

### Preliminary communication

## USES OF Si—N BONDS IN SYNTHESIS: EASY ROUTES TO FUNCTIONAL PROTECTED PRIMARY AMINES FROM PROPARGYLIC SILYLAMINES

ROBERT J.P. CORRIU, VILAM HUYNH, JAVED IQBAL and JOEL J.E. MOREAU

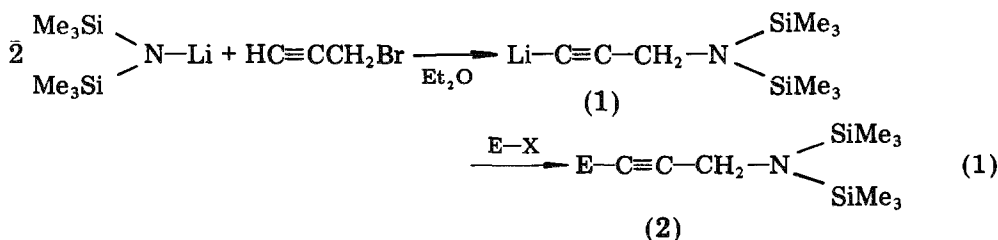
*Laboratoire des Organométalliques, Laboratoire associé au CNRS No. 349, Université des Sciences et Techniques du Languedoc, Place Eugène Bataillon, F-34060 Montpellier Cedex (France)*

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### Summary

The readily obtained *N,N*-bis(trimethylsilyl)-propargylic amines are shown to be useful precursors of various functional protected primary amines. It readily gives rise to *N,N*-bis(trimethylsilyl)dienamines, 2-aza-1,3,5-hexatrienes,  $\alpha$ -allenic amines, substituted allylamines and lactams.

We recently reported a simple synthesis of silyl-protected  $\alpha$ -acetylenic amines by use of lithium *N,N*-bis(trimethylsilyl)aminomethylacetylide **1** (eq. 1) [1].

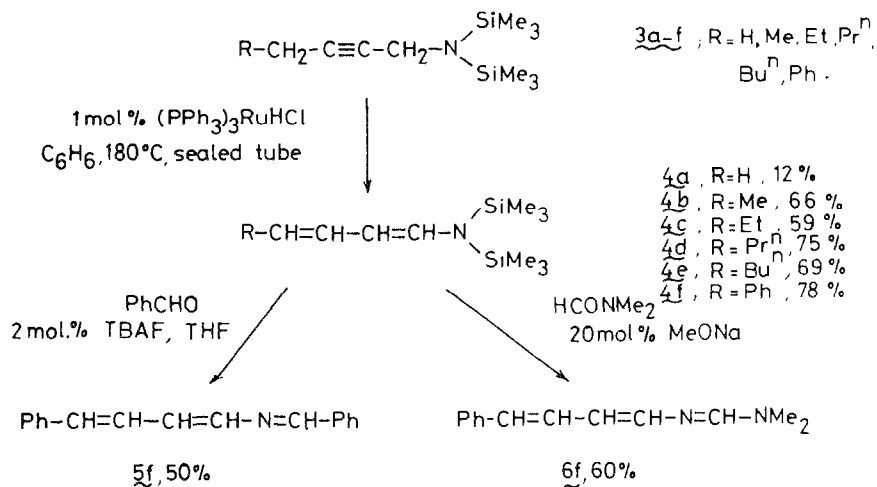


(E—X = R—X, aldehyde, ketone, enone...)

The functional  $\alpha$ -acetylenic amines **2** obtained can be used to provide easy access to various functional protected amines, and we report below easy routes to the new bis(silyl)dienamines and to allenic, allylic amines and lactams.

