrm{C}_{45} \mathrm{H}_{57} \mathrm{~N}_{9} \mathrm{O}_{6}$ | 13 | 20 | 5.2 | A-T (2.5) | 0.0020 |
| $4 c^{\text {h }}$ | 113-120 (250-252 dec) | - | ${ }^{\text {a }}$ |  | 1.8 | 5 | none | 0.50 |
| 9 g | 162-163 (262-264 dec) | 64 | $\mathrm{C}_{41} \mathrm{H}_{49} \mathrm{Ng}_{9} \mathrm{O}_{6}$ | 1.9 | 1.9 | 1.9 | none | 8.9 |
| 9h | 221-222 (250-251) | 45 | $\mathrm{C}_{43} \mathrm{H}_{53} \mathrm{~N}_{9} \mathrm{O}_{6}$ | 13 | 4.4 | 8.3 | A-T (1.6) | 0.80 |
| $4{ }^{\text {h }}$ |  | - | ${ }^{\text {a }}$ | - | 3.4 | - | none | 2.0 |
| 9 i | 196-197 (248-249 dec) | 25 | $\mathrm{C}_{41} \mathrm{H}_{49} \mathrm{Ng}_{9} \mathrm{O}_{2}$ | 12 | 13 | 6.3 | A-T (1.9) | 2.0 |
| 9 j | 166-167 (240-242 dec) | 50 | $\mathrm{C}_{43} \mathrm{H}_{53} \mathrm{~N}_{9} \mathrm{O}_{2}$ | 1260 | 1400 | 43 | A-T (29) | 0.23 |
| 5 |  | - | - | 2.8 | 150 | 23 | G-C (8.2) | 0.007 |
| 13a | 135-136 (258-260 dec) | 39 | $\mathrm{C}_{41} \mathrm{H}_{49} \mathrm{~N}_{9} \mathrm{O}_{4}$ | 8.4 | 1.5 | 3.8 | A-T (2.2) | 0.41 |
| 13b | 112-113 (235-236 dec) | 48 | $\mathrm{C}_{43} \mathrm{H}_{53} \mathrm{Ng}_{9} \mathrm{O}_{4}$ | 9.3 | 10 | 7.5 | none | 0.00043 |
| 13c | 152-153 (255-257 dec) | 67 | $\mathrm{C}_{43} \mathrm{H}_{51} \mathrm{~N}_{11} \mathrm{O}_{8}$ | 19 | 14 | 23 | none | $<0.0001$ |
| $4 b^{\text {h }}$ |  | - |  | - | 7.4 | - | A-T (1.3) | 0.19 |
| 13d | 160-162 (250-252 dec) | 35 | $\mathrm{C}_{43} \mathrm{H}_{53} \mathrm{~N}_{9} \mathrm{O}_{6}$ | 1.3 | 1.8 | 0.88 | A-T (1.5) | 3.3 |
| 13e | 160-161 (230-233 dec) | 40 | $\mathrm{C}_{45} \mathrm{H}_{57} \mathrm{~N}_{9} \mathrm{O}_{6}$ | 26 | 6.2 | 14 | A-T (1.9) | 0.039 |
| 13f | 190-192 (254-255) | 78 | $\mathrm{C}_{41} \mathrm{H}_{49} \mathrm{~N}_{9} \mathrm{O}_{6}$ | 1.3 | 2.0 | 1.4 | none | > 10 |
| 13 g | 186-188 (270-271) | 49 | $\mathrm{C}_{43} \mathrm{H}_{53} \mathrm{~N}_{9} \mathrm{O}_{6}$ | 13 | 12 | 7.4 | A-T (1.8) | 6.9 |
| 13h | 138-139 (255-257 dec) | 57 | $\mathrm{C}_{41} \mathrm{H}_{49} \mathrm{~N}_{9} \mathrm{O}_{2}$ | 2.1 | 5.3 | 0.72 | A-T (2.9) | 0.47 |
| 13i | 130-131 (250-252 dec) | 25 | $\mathrm{C}_{43} \mathrm{H}_{53} \mathrm{~N}_{9} \mathrm{O}_{2}$ | 66 | 12 | 9.4 | A-T (7.0) | 0.047 |
| Mx |  |  |  |  | 34 |  |  | 0.010 |

${ }^{\text {a }}$ In parentheses hydrochlorides melting points, dec = decomposition. ${ }^{\mathrm{b}}$ Analyses for $\mathrm{C}, \mathrm{H}$, and N. ${ }^{\mathrm{c}} \mathrm{CT}$-DNA, AT, and GC refer to calf thymus DNA, [poly(dA-dT)] $]_{2}$, and [poly(dG-dC) $]_{2}$, respectively. ${ }^{\mathrm{d}} \mathrm{K}_{\mathrm{app}}=1.26 / \mathrm{C}_{50} \times 10^{7}$ in which 1.26 is the concentration ( $\mu \mathrm{M}$ ) of ethidium in ethidium-DNA complex, $\mathrm{C}_{50}$ is drug concentration $(\mu \mathrm{M})$ to effect $50 \%$ drop in fluorescence of bound ethidium, and $10^{7}$ is the value of $\mathrm{K}_{\text {app }}$ assumed for ethidium in the complex. e The binding site preference is considered to be significant only for [AT]/[GC] ratio differing by > $30 \%$ from the sequence-neutral unity value (i.e., $<0.7$ or >1.3). In parentheses the values of [AT]/[GC] ratio ([GC]/[AT] for 5). ${ }^{\text {f Drug }}$ concentration required to inhibit cell growth by $50 \%$, all assays were performed in triplicate. 9 I solated as hydrochloride salts. ${ }^{h}$ Data from ref 7. ' Data from ref 8.
parison of the related pairs $\mathbf{9 a}, \mathbf{i}, \mathbf{9 b}, \mathbf{j}, \mathbf{1 3 a}, \mathbf{h}$, and $\mathbf{1 3 b}, \mathbf{i}$ indicate that the "acridinic" chromophore generally yields higher affinity than the "acridonic" one. The most relevant difference is shown by the pair 9b,j, 9j being 220 times more affinic than 9b. (f) Effect of the linker position: contrasting results can be observed by comparing compounds 9 (linker in positions 4,4') with related 13 (linker in positions 1,1'). The rank affinity order for unsubstituted compounds is 9a > 13a, 9b < $\mathbf{1 3 b}, 9 \mathrm{i}>13 \mathrm{~h}$, and $\mathbf{9 j}>13 \mathrm{i}$; for methoxy derivatives is $\mathbf{9 e}, \mathbf{f}>\mathbf{1 3 d}, \mathbf{e}$, while for hydroxy is $\mathbf{9 g}, \mathbf{h}<\mathbf{1 3 f}, \mathbf{g}$.

