

## COMPLEXES OF BIDENTATE GROUP VB CHELATES

### XIX \*. MASS SPECTRAL STUDIES ON SOME METHYL-SUBSTITUTED BIDENTATE GROUP VB DONOR LIGANDS

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#### Summary

The fragmentation patterns of seven bidentate and two tridentate Group VB donor ligands are reported. For these methyl-substituted ligands the initial fragmentation is essentially loss of methyl radicals. Phenyl, but not methyl, migration occurs for the bis(phenylmethylarsino)alkanes, but methyl migration is evident in the all aliphatic bidentate chelates. The tridentate ligands  $\text{PhE}-(\text{CH}_2\text{CH}_2\text{CH}_2\text{AsMe}_2)_2$  ( $\text{E} = \text{P}, \text{As}$ ) show similar fragmentation routes, but are more complex. The long chain diarsine,  $\text{Me}_2\text{As}(\text{CH}_2)_{12}\text{AsMe}_2$ , exhibits a tendency to cyclise and lose  $\text{C}_2\text{H}_4$  fragments progressively.

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#### Introduction

There is increasing interest in the mass spectral fragmentation patterns of bi- and multi-dentate ligands. In addition to their inherent interest, mass spectra of ligands have been recorded in an attempt to elucidate the difficulties in the synthesis of various diarsines [2], to identify the products from the attempted preparations of several ditertiary stibines [3], and to examine the possibility of a correlation between the electron impact fragmentation of thioether ligands and metal catalysed *S*-dealkylation reactions [4]. Previous studies have reported the mass spectra of  $\text{EPh}_3$  [5],  $\text{Ph}_2\text{E}(\text{CH}_2)_n\text{EPh}_2$  [6,7], *cis*- and *trans*- $\text{Ph}_2\text{E}(\text{CH}=\text{CHEPh}_2)$  [6] ( $\text{E} = \text{P}, \text{As}$ ), and the range of *o*-phenylene ligands  $o\text{-C}_6\text{H}_4(\text{E}'\text{Ph}_2)(\text{E}''\text{Ph}_2)$  [8]. Alkyl substituted ligands have been much less studied although the fragmentation of *o*-phenylenebis(dimethylarsine) [2,9], and *cis*-1,2-bis(dimethylarsino)-ethylene [9] have been briefly discussed.

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\* For Part XVIII see ref. 1.

