

Preliminary communication

PENTAMETHYLCYCLOPENTADIENYL-RHODIUM AND -IRIDIUM COMPLEXES

XXIII*. DI- μ -CARBONYLBIS(PENTAMETHYLCYCLOPENTADIENYL-RHODIUM)

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(Received November 27th, 1978)

Summary

The synthesis and properties of the novel dinuclear rhodium carbonyl complex $[\text{Rh}_2(\eta^5\text{-C}_5\text{Me}_5)_2(\text{CO})_2]$ are reported.

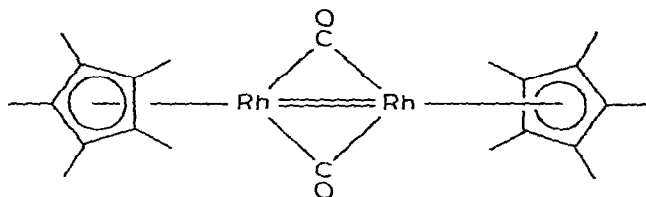
The current widespread interest in cyclopentadienylrhodium carbonyl cluster complexes [2–7] prompts us to communicate the preparation of a new dinuclear complex $[\text{Rh}_2(\text{C}_5\text{Me}_5)_2(\text{CO})_2]$ (I). The cyclopentadienylcobalt analogue $[\text{Co}_2(\text{C}_5\text{H}_5)_2(\text{CO})_2]$ has been obtained [8,9], but there has so far been no report of a corresponding rhodium complex.

We find that, although the yellow $[\text{Rh}(\text{C}_5\text{Me}_5)(\text{CO})_2]$ can, as described [10], be sublimed unchanged at 10^{-3} mmHg and $30\text{--}40^\circ\text{C}$, when it is heated (1 h/ $80\text{--}85^\circ\text{C}$) at higher pressure (10–20 mmHg) a new dark blue complex is obtained which proved to be $[\text{Rh}_2(\text{C}_5\text{Me}_5)_2(\text{CO})_2]$ (I).

Complex I was characterised by elemental analysis (Found: C, 49.7; H, 5.4. $\text{C}_{22}\text{H}_{30}\text{O}_2\text{Rh}_2$ calcd.: C, 49.6; H, 5.7%), by its mass spectrum which indicated it to be dinuclear (m/e 532 $[\text{Rh}_2(\text{C}_5\text{Me}_5)_2(\text{CO})_2]^+$, 504 $[\text{Rh}_2(\text{C}_5\text{Me}_5)_2(\text{CO})]^+$, and 476 $[\text{Rh}_2(\text{C}_5\text{Me}_5)_2]^+$), as well as by ^1H and ^{13}C NMR spectroscopy (^1H NMR (CDCl_3), δ at 1.65 ppm (s, C_5Me_5); ^{13}C NMR (C_6D_6) δ 9.3 (C_5Me_5) and 101.1 ppm (d, $J(\text{C-Rh})$ 6 Hz; $\underline{\text{C}}_5\text{Me}_5$). The IR spectrum showed a strong $\nu(\text{CO})$ band at 1732 cm^{-1} , consistent with the presence of bridging carbonyls. The IR spectrum of $[\text{Co}_2(\text{C}_5\text{H}_5)_2(\text{CO})_2]$ showed $\nu(\text{CO})$ at 1790 cm^{-1} [8] and, by analogy, we formulate I as shown with a rhodium–rhodium double bond. The colour arises from a charge-transfer band at 582 nm (in dichloromethane).

Complex I is stable to air in the solid for a day but decomposes very rapidly in solution on exposure to air to give a mixture, the water-soluble component of which showed a strong broad band in the IR spectrum at 1602 cm^{-1} consistent with the presence of a bidentate carbonato complex [11]. An analogous

*For part XXII see ref. 1.



(I)

material was obtained by reaction of $[\text{Rh}_2(\text{C}_5\text{Me}_5)_2\text{Cl}_4]$ and silver carbonate in water.

When complex I was treated with CO (3 h/20°C/1 atm) in toluene the solution turned from dark blue to brown. Evaporation of the solvent left a brown solid, the IR spectrum of which showed the presence of ca. 30% $[\text{Rh}(\text{C}_5\text{Me}_5)(\text{CO})_2]$ together with (ca. 5%) I and other materials showing $\nu(\text{CO})$ at 1793 and 1947 cm^{-1} .

We thank Johnson Matthey Ltd. for the loan of rhodium chloride and the S.R.C. and Johnson Matthey for the award of a CASE studentship to A.N.

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