The binding behavior of target compounds with synthetic polynucleotides reflects what observed for CTDNA. The observed binding site preference indicates a frequent, remarkable in some cases, preference for binding to AT-rich duplexes. Previously, in contrast to other acridine/acridone derivatives, we have noted a moderate AT preference for monomers 4 related to "acridonic" dimers. ${ }^{7}$ Compounds with "acridinic" chromophore, $\mathbf{9 i}, \mathbf{j}$ and $\mathbf{1 3 h}, \mathbf{i}$, show AT $\gg$ GC preference, in contrast with related monomer 5 endowed with GC > AT preference, suggesting a different binding mechanism between "acridinic" dimers and monomer 5 .

Cytotoxic Activity. In vitro cytotoxic potencies of target compounds $\mathbf{9 a} \mathbf{- j}$ and 13a-i, of intermediates $\mathbf{8 a}, \mathbf{b}$, of related monomers $\mathbf{4 a}-\mathbf{d}$ and $\mathbf{5}$, and of reference drug $\mathrm{M} x$ against human colon adenocarcinoma cell line (HT29) are tabulated in Table 1. The results indicate that compounds 9 f and 13b,c possess excellent antiproliferative activity, with $\mathrm{IC}_{50}$ values in the Iow/sub nM range, being remarkably more potent than Mx itself.

The following remarks can be made:
(i) Regarding the linker: (a) In the unsubstituted $(X=H)$ subseries $\mathbf{9 a}-\mathbf{d}$ with four different linkers, $\mathbf{9 b}$, with $\mathrm{Y}=-\left(\mathrm{CH}_{2}\right)_{3} \mathrm{~N}\left(\mathrm{CH}_{3}\right)\left(\mathrm{CH}_{2}\right)_{3}$-, appears to be the most active ( $\mathrm{IC}_{50}$ in the high nM range); on the other hand, 9c, does not inhibit the 50\% cell growth even at highest concentration ( $10 \mu \mathrm{M}$ ) tested; 9a,d, with $\mathrm{IC}_{50}$ in sub $\mu \mathrm{M}$ range, are in the middle. (b) Comparison of the homologous pairs (8a,b, 9a,b, 9e,f, 9g,h, 9i,j, 13a,b, 13d,e, $\mathbf{1 3 f}, \mathbf{g}$, and $\mathbf{1 3 h}, \mathbf{i}$ ) dearly indicates that the best results are always obtained with $Y=-\left(\mathrm{CH}_{2}\right)_{3} \mathrm{~N}\left(\mathrm{CH}_{3}\right)\left(\mathrm{CH}_{2}\right)_{3}-$ either at the $4,4^{\prime}$ or $1,1^{\prime}$ positions, according to what was observed for similar derivatives. ${ }^{5}$ (c) Analysis of the linker position leads to contrasting results: in the pairs with $X=H$ ( $9 \mathbf{a}, \mathbf{b}, \mathbf{9 i}, \mathbf{j}, \mathbf{1 3 a}, \mathbf{b}$, and 13h,i, respectively), compounds 13 (linker at $1,1^{\prime}$ ) are more cytotoxic than 9 (linker at $4,4^{\prime}$ ). F or the pairs with $\mathrm{X}=\mathrm{OMe}$ or $\mathrm{OH}(\mathbf{9 e}, \mathbf{f}$, $\mathbf{9 g}, \mathbf{h}, \mathbf{1 3 d}, \mathbf{e}$, and $13 \mathbf{f}, \mathbf{g}$, respectively), the opposite is observed ( 9 more cytotoxic than 13).
(ii) Regarding the side chains of subseries 9, the $\mathrm{C}_{50}$ comparison of related pairs $\mathbf{8 a}, \mathbf{b}$ and $\mathbf{9 a}, \mathbf{b}$ underlines the importance of the basic side chains in $1,1^{\prime}$ for the cytotoxic activity. The intermediate 8a,b, without these side chains, are 6-11 times less active than corresponding target $\mathbf{9 a}, \mathbf{b}$, bearing the side chains at $\mathbf{1 , 1} \mathbf{1}^{\prime}$.
(iii) In respect to substituent groups at the 7,7' positions, it can be noted that the nature of the substituents remarkably influences the activity, but with a different trend in subseries 9 and 13. In subseries 9, the cytotoxicity rank order of derivatives with the same linker is $\mathbf{9 f}(X=O M e) \gg \mathbf{9 b}(X=H) \gg \mathbf{9 h}(X=O H)$

Table 2. Percent Growth of Some NCI Cell Lines Exposed 48 h at Three Increasing Concentrations ( $10^{-8}, 10^{-6}$, and $10^{-4}$ or $10^{-5} \mathrm{M}$ ) of Selected Compounds. The Negative Values Indicate the Percent of Cell Killed

|  | 9b |  |  | 97 |  |  | 13b |  |  | 13c |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| cell line | $10^{-8}$ | $10^{-6}$ | $10^{-4}$ | $10^{-8}$ | $10^{-6}$ | $10^{-4}$ | $10^{-8}$ | $10^{-6}$ | $10^{-5}$ | $10^{-8}$ | $10^{-6}$ | $10^{-5}$ |
| leucemia: RPMI-8226 | 32 | -23 | -46 | 4 | -36 | -44 | 16 | -37 | -49 | 12 | -44 | -58 |
| lung-NSC: $\mathrm{NCI}-\mathrm{H} 460$ | 42 | 11 | -86 | 9 | -10 | -90 | 12 | -52 | -90 | 28 | -31 | -95 |
| colon: HCT-116 | 20 | -14 | -90 | 17 | -23 | -97 | 20 | -35 | -96 | 31 | -5 | -94 |
| cns: SNB-19 | 44 | 4 | -88 | 38 | 3 | -85 | 38 | -20 | -55 | 42 | -7 | -96 |
| melanoma: LOX IMVI | 26 | -23 | -82 | 31 | -2 | -73 | 30 | -65 | -93 | 28 | -40 | -93 |
| ovarian: OVCAR-4 | 19 | -22 | -97 | 38 | -21 | -96 | 46 | -47 | -74 | 67 | 16 | -72 |
| renal: 786-0 | 47 | 13 | -88 | 28 | 1 | -98 | 30 | -26 | -98 | 41 | 6 | -95 |
| prostate: DU-145 | 52 | 5 | -100 | 25 | -25 | -77 | 11 | -31 | -72 | 55 | -2 | -93 |
| breast: MCF 7 | 20 | -8 | -78 | 18 | 0 | -85 | 22 | -6 | -86 | 26 | 0 | -86 |
| mean of the cell lines | 34 | -6 | -84 | 23 | -13 | -83 | 25 | -35 | -79 | 32 | -28 | -87 |