TABLE 1  
FRAGMENTATION PATTERNS OF COMPOUNDS I-IV

Me <sub>2</sub> PCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> PMe <sub>2</sub> (I)			Me <sub>2</sub> AsCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> PMe <sub>2</sub> (II)			Me <sub>2</sub> AsCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> AsMe <sub>2</sub> (III)			Me <sub>2</sub> SbCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> SbMe <sub>2</sub> (IV) <sup>a</sup>		
Mass	Rel. Int.	Fragment	Mass	Rel. Int.	Fragment	Mass	Rel. Int.	Fragment	Mass	Rel. Int.	Fragment
149	100	C <sub>6</sub> H <sub>15</sub> P <sub>2</sub>	193	100	C <sub>6</sub> H <sub>15</sub> AsP	237	100	C <sub>6</sub> H <sub>15</sub> As <sub>2</sub>	329	49.8	C <sub>6</sub> H <sub>15</sub> Sb <sub>2</sub>
194	25.0	C <sub>5</sub> H <sub>12</sub> P <sub>2</sub>	178	59.2	C <sub>5</sub> H <sub>12</sub> AsP	222	24.2	C <sub>5</sub> H <sub>12</sub> As <sub>2</sub>	314	3.4	C <sub>5</sub> H <sub>12</sub> Sb <sub>2</sub>
183	3.0	C <sub>5</sub> H <sub>11</sub> P <sub>2</sub>	151	9.4	C <sub>3</sub> H <sub>9</sub> AsP	195	19.3	C <sub>3</sub> H <sub>9</sub> As <sub>2</sub>	302	2.0	C <sub>4</sub> H <sub>12</sub> Sb <sub>2</sub>
122	6.0	C <sub>4</sub> H <sub>12</sub> P <sub>2</sub>	136	9.5	C <sub>2</sub> H <sub>6</sub> AsP	180	3.3	C <sub>2</sub> H <sub>6</sub> As <sub>2</sub>	287	19.5	C <sub>3</sub> H <sub>9</sub> Sb <sub>2</sub>
119	1.7	C <sub>4</sub> H <sub>9</sub> P <sub>2</sub>	131	10.7	C <sub>4</sub> H <sub>8</sub> As	165	6.0	CH <sub>3</sub> As <sub>2</sub>	272	4.0	C <sub>2</sub> H <sub>6</sub> Sb <sub>2</sub>
107	10.6	C <sub>3</sub> H <sub>9</sub> P <sub>2</sub>	121	3.2	CH <sub>3</sub> AsP	131	14.2	C <sub>4</sub> H <sub>8</sub> As	257	6.3	CH <sub>3</sub> Sb <sub>2</sub>
104	1.4	C <sub>3</sub> H <sub>6</sub> P <sub>2</sub>	119	5.0	C <sub>3</sub> H <sub>8</sub> As	120	2.1	C <sub>3</sub> H <sub>9</sub> As	242	1.9	Sb <sub>2</sub>
103	2.4	C <sub>3</sub> H <sub>12</sub> P	106	13.6	C <sub>2</sub> H <sub>6</sub> As	117	3.4	C <sub>3</sub> H <sub>6</sub> As	117	30.3	C <sub>4</sub> H <sub>8</sub> Sb
89	4.8	C <sub>4</sub> H <sub>10</sub> P	103	19.2	C <sub>5</sub> H <sub>12</sub> P } C <sub>2</sub> H <sub>4</sub> As }	105	24.3	C <sub>2</sub> H <sub>6</sub> As	166	13.6	C <sub>3</sub> H <sub>9</sub> Sb
88	2.4	C <sub>4</sub> H <sub>9</sub> P				103	26.4	C <sub>2</sub> H <sub>4</sub> As	151	100.0	C <sub>2</sub> H <sub>6</sub> Sb
87	7.0	C <sub>4</sub> H <sub>8</sub> P	91	5.7	CH <sub>4</sub> As	91	7.2	CH <sub>4</sub> As	149	16.8	C <sub>2</sub> H <sub>4</sub> Sb
79	2.4	CH <sub>3</sub> P	90	6.5	CH <sub>3</sub> As	90	6.8	CH <sub>3</sub> As	136	40.4	CH <sub>3</sub> Sb
77	3.3	CH <sub>3</sub> P <sub>2</sub>	89	11.8	CH <sub>2</sub> As	89	13.6	CH <sub>2</sub> As	135	17.9	CH <sub>2</sub> Sb
76	23.9	C <sub>3</sub> H <sub>9</sub> P	76	13.2	AsH	75	0.8	As	121	21.4	Sb
						41	8.4	C <sub>3</sub> H <sub>5</sub>	41	16.1	C <sub>3</sub> H <sub>5</sub>
75	17.4	C <sub>3</sub> H <sub>8</sub> P	75	16.9	C <sub>3</sub> H <sub>9</sub> P As	41	8.4	C <sub>3</sub> H <sub>5</sub>			
73	9.9	C <sub>3</sub> H <sub>6</sub> P	61	13.0	C <sub>2</sub> H <sub>6</sub> P						
62	12.8	C <sub>2</sub> H <sub>7</sub> P	59	11.0	C <sub>2</sub> H <sub>4</sub> P						
61	19.6	C <sub>2</sub> H <sub>6</sub> P	46	3.3	CH <sub>3</sub> P						
59	17.3	C <sub>2</sub> H <sub>4</sub> P	46	8.8	CH <sub>2</sub> P						
57	10.8	C <sub>2</sub> H <sub>2</sub> P	41	18.3	C <sub>3</sub> H <sub>5</sub>						
45	15.7	CH <sub>2</sub> P									
41	13.6	C <sub>3</sub> H <sub>5</sub>									

<sup>a</sup> Only <sup>121</sup>Sb fragments listed.

