

**Preliminary communication**

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**CYCLOPENTADIENYL-RUTHENIUM AND -OSMIUM CHEMISTRY.  
 PREPARATION AND REACTIONS OF SOME TRIMETHYLPHOSPHINE  
 COMPLEXES**

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(Received February 2nd, 1981)

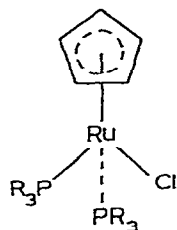
**Summary**

Cationic complexes  $[\text{Ru}(\text{L})(\text{PMe}_3)_2(\eta\text{-C}_5\text{H}_5)]^+$  (L = olefin, allene, diene, alkyne,  $\text{CS}_2$ ) have been prepared from  $\text{RuCl}(\text{PMe}_3)_2(\eta\text{-C}_5\text{H}_5)$  and L in the presence of  $\text{NH}_4\text{PF}_6$ . Oxidative addition of HCl or  $\text{Cl}_2$  to the chloro complex affords the organoruthenium(IV) complexes  $[\text{RuXCl}(\text{PMe}_3)_2(\eta\text{-C}_5\text{H}_5)]^+$  (X = H or Cl, respectively); related reactions have given  $[\text{HOsBr}(\text{PR}_3)_2(\eta\text{-C}_5\text{H}_5)]^+$  (R = Me and Ph).

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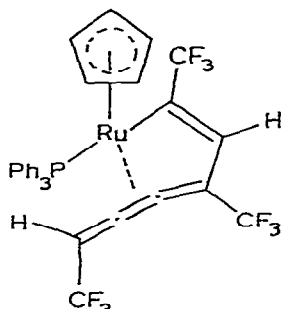
Our studies of the chemistry of  $\text{RuCl}(\text{PPh}_3)_2(\eta\text{-C}_5\text{H}_5)$  (I) and its derivatives have uncovered a wealth of unusual chemistry [1], resulting in the synthesis of such complexes as the 1,3,4,5-hexatetraenyl complex II [2], and the vinylidene derivative III [3]. Many reactions of I are facilitated by the ready loss of the bulky  $\text{PPh}_3$  ligand. It was thus of interest to investigate the chemistry of the trimethylphosphine analogue IV; the smaller size of the  $\text{PMe}_3$  ligand should result in its being attached more tightly, and its increased basicity should result in the reactivity of IV towards unsaturated molecules, for example, being considerably higher than that of I.

Complex IV is readily made by tertiary phosphine exchange with I in light petroleum at  $100^\circ\text{C}$ ; the complex forms orange crystals, m.p.  $130^\circ\text{C}$ , and is accompanied by white  $[\text{Ru}(\text{PMe}_3)_3(\eta\text{-C}_5\text{H}_5)]\text{Cl}$  (V), formed by displacement of chloride by a third molecule of the highly nucleophilic  $\text{PMe}_3$ . The tighter bonding of the  $\text{PMe}_3$  ligand is demonstrated by the considerably shorter Ru—P distances: 2.273 Å in IV versus 2.335 Å in I [4].

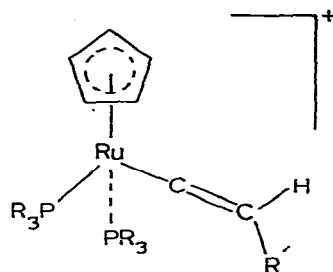


(I, R = Ph ;

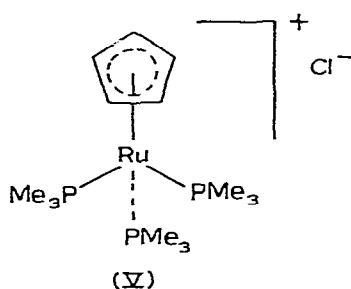
IV, R = Me )



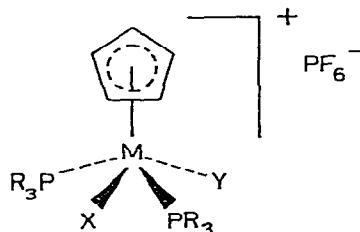
(II)



	R	R'
III	Ph	Ph
XV	Me	H
XVI	Me	Ph



(V)