and $\mathbf{9 a}(\mathrm{X}=\mathrm{H})>\mathbf{9 e}(\mathrm{X}=\mathrm{OMe}) \gg \mathbf{9 g}(\mathrm{X}=\mathrm{OH})$. In subseries 13, the cytotoxicity rank order of derivatives with the same linker is $\mathbf{1 3 c}\left(\mathrm{X}=\mathrm{NO}_{2}\right)>13 \mathrm{~b}(\mathrm{X}=\mathrm{H})$ $\gg 13 \mathrm{~d}(\mathrm{X}=\mathrm{OMe}) \gg 13 \mathrm{~g}(\mathrm{X}=\mathrm{OH})$ and 13a $(\mathrm{X}=\mathrm{H}) \gg$ 13d $(X=O M e)>13 g(X=O H)$. In general, can be remarked that hydroxy derivatives are always the least cytotoxic target drugs, while 13c, with $\mathrm{X}=\mathrm{NO}_{2}$, the linker at $1,1^{\prime}$ and $Y=-\left(\mathrm{CH}_{2}\right)_{3} \mathrm{~N}\left(\mathrm{CH}_{3}\right)\left(\mathrm{CH}_{2}\right)_{3^{-}}$, is the most active compound. Unfortunately, it was not possible synthesize the corresponding derivative for 9 group.
(iv) Regarding the chromophore: comparing the I $\mathrm{C}_{50}$ of $\mathbf{9 a}, \mathbf{b}$ and $\mathbf{1 3} \mathbf{a}, \mathbf{b}$ (9-acridone chromophore) with that of the corresponding $9 \mathrm{ij}, \mathbf{j}$ and $\mathbf{1 3 h}, \mathrm{i}$ (acridine chromophore), the activity rank order of related pairs is $\mathbf{9 a}>\mathbf{9 i}, \mathbf{9 b}>\mathbf{9 j}, \mathbf{1 3 a} \cong \mathbf{1 3 h}$, and $\mathbf{1 3 b} \gg \mathbf{1 3 i}$. It is clear that in both the subseries 9 and 13 the best results are obtained with the "acridonic" chromophore, in contrast to that observed with related monomers 4a and 5. However, it should be noted that monomers 4a-d generally possess antiproliferative activity less than or much less than that of related bis compounds with $\mathrm{Y}=$ $-\left(\mathrm{CH}_{2}\right) \mathrm{N}\left(\mathrm{CH}_{3}\right)\left(\mathrm{CH}_{2}\right)_{3^{-}}$, but greater than that of related bis compounds with $\mathrm{Y}=-\left(\mathrm{CH}_{2}\right)_{2} \mathrm{~N}\left(\mathrm{CH}_{3}\right)\left(\mathrm{CH}_{2}\right)_{2}$-; instead, monomer 5 is more active than all the related bis $\mathbf{9 i}, \mathbf{j}$ and $\mathbf{1 3 h}$,i. A possible explanation can derive from the DNA binding behavior: the "acridonic" bis derivatives, more active than the corresponding monomers 4, are also more DNA affinic; in the "acridinic" subseries, 9i,j and $\mathbf{1 3 h} \mathbf{i} \mathbf{i}$, only $9 \mathbf{j}$ is more DNA affinic than corresponding monomer 5, but with a very remarkable AT preference, while 5 has a noticeable GC preference. So, the opposite binding site preference may influence the potency of cytotoxic activity.

Generally, there is no quantitative correlation between $\mathrm{IC}_{50}$ and $\mathrm{K}_{\text {app }}$ values. However, some generalizations can be made: (i) in the homologous pairs, the shortest linker, $\mathrm{Y}=-\left(\mathrm{CH}_{2}\right)_{2} \mathrm{~N}\left(\mathrm{CH}_{3}\right)\left(\mathrm{CH}_{2}\right)_{2}$-, al ways corresponds to an inferior cytotoxicity and DNA affinity in respect to the longest linker, $Y=-\left(\mathrm{CH}_{2}\right)_{3} \mathrm{~N}\left(\mathrm{CH}_{3}\right)$ -$\left(\mathrm{CH}_{2}\right)_{3}-$; (ii) compound $\mathbf{9 c}$, the weakest DNA ligand, is not cytotoxic; (iii) the most active derivatives, $9 f$ and 13b,c, also possess very high $K_{\text {app }}$ values (range 1-2× $10^{8}$ ); (iv) the hydrophilic hydroxy groups at 7,7' lead to compounds, $\mathbf{9 g}, \mathbf{h}$ and $\mathbf{1 3 f}, \mathbf{g}$, endowed with a good DNA affinity, but with a poor cytotoxicity, indicating that other factors, e.g., cellular uptake, may influence the activity in these cases.

Compounds 9b,f and 13b,c were selected for the National Cancer Institute ( NCI ) screening on 60 human tumor cell lines. This screen is designed to discover the spectrum of activity and, eventually, selectivity of drugs.

