

Preliminary communication

Cyclooct-1-ene-5-yne tetraruthenium undecacarbonyl

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SUMMARY

$\text{Ru}_4(\text{CO})_{11}\text{C}_8\text{H}_{10}$ has the four metal atoms in a "butterfly" conformation; the carbonyl groups are all terminally bonded, the cyclooct-1-ene-5-yne ligand being coordinated to the four ruthenium atoms via " σ " " π " and " μ " bonds.

The reactions of cycloocta-1,5-diene with the tetrameric cluster, $\alpha\text{-H}_4\text{Ru}_4(\text{CO})_{12}$, are summarised in the accompanying report¹. Two of the products, $\text{Ru}_4(\text{CO})_{11}\text{C}_8\text{H}_{10}$ and $\text{H}_2\text{Ru}_3(\text{CO})_9\text{C}_8\text{H}_{12}$, have been examined by X-ray diffraction methods; the analysis of the tetranuclear complex is essentially complete and is described here.

The crystals (cyclohexane) are monoclinic, space group $P2_1/n$, with $a = 10.499(2)$, $b = 14.764(3)$, $c = 14.724(3)$ Å, $\beta = 90.88^\circ$; $Z = 4$, $D_c = D_m = 2.43 \text{ g}\cdot\text{cm}^{-3}$. 2157 reflexions with $F_{\text{obs}}^2 > 3.0 \sigma(F^2)$ have been observed by four-circle diffractometry (Mo-K α , graphite monochromator) and heavy atom Fourier methods have provided the structural arrangement; least squares analysis has converged R to 0.059 for the data which are presently uncorrected for (small) absorption effects. The stereochemistry is shown in Fig.1; e.s.d.'s in the bond lengths average 0.002 Å (Ru-Ru), 0.015 Å (Ru-C) and 0.025 Å (C-C and C-O).

The geometry is very similar to that of $\text{Co}_4(\text{CO})_{10}(\text{C}_2\text{H}_5\text{C}\equiv\text{CC}_2\text{H}_5)^2$ in that the four metal atoms are in a butterfly conformation; each acetylenic carbon atom C_1 and C_2 forms one metal-carbon " σ " bond of average length 2.16 Å (to Ru_1 and Ru_2) and two " μ " bonds of length 2.21 Å (to Ru_3 and Ru_4)^{*}. Only the "hinge" bond ($\text{Ru}_1\text{-Ru}_2$)

^{*} These symmetry designations of the bonds are intended only to be descriptive and consistent with earlier suggestions²; the symmetry of the individual bonds is low.

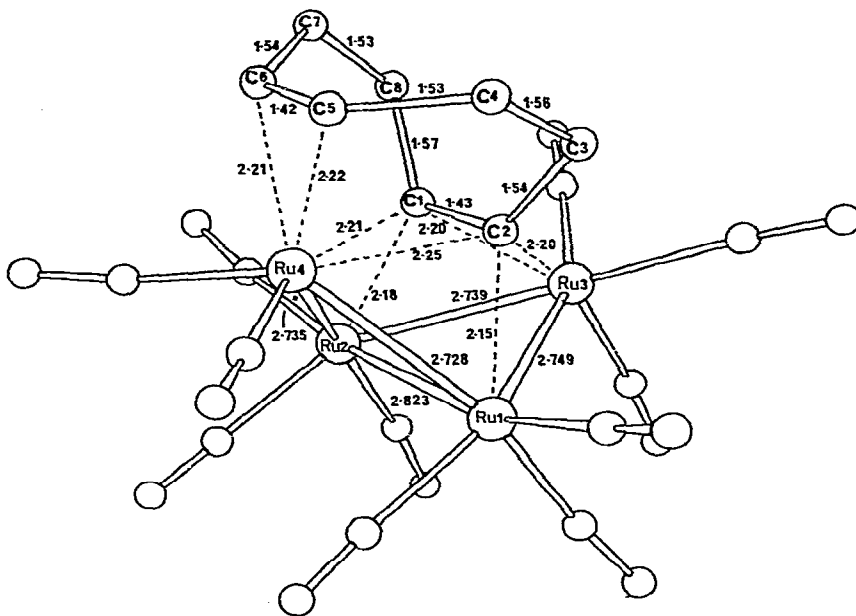


Fig.1. Stereochemistry of $\text{Ru}_4(\text{CO})_{11}\text{C}_8\text{H}_{10}$.

has a bond length which is significantly different from the average length (2.47 Å) of the remaining metal-metal bonds; the dihedral angle between the two planes containing $\text{Ru}_1-\text{Ru}_2-\text{Ru}_4$ and $\text{Ru}_1-\text{Ru}_2-\text{Ru}_3$ respectively is 112° .

The conformation of the macrocyclic ligand is obviously dictated by the bonding and steric requirements of the cluster. The equality, within experimental error, of the bonds, C_1-C_2 and C_5-C_6 , is evidence for considerable charge transfer between the cluster and the acetylenic bond. In the present complex the "μ" bonds are only 0.05 Å longer than the metal-carbon "σ" bonds, the corresponding difference being 0.09 Å in $\text{Co}_4(\text{CO})_{10}(\text{C}_2\text{H}_5\text{C}\equiv\text{CC}_2\text{H}_5)$ and up to 0.25 Å in several benzyne complexes of osmium clusters³; the variations can be related, *inter alia*, to the nature of the other ligands coordinated to the metals, and to the need to preserve a high degree of cyclic delocalisation in coordinated benzyne.

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