

## ORGANOMETALLIC COMPOUNDS

### XXXVII\*. MASS SPECTRA OF MIXED TETRAALKYLTINS AND OF TRIMETHYLTIN HALIDES

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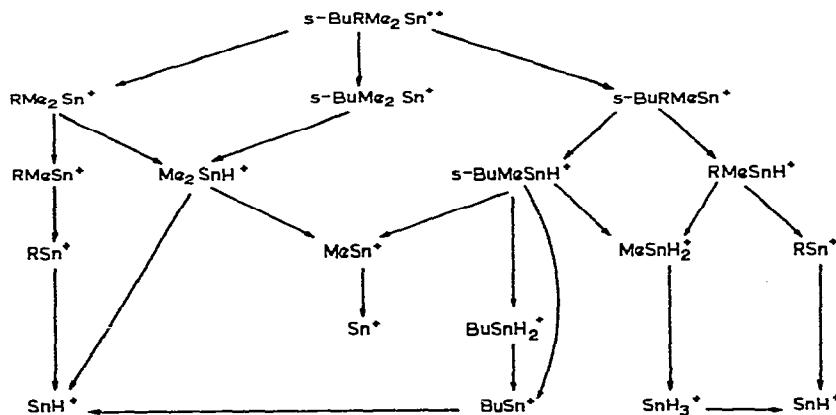
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#### SUMMARY

The mass spectra obtained for a series of compounds of alkyldimethyl-sec-butyltin, alkyldimethyl( $\alpha$ -methylbenzyl)tin, alkyltriisobutyltin and trimethyltin halides confirm the fragmentation rules described earlier.

Diastereotopic nonequivalence has been described for molecules of the type  $R'Me_2SnCHY(CH_3)$  with  $Y = C_2H_5, C_6H_5$  and with various  $R'$  groups<sup>2</sup>. The purpose of this paper is to describe the mass spectra of these compounds, and to compare the obtained results with the conclusions of previous studies<sup>3</sup>.

Table 1 gives the observed fragmentations for a series of alkyl- and phenyldimethyl-sec-butyltins. Table 2 presents the mass spectra of various dibutyldimethyltins. The results may be summarized in the following possible fragmentation pattern:



(continued on p. 284)

\* For part XXXVI see ref. 1.

TABLE I  
70 EV MASS SPECTRA OF A SERIES OF  $\text{RM}_{2}\text{Ts-BuSn}$  COMPOUNDS

(a), (b), (c) : In a given column, (a) is indistinguishable from (a), (b) from (b), (c) from (c).

There are also tin-containing fragment-ions:

- (d) at  $m/e$  165 (intensity 1.0) (g) at  $m/e$  185 (intensity 2.1);  $m/e$  145 ( $\text{SnC}_2\text{H}^+$ ) (intensity 2)
- (e) at  $m/e$  165 (intensity 1.6) (h) at  $m/e$  165 ( $\text{MeSn}^+$ ) (2.3);  $m/e$  197 ( $\text{PhSn}^+$ ) (0.3);  $m/e$  226 [ $\text{Me}(\text{Ph})\text{CH}_2\text{Sn}^+$ ] (0.2)
- (f) at  $m/e$  189 (intensity 9.4) (i) at  $m/e$  165 (2.7);  $m/e$  197 (5.8);  $m/e$  213 ( $\text{PhMe}_2\text{Sn}^+$ ) (3.1);  $m/e$  227 ( $\text{PhMe}_2\text{Sn}^+$ ) (23.2);  $m/e$  303 (0.2)