The data from this assay can be presented in several different formats. Since it is not practical to report all experimental data available, we choose to describe in Table 2, in one of the possible formats, the antiproliferative activity of selected compounds against one cell line of each NCI subpanel and the mean of the activity on these nine cell lines. So, for each compound, we report the percent growth of some cell lines exposed 48 h at three different increasing concentrations ( $10^{-8}$, $10^{-6}$, and $10^{-4}$ or $10^{-5} \mathrm{M}$, respectively) of drug. Positive values represent the percent growth of each cell line in respect to the untreated control (100\% growth) and give an indication of the cytostatic action. Negative values represent the percent of death cells in respect to the initial number and give an indi cation of the derivative's cell killing capacity. The data show that all the compounds explicate a strong cytostatic action (70-80\% average of cell growth inhibition) at 10 nM concentration; the differences are not so marked as observed with the HT29 cell line at 144 h of drug exposure, the antiproliferative activity average being in the order: 9f $\cong \mathbf{1 3 b}>\mathbf{1 3} \cong \cong \mathbf{9 b}$. At $10^{-6} \mathrm{M}$ concentration there is not only a general stop of cellular growth, but often it begins a significant reduction of the initial cell number (cell killing): the activity rank order is 13b > 13c > $\mathbf{9 f}>\mathbf{9 b}$, reflecting the rank observed with HT29 cell line. A very marked cell killing capacity ( $80-90 \%$ ) is shown by $\mathbf{9 b}, \mathbf{f}$ at $10^{-4} \mathrm{M}$ and by $\mathbf{1 3 b}, \mathbf{c}$ at $10^{-5} \mathrm{M}$. It is worth noting the potency and the broad spectrum of activity of all four compounds. Indeed, a COMPARE analysis ${ }^{16}$ was performed with these compounds to check whether they resemble previously identified anticancer drugs. The results are similar, with $\mathbf{1 3 b}$ and 13c being more closely related to each other than the other two sets of compounds. However, the highest correlation (correlation coefficient 0.77) was found for 9f with rapamycin, an antibiotic with immunosuppressant activity. Other very toxic immunosuppressant or antiblastic agents with various mechanisms of action were found with this analysis, consistent with the high cytotoxicity exhibited by this class of bis-acridine derivatives.

## Conclusions

From the present study we can conclude the following: (i) Our expectation, that the outstanding biological response shown by the monomers 4 and 5 could be increased with bis derivatives 9 and 13, was fulfilled for target compounds with "acridonic" chromophores. In fact, 9b,f,h and 13b,c,e exhibit enhanced cytotoxic activity and higher DNA affinity than corresponding
monomers. (ii) The most potent compounds, 9b,f and 13b,c, represent new leads in the field of anticancer acridine derivatives, being endowed with excellent DNA affinity, a broad spectrum of activity, and remarkable cytotoxic potency. (iii) However, further studies, to better understand the mechanism of action of these bis derivatives, are needed.

## Experimental Section

Synthetic Chemistry. Melting points were determined on a Büchi 510 apparatus and are uncorrected. Thin-layer chromatography (TLC) was accomplished using plates precoated with silica gel 60 F-254 (Merck). All ${ }^{1} \mathrm{H}$ NMR spectra were recorded on a Varian VXR 300 instrument. Chemical shifts are reported as $\delta$ values (ppm) downfield from internal Measi in the solvent shown. The following NMR abbreviations are used: br (broad), s (singlet), d (doublet), t (triplet), m (multiplet), ar (aromatic proton), ex (exchangeable with $\mathrm{D}_{2} \mathrm{O}$ ). Elemental analyses were performed on a Model 1106 elemental analyzer (Carlo Erba Strumentazione).

1,7-Bis(1-chloro-9-oxo-9,10-dihydroacridine-4-carbonyl)-4-methyl-1,4,7-triazaheptane (8a). Example of General Procedure for the Preparation of $\mathbf{8 a - f}$. To a mixture of $6^{9}(0.5 \mathrm{~g}, 1.83 \mathrm{mmol})$ and triethylamine ( $0.26 \mathrm{~mL}, 1.83 \mathrm{mmol}$ ) in $\mathrm{CHCl}_{3}(20 \mathrm{~mL})$ was added a solution of CICOOEt $(0.26 \mathrm{~mL}$, 1.83 mmol ) in $\mathrm{CHCl}_{3}(10 \mathrm{~mL})$ dropwise at $0^{\circ} \mathrm{C}$. After stirring at room temperature for 1 h , the mixture was again cooled at $0^{\circ} \mathrm{C}$ and $\mathrm{N}^{2}$-methyldiethylentriamine ( $0.12 \mathrm{~mL}, 0.915 \mathrm{mmol}$ ) was added. The resulting mixture was stirred for 4 h at room temperature. The precipitate solid was filtered and washed with boiling $\mathrm{MeOH}(2 \times 10 \mathrm{~mL})$ to yield pure 8 a as hydrochloride salt: ${ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ) $\delta 2.80-3.11\left(\mathrm{~m}, 7 \mathrm{H}, \mathrm{CH}_{3}\right.$ $\left.+2 \times \mathrm{CH}_{2}\right), 3.53-3.75\left(\mathrm{~m}, 4 \mathrm{H}, 2 \times \mathrm{CH}_{2}\right), 7.09(\mathrm{~d}, 2 \mathrm{H}, \mathrm{ar}), 7.15$ (t, 2H, ar), 7.35 (d, 2H, ar), 7.52 (t, 2H, ar), 8.00-8.11 (m, 4H, ar), 9.20 (br t, $2 \mathrm{H}, 2 \times \mathrm{C}-\mathrm{NH}, \mathrm{ex}), 9.68$ (br s, $1 \mathrm{H}, \mathrm{ex}), 12.50$ (s, $2 \mathrm{H}, 10-\mathrm{H}+10^{\prime}-\mathrm{H}$, ex).

Derivatives $\mathbf{8 b}-\mathbf{f}$ were prepared in a similar manner from the suitable 1-chloroacridine-4-carboxylic acid and the appropriate amine.

1,7-B is\{ 1-[(2-(dimethylamino)ethyl)ami no]-9-0xo-9,10-dihydroacridine-4-carbonyl\}-4-methyl-1,4,7-triazaheptane (9a). Example of General Procedure for the Preparation of $9 \mathbf{a}-\mathbf{f}$. The hydrochloride salt of $\mathbf{8 a}(0.15 \mathrm{~g}, 0.23$ mmol) was refluxed in N,N-dimethylethylendiamine ( 2 mL ) for 1 h . After cooling, the mixture was partitioned between $\mathrm{CHCl}_{3}(2 \times 20 \mathrm{~mL})$ and an excess of 1 M aqueous $\mathrm{Na}_{2} \mathrm{CO}_{3}(30$ mL ). The organic layer was worked up to give a residue which was flash-chromatographed on silica gel column eluted with $\mathrm{CHCl}_{3} / \mathrm{MeOH} / \mathrm{NH}_{3}(1: 1: 0.01 \mathrm{v} / \mathrm{v})$ to give pure 9a: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 2.31\left(\mathrm{~s}, 12 \mathrm{H}, 4 \times \mathrm{CH}_{3}\right), 2.50\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.60(\mathrm{t}$, $\left.4 \mathrm{H}, 2 \times \mathrm{CH}_{2}\right), 2.77-2.87\left(\mathrm{~m}, 4 \mathrm{H}, 2 \times \mathrm{CH}_{2}\right), 3.54-3.65(\mathrm{~m}, 4 \mathrm{H}$, $2 \times \mathrm{CH}_{2}$ ), $5.80(\mathrm{~d}, 2 \mathrm{H}, \mathrm{ar}), 6.78(\mathrm{br} \mathrm{t}, 2 \mathrm{H}, 2 \times \mathrm{CONH}, \mathrm{ex}), 7.10-$ 7.21 (m, 4H, ar), 7.41-7.52 (m, 4H, ar), 8.28 (d, 2H, ar), 10.85 (br t, 2H, $2 \times 1-\mathrm{NH}$, ex), 13.18 (s, 2H, 10-H + 10'-H, ex).