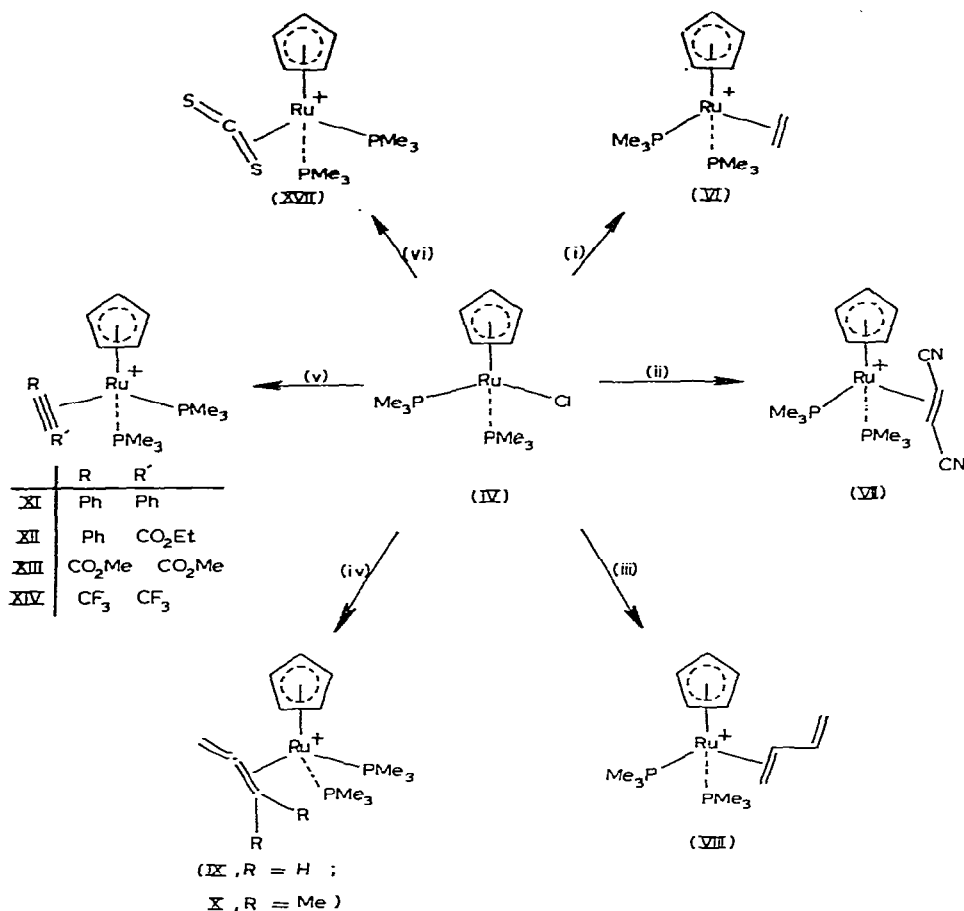


	M	R	X	Y
XVIII	Ru	Me	H	Cl
XIX	Os	Me	H	Br
XX	Ru	Me	Cl	Cl
XXI	Os	Ph	H	Br

The increased reactivity of IV towards olefins and alkynes (un) is demonstrated by the ready formation of the cationic complexes  $[\text{Ru}(\text{un})(\text{PMe}_3)_2-(\eta\text{-C}_5\text{H}_5)]\text{PF}_6$  in reactions carried out in the presence of  $\text{NH}_4\text{PF}_6$  (Scheme 1). Thus complexes with ethylene (VI) [white;  $^1\text{H NMR}^* \delta$  2.48 t,  $J(\text{HP})$  3.8 Hz,  $\text{C}_2\text{H}_4$ ], fumaronitrile (VII) [red;  $\nu(\text{CC})$  1600w,  $\nu(\text{CN})$  2209s  $\text{cm}^{-1}$ ;  $^1\text{H NMR} \delta$  6.78, 6.87, =CH], 1,3-butadiene (VIII) [white;  $\nu(\text{CC})$  1619w  $\text{cm}^{-1}$ ], allene (IX) [white;  $\nu(\text{CC})$  1725w, 1685w  $\text{cm}^{-1}$ ;  $^1\text{H NMR} \delta$  5.50 m, =CH<sub>2</sub>; 6.23 m, coordinated =CH<sub>2</sub>], 1,1-dimethylallene (X) [white;  $^1\text{H NMR} \delta$  1.83 s,  $\text{CMe}_2$ ; 5.35 m, =CH<sub>2</sub>], diphenylacetylene (XI) [yellow;  $\nu(\text{CC})$  1900w  $\text{cm}^{-1}$ ], ethyl phenylpropiolate (XII) [yellow;  $\nu(\text{CC})$  1880w, 1852w;  $\nu(\text{CO})$  1698vs  $\text{cm}^{-1}$ ], dimethyl acetylenedicarboxylate (XIII) [yellow;  $\nu(\text{CC})$  1865s;  $\nu(\text{CO})$  1696vs  $\text{cm}^{-1}$ ], and hexafluorobut-2-yne (XIV) [orange-yellow;  $\nu(\text{CC})$  1863m  $\text{cm}^{-1}$ ] have been prepared and suitably characterised\*\*, whereas we have been unable

\*In ppm.