Fragment	$R = \text{Et}$	$Pr$	$i\text{-Pr}$	$2\text{-Pentyl}$	$3\text{-Pentyl}$	$\text{tert-Pentyl}$	$\text{Cyclohexyl}$	$P_{\text{Ph}}$	$\text{Benzyl}$	$\text{Naphthyl}$	(f)
$\text{Sn}^+$	3.4	4.6	3.4	3.1	4.1	3.3	3.8	3.9	11.6	5.8	2.1
$\text{SnH}^+$	5.1	7.7	7.7	9.9	6.2	4.8	4.5	6.6	2.3	3.3	2.8
$\text{SnH}_3^+$	0.4	0.2	0.3	0.3	0.2	0.3	0.3	0.3			
$\text{MeSn}^+$	11.7	19.8	16.7	1.1	3.3	14.8	12.7	16.4	10.7	0.4	9.2
$\text{MeSnH}_2^+$	3.7	4.9	5.0	18.4	18.1	2.2	3.6	3.4	17.1	0.1	0.1
$\text{Me}_2\text{Sn}^+$	1.2	0.4	0.7	0.8	1.2	0.9	2.2	1.0	1.5	1.7	0.8
$\text{Me}_2\text{SnH}^+$	32.2(a)	40.2	38.2	45.4	43.9	43.2	40.5	40.2	2.5	33.0	6.7
$\text{BuSn}^+$	2.5	0.7	0.5	0.5	0.5	0.4	0.3	0.4	0.4	0.4	0.2
$\text{BuSnH}_2^+$	21.3(b)	2.5(a)	2.5(a)	18.5(b)	2.2(b)	2.1(a)	0.1	0.8	0.8	0.3	0.2
$\text{MeBuSnH}^+$	14.0(b)	1.1	1.1	7.5(b)	7.6(b)	1.9(a)	19.9	8.1	4.4	15.6	1.6
$\text{Me}_2\text{BuSn}^+$	6.7(c)	2.7	3.1	7.5(b)	7.6(b)	1.9(a)	1.9(a)	0.3	0.4	17.6	4.8
$\text{RSn}^+$	1.5	1.2	1.1					1.9	0.4		1.0
$\text{RSnH}_2^+$	32.2(a)	0.3	0.3	2.2(a)	2.1(a)	1.9(a)	0.3	1.2			
$\text{MeRSnH}^+$	7.1	2.5(a)	2.5(a)	7.5(b)	7.6(b)	1.5	9.3	7.3	0.8	1.2	
$\text{Me}_2\text{RSn}^+$	21.3(b)	14.0(b)	18.5(b)	10.4	9.5	4.7	14.5	0.04	1.0	39.7	35.6
$\text{BuRSnH}^+$	6.7(c)	1.1(c)	1.2(c)	1.0	0.9	1.5	0.9	0.4	1.0	0.1	0.1
$\text{MeBRSn}^+$	1.7	0.3	0.5	0.2	0.1	0.7	0.7	0.8	0.1	1.1	0.1
$\text{Me}_2\text{BURSn}^+$	0.9										

TABLE 2

70 eV MASS SPECTRA OF A SERIES OF RMe<sub>2</sub>-s-BuSn COMPOUNDS

<i>Fragment</i>	<i>R=n-Bu</i>	<i>R=i-Bu</i>	<i>R=t-Bu</i>
Sn <sup>+</sup>	4.6	2.0	4.8
SnH <sup>+</sup>	7.0	14.7	3.2
SnH <sub>3</sub> <sup>+</sup>	0.3	0.2	—
MeSn <sup>+</sup>	15.6	18	15.3
MeSnH <sub>2</sub> <sup>+</sup>	5.1	4.2	2.6
Me <sub>2</sub> Sn <sup>+</sup>	0.4	30.8	1.8
Me <sub>2</sub> SnH <sup>+</sup>	40.9	40.3	46.7
BuSn <sup>+</sup>	1.1	1.0	0.6
MeBuSnH <sup>+</sup>	3.9	3.6	2.9
Me <sub>2</sub> BuSn <sup>+</sup>	20	14.1	20.8
Bu <sub>2</sub> MeSn <sup>+</sup>	1.0	0.7	1.3
Bu <sub>2</sub> Me <sub>2</sub> Sn <sup>+</sup>	0.1	0.3	0.3

