

### Preliminary communication

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## MERCURY—MERCURY SPIN—SPIN COUPLING IN AN ORGANOMERCURY COMPOUND

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### Summary

The  $^{199}\text{Hg}$  NMR spectrum of 2-methoxy-1-nitro-3,5-bis(trifluoroacetato-mercurio)benzene, prepared by mercuriation of 1-methoxy-2-nitrobenzene with mercuric trifluoroacetate in trifluoroacetic acid, shows an AB system attributable to  $^{199}\text{Hg}^{199}\text{Hg}$  coupling with  $^4J(^{199}\text{Hg}^{199}\text{Hg})$  2163 Hz.

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Although spin—spin coupling is well established between  $^{199}\text{Hg}$  (spin  $I = \frac{1}{2}$ ) and isotopes of other Main Group elements with spin  $I = \frac{1}{2}$ , e.g.  $^{29}\text{Si}$  [1],  $^{117,119}\text{Sn}$  [2], the first example of  $^{199}\text{Hg}^{199}\text{Hg}$  coupling has only recently been observed, viz. for the compound  $\text{Hg}_3(\text{AsF}_6)_2$  in liquid sulfur dioxide [3]. The coupling was observed as the inner two lines of an AB system (the outer lines being too weak for observation) for which  $^1J(^{199}\text{Hg}^{199}\text{Hg})$  was calculated to be ca. 140 000 Hz [3]. The assignment was confirmed by measurements at two different field strengths. We now report the first example of  $^{199}\text{Hg}^{199}\text{Hg}$  coupling in an organomercury compound with observation of a complete AB system.

Reaction of mercuric trifluoroacetate with 1-methoxy-2-nitrobenzene (mol ratio 2/1) in refluxing trifluoroacetic acid for 2 h yielded a yellow precipitate of 1-methoxy-2-nitro-4-trifluoroacetatomercuriobenzene (36%). The filtrate was heated under reflux for a further 8 h giving a yellow precipitate of 2-methoxy-1-nitro-3,5-bis(trifluoroacetatomercurio)benzene, **1** (16%), which was analytically pure with a  $^1\text{H}$  NMR spectrum [ $(\text{CD}_3)_2\text{SO}$ ]: 7.84, d, with satellites  $^3J(^{199}\text{HgH})$  233 Hz, 1H, H(6); 7.73, d, with satellites  $^3J(^{199}\text{HgH})$  229 Hz, 209 Hz, 1H, H(4); 3.91, s, 3H, OMe.

The proton-decoupled  $^{199}\text{Hg}$  NMR spectrum of **1** in  $(\text{CD}_3)_2\text{SO}$  is shown in Fig. 1. The major peaks are attributable to molecules with only one  $^{199}\text{Hg}$  atom ( $\delta$  (A) Hg(3);  $\delta$  (B) Hg(5)), and  $\nu_1, \nu_2, \nu_3, \nu_4$  are an AB system from molecules in

which both mercury atoms are  $^{199}\text{Hg}$  isotopes. The  $^{199}\text{Hg}^{199}\text{Hg}$  coupling constants (below) are of similar magnitude to the chemical shift difference between  $\delta(A)$  and  $\delta(B)$ , thus second order effects give a highly distorted AB pattern. From the spectrum with a JEOL FX200 instrument, an observed value of  $^4J(^{199}\text{Hg}^{199}\text{Hg})$  2163 Hz was obtained compared with a calculated value ( $\delta(A)$ ,  $\delta(B)$ ,  $\nu_3$ ,  $\nu_2$  determined from the experimental;  $\nu_1$ ,  $\nu_4$  values by  $\delta(A) - \delta(B) = \sqrt{(\nu_4 - \nu_1)(\nu_3 - \nu_2)}$

[7] with  $J = \frac{\nu_1 - \nu_4}{2} - \frac{\nu_2 - \nu_3}{2}$ ) of 2110 Hz. Confirmation of mercury—mercury

coupling was provided by relative changes in the chemical shifts of  $\nu_2$  and  $\nu_3$  when the spectrum was recorded with a JEOL FX100 spectrometer. In this case, no lines due to  $\nu_1$  and  $\nu_4$  were observed, but values calculated as above give  $^4J(^{199}\text{Hg}^{199}\text{Hg})$  2155 Hz in good agreement with the observed value (JEOL FX200).

A programme of synthesis of other unsymmetrically disubstituted benzenes is underway in particular to establish values of  $^3J(^{199}\text{Hg}^{199}\text{Hg})$  and  $^5J(^{199}\text{Hg}^{199}\text{Hg})$ . The present  $^4J(^{199}\text{Hg}^{199}\text{Hg})$  coupling is expectedly considerably smaller than the reported  $^1J(^{199}\text{Hg}^{199}\text{Hg})$  coupling [3].

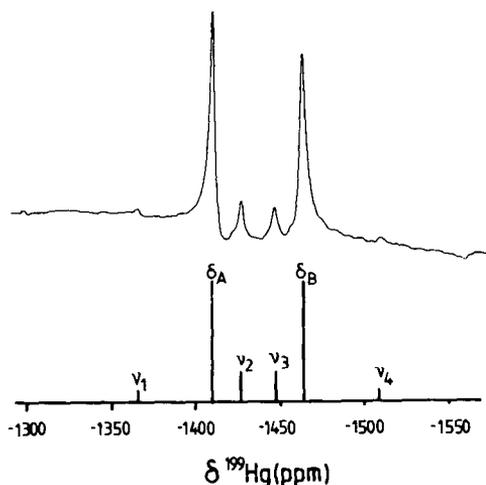


Fig. 1. Proton decoupled  $^{199}\text{Hg}$  NMR spectrum of 1 at 35.64 MHz referenced to neat  $(\text{CH}_3)_2\text{Hg}$ . The above spectrum was recorded with an internal  $d_6$  dimethylsulphoxide lock on the sample which was spun. The spectral width was 20 kHz. A delay time of 50  $\mu\text{s}$  caused the rolling baseline, however longer delay times made accurate phasing of the spectrum impossible. Chemical shift values were measured relative to 1 M phenylmercuric acetate in dimethylsulphoxide [4], the sample replacement method being used. Subtraction of 1437 ppm gives a value referenced to neat dimethylmercury [5]. The poorly phased signal at ca. -1560 ppm is an impurity due to 1-methoxy-2-nitro-3,4-bis(trifluoroacetatomercurio)benzene where the two Hg atoms have coincident chemical shifts. The identity of the species which gives rise to the small resonance at ca. -1330 ppm has not yet been established [6].

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