Experimental

Attempted preparation of di-n-butylvinyltin hydroxide. A caustic solution of sodium hydroxide (9.0 g, 0.22 mole) in 200 ml of water was added, dropwise, to a clear solution of vinyldibutyltin chloride (59.08 g, 0.20 mole) in 200 ml of ether. A reaction temperature of 30° C was maintained. Fine white solids formed during the caustic addition. The reaction was then stirred for 1 h at room temperature. The phases were separated and the white solids were filtered from each and air dried. The aqueous phase was discarded and the organic phase was dried with anhydrous sodium sulfate. The ether was stripped to a pot temperature of 60° C at 15 mm pressure. The residue was filtered to remove white solids that had formed during stripping. The now clear liquid formed solids upon storage and was filtered clear at various intervals, until no more solids formed. This clear organic liquid weighed 23 g (79%, 0.08 moles). Its IR spectrum was identical with that of di-*n*-butyldivinyltin. (Found: Sn, 41.09. $C_{12}H_{24}Sn$ calcd.: Sn, 41.36%.)

The combined white solids, after washing with acetone and drying weighed 10 g (40%, 0.04 moles). Its IR spectrum was identical with that of di-*n*-butyltin oxide. (Found: Sn, 47.44. $C_8H_{18}OSn$ calcd.: Sn, 47.69%.)

Research Laboratory, M&T Chemicals Inc., Rahway, N.J. (U.S.A.)

WM. J. CONSIDINE
J. J. VENTURA
B. G. KUSHLEFSKY
A. ROSS

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ERRATUM

J. Organometal. Chem., 1(1963) 97, in Table 2, third column, the last figure (35.12) should read (35.81).