TABLE 3

## 70 eV MASS SPECTRA OF A SERIES OF RMe<sub>2</sub>[Ph(CH<sub>3</sub>)CH]Sn COMPOUNDS

There are also tin-containing fragment-ions:

(g) at  $m/e$  213 (intensity 0.9) and  $m/e$  185 (intensity 1.9) (i) at  $m/e$  185 (intensity 3.0)

(h) at  $m/e$  277 (intensity 0.2)

(i) at  $m/e$  185 (intensity 3.0)

(j) at  $m/e$  260 (intensity 0.2) and  $m/e$  185 (intensity 1.1)

<i>Fragment<sup>a</sup></i>	<i>R = Et</i>	<i>Pr</i> ( <i>g</i> )	<i>i-Pr</i> ( <i>h</i> )	<i>i-Bu</i>	<i>s-Bu</i> ( <i>i</i> )	<i>t-Bu</i>	<i>i-Pentyl</i>	<i>Ph</i>	<i>Neophyl</i>
$\text{Sn}^+$	4.9	4.0	4.7	3.8	2.5	1.8	3.0	7.2	
$\text{SnH}^+$	6.6	5.3	6.1	5.8	5.5	3.5	5.3	2.3	
$\text{MeSn}^+$	12.6	12.4	12.9	13.3	12.7	9.9	14.2	6.1	11.7
$\text{MeSnH}_2^+$	0.7	1.7	0.6	0.7	1.0	0.3	0.8		
$\text{Me}_2\text{Sn}^+$								3.4	1.2
$\text{Me}_2\text{SnH}^+$	29.7	44	33.6	51.3	44	26.9	52		1.7
$\text{Me}_3\text{Sn}^+$					1.0	1.3	4.6	0.5	4.6
$\text{PhSn}^+$	1.9				1.4	1.5	1.7	1.3	14.8(a) 8.1
$\text{PhenSn}^+$	1.1	1.0	1.6	1.0	1.7	2.1	0.2		30.7
$\text{PhenSnH}_2^+$		0.2			0.7	0.2			
$\text{MePhenSn}^+$			0.2		0.3	0.3		1.3	
$\text{MePhenSnH}^+$	0.1	0.1	0.2	0.4	0.2	0.3	0.3	3.2	
$\text{Me}_2\text{PhenSn}^+$	1.7(a)	1.2	4.3		2.0	19.4	1.6	5.0	0.8
$\text{RSn}^+$	0.9	1.0	0.9	0.4	0.3		0.4	14.8(a)	1.7
$\text{RSnH}_2^+$		0.8	0.1						
$\text{RMeSnH}^+$	6.7	1.0	1.3	0.7	1.3	0.7	0.5	1.5	0.8
$\text{RMe}_2\text{Sn}^+$	29.4	21.5	30	18.2	18.6	20.9	18.8	52.8	37.5
$\text{RPhenSn}^+$								0.2	
$\text{RPhenSnH}^+$	1.7(a)								
$\text{RPhenMeSn}^+$	1.5	0.9	1.2	0.5	0.8	2.2	0.7	0.8	
$\text{RPhenMe}_2\text{Sn}^+$	2.2	2.2	2.2	1.2	1.5	3.7	0.7	1.3	

<sup>a</sup> Phen = Ph(CH<sub>3</sub>)CH.

