

Preliminary communication

Synthesis and X-ray structure of bis(π -cyclopentadienyl)pentacarbonyl-rhenium

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We wish to report the synthesis and structure of the title compound, $(\eta^5\text{-C}_5\text{H}_5)_2\text{-Re}_2(\text{CO})_5$ (I), which is closely related to, but much more stable than, diosmium enneacarbonyl¹.

Ultraviolet irradiation of a cyclohexane solution (0.80 g in 100 ml, nitrogen atmosphere) of $\pi\text{-C}_5\text{H}_5\text{Re}(\text{CO})_3$ for 2.5 h resulted in a yellow solution. Solvent was removed at reduced pressure and the solid sublimed to yield 0.60 g of starting material. Recrystallization of the residue from dichloromethane afforded yellow needles of I (0.08 g, m.p. 138–140° with decomposition). The mass spectrum of the compound exhibits the molecular ion, with the most intense peak of this multiplet at m/e 642. A satisfactory elemental analysis has also been obtained. The NMR spectrum showed a single sharp peak at τ 4.65, and carbonyl stretching bands were observed in the infrared spectrum at 1992 (0.7), 1956 (10.0), 1923 (9.0), 1904 (1.2) and 1740 (4.5) cm^{-1} (cyclohexane, relative intensities on a transmittance scale in parentheses).

Compound I crystallizes in the monoclinic system, space group $P2_1/n$, with a , 9.529 (1); b , 13.934 (2); c , 11.521 (2); β , 97.290 (4). Intensity data were collected on a Picker manual four-circle diffractometer with crystal monochromatized Mo-K α radiation for $2\theta \leq 45^\circ$. Of the 2034 reflections measured in the asymmetric unit, 1390 with $F \geq 5\sigma(F)$ were treated for Lorenz, polarization, and absorption effects and employed in the subsequent calculations.

The structure was solved by Patterson and Fourier techniques and refined by full-matrix least squares with rhenium atoms anisotropic, cyclopentadienes as hindered rotors², all other atoms isotropic and anomalous dispersion included, to a conventional R factor of 4.8% with 77 parameters. A detailed report of the structure will be made later. Figure 1 shows the determined geometry of $(\text{C}_5\text{H}_5)_2\text{Re}_2(\text{CO})_5$.

The principal stereochemical feature of I is the ketonic carbonyl group symmetrically bridging the 2.957 (1) Å Re–Re bond. This is the largest carbonyl-bridged metal-metal separation yet observed, and is distinctly greater than carbonyl-bridged metal-metal distances in the related compounds $(\text{C}_5\text{H}_5)_2\text{Rh}_2(\text{CO})_3$ (Rh–Rh, 2.681 Å)³, $(\text{C}_5\text{H}_5)_2\text{Ru}_2(\text{CO})_4$ (Ru–Ru, 2.735 Å)⁴, and $\text{Pt}_4(\text{PPh}_3)_4(\text{CO})_5$ (Pt–Pt, 2.75–2.79 Å)⁵. It is also distinctly longer than the bridged metal-metal distances in $\text{Ir}_4(\text{CO})_{10}(\text{PPh}_3)_2$ and $\text{Ir}_4(\text{CO})_9(\text{PPh}_3)_3$ (2.73 Å)⁶. The *trans* arrangement of the cyclopentadienyl groups in I and resulting C_2 molecular symmetry are consistent with the solution infrared spectrum.

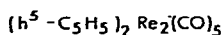
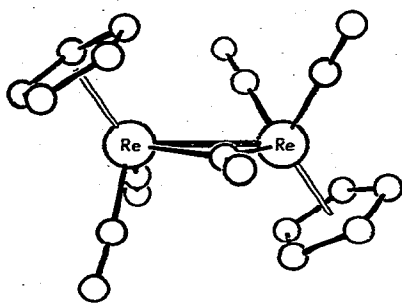
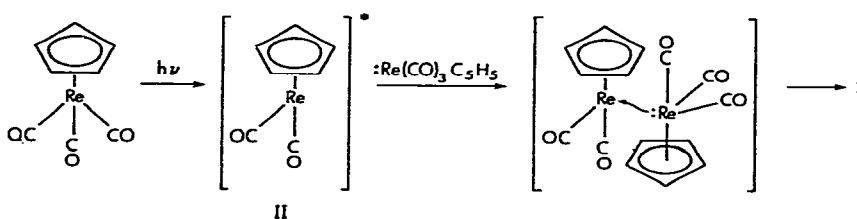


Fig. 1. The molecular structure of bis(η -cyclopentadienyl)pentacarbonyl dirhenium. Selected bond lengths (\AA) are: Re-Re, 2.957 (1); Re-CO (bridging), 2.06 (2); Re-CO (terminal), 1.85 (2); Re-C (C_5H_5), 2.24-2.34 (1). Selected bond angles are Re-CO-Re, $91.7 (8)^\circ$; terminal CO-Re-terminal CO, $85 (1)^\circ$.

We postulate that I is formed in the following sequence of steps:



The basicity of the metal atom in $\text{C}_5\text{H}_5\text{Re}(\text{CO})_3$ suggests that it may coordinate to the unsaturated intermediate II, followed by rearrangement of a terminal carbonyl to bridging to give I. This viewpoint is similar to the donor-acceptor interaction demonstrated by Deubzer and Kaesz⁷, although subsequent rearrangement did not occur in the case they examined. We have found that I reacts with triphenylphosphine to give equimolar amounts of $\pi\text{-C}_5\text{H}_5\text{Re}(\text{CO})_3$ and $\pi\text{-C}_5\text{H}_5\text{Re}(\text{CO})_2\text{P}(\text{C}_6\text{H}_5)_3$, a process which can be regarded as a displacement of the donor $\pi\text{-C}_5\text{H}_5\text{Re}(\text{CO})_3$ by the stronger donor $\text{P}(\text{C}_6\text{H}_5)_3$.

An exactly similar mechanism would account for the formation of $\text{Os}_2(\text{CO})_9$ upon irradiation of $\text{Os}(\text{CO})_5$ at low temperature. The $\text{C}_5\text{H}_5\text{Re}(\text{CO})_2$ and $\text{Os}(\text{CO})_4$ moieties are equivalent from the effective atomic number viewpoint, and the structure now established for I must be considered to lend support to the single-bridged structure inferred from infrared evidence for $\text{Os}_2(\text{CO})_9$ ¹

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