Here we report the mass spectra of a series of predominantly alkyl substituted ligands: 1,3-bis(dimethylphosphino) propane (dmp) (I), 1,3-bis(dimethylarsino)propane (dma) (II), (3-dimethylphosphinopropyl)dimethylarsine (dmap) (III), 1,3-bis(dimethylstibino)propane (dmsb) (IV), 1,2-bis(phenylmethylarsino)ethane (dase) (V), 1,3-bis(phenylmethylarsino)propane (dasp) (VI), bis(3-dimethylarsinopropyl)phenylarsine (tasp) (VII), bis(3-dimethylarsinopropyl)phenylphosphine (tap) (VIII), and 1,12-bis(dimethylarsino)dodecane (dmd) (IX) \*.

## Experimental

The 70 eV mass spectra were obtained using an A.E.I. MS 902 mass spectrometer linked to a DS 30 data system. Spectra were recorded at a resolving power of 2000, and atomic compositions confirmed by accurate mass measurements (RP = 10000).

The ligands were obtained by literature methods: dmp [10], dma [11], dmap [10], dmsb [12], dase [13], dasp [14], tasp [15], tap [16], and dmd by a method [17] similar to that used to prepare dma.

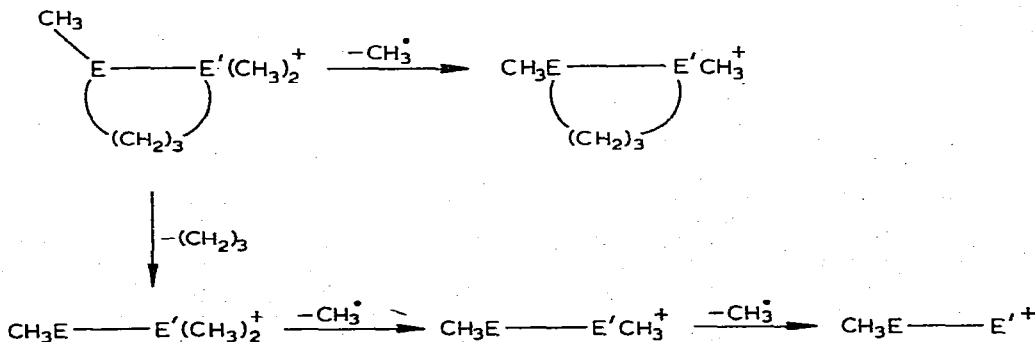
## Results and discussion

The prominent ions in the mass spectra of these ligands are listed in Tables 1 (I-IV), 2 (V and VI), 3 (VII and VIII) and 4 (IX).

### Ligands I-IV (Table 1)

These ligands  $\text{Me}_2\text{E}(\text{CH}_2)_3\text{E}'\text{Me}_2$  do not exhibit a parent ion in their spectra, the heaviest fragment observed corresponding to  $\text{P}-\text{CH}_3^+$ . For I-III the ion having  $m/e \text{P}-\text{CH}_3^+$  is also the base peak, but for IV the base peak corresponds to  $m/e 151, \text{Me}_2\text{Sb}^+$ . All four spectra show the propensity of this type of ligand to lose methyl radicals from the parent and subsequent daughter ions. Also all four spectra exhibit ions containing two Group VB atoms,  $\text{P}-\text{CH}_3^+$ ,  $\text{P}-2\text{CH}_3^+$ ,  $\text{C}_3\text{H}_9\text{EE}'^+$ ,  $\text{C}_2\text{H}_6\text{EE}'^+$ ,  $\text{CH}_3\text{EE}'^+$ , etc., suggesting that there is a linking of the pnictogen atoms probably after loss of a methyl radical from the parent as shown in Scheme 1.

SCHEME 1



\*  $\text{Me}_2\text{P}(\text{CH}_2)_3\text{PMe}_2$ , I;  $\text{Me}_2\text{As}(\text{CH}_2)_3\text{AsMe}_2$ , II;  $\text{Me}_2\text{P}(\text{CH}_2)_3\text{AsMe}_2$ , III;  $\text{Me}_2\text{Sb}(\text{CH}_2)_3\text{SbMe}_2$ , IV;  $\text{PhMeAs}(\text{CH}_2)_2\text{AsMePh}$ , V;  $\text{PhMeAs}(\text{CH}_2)_3\text{AsMePh}$ , VI;  $\text{PhAs}\{(\text{CH}_2)_3\text{AsMe}_2\}_2$ , VII;  $\text{PhP}\{(\text{CH}_2)_3\text{AsMe}_2\}_2$ , VIII;  $\text{Me}_2\text{As}(\text{CH}_2)_{12}\text{AsMe}_2$ , IX.