### Isomerisation to *N,N*-bis(trimethylsilyl)dienamines

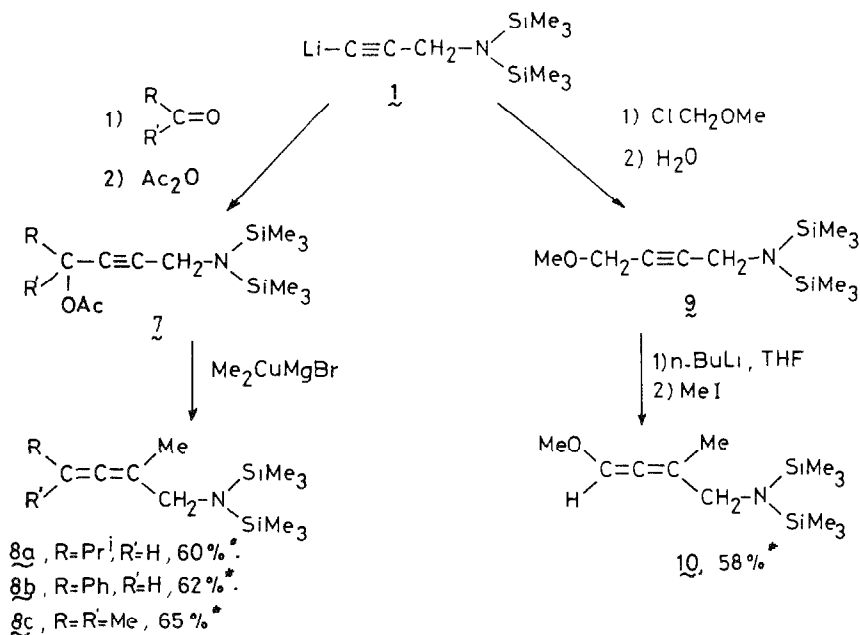
We found that the ruthenium complex  $(\text{PPh}_3)_3\text{RuHCl}$ , which is known to isomerize propargylic silyl ether [2], was also effective for the isomerization of  $\alpha$ -acetylenic silylamines. As shown in Scheme 1, the amines **3** (readily obtained



SCHEME 1. Isomerisation of *N,N*-bis(trimethylsilyl)- $\alpha$ -acetylenic amines catalysed by  $(\text{PPh}_3)_3\text{RuHCl}$ .

by alkylation of the lithium acetylide **1** [1]) were converted into conjugated dienamines by heating in benzene solution at  $180^\circ\text{C}$  in sealed tubes in the presence of 1 mole percent of  $(\text{PPh}_3)_3\text{RuHCl}$ .

Good yields of the isomerized product were obtained in most cases, except for but-2-ynylamine **3a**, for which 80% of the starting material was recovered after prolonged heating. The dienamines **4a**–**4f** were formed as a mixture of *Z* and *E* isomers which were not separated. They are very stable and are similar in reactivity to *N,N*-bis(silyl)enamines [3]. Nucleophilic activation of the silicon—



SCHEME 2. Access to silyl protected  $\alpha$ -allenic amines (\*, non-optimized yields calculated from  $\text{HC}\equiv\text{CCH}_2\text{N}(\text{SiMe}_3)_2$ ).

nitrogen bond by fluoride ion or sodium methoxide allowed reaction of benzaldehyde or dimethylformamide. In this case 2-aza-1,3,5-hexatrienes **5f–6f** were formed.

#### Conversion to $\alpha$ -allenic amines

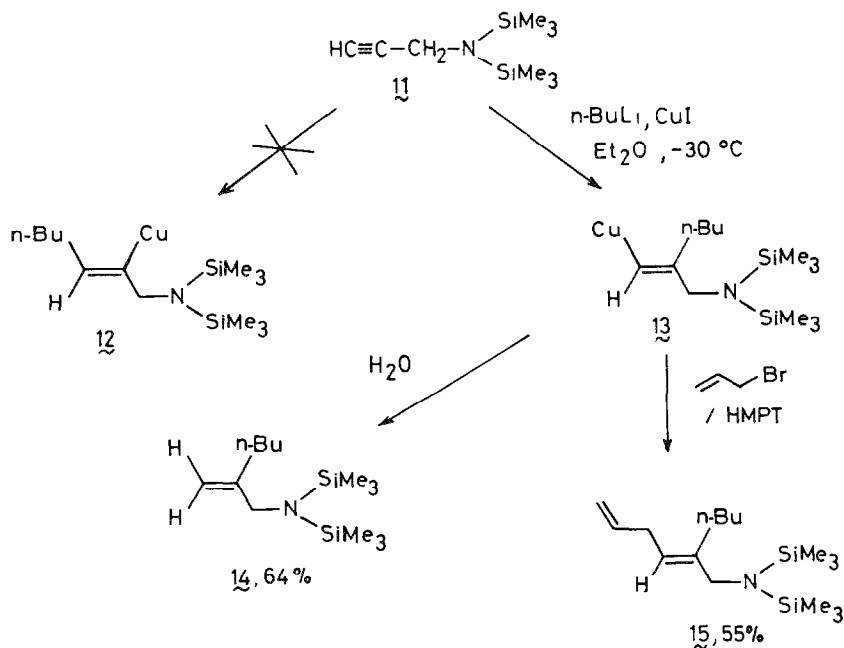
As shown in Scheme 2,  $\alpha$ -allenic amines [4], are readily obtained from functional  $\alpha$ -acetylenic amines.