TABLE 4

## 70 eV MASS SPECTRA OF ALKYLTRIISOBUTYLTINS

Fragment	$R = Me$	$R = Pr^a$	$R = i-Pr^b$
$Sn^+$	2.3	3.3	3.5
$SnH^+$	5.9	9.5	9.7
$SnH_3^+$	0.2	2.0	1.7
$MeSn^+$	10.4	0.3	5.7
$MeSnH_2^+$	13.4	0.4	
$EtSn^+$	0.4	0.5	0.7
$EtSnH_2^+$	8.8	1.3	
$PrSn^+$	0.2	1.5	1.6
$PrSnH_2^+$	1.0	6.4	9.8
$BuSn^+$	2.4	3.8	3.7
$BuSnH_2^+$	1.9	18.8	18.6
$BuMeSnH^+$	32	1.2	0.7
$BuEtSnH^+$	4.5	2.3	1.4
$BuPrSnH^+$	4.0	12.0	12.0
$Bu_2SnH^+$	1.4	21.8	17.7
$Bu_2SnMe^+$	14.1	0.4	0.4
$Bu_2SnEt^+$	0.1	0.8	0.6
$Bu_2SnPr^+$		4.5	3.7
$Bu_3Sn^+$	0.4	9.2	8.3
$Bu_3SnMe^{++}$	0.07		0.2
$Bu_3SnPr^{++}$			

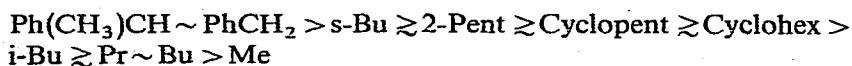
<sup>a</sup> There is also a tin-containing fragment-ion corresponding to  $BuPrSn^+$  (0.2). <sup>b</sup> There is also a tin-containing fragment-ion corresponding to  $BuMeSn^+$  (0.4).

TABLE 5

70 eV MASS SPECTRA OF TRIMETHYLTIN HALIDES  $Me_3SnX$ 

Fragment	$X = F$	$X = Cl$	$X = Br$	$X = I$
$Me_3SnX^{++}$	1.9	0.5	3.5	4.9
$Me_2SnX^+$	41.4	40.7	47.3	19.7
$MeSnX^+$	5.2	6.6	9.4	6.0
$SnX^+$	14.5	12.2	6.0	12.3
$Me_3Sn^+$	17.8	18.6	10.9	31.6
$Me_2SnH^-$	0.06	0.5	0.5	0.2
$Me_2Sn^-$	2.5	5.9	5.1	2.6
$MeSn^-$	9.8	8.7	9.5	14.8
$CH_3Sn^+$	1.9	2.5	1.8	3.6
$SnH^+$	1.3	1.5	1.7	2.1
$Sn^+$	5.7	5.7	5.8	7.7

Mass spectra at lower potentials (see *e.g.* fig. 1 to 6) confirm the sequence for the ease of cleavage of a carbon–tin bond of the molecular ion previously described<sup>3</sup>, *viz.*



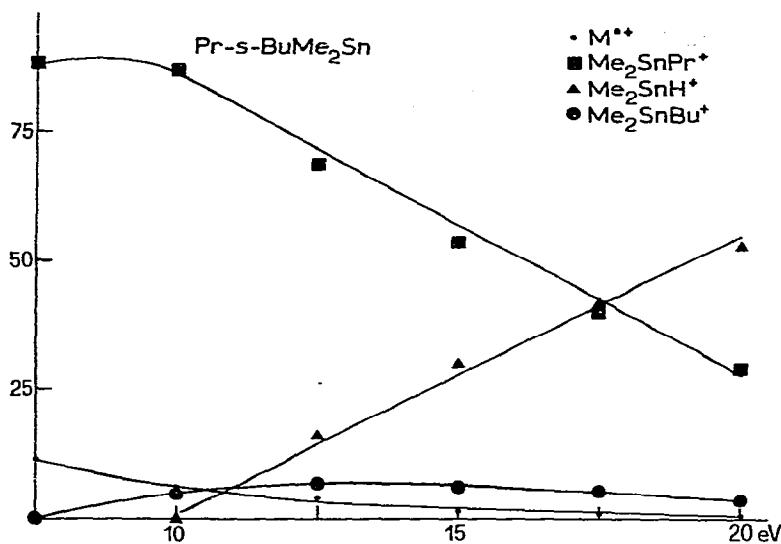


Fig. 1.