The cleavage of an  $-E(CH_3)_2$  fragment from the backbone is also evident by the occurrence of  $C_4H_8E^+$  ions. The presence of  $C_7H_9E^+$  ions show that the methyl groups can migrate from one pnictogen atom to the other in the compound as previously noted [4,6]. In unsymmetrical ligands the migration only occurs from the heavier pnictogen atom to the lighter since the spectrum of ligand II exhibit a peak at  $m/e$  86,  $(CH_3)_3P^+$  but not at  $m/e$  120,  $(CH_3)_3As^+$ .

### Ligands V and VI (Table 2)

The fragmentation patterns of the two  $\alpha,\omega$ -bis(phenylmethylarsino)alkanes show interesting differences. Ligand V exhibits a parent ion  $m/e$  362 and the base peak at  $m/e$  167,  $C_7H_8As^+(PhMeAs^+)$ . Ligand VI resembles the tetramethyl analogue II in that the heaviest fragment observed is  $m/e$  361 which is also the base peak,  $P-CH_3^+$ . Ligand V unlike ligand VI drops out the backbone to give  $m/e$  334,  $C_{14}H_{16}As_2^+$  which then loses a methyl radical to give  $m/e$  319,  $C_{13}H_{16}As_2^+$ , or undergoes As-As bond fission to give  $m/e$  167,  $C_7H_8As^+$ . Phenyl migration also occurs since a prominent ion occurs at  $m/e$  229,  $C_{12}H_{10}As^+(Ph_2As^+)$ , the subsequent decomposition of which is a characteristic feature of the spectra of arylarsines [5,6]. In the spectra of both V and VI there is no evidence for methyl migration.

A comparison of the fragmentation patterns of V and of  $Ph_2As(CH_2)_2AsPh_2$  [7] does not provide any clue to the reasons for the instability of  $Me_2As(CH_2)_2AsMe_2$  [18], the tendency to eliminate the backbone is greater for both ligands than for the trimethylene linked II and VI but partial replacement of phenyl by methyl groups [ $Ph_2As(CH_2)_2AsPh_2$  vs.  $MePhAs(CH_2)_2AsPhMe$ ] does not cause any profound change in the spectra.

TABLE 2  
FRAGMENTATION PATTERNS OF COMPOUNDS V AND VI

$\begin{array}{c} \text{Ph} \quad \quad \quad \text{Ph} \\ \diagdown \quad \quad \diagup \\ \text{AsCH}_2\text{CH}_2\text{As} \\ \diagup \quad \quad \diagdown \\ \text{Me} \quad \quad \quad \text{Me} \end{array}$			$\begin{array}{c} \text{Ph} \quad \quad \quad \text{Ph} \\ \diagdown \quad \quad \diagup \\ \text{AsCH}_2\text{CH}_2\text{CH}_2\text{As} \\ \diagup \quad \quad \diagdown \\ \text{Me} \quad \quad \quad \text{Me} \end{array}$		
Mass	Rel. Int.	Fragment	Mass	Rel. Int.	Fragment
362	13.0	$C_{16}H_{20}As_2$	361	100	$C_{16}H_{19}As_2$
347	3.4	$C_{15}H_{17}As_2$	319	2.4	$C_{13}H_{13}As_2$
334	15.2	$C_{14}H_{16}As_2$	299	8.0	$C_{11}H_{17}As_2$
319	3.6	$C_{13}H_{13}As_2$	229	6.8	$C_{12}H_{10}As$
244	17.3	$C_{13}H_{13}As$	227	6.2	$C_{12}H_8As$
229	32.6	$C_{12}H_{10}As$	193	6.2	$C_9H_{10}As$
227	13.4	$C_{12}H_8As$	180	5.4	$C_2H_6As_2$
182	4.4	$C_8H_{11}As$	167	36.1	$C_7H_8As$
167	100	$C_7H_8As$	165	10.5	$CH_3As_2, C_7H_6As$
165	10.7	$CH_3As_2, C_7H_6As$	154	3.3	$C_{12}H_{10}$
154	10.6	$C_{12}H_{10}$	152	12.1	$C_{12}H_8, C_6H_5As$
152	22.3	$C_6H_5As, C_{12}H_8$	151	8.5	$C_6H_4As$
151	26.7	$C_6H_4As$	117	6.0	$C_3H_6As$
91	97.2	$C_7H_7, CH_4As$	104	3.8	$C_2H_5As$
89	9.8	$CH_2As$	91	64.9	$C_7H_7, CH_4As$
78	11.7	$C_6H_6$	89	5.6	$CH_2As$
77	15.3	$C_6H_5$	78	10.1	$C_6H_6$
75	2.3	As	77	10.7	$C_6H_5$
			75	0.3	As