Derivatives $\mathbf{9 b}-\mathbf{f}$ were prepared in a similar manner from the corresponding intermediates $\mathbf{8 b}-\mathbf{f}$.

1,7-Bis\{ 1-[(2-(dimethylami no)ethyl )amino]-7-hydroxy-9-0xo-9,10-dihydroacridine-4-carbonyl\}-4-methyl-1,4,7triazaheptane (9g). Example of General Procedure for the Preparation of $\mathbf{9 g}, \mathbf{h}$. 9 e ( $0.13 \mathrm{~g}, 0.16 \mathrm{mmol}$ ) was suspended in aqueous $\mathrm{HBr} 48 \%$ ( 3 mL ) and refluxed for 1 h . The mixture was cooled at room temperature and partitioned between $\mathrm{CHCl}_{3}(2 \times 20 \mathrm{~mL})$ and an excess of 1 M aqueous $\mathrm{Na}_{2} \mathrm{CO}_{3}(20 \mathrm{~mL})$. The organic layer was worked up to give a residue which was flash-chromatographed on silica gel column eluted with $\mathrm{CHCl}_{3} / \mathrm{MeOH} / \mathrm{NH}_{3}(1: 1: 0.01 \mathrm{v} / \mathrm{v})$ to give pure $\mathbf{9 g}$ : ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{DMSO}_{\mathrm{d}}\right) \delta 2.23\left(\mathrm{~s}, 12 \mathrm{H}, 4 \times \mathrm{CH}_{3}\right), 2.34\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$, 2.51-2.74 (m, 8H, $\left.4 \times \mathrm{CH}_{2}\right), 3.17-3.22\left(\mathrm{~m}, 4 \mathrm{H}, 2 \times \mathrm{CH}_{2}\right)$, 3.42-3.46 (m, 4H, $2 \times \mathrm{CH}_{2}$ ), $6.05(\mathrm{~d}, 2 \mathrm{H}, \mathrm{ar}), 7.16$ (d, 2H, ar), 7.41-7.50 (m, 4H, ar), 7.89 (d, 2H, ar), 8.22 (br t, 2H, $2 \times$ CONH, ex), 9.65 (br s, 2H, $2 \times \mathrm{OH}$, ex), 10.83 (br t, 2H, $2 \times$ 1-NH, ex), 13.62 (s, 2H, 10-H + 10'-H, ex).

Derivative 9h was prepared in a similar manner from 9 f.
1,7-Bis\{1-[(2-(dimethylamino)ethyl)amino]acridine-4carbonyl \}-4-methyl-1,4,7-triazaheptane (9i). Example of General Procedure for the Preparation of $9 \mathrm{i}, \mathrm{j}$. The hydrochloride salt of $9 \mathbf{a}(0.35 \mathrm{~g}, 0.42 \mathrm{mmol})$ was refluxed in $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}(3: 1,30 \mathrm{~mL})$. Portions of aluminum foil ( 0.35 g ) were amalgamated in a solution of $\mathrm{HgCl}_{2}(1.3 \mathrm{~g})$ in EtOH (25 mL ) and added to the above boiling sol ution over 30 min . The mixture was refluxed for other 30 min and then filtered and the solid washed with hot EtOH ( 20 mL ). The filtrate was diluted with $\mathrm{H}_{2} \mathrm{O}(50 \mathrm{~mL})$, and $\mathrm{FeCl}_{3}(1.2 \mathrm{~g})$ was added. The resulting mixture was partitioned between $\mathrm{CHCl}_{3}(3 \times 30 \mathrm{~mL})$ and an excess of 1 M aqueous $\mathrm{Na}_{2} \mathrm{CO}_{3}(30 \mathrm{~mL})$. The organic layer was worked up to give a residue which was flashchromatographed on silica gel column eluted with $\mathrm{CHCl}_{3} /$ $\mathrm{MeOH} / \mathrm{NH}_{3}(1: 1: 0.01 \mathrm{v} / \mathrm{v})$ to give pure $9 \mathrm{i}:{ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta$ $2.46\left(\mathrm{~s}, 12 \mathrm{H}, 4 \times \mathrm{CH}_{3}\right), 2.62\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.75-2.90(\mathrm{~m}, 4 \mathrm{H}$, $\left.2 \times \mathrm{CH}_{2}\right), 2.90-3.06\left(\mathrm{~m}, 4 \mathrm{H}, 2 \times \mathrm{CH}_{2}\right), 3.30-3.44(\mathrm{~m}, 4 \mathrm{H}, 2 \times$ $\left.\mathrm{CH}_{2}\right), 3.81-3.97\left(\mathrm{~m}, 4 \mathrm{H}, 2 \times \mathrm{CH}_{2}\right), 5.74(\mathrm{br} \mathrm{t}, 2 \mathrm{H}, 2 \times \mathrm{CONH}$, ex), 6.27 (d, 2H, ar), 7.07 (t, 2H, ar), 7.32 (t, 2H, ar), 7.51 (d, $2 \mathrm{H}, \mathrm{ar}), 7.71$ (d, 2H, ar), 8.32 (s, 2H, ar), 8.50 (d, 2H, ar), 11.60 (s, 2H, $2 \times 1-\mathrm{NH}$, ex).

