

SHORT
COMMUNICATIONS

Two Approaches to One-Pot Assembly of 1-Alkyl-2-alkylsulfanyl-5-methoxy-3-neopentylpyrroles

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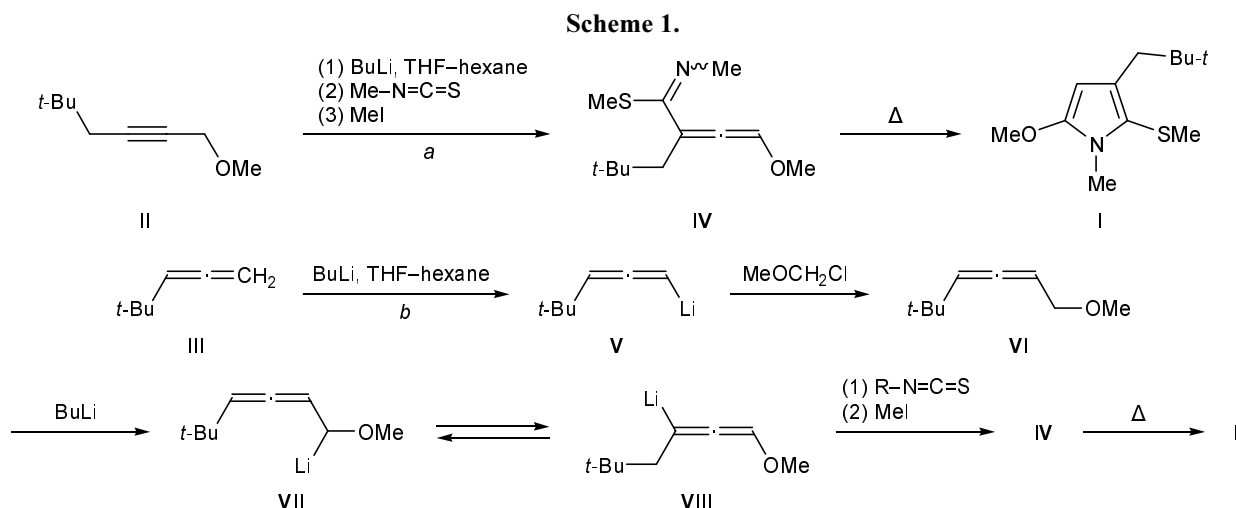
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The importance of pyrrole derivatives for living nature (heme, chlorophyll, bile pigments, pyrrolizidine and indolizidine alkaloids, vitamin B₁₂, etc.) and practice (pharmaceuticals, conducting polymers, molecular optics, electronics, etc.) is difficult to overestimate [1]. Due to broad spectrum of biological activity and diversity of applications, synthesis of pyrrole structures is the subject of extensive studies [2, 3].

We recently showed [3–6] that reactions of heterocumulenes with unsaturated carbanions generated *in situ* from accessible alkynes and 1,2-dienes by the action of superbases (BuLi, BuLi-*t*-BuOK [7]) opens a simple and convenient route to 1-aza-1,3,4-triene systems which are promising precursors of practically important nitrogen- and sulfur-containing heterocycles, including pyrroles. We also found that the cyclization direction strongly depends on the initial reactant structure (together with reaction conditions) [4, 5].

While continuing studies in this line, we have discovered a convenient method for the synthesis of hitherto unknown 1-alkyl-2-alkylsulfanyl-5-alkoxy-3-neopentylpyrroles **I** using accessible isothiocyanates, e.g., methyl isothiocyanate, and 1-methoxy-5,5-dimethylhex-2-yne (**II**) or 4,4-dimethylpenta-1,2-diene (**III**) as key building blocks. 4,4-Dimethylpenta-1,2-diene (**III**) was used by us previously in the newly developed syntheses of alkylsulfanyl- or trialkylsiloxy-substituted 5-*tert*-butylpyrroles, 5-*tert*-butylpyridines, 3-*tert*-butyl-2,3-dihydropyridines, 4-neopentylquinolines, and 5-*tert*-butyl-2-aminothiophenes [3, 4].

Methyl *N*-methyl-4-methoxy-2-neopentylbuta-2,3-dienimidothioate (**IV**) which is necessary as precursor of pyrrole **I** may be obtained in two ways. The first of these (path *a*) includes lithiation of 1-methoxy-5,5-dimethylhex-2-yne (**II**), followed by addition of the carbanion thus formed to methyl isothiocyanate and



S-alkylation of the adduct. According to the second path (*b*), lithiation of 4,4-dimethylpenta-1,2-diene (**III**) gives lithium derivative **V** which is brought into condensation with chloro(methoxy)methane to obtain 1-methoxy-5,5-dimethylhexa-2,3-diene (**VI**); the latter is converted into lithium derivative **VII** or **VIII**, and the subsequent addition to the reaction mixture of methyl isothiocyanate and methyl iodide leads to formation of compound **IV**. 1-Aza-1,3,4-triene **IV** undergoes intramolecular ring closure to pyrrole **I** on heating with 100% selectivity. Both reaction sequences are accomplished in one preparative step via successive addition of the corresponding reactants (Scheme 1).