TABLE 3

## FRAGMENTATION PATTERNS OF COMPOUNDS VII AND VIII

Me <sub>2</sub> AsCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> As(Ph)CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> AsMe <sub>2</sub> (VII)				Me <sub>2</sub> AsCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> P(Ph)CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> AsMe <sub>2</sub> (VIII)							
Mass	Rel. Int.	Fragment	Mass	Rel. Int.	Fragment	Mass	Rel. Int.	Fragment			
431	26.7	C <sub>15</sub> H <sub>26</sub> As <sub>3</sub>	90	11.0	CH <sub>3</sub> As	387	100	C <sub>13</sub> H <sub>26</sub> As <sub>2</sub> P	140	5.6	C <sub>8</sub> H <sub>13</sub> P
401	1.2	C <sub>13</sub> H <sub>20</sub> As <sub>3</sub>	89	13.9	CH <sub>2</sub> As	357	9.0	C <sub>13</sub> H <sub>20</sub> As <sub>2</sub> P	137	2.7	C <sub>8</sub> H <sub>10</sub> P
362	2.1	C <sub>10</sub> H <sub>17</sub> As <sub>3</sub>	78	21.0	C <sub>6</sub> H <sub>6</sub>	300	6.4		133	6.9	C <sub>4</sub> H <sub>10</sub> As
300	39.6	C <sub>11</sub> H <sub>18</sub> As <sub>2</sub>	77	10.5	C <sub>6</sub> H <sub>5</sub>	283	42.8	C <sub>13</sub> H <sub>21</sub> AsP	131	22.0	C <sub>4</sub> H <sub>8</sub> As
299	100.0	C <sub>11</sub> H <sub>17</sub> As <sub>2</sub>	75	2.6	As	281	23.8	C <sub>13</sub> H <sub>19</sub> AsP	127	2.8	C <sub>4</sub> H <sub>4</sub> As
284	10.2	C <sub>10</sub> H <sub>14</sub> As <sub>2</sub>				279	14.0	C <sub>13</sub> H <sub>17</sub> AsP	125	8.1	C <sub>4</sub> H <sub>2</sub> As
257	8.4	C <sub>8</sub> H <sub>11</sub> As <sub>2</sub>				267	10.0	C <sub>12</sub> H <sub>17</sub> AsP	123	7.7	C <sub>7</sub> H <sub>8</sub> P
227	12.7	C <sub>6</sub> H <sub>5</sub> As <sub>2</sub>				255	18.2	C <sub>11</sub> H <sub>17</sub> AsP	122	6.1	C <sub>7</sub> H <sub>7</sub> P
221	6.2	C <sub>5</sub> H <sub>11</sub> As <sub>2</sub>				248	4.3		121	10.6	CH <sub>3</sub> AsP
207	2.7	C <sub>4</sub> H <sub>9</sub> As <sub>2</sub>				241	3.0	C <sub>10</sub> H <sub>15</sub> AsP	120	6.6	C <sub>3</sub> H <sub>9</sub> As
193	3.3	C <sub>3</sub> H <sub>7</sub> As <sub>2</sub>				240	7.8	C <sub>10</sub> H <sub>14</sub> AsP	119	5.6	C <sub>3</sub> H <sub>8</sub> As
189	3.1	C <sub>3</sub> H <sub>3</sub> As <sub>2</sub>				239	3.8	C <sub>10</sub> H <sub>13</sub> AsP	117	6.0	C <sub>3</sub> H <sub>6</sub> As