The reaction of methyl-cuprate with the propargylic acetates [5] **7** obtained by condensation of carbonyl compounds with the lithium acetylide **1** gave the substituted  $\alpha$ -allenic amines **8a–8c** in 60 to 65% yield based on the lithium reagent **1**. The  $\alpha$ -aminoallenic ether **10** was also readily prepared in 58% from **1** via the metallation  $\alpha$  to the oxygen atom [6] and alkylation of the acetylenic amino ether **9**.

#### Addition of organocopper reagents

We also examined the reaction of organocopper reagents with the propargylamine **11** and the acetylenic amino ester **16** which were prepared respectively in 80 and 60% yields in one-pot reactions from propargyl bromide according to eq. 1. As shown in Scheme 3, substituted allylamines [7,8] have been readily obtained.

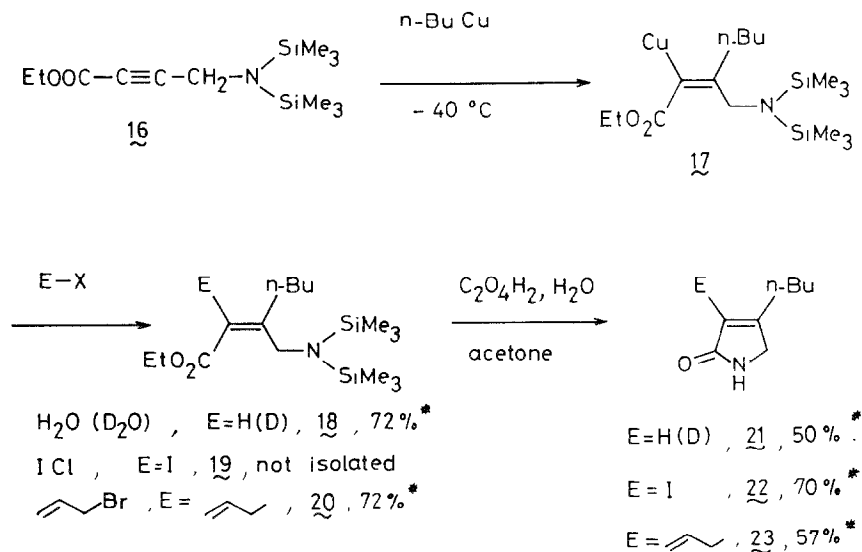
The regioselectivity of the addition of reagents to propargylic dialkylamines was reported to result in formation of the linear or branched adduct depending on the solvent ( $\text{Et}_2\text{O}$  or THF) used [9]. In contrast to the reaction with the dialkyl analogues, addition to *N,N*-bis(silyl)propargylamine **11** gave exclusively the branched adduct, and allowed isolation of allylic amines **14** and **15**. Under the same reaction conditions the reverse regioselectivity was observed in the case



SCHEME 3. Preparation of substituted allylamines.

of *N,N*-dialkylpropargylamines. In agreement with the previous interpretation [9] the low nucleophilicity of the  $(\text{Me}_3\text{Si})_2\text{N}$  group may account for a lack of intramolecular stabilisation of the linear adduct **12**.

Addition of butylcopper to the acetylenic ester **16** was also carried out (Scheme 4). Quenching of the intermediate **17** with electrophiles ( $\text{H}_2\text{O}$ ,  $\text{D}_2\text{O}$ ,  $\text{ICl}$  or  $\text{CH}_2=\text{CHCH}_2\text{Br}$ ) led to the expected *cis*-adducts [10] **18**–**20**. These underwent cyclization to lactams **21**–**23** upon desilylation by oxalic acid; yields were 50 to 70% based on the starting ester **16**.



SCHEME 4. Preparation of substituted 3-pyrrolin-2-ones. (\*, non optimized yields calculated from **16**).

## References

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