Derivative $\mathbf{9 j}$ was prepared in a similar manner from $\mathbf{9 b}$.
1,7-Bis\{4-[N-(2-(dimethylamino)ethyl)carbamoyl]-9-oxo-9,10-dihydroacridin-1-yl\}-4-methyl-1,4,7-triazaheptane (13a). Example of General Procedure for the Preparation of $13 \mathrm{a}-\mathbf{e} . \mathrm{N}^{4}$-[2-(Dimethylamino)ethyl]-1-chloro-9-oxo-9,10-dihydro-4-acridinecarboxamide ${ }^{7}$ (10, $0.5 \mathrm{~g}, 1.45$ mmol), $\mathrm{N}^{2}$-methyldiethylentriamine ( $0.1 \mathrm{~mL}, 0.72 \mathrm{mmol}$ ), and triethylamine $(0.5 \mathrm{~mL})$ were stirred in 2-ethoxyethanol ( 10 mL ) at $120{ }^{\circ} \mathrm{C}$ for 5 h . The resulting mixture was partitioned between $\mathrm{CHCl}_{3}(2 \times 30 \mathrm{~mL})$ and an excess of 1 M aqueous $\mathrm{Na}_{2} \mathrm{CO}_{3}(30 \mathrm{~mL})$. The organic layer was worked up to give a residue which was flash-chromatographed on a silica gel column eluted with $\mathrm{CHCl}_{3} / \mathrm{MeOH} / \mathrm{NH}_{3}(1: 1: 0.01 \mathrm{v} / \mathrm{v})$ to give pure 13a: ${ }^{1} \mathrm{H} N M R\left(\mathrm{CDCl}_{3}\right) \delta 2.32\left(\mathrm{~s}, 12 \mathrm{H}, 4 \times \mathrm{CH}_{3}\right), 2.48(\mathrm{~s}$, $\left.3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.59\left(\mathrm{t}, 4 \mathrm{H}, 2 \times \mathrm{CH}_{2}\right), 2.83\left(\mathrm{t}, 4 \mathrm{H}, 2 \times \mathrm{CH}_{2}\right), 3.34-$ $3.44\left(\mathrm{~m}, 4 \mathrm{H}, 2 \times \mathrm{CH}_{2}\right), 3.45-3.55\left(\mathrm{~m}, 4 \mathrm{H}, 2 \times \mathrm{CH}_{2}\right), 6.16(\mathrm{~d}$, $2 \mathrm{H}, \mathrm{ar}), 6.88$ (brt, 2H, $2 \times \mathrm{CONH}, \mathrm{ex}$ ), 7.10 (7, 2H, ar), 7.25 (d, 2H, ar), 7.49-7.60 (m, 4H, ar), 8.20 (d, 2H, ar), 10.97 (br t, $2 \mathrm{H}, 2 \times 1-\mathrm{NH}$, ex), 13.20 (s, 2H, 10-H + 10'-H, ex).

Derivative 13b-e were prepared in a similar manner from the appropriate acridine-4-carboxamide and the suitable triamine.

1,7-B is $\{4-[\mathrm{N}$-(2-(dimethylamino)ethyl )carbamoyl]-7-hydroxy-9-0xo-9,10-dihydroacridin-1-yl\}-4-methyl-1,4,7triazaheptane (13f). Example of General Procedure for the Preparation of $\mathbf{1 3 f}, \mathrm{g}$. 13d ( $0.28 \mathrm{~g}, 0.35 \mathrm{mmol}$ ) was suspended in aqueous $\mathrm{HBr} 48 \%$ ( 3 mL ) and refluxed for 1 h . The mixture was cooled at room temperature and partitioned between $\mathrm{CHCl}_{3}(2 \times 20 \mathrm{~mL})$ and an excess of 1 M aqueous $\mathrm{Na}_{2} \mathrm{CO}_{3}(20 \mathrm{~mL})$. The organic layer was worked up to give a residue which was flash-chromatographed on silica gel column el uted with $\mathrm{CHCl}_{3} / \mathrm{MeOH} / \mathrm{NH}_{3}(1: 1: 0.01 \mathrm{v} / \mathrm{v})$ to give pure 13f: ${ }^{1} \mathrm{H}$ NMR (DMSO- $\mathrm{d}_{6}$ ) $\delta 2.25\left(\mathrm{~s}, 12 \mathrm{H}, 4 \times \mathrm{CH}_{3}\right), 2.38-2.50(\mathrm{~m}$, $\left.7 \mathrm{H}, \mathrm{CH}_{3}+2 \times \mathrm{CH}_{2}\right), 2.75-2.93\left(\mathrm{~m}, 4 \mathrm{H}, 2 \times \mathrm{CH}_{2}\right), 3.32-3.77$ $\left(\mathrm{m}, 8 \mathrm{H}, 4 \times \mathrm{CH}_{2}\right), 6.31(\mathrm{~d}, 2 \mathrm{H}, \mathrm{ar}), 7.10(\mathrm{~d}, 2 \mathrm{H}, \mathrm{ar}), 7.41-7.54$ (m, 4H, ar), 7.98 (d, 2H, ar), 8.30 (br t, 2H, $2 \times$ CONH, ex), 9.64 (br s, $2 \mathrm{H}, 2 \times \mathrm{OH}, \mathrm{ex}), 10.96$ (br t, $2 \mathrm{H}, 2 \times 1-\mathrm{NH}, \mathrm{ex})$, 13.69 (s, 2H, 10-H + 10'-H, ex).