180	3.1	C <sub>2</sub> H <sub>6</sub> As <sub>2</sub>				226	7.4	C <sub>9</sub> H <sub>12</sub> AsP	116	13.0	C <sub>3</sub> H <sub>5</sub> As or C <sub>6</sub> H <sub>13</sub> P
179	7.3	C <sub>2</sub> H <sub>5</sub> As <sub>2</sub>				225	7.4	C <sub>9</sub> H <sub>11</sub> AsP	115	5.4	C <sub>6</sub> H <sub>12</sub> P
167	29.3	C <sub>7</sub> H <sub>8</sub> As				210	4.1	C <sub>4</sub> H <sub>10</sub> As <sub>2</sub>	109	22.1	C <sub>6</sub> H <sub>6</sub> P
165	15.2	CH <sub>3</sub> As <sub>2</sub>				207	5.2	C <sub>4</sub> H <sub>9</sub> As <sub>2</sub>	108	6.9	C <sub>6</sub> H <sub>5</sub> P
153	19.7	C <sub>6</sub> H <sub>6</sub> As				199	2.1	C <sub>7</sub> H <sub>9</sub> AsP	107	32.6	C <sub>6</sub> H <sub>4</sub> P
152	12.2	C <sub>6</sub> H <sub>5</sub> As									
151	8.7	C <sub>6</sub> H <sub>4</sub> As } HAs <sub>2</sub>									
150	5.0	As <sub>2</sub>				186	4.8	C <sub>6</sub> H <sub>8</sub> AsP	105	24.3	C <sub>7</sub> H <sub>6</sub> As
147	25.9	C <sub>5</sub> H <sub>12</sub> As				183	8.6	C <sub>6</sub> H <sub>5</sub> AsP	103	20.4	C <sub>2</sub> H <sub>4</sub> As
146	5.4	C <sub>5</sub> H <sub>11</sub> As				181	4.3	C <sub>2</sub> H <sub>7</sub> As <sub>2</sub>	102	5.9	C <sub>2</sub> H <sub>3</sub> As
131	15.0	C <sub>4</sub> H <sub>8</sub> As				179	5.1	C <sub>2</sub> H <sub>5</sub> As <sub>2</sub>	101	10.9	C <sub>2</sub> H <sub>2</sub> As
127	2.0	C <sub>4</sub> H <sub>4</sub> As				178	2.6	C <sub>2</sub> H <sub>4</sub> As <sub>2</sub> } C <sub>5</sub> H <sub>12</sub> AsP }	91	25.0	CH <sub>4</sub> As, C <sub>7</sub> H <sub>7</sub>
125	3.1	C <sub>4</sub> H <sub>2</sub> As				167	4.3	C <sub>7</sub> H <sub>8</sub> As	90	30.1	CH <sub>3</sub> As
117	10.7	C <sub>3</sub> H <sub>6</sub> As				165	3.0	CH <sub>3</sub> As <sub>2</sub>	89	30.0	CH <sub>2</sub> As
105	31.0	C <sub>2</sub> H <sub>6</sub> As				164	10.2	C <sub>4</sub> H <sub>10</sub> AsP	88	9.0	C <sub>4</sub> H <sub>9</sub> P
103	22.8	C <sub>2</sub> H <sub>4</sub> As				163	6.0	C <sub>4</sub> H <sub>9</sub> AsP	78	11.3	C <sub>6</sub> H <sub>6</sub>
91	36.4	CH <sub>4</sub> As } C <sub>7</sub> H <sub>7</sub>				151	5.6	C <sub>6</sub> H <sub>4</sub> As	77	8.6	C <sub>6</sub> H <sub>5</sub>
						150	9.6	C <sub>8</sub> H <sub>11</sub> P, As <sub>2</sub>	75	7.1	As
						149	24.4	C <sub>9</sub> H <sub>10</sub> P			
						147	8.6	C <sub>5</sub> H <sub>12</sub> As			
						146	3.3	C <sub>5</sub> H <sub>11</sub> As			

