

## Errata

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*J. Organometal. Chem.*, Vol. 112, No. 2 (June 1st, 1976)

Page 148, line 29 should read:

$d-d(x^2-y^2 \leftarrow xz, yz)$  transition and is sensitive to the bonding characteristics

*J. Organometal. Chem.*, Vol. 120, No. 1 (October 26th, 1976)

Page C17, the footnotes to Table 1 should read:

<sup>a</sup> For the irreversible processes  $(Ep)_{OX}$  is given, measured at a scan rate of 100 mV/s; all potentials are vs. a calomel electrode 1 M in LiCl.

<sup>b</sup> For the known reversible process  $[Ni\{S_2C_2(CN)_2\}_2]^{2-} \rightleftharpoons [Ni\{S_2C_2(CN)_2\}_2]^{1-}$   $(ip)_{OX} = 3.5 \mu A$ ,  $(ip)_{RED} = 3.8 \mu A$ .

<sup>c</sup> R = reversible, IR = chemically irreversible.