Derivative $\mathbf{1 3} \mathbf{g}$ was prepared in a similar manner from $\mathbf{1 3 e}$.
1,7-Bis\{4-[N-(2-(dimethylami no)ethyl)carbamoyl]ac-ridin-1-yl\}-4-methyl-1,4,7-triazaheptane (13h). Example of General Procedure for the Preparation of $\mathbf{1 3 h}, \mathrm{i}$. The hydrochloride salt of $13 \mathrm{a}(0.19 \mathrm{~g}, 0.26 \mathrm{mmol})$ was refluxed in $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}(3: 1,15 \mathrm{~mL})$. Portions of aluminum foil ( 0.19 g ) were amalgamated in a solution of $\mathrm{HgCl}_{2}(0.76 \mathrm{~g})$ in EtOH ( 18 mL ) and added to the above boiling sol ution over 30 min . The mixture was refluxed for other 30 min and then filtered and the solid washed with hot EtOH ( 20 mL ). The filtrate was diluted with $\mathrm{H}_{2} \mathrm{O}(40 \mathrm{~mL})$, and $\mathrm{FeCl}_{3}(0.6 \mathrm{~g})$ was added. The resulting mixture was partitioned between $\mathrm{CHCl}_{3}(3 \times 30 \mathrm{~mL})$ and an excess of 1 M aqueous $\mathrm{Na}_{2} \mathrm{CO}_{3}(30 \mathrm{~mL})$. The organic layer was worked up to give a residue which was flash-
chromatographed on silica gel column eluted with $\mathrm{CHCl}_{3} /$ $\mathrm{MeOH} / \mathrm{NH}_{3}(1: 1: 0.01 \mathrm{v} / \mathrm{v})$ to give pure $\mathbf{1 3 h}:{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ $\delta 2.45\left(\mathrm{~s}, 12 \mathrm{H}, 4 \times \mathrm{CH}_{3}\right), 2.57\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.70(\mathrm{t}, 4 \mathrm{H}, 2 \times$ $\left.\mathrm{CH}_{2}\right), 2.90-3.02\left(\mathrm{~m}, 4 \mathrm{H}, 2 \times \mathrm{CH}_{2}\right), 3.40-3.53\left(\mathrm{~m}, 4 \mathrm{H}, 2 \times \mathrm{CH}_{2}\right)$, $3.68-3.82\left(\mathrm{~m}, 4 \mathrm{H}, 2 \times \mathrm{CH}_{2}\right), 5.80(\mathrm{br} \mathrm{t}, 2 \mathrm{H}, 2 \times \mathrm{CONH}, \mathrm{ex})$, 6.48 (d, 2H , ar), 6.97-7.17 (m, 4H, ar), 7.58 (t, 2H, ar), 7.78 ( $\mathrm{d}, 2 \mathrm{H}, \mathrm{ar}$ ), $8.28(\mathrm{~s}, 2 \mathrm{H}, \mathrm{ar}), 8.73(\mathrm{~d}, 2 \mathrm{H}, \mathrm{ar}), 11.65(\mathrm{br} \mathrm{t}, 2 \mathrm{H}$, $2 \times 1-\mathrm{NH}$, ex).

Derivative 13i was prepared in a similar manner from 13b.
Biophysical Evaluation. 1. DNA Unwinding. To test the drug's ability to unwind DNA, we performed a direct assay employing negative supercoiled pBR322 and PPAC as control. Plasmid $(0.25 \mu \mathrm{~g})$ was incubated in TE buffer, pH 8.0, with $0.1,1,10$, and $100 \mu \mathrm{M}$ of either compound, for 1 h at room temperature. After the incubation period, the complex was directly loaded on a $1 \%$ agarose gel in Tris-borate 50 mM , EDTA 1 mM , and run at $5 \mathrm{~V} / \mathrm{cm}$ for 4 h . The gel was then stained with ethidium bromide to visualize the change in plasmid mobility upon complex formation.
2. Fluorescence Binding Studies. The fluorometric assays have been described previously. ${ }^{13}$ The $\mathrm{C}_{50}$ values for ethidium displacement from CT-DNA and from synthetic [poly-(dA-dT) $]_{2}(A T)$ and $[p o l y(d G-d C)]_{2}$ (GC) ol igonucl eotides were determined using aqueous buffer ( $10 \mathrm{mM} \mathrm{Na} \mathrm{NPO}_{4}, 10 \mathrm{mM}$ $\mathrm{NaH}_{2} \mathrm{PO}_{4}, 1 \mathrm{mM}$ EDTA, pH 7.0) containing $1.26 \mu \mathrm{M}$ ethidium bromide and $1 \mu \mathrm{M}$ CT-DNA, AT, and GC, respectively. ${ }^{13,14}$

All measurements were made in $10-\mathrm{mm}$ quartz cuvettes at $20^{\circ} \mathrm{C}$ using a Perkin-EImer LS5 instrument (excitation at 546 nm ; emission at 595 nm ) following serial addition of aliquots of a stock drug solution ( $\sim 5 \mathrm{mM}$ in DMSO). The $\mathrm{C}_{50}$ values are defined as the drug concentrations which reduce the fluorescence of the DNA-bound ethidium by $50 \%$ and are cal culated as the mean from three determinations.
3. In Vitro Cytotoxicity. Human Colon Adenocarcinoma Experimental Protocol. Establishment details of human col on adenocarcinoma carcinoma cell lines (HT29, LoVo sensitive, and LoVo/Dx resistant) have been previously described. ${ }^{17-19}$ Drug solutions of appropriate concentration were added to a culture containing HT29 cells at $2.5 \times 10^{4}$ cells $/ \mathrm{mL}$ of medium ${ }^{17}$ or to a culture containing LoVo or LoVo/ Dx cells at $2.5 \times 10^{5}$ cell $/ / \mathrm{mL}$ of medium. ${ }^{18}$ All assays were performed in triplicate.

Supporting Information Available: Detailed information on target compounds ( ${ }^{1} \mathrm{H}$ NMR, purification procedure). This material is available free of charge via the Internet at http://pubs.acs.org.