TABLE 4  
FRAGMENTATION PATTERN OF  $\text{Me}_2\text{As}(\text{CH}_2)_{12}\text{AsMe}_2$  (IX)

Mass	Rel. Int.	Fragment	Mass	Rel. Int.	Fragment
378	1.7	$\text{C}_{16}\text{H}_{36}\text{As}_2$	133	3.5	$\text{C}_4\text{H}_{10}\text{As}$
377	1.7	$\text{C}_{16}\text{H}_{35}\text{As}_2$	131	1.5	$\text{C}_4\text{H}_8\text{As}$
363	100.0	$\text{C}_{15}\text{H}_{33}\text{As}_2$	120	3.3	$\text{C}_3\text{H}_9\text{As}$
349	6.1	$\text{C}_{14}\text{H}_{31}\text{As}_2$	119	5.4	$\text{C}_3\text{H}_8\text{As}$
347	2.1	$\text{C}_{14}\text{H}_{29}\text{As}_2$	117	4.3	$\text{C}_3\text{H}_6\text{As}$
335	21.1	$\text{C}_{13}\text{H}_{29}\text{As}_2$	107	6.9	$\text{C}_8\text{H}_{11}$
321	1.8	$\text{C}_{12}\text{H}_{27}\text{As}_2$	106	13.8	$\text{C}_2\text{H}_7\text{As}$
307	4.0	$\text{C}_{11}\text{H}_{25}\text{As}_2$	105	31.6	$\text{C}_2\text{H}_6\text{As}$
293	1.9	$\text{C}_{10}\text{H}_{23}\text{As}_2$	103	9.6	$\text{C}_2\text{H}_4\text{As}$
279	2.8	$\text{C}_9\text{H}_{21}\text{As}_2$	101	2.3	$\text{C}_2\text{H}_2\text{As}$
274	2.1	$\text{C}_{14}\text{H}_{31}\text{As}$	95	3.7	$\text{C}_7\text{H}_{11}$
273	1.3	$\text{C}_{14}\text{H}_{30}\text{As}$	91	2.8	$\text{CH}_4\text{As}$
272	1.4	$\text{C}_{14}\text{H}_{29}\text{As}$	90	4.2	$\text{CH}_3\text{As}$
265	1.2	$\text{C}_8\text{H}_{19}\text{As}_2$	89	4.8	$\text{CH}_2\text{As}$
251	2.6	$\text{C}_7\text{H}_{17}\text{As}_2$	83	4.9	$\text{C}_6\text{H}_{11}$
245	2.5	$\text{C}_{12}\text{H}_{26}\text{As}$	81	4.1	$\text{C}_6\text{H}_9$
237	2.8	$\text{C}_6\text{H}_{15}\text{As}_2$	75	2.0	As
231	2.1	$\text{C}_{11}\text{H}_{24}\text{As}$	71	3.8	$\text{C}_5\text{H}_{11}$
217	1.1	$\text{C}_{10}\text{H}_{12}\text{As}$	69	18.9	$\text{C}_5\text{H}_9$
210	16.3	$\text{C}_4\text{H}_{12}\text{As}_2$	67	7.0	$\text{C}_5\text{H}_7$
201	2.9	$\text{C}_9\text{H}_{18}\text{As}$	57	13.6	$\text{C}_4\text{H}_9$
196	1.5	$\text{C}_3\text{H}_{10}\text{As}_2$	56	5.7	$\text{C}_4\text{H}_8$
195	2.4	$\text{C}_3\text{H}_9\text{As}_2$	55	41.5	$\text{C}_4\text{H}_7$
189	0.7	$\text{C}_8\text{H}_{18}\text{As}$			
187	2.2	$\text{C}_8\text{H}_{16}\text{As}$			
182	2.4	$\text{C}_2\text{H}_8\text{As}_2$			
181	1.7	$\text{C}_8\text{H}_{10}\text{As}$			
175	1.5	$\text{C}_7\text{H}_{16}\text{As}$			
161	1.4	$\text{C}_6\text{H}_{14}\text{As}$			
159	1.3	$\text{C}_6\text{H}_{12}\text{As}$			