It should be noted that heating of methoxy-substituted 1-aza-1,3,4-trienes obtained from aliphatic isothiocyanates and α -lithiated methoxyallene is known [3] to give mixtures of 3-methoxypyrroles and 5-methoxy-2,3-dihydropyridines whose ratio depends on the initial isothiocyanate. In order to avoid formation of 2,3-dihydropyridines, the cyclization is usually performed in the presence of CuBr which acts as catalyst specific for cyclization of 1-aza-1,3,4-triene systems to pyrrole derivatives [6].

Obviously, analogous reactions using various isothiocyanates, α -chloro ethers, and alkylating agents should give rise to a new family of difficultly accessible 5-alkoxy-2-alkylsulfanyl-3-neopentylpyrroles as potential biologically active substances.

3-(2,2-Dimethylpropyl)-5-methoxy-1-methyl-2-methylsulfanylpyrrole (I). *a.* A solution of 59 mmol of butyllithium in 37 ml of hexane and 50 ml of THF was cooled to -75°C , and 8 g (57 mmol) of 1-methoxy-5,5-dimethylhex-2-yne (**II**) was added. The mixture was stirred for 15 min at -60 to -50°C , cooled to -100°C , and a solution of 3.9 g (53 mmol) of methyl isothiocyanate in 10 ml of diethyl ether was quickly added. The mixture was stirred for ~ 10 min at -60 to -40°C , 12 g (83 mmol) of methyl iodide was added, and the mixture was allowed to warm up to 20°C , treated with water, and extracted with diethyl ether (2×50 ml). The organic phases were combined, dried over K_2CO_3 , and evaporated on a rotary evaporator, and the residue was distilled under reduced pressure. Yield 8 g (70.5%), purity $\sim 99\%$ (GLC), bp $\sim 110^{\circ}\text{C}$ (~ 0.8 mm). ^1H NMR spectrum, δ , ppm: 5.18 s (1H, 4-H), 3.80 s (3H, OMe), 3.47 s (3H, NMe), 2.48 s (2H, CH_2), 2.10 s (3H, SMe), 0.95 s (9H, *t*-Bu). ^{13}C NMR spectrum, δ_{C} , ppm: 146.00 (C^2), 125.88 (C^5), 112.36 (C^3), 85.78 (C^4), 56.97 (OMe), 40.85 (NMe), 31.61 (CMe_3), 29.62 (CMe_3), 28.32 (CH_2), 20.73 (SMe).

Found, %: C 63.61; H 9.43; N 5.87; S 14.15. $\text{C}_{12}\text{H}_{21}\text{NOS}$. Calculated, %: C 63.39; H 9.31; N 6.16; S 14.10.

b. A solution of 112 mmol of butyllithium in 70 ml of hexane was added at 20°C to a solution of 9.6 g (100 mmol) of 4,4-dimethylpenta-1,2-diene (**III**) in 50 ml of diethyl ether. The mixture was stirred for 30 min at $\sim 40^{\circ}\text{C}$, 50 ml of diethyl ether was added, the mixture was cooled to -20°C , and 10 g (124.2 mmol) of chloro(methoxy)methane was added. When the exothermic reaction was complete, the solvent was removed on a rotary evaporator. Tetrahydrofuran, 70 ml, was added to the residue (a suspension), the mixture was cooled to -100°C , and 109 mmol of butyllithium in 68 ml of hexane was added. The mixture warmed up to -35°C . It was cooled again to -100°C , and a solution of 8 g (109 mmol) of methyl isothiocyanate in 20 ml of THF was quickly added. The mixture was allowed to warm up to -25°C , 20 g (138 mmol) of methyl iodide was added, and the mixture was warmed to 20°C and treated with water. The organic phase was separated and dried over K_2CO_3 , the solvent was removed on a rotary evaporator, and the residue was distilled under reduced pressure. Yield 18.8 g (82.8%), purity $\sim 90\%$ (GLC), bp $\sim 110^{\circ}\text{C}$ (~ 0.8 mm), $n_{\text{D}}^{20} = 1.5185$. ^1H NMR spectrum (90 MHz, CCl_4), δ , ppm: 5.04 s (1H, 4-H), 3.75 s (3H, OMe), 3.45 s (3H, NMe), 2.40 s (2H, CH_2), 2.08 s (3H, SMe), 0.95 s (9H, *t*-Bu).

