### SHORT COMMUNICATIONS

# Spiro Recyclization of Pyrrolo[2,1-c][1,4]benzoxazine-1,2,4-triones by the Action of Acyclic Enamines

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Reactions of substituted pyrrolo[2,1-c][1,4]benzoxazine-1,2,4-triones with acyclic enamines were not described previously. We performed reactions of 3-aroyl-2,4-dihydro-1H-pyrrolo[2,1-c][1,4]benzoxazine-1,2,4-triones Ia and Ib [1] with ethyl 3-arylaminobut-2-enoates IIa and IIb at a ratio of 1:1 in anhydrous benzene by heating the reaction mixture for a short time (4–5 min) at the boiling point and obtained the corresponding ethyl 4-aroyl-7-aryl-3-hydroxy-1-ohydroxyphenyl-8-methyl-2,6-dioxo-1,7-diazaspiro-[4.4]nona-3,8-diene-9-carboxylates IIIa and IIIb in almost quantitative yield. The spectral parameters of spiro compounds IIIa and IIIb were analogous to the corresponding parameters of model hexahydroindole-3-spiro-2'-dihydropyrroles whose structure was proved by X-ray analysis [2].

Presumably, the first reaction step involves addition of the activated  $\beta$ -CH group of the enamino fragment in ester **IIa** or **IIb** to the carbon atom in position 3a of pyrrolobenzoxazinetrione **Ia** or **Ib**, as was reported for reactions of the latter with mononucleophiles. Next follows closure of pyrrole ring via intramolecular attack by the side-chain amino group on the lactone carbonyl carbon atom in the benzoxazine ring, which is accompanied by cleavage of the oxazine ring at the  $C^4$ – $O^5$  bond. It should be emphasized that the revealed reaction is a fairly rare example of regioselective synthesis of difficultly accessible spirobipyrrole system with purposefully variable functional substituents in several positions of both pyrrole rings.

## Ethyl 3-hydroxy-1-o-hydroxyphenyl-4-p-methoxybenzoyl-8-methyl-2,6-dioxo-7-p-tolyl-1,7-diaza-

I, X = MeO (a), Br (b); II, Y = Me (a), Cl (b); III, X = MeO, Y = Me (a); X = Br, Y = Cl (b).

**spiro[4.4]nona-3,8-diene-9-carboxylate (IIIa).** A solution of 1 mmol of compound **Ia** and 1 mmol of enamine **IIa** in 10 ml of anhydrous benzene was heated for 4 min under reflux. The mixture was cooled, and the precipitate was filtered off. Yield 96%, mp 198–199°C (from ethyl acetate). IR spectrum, v, cm<sup>-1</sup>: 3200 br (OH); 1756 (COOEt); 1701, 1698 (C<sup>2</sup>=O, C<sup>6</sup>=O); 1653 (COAr). <sup>1</sup>H NMR spectrum, δ, ppm: 1.20 t (3H, C**H**<sub>3</sub>CH<sub>2</sub>, J = 7.1 Hz), 2.01 s (3H, Me), 2.37 s (3H, C<sub>6</sub>H<sub>4</sub>**Me**-p), 3.86 s (3H, OMe), 4.04 q (2H, CH<sub>3</sub>C**H**<sub>2</sub>, J = 7.1 Hz), 6.80–7.84 m (12H, C<sub>6</sub>H<sub>4</sub>), 9.78 s (1H, OH), 12.13 br.s (1H, 3-OH). Found, %: C 67.63; H 5.00; N 4.95. C<sub>32</sub>H<sub>28</sub>N<sub>2</sub>O<sub>8</sub>. Calculated, %: C 67.60; H 4.96; N 4.93.

Ethyl 7-p-bromophenyl-4-p-chlorobenzoyl-3-hydroxy-1-o-hydroxyphenyl-8-methyl-2,6-dioxo-1,7-diazaspiro[4.4]nona-3,8-diene-9-carboxylate (IIIb) was synthesized in a similar way from compounds **Ib** and **IIb**. Yield 95%, mp 262–264°C (from ethyl acetate). IR spectrum, v, cm<sup>-1</sup>: 3360 br, 3100 br (OH); 1758 (COOEt); 1722 (C<sup>2</sup>=O, C<sup>6</sup>=O); 1654 (COAr). <sup>1</sup>H NMR spectrum, δ, ppm: 1.21 t (3H,

CH<sub>3</sub>CH<sub>2</sub>, J = 7.1 Hz), 1.99 s (3H, Me), 4.04 q (2H, CH<sub>3</sub>CH<sub>2</sub>, J = 7.1 Hz), 6.80–8.00 m (12H, C<sub>6</sub>H<sub>4</sub>), 9.93 s (1H, OH), 12.82 br.s (1H, 3-OH). Found, %: C 56.50; H 3.44; N 4.38. C<sub>30</sub>H<sub>22</sub>BrClN<sub>2</sub>O<sub>7</sub>. Calculated, %: C 56.49; H 3.48; N 4.39.

The IR spectra were recorded on a UR-20 spectrophotometer from samples dispersed in mineral oil. The <sup>1</sup>H NMR spectra were recorded on a Bruker WP-400 instrument from solutions in DMSO-d<sub>6</sub> using tetramethylsilane as internal reference. The purity of the products was checked by TLC on Silufol plates using ethyl acetate as eluent; development with iodine vapor.

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#### REFERENCES

- 1. Maslivets, A.N., Mashevskaya, I.V., Krasnykh, O.P., Shurov, S.N., and Andreichikov, Yu.S., *Zh. Org. Khim.*, 1992, vol. 28, p. 2545.
- Bannikova, Yu.N., Maslivets, A.N., and Aliev, Z.G., *Russ. J. Org. Chem.*, 2004, vol. 40, p. 1791.