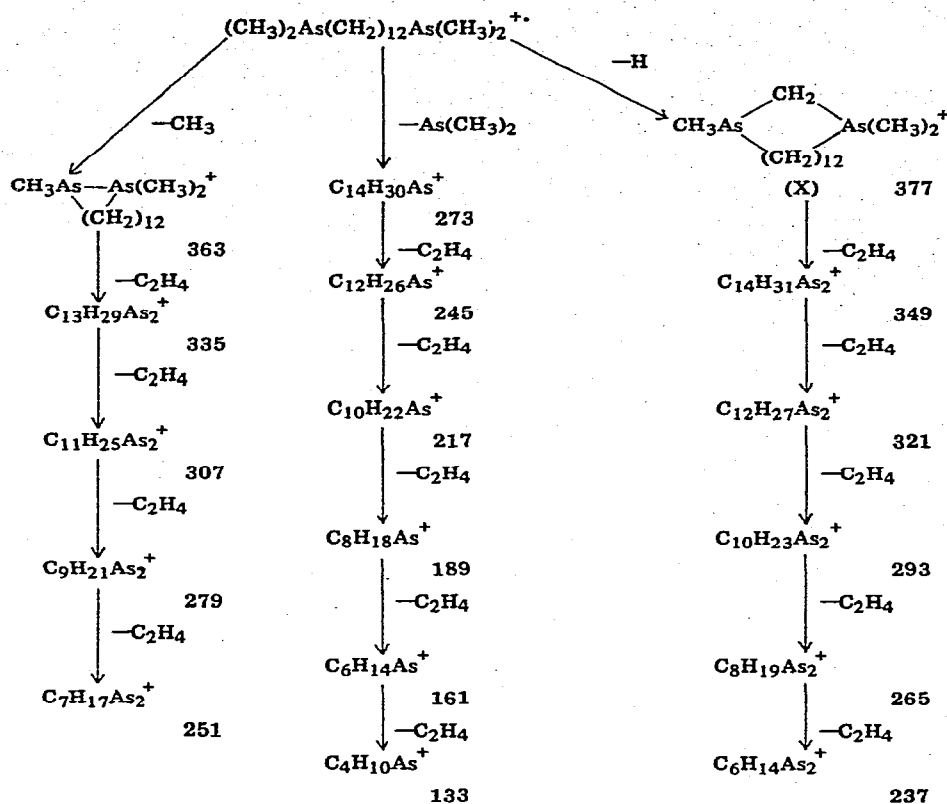
### Ligands VII and VIII (Table 3)

The fragmentation patterns of the two tridentates are complex but show similar fragmentation routes to the bidentates previously discussed. As in the trimethylene backbone bidentates no parent ion is seen, the heaviest fragment being  $P - \text{CH}_3^+$ . In the case of VIII this ion is also the base peak  $m/e$  387, however, the base peak for VII occurs at  $m/e$  299 corresponding to  $P - (\text{CH}_2)_3\text{-AsMe}_2^+$ . This is a reflection of the relatively stronger C—P bond against the C—As bond. The loss of  $-(\text{CH}_2)_3\text{AsMe}_2$  from the parent does occur for VIII but with a relative intensity of 18.2%,  $m/e$  255. All expected ions of lower  $m/e$  are found.

### Ligand IX (Table 4)

The ligand 1,12-bis(dimethylarsino)dodecane gives a complex fragmentation pattern. The parent ion is seen at low relative intensity  $m/e$  378, 1.7%, and analogous to II and VI the base peak corresponds to  $P - \text{CH}_3^+$ ,  $m/e$  363. A weak  $P - 1$  peak occurs probably corresponding to the cyclic ion (X), which then loses ethylene progressively. Two other series of progressive ethylene loss are seen beginning with the ions  $P - \text{CH}_3^+$ ,  $m/e$  363, and  $P - \text{As}(\text{CH}_3)_2^+$ ,  $m/e$  273. (Scheme 2). The other most prominent arsenic containing fragments are  $m/e$  210,  $(\text{CH}_3)_2\text{AsAs}(\text{CH}_3)_2^+$  and  $m/e$  105,  $(\text{CH}_3)_2\text{As}^+$ .

## SCHEME 2

FRAGMENTATION OF  $\text{Me}_2\text{As}(\text{CH}_2)_{12}\text{AsMe}_2$ 

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