The ^1H and ^{13}C NMR spectra were recorded on Bruker DPX-400 (400.13 and 100.61 MHz, respectively; ~ 5 – 10% solutions in CDCl_3 ; HMDS) and Varian EM-390 spectrometers (90 MHz; $\sim 20\%$ solutions in CCl_4 ; TMS). 1-Methoxy-5,5-dimethylhex-2-yne (**II**) and 4,4-dimethylpenta-1,2-diene (**III**) were synthesized by the procedures described in [8]. The reactions were carried out under nitrogen using liquid nitrogen as cooling agent. Tetrahydrofuran was purified by treatment with mechanically dispersed KOH (~ 50 g/l), followed by distillation over LiAlH_4 in the presence of benzophenone under nitrogen. Butyllithium (a ~ 1.6 M solution in hexane), methyl isothiocyanate, and the other reagents and solvents were commercial products.

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REFERENCES

1. Jones, R.A. and Bean, G.P., *The Chemistry of Pyrroles*, London: Academic, 1977; *The Chemistry of Heterocyclic*

- Compounds. Pyrroles. Part Two. The Synthesis, Reactivity, and Physical Properties of Substituted Pyrroles*, Jones, R.A., Ed., New York: Wiley, 1992.
2. Ferreira, V.F., de Souza, M.C.B.V., Cunha, A.C., Pereira, L.O.R., and Ferreira, M.L.G., *Org. Prep. Proced. Int.*, 2001, vol. 33, p. 411.
 3. Nedolya, N.A., *Ph.D. Thesis of Utrecht University*, Utrecht (The Netherlands), 1999; Brandsma, L., *Eur. J. Org. Chem.*, 2001, p. 4569; Brandsma, L. and Nedolya, N.A., *Synthesis*, 2004, p. 735.
 4. Nedolya, N.A., Brandsma, L., and Trofimov, B.A., *Mendeleev Commun.*, 1997, p. 92; Nedolya, N.A., Brandsma, L., de Lang, R.-J., and Trofimov, B.A., *Russ. J. Org. Chem.*, 1997, vol. 33, p. 580; Nedolya, N.A., Brandsma, L., de Lang, R.-J., and Trofimov, B.A., *Russ. J. Org. Chem.*, 1997, vol. 33, p. 1361; Tarasova, O.A., Klyba, L.V., Vvedensky, V.Yu., Nedolya, N.A., Trofimov, B.A., Brandsma, L., and Verkruijsse, H.D., *Eur. J. Org. Chem.*, 1998, p. 253; Nedolya, N.A., Brandsma, L., Zinov'eva, V.P., and Trofimov, B.A., *Russ. J. Org. Chem.*, 1998, vol. 34, p. 1494; Nedolya, N.A. and Brandsma, L., *Russ. J. Org. Chem.*, 2003, vol. 39, p. 609.
 5. Brandsma, L., Nedolya, N.A., Verkruijsse, H.D., Owen, N.L., Li Du, and Trofimov, B.A., *Tetrahedron Lett.*, 1997, vol. 38, p. 6905; Brandsma, L., Nedolya, N.A., and Trofimov, B.A., *Eur. J. Org. Chem.*, 1999, p. 2663; Brandsma, L., Nedolya, N.A., and Tolmachev, S.V., *Khim. Geterotsikl. Soedin.*, 2002, p. 60; Nedolya, N.A., Brandsma, L., and Trofimov, B.A., *Khim. Geterotsikl. Soedin.*, 2002, p. 1396; Nedolya, N.A., Schlyakhtina, N.I., Klyba, L.V., Ushakov, I.A., Fedorov, S.V., and Brandsma, L., *Tetrahedron Lett.*, 2002, vol. 43, p. 9679; Tarasova, O.A., Nedolya, N.A., Brandsma, L., and Albanov, A.I., *Tetrahedron Lett.*, 2004, vol. 45, p. 5881.
 6. Nedolya, N.A., Brandsma, L., Tarasova, O.A., Verkruijsse, H.D., and Trofimov, B.A., *Tetrahedron Lett.*, 1998, vol. 39, p. 2409.
 7. Brandsma, L. and Verkruijsse, H., *Preparative Polar Organometallic Chemistry*, Berlin: Springer, 1987, vol. 1.
 8. Brandsma, L., *Preparative Acetylenic Chemistry*, Amsterdam: Elsevier, 1988, p. 68; Brandsma, L. *Best Synthetic Methods. Synthesis of Acetylenes, Allenes and Cumulenes. Methods and Techniques*, Amsterdam: Elsevier, 2004, p. 244.