

Preliminary communication

FORMATION OF ARENE(CARBENE)RUTHENIUM COMPLEXES VIA VINYLIDENERUTHENIUM INTERMEDIATES

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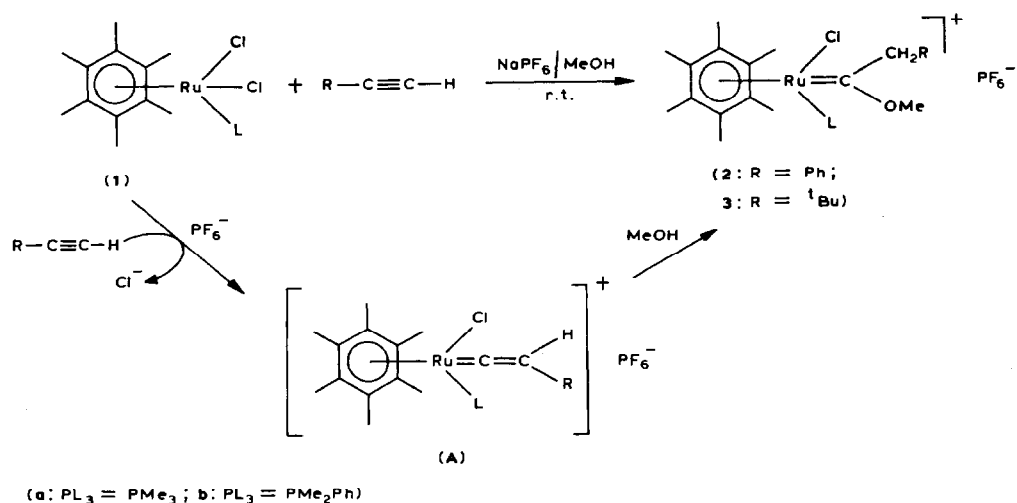
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Summary

$\text{RuCl}_2(\text{PR}_3)(\eta^6\text{-C}_6\text{Me}_6)$ ($\text{PR}_3 = \text{PMe}_3; \text{PMe}_2\text{Ph}$) reacts at room temperature with alkyne $\text{RC}\equiv\text{CH}$ and NaPF_6 in methanol to produce the (alkoxy)alkylcarbeneruthenium complexes $[\text{Ru}(\text{=C}(\text{OMe})\text{CH}_2\text{R})\text{Cl}(\text{PR}_3)(\eta^6\text{-C}_6\text{Me}_6)]^+\text{PF}_6^-$ ($\text{R} = \text{Ph}, \text{'Bu}, \text{H}$) via a vinylideneruthenium intermediate. Use of 4-hydroxybutyne-1 leads to intramolecular cyclisation and formation of $[\text{Ru}(\text{=CCH}_2\text{CH}_2\text{CH}_2\text{O})\text{Cl}(\text{PMe}_3)(\text{C}_6\text{Me}_6)]\text{PF}_6$.

Areneruthenium(II) complexes such as $\text{RuCl}_2(\text{PR}_3)(\eta^6\text{-C}_6\text{Me}_6)$ (**1**) have recently been shown [1] to be efficient catalyst precursors for the addition of carbamates to terminal alkynes to give vinyl carbamates. The regioselectivity of the addition suggested that the activation of the alkyne might proceed via a vinylideneruthenium intermediate. Although the $\text{M}(\eta^2\text{-alkyne}) \rightarrow \text{M}(\eta^1\text{-vinylidene})$ rearrangement is well documented [2], especially in the cyclopentadienylruthenium(II) series [3], it has not been observed with areneruthenium(II) derivatives. Our attempts to isolate vinylideneruthenium complexes from **1** were initially unsuccessful, but we have found now that these intermediates can be trapped with alcohol to produce new (alkoxy)alkylideneruthenium complexes containing arene ligands.