\*\* All new complexes have satisfactory elemental analyses, and characteristic resonances for the  $\text{C}_5\text{H}_5$  and  $\text{PMe}_3$  ligands in their  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra.



## SCHEME 1

Reagents (i)  $C_2H_4$ ; (ii) *trans*- $CH(CN)=CH(CN)$ ; (iii)  $CH_2=CHCH=CH_2$ ; (iv)  $CH_2=C=CR_2$ ; (v)  $RC\equiv CR'$ ; (vi)  $CS_2$ ; all reactions carried out in the presence of  $NH_4PF_6$ .

to prepare analogous complexes containing  $PPh_3$  from I. We note that the complexes  $[Ru(un)(dppe)(\eta-C_5H_5)]PF_6$  ( $un = C_2H_4, CH_2=CHMe, CH_2=CHPh, CH_2=CHCO_2Me$  and  $CH_2=CHCH=CH_2$ ,  $dppe = Ph_2PCH_2CH_2PPh_2$ ) have been briefly reported [5]. As found with the  $PPh_3$  series, however, 1-alkynes react with IV to give the vinylidene complexes, such as yellow  $[Ru(\eta^1-C=CH_2)(PMe_3)_2(\eta-C_5H_5)]PF_6$  (XV) [ $\nu(CC)$  1749m, 1632w  $cm^{-1}$ ;  $^1H$  NMR  $\delta$  5.35 t,  $J(HP)$  7.2 Hz,  $=CH_2$ ] and deep red  $[Ru(\eta^1-C=CHPh)(PMe_3)_2(\eta-C_5H_5)]PF_6$  (XVI). A further indication of the reactivity of IV is given by the reaction with  $CS_2$ , which affords orange  $[Ru(\eta^2-S=C=S)(PMe_3)_2(\eta-C_5H_5)]PF_6$  (XVII) [ $\nu(CS)$  824s  $cm^{-1}$ ].

This high reactivity suggested that IV and the related osmium complex should undergo oxidative addition reactions to form ruthenium(IV) or osmium(IV) derivatives. The reactions between these complexes and either  $HPF_6$  or  $Cl_2$  gave directly  $[HMX(PMe_3)_2(\eta-C_5H_5)]PF_6$  [ $M = Ru, X = Cl$ ]

(XVIII); M = Os, X = Br (XIX)] and  $[\text{RuCl}_2(\text{PMe}_3)_2(\eta\text{-C}_5\text{H}_5)]\text{PF}_6$  (XX), respectively. Complexes XVIII and XIX are readily characterised on the basis of characteristic MH resonances at  $\delta$   $-9.7\tau$  and  $-14.9\tau$ , respectively, and weak  $\nu(\text{MH})$  bands around  $2040\text{ cm}^{-1}$ . These metal(IV) complexes also have unusually low-field  $\text{C}_5\text{H}_5$  resonances, at  $\delta$  ca. 6.0. In the  $\text{PPh}_3$  series, only  $[\text{HOsBr}(\text{PPh}_3)_2(\eta\text{-C}_5\text{H}_5)]\text{PF}_6$  (XXI) has been obtained, and characterised by the OsH resonance at  $\delta$   $-12.5\tau$  [ $J(\text{HP})$  34 Hz]. Addition of dibromine to  $\text{RuBr}(\text{CO})_2(\eta\text{-C}_5\text{Me}_4\text{Et})$  gives  $\text{RuBr}_3(\text{CO})(\eta\text{-C}_5\text{Me}_4\text{Et})$  in an oxidative elimination reaction. In this case, it is interesting to note that a CO group is lost, the high electron density at the metal atom being supplied by the strongly electron-donating polyalkylcyclopentadienyl group [6].

### Acknowledgement

We thank the Australian Research Grants Committee for support of this work.

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*Note added in proof.* Shortly after submission of this Communication, the paper by P.M. Treichel and D.A. Komar, "Syntheses of olefin and acetylene complexes  $[\text{Ru}(\text{C}_5\text{H}_4)(\text{PMe}_3)_2(\text{un})]\text{PF}_6$ " (*Inorg. Chim. Acta*, 42 (1980) 277) was received in the authors' library. This describes the preparation of complexes XI, XIII and XVI, and of several related derivatives.