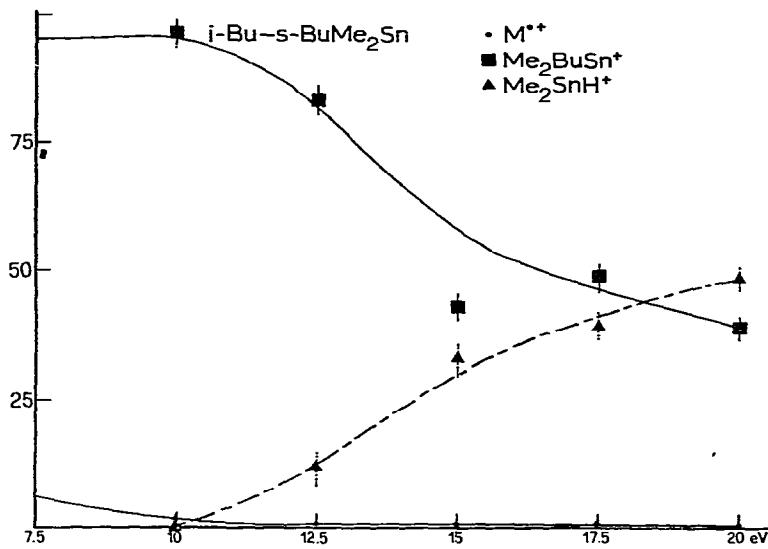


Fig. 2.

Table 3 shows the mono-isotopic spectra of alkyl- and phenyldimethyl-( $\alpha$ -methylbenzyl)tin. The same type of fragmentation pattern describes the fate of the molecular ion. The mass spectra of alkyltriisobutyltins are presented in Table 4, and show rather unexpected fragments corresponding to cleavage of carbon-carbon bonds (up to 19% of the sum of the intensities of all fragments for  $i\text{-Bu}_3SnMe$ ). All the expected fragments are present.

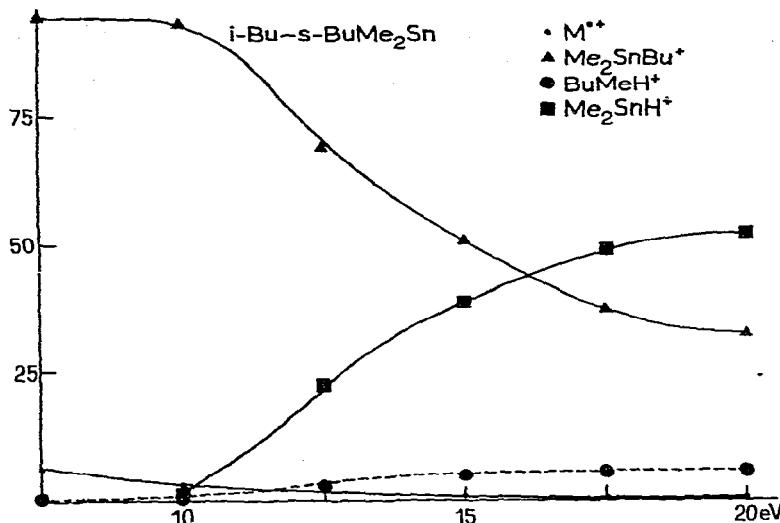


Fig. 3.