## References

(1) Nitiss, J . L.; Zhou, J.; Rose, A., H siung, Y.; Gale, K. C.; Osheroff, N. The Bis(naphthalimide) DMP-840 Causes Cytotoxicity by Its Action against Eukaryotic Topoisomerase II. Biochemistry 1998, 37, 3078-3085.
(2) O'Reilly, S.; Baker, S. D., Sartorius, S.; Rowinsky, E. K.; Finizio, M.; Lubiniecki, G. M.; Grochow, L. B.; Gray, J . E.; Pieniaszek, H. J .; Donehower, R. C. A Phase I and Pharmacologic Study of DMP 840 Administered by 24-H our Infusion. Ann. Oncol. 1998, 9, 101-104.
(3) Bousquet, P. F.; Brana, M. F.; Conlon, D.; Fitzgerald, K. M.; Perron, D.; Cocchiaro, C.; Miller, R.; Moran, M.; George, J .; Qian, X. D. Preclinical Evaluation of LU 79553: a Novel Bis-Naphthal imide with Potent Antitumor Activity. Cancer Res. 1995, 55, 1176-1180.
(4) Cholody, W. M.; Hernandez, L.; Hassner, L.; Scudiero, D. A.; Djurickovic, D. B.; Michejda, C. J. Bisimidazoacridones and Related Compounds: New Antineoplastic Agents with High Selectivity against Colon Tumors. J. Med. Chem. 1995, 38, 3043-3052.
(5) Gamage, S. A.; Spicer, J . A.; Atwell, G. J .; Finlay, G. J .; Baguley, B. C.; Denny, W. A. Structure-Activity Relationships for Substituted Bis(acridine-4-carboxamides): A New Class of Anticancer Agents. J. Med. Chem. 1999, 42, 2383-2393.
(6) Gamage, S. A.; Spicer, J. A.; Finlay, G. J.; Stewart, A. J.; Charlton, P.; Baguley, B. C.; Denny, W. A. Dicationic Bis(9-methylphenazine-1-carboxamides): Relationships between Biological Activity and Linker Chain Structure for a Series of Potent TopoisomeraseTargeted Anticancer Drugs. J . Med. Chem. 2001, 44, 1407-1415.
(7) Antonini, I.; Polucci, P.; J enkins, T. C.; Kelland, L. R.; Menta, E.; Pescalli, N.; Stefanska, B.; Mazerski, J .; Martelli, S. 1-[( $\omega$ -Aminoalkyl)amino]-4-[N-( $\omega$-aminoalkyl)carbamoyl]-9-oxo-9, 10dihydroacridines as Intercalating Cytotoxic Agents: Synthesis, DNA Binding, and Biological Evaluation. J. Med. Chem. 1997, 40, 3749-3755.
(8) Antonini, I.; Polucci, P.; Kelland, L. R.; Spinelli, S.; Pescalli, N.; Martelli, S. N4-( $\omega$-Aminoal kyl)-1-[( $\omega$-aminoalkyl)amino]-4-acridinecarboxamides: Novel, Potent, Cytotoxic, and DNA-Binding Agents. J. Med. Chem. 2000, 43, 4801-4805.
(9) Rewcastle, G. W.; Denny, W. A. The Synthesis of Substituted 9-Oxoacridan-4-carboxylic Acids; Part 3. The Reaction of Methyl Anthranilates with Diphenyliodonium-2-carboxylates. Synthesis 1985, 220-222.
(10) Antonini, I.; Polucci, P.; Cola, D.; Palmieri, G. F.; Martelli, S. Synthesis of 7-Oxo-7H-benzo[e]perimidine-4-carboxamides as Potential Antitumor Drugs. II Farmaco 1992, 47, 1385-1393.
(11) Braña, M. F.; Castellano, J. M.; Perron, D.; Maher, C.; Conlon, D.; Bousquet, P. F.; Gorge, J.; Qian, X.-D.; Robinson, S. P. Chromophore-Modified Bis-Naphthalimides: Synthesis and Antitumor Activity of Bis-Dibenz[de,h]isoquinoline-1,3-diones. J. Med. Chem. 1997, 40, 449-454.
(12) Antonini, I.; Polucci, P.; Magnano, A.; Gatto, B.; Palumbo, M.; Menta, E.; Pescalli, N.; Martelli, S. 2,6-Di( $\omega$-aminoalkyl)-2,5,6,7tetrahydropyrazol o[3,4,5-mn]pyrimido[5,6,1-de]acridine-5,7-diones: Novel, Potent, Cytotoxic, and DNA-Binding Agents. J. Med. Chem. 2002, 45, 696-702.
(13) (a) McConnaughie, A. W.; J enkins, T. C. Novel AcridineTriazenes as Prototype Combilexins: Synthesis, DNA Binding and Biological Activity. J. Med. Chem. 1995, 38, 3488-3501. (b) J enkins, T. C. Optical Absorbance and Fluorescence Techniques for Measuring DNA-Drug Interactions. In Methods in M olecular Biology, Vol. 90: Drug-DNA Interaction Protocols; Fox, K. R., Ed.; Humana Press: Totawa, NJ , 1997; Ch 14, pp 195-218.
(14) (a) M organ, A. R.; Lee, J. S.; Pulleyblank, D. E.; Murray, N. L.; Evans, D. H. Review: Ethidium Fluorescence Assays. Part 1. Physicochemical Studies. Nucleic Acids Res. 1979, 7, 547-569. (b) Baguley, B. C.; Denny, W. A.; Atwell, G. J.; Cain, B. F. Potential Antitumor Agents. 34. Quantitative Relationships between DNA Binding and Molecular Structure for 9-Anilinoacridines Substituted in the Anilino Ring. J. Med. Chem. 1981, 24, 170-177.
(15) Bailly, C.; Pommery, N.; Houssin, R.; Hénichart, J .-P. Design, Synthesis, DNA Binding, and Biological Activity of a Series of DNA Minor Groove-Binding Intercalating Drugs. J . Pharm. Sci. 1989, 78, 910-917.
(16) Boyd, M. R. Status of the NCI Preclinical Antitumor Drug Discovery Screen. In Cancer: Principles and Practice of Oncol ogy Updates; DeVita, V. T., J r.; Hellman, S.; Rosenberg, S. A., Eds.; Lippincott: Philadelphia, PA, 1989; Vol. 3, pp 1-12.
(17) Antonini, I.; Polucci, P.; Cola, D.; Bontemps-Gracz,M.; Pescalli, N.; Menta, E.; Martelli, S. Pyrimido[4,5,6-kl]acridines, a New Class of Potential Anticancer Agents. Synthesis and Biological Evaluation. Anti-Cancer Drug Des. 1996, 11, 339-349.
(18) Krapcho, A. P.; Petry, M. E.; Getahun, Z.; Landi, J.J .; Stallman, J.; Polsenberg, J. F.; Gallagher, C. E.; Maresch, M. J.; Hacker, M. P.; Giuliani, F.' C.; Beggolin, G.; Pezzoni, G.; Menta, E.; Manzotti, C.; Oliva, A.; Spinelli, S.; Tognella, S. 6,9-Bis[(amino-alkyl)amino]benzo[g]lisoquinoline-5, 10-diones. A Novel Class of Chromophore-M odified Antitumor Anthracene-9,10-diones: Synthesis and Antitumor Evaluation. J . Med. Chem. 1994, 37, 828837.
(19) Grandi, M.; Geroni, C.; Giuliani, F. C. I solation and Characterization of a Human Colon Adenocarcinoma Cell Line Resistant to Doxorubicin. Br. J. Cancer 1986, 54, 515-518.
J M 030820X


[^0]:    * Address correspondence to this author. Telephone + 390737402235; fax + 390737637345; e-mail: ippolito.antonini@unicam.it.
    † University of Camerino.
    § University of Padova.
    \# Novuspharma SpA.
    $\perp$ Present address: Department of Chemistry, 75/B1, Discovery Research Oncology, Pharmacia Corporation, Viale Pasteur 10, 20014 Nerviano (MI).