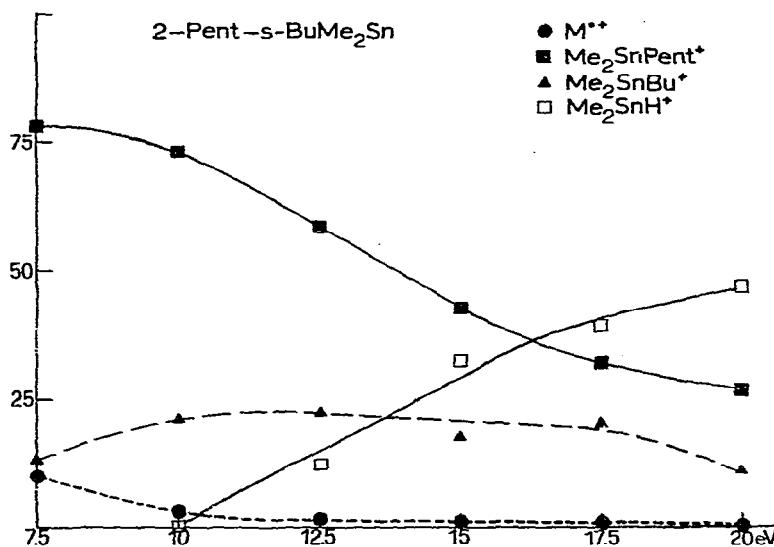


Fig. 4.

The 70 eV mass spectra of trimethyltin halides, which have been synthesized for a  $\gamma$ -spectroscopy investigation<sup>4</sup>, are summarized in Table 5. All the expected<sup>2</sup> fragments are present.

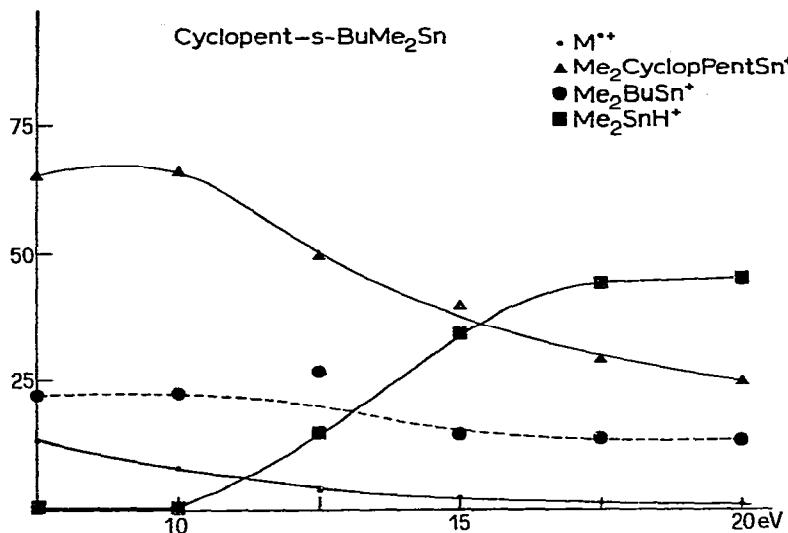


Fig. 5.

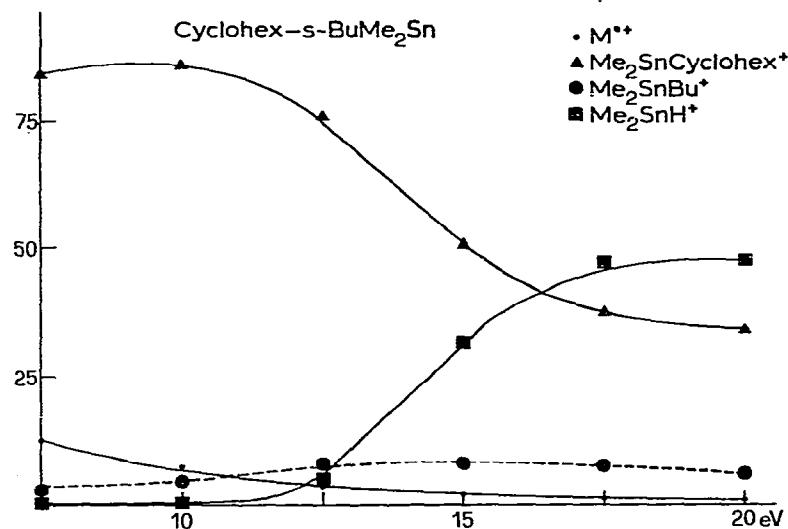


Fig. 6.

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