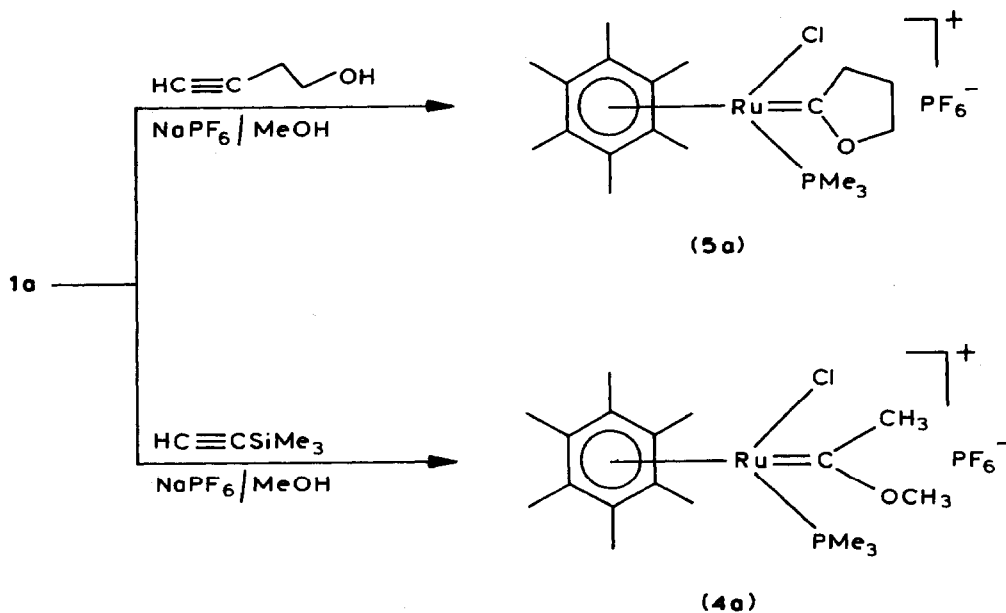
To a suspension of 1 mmol (0.41 g) of **1a** [4] in 30 ml of methanol were added 1 mmol of phenylacetylene and 1 mmol NaPF_6 . On stirring at room temperature a yellow precipitate formed rapidly, and this was recrystallized from dichloromethane/ether to give a 60% (0.395 g) yield of **2a** (Scheme 1). Only one isomer was observed by NMR spectroscopy. The ^1H NMR spectrum showed an AB system for the methylene protons, a signal consistent with a chiral ruthenium centre. In ^{13}C $\{^1\text{H}\}$ NMR, the carbene-carbon resonance appeared at low field, as expected for a cationic carbene complex, and at 223 K only the resonance due to the $\text{CH}_2\text{-C}(\text{Ph})$ carbon nuclei disappeared, indicating a coalescence related to the rotation around the $(\text{Ru})\text{CCH}_2\text{C}(\text{Ph})$ bonds. (^1H NMR (300.13 MHz, CD_2Cl_2 , 300 K) $\delta(\text{ppm})$: 7.35 (m, Ph), 5.04 and 4.50 (AB system, CH_2Ph , $^2J(\text{HH})$ 13 Hz), 4.59 (s, OMe), 2.01 (s,



SCHEME 1

C₆Me₆, 1.44 (d, PMe₃, ²J(PH) 10.6 Hz); ³¹P NMR (32.38 MHz, CD₃COCD₃, 309 K) δ (ppm): 6.48 (s, PMe₃), 145.1 (sept, PF₆⁻); ¹³C {¹H} NMR (75.47 MHz, CD₂Cl₂, 300 K) δ (ppm): 323.0 (d, Ru=C, ²J(PC) 20.6 Hz), 108.1 (s, C₆Me₆), 68.0 (s, OMe), 56.7 (s, CH₂), 16.6 (s, C₆Me₆), 16.2 (d, PMe₃, ¹J(PC) 35 Hz).

Under similar conditions reaction of complex **1b** [4] with phenylacetylene gave **2b** in 58% yield. The carbeneruthenium complex **3a** (56%) was obtained by reaction of



SCHEME 2

TABLE 1
NMR DATA FOR CARBENE-RUTHENIUM COMPLEXES 2-5

Complex	¹ H NMR (300.13 MHz) δ(=C-CH ₂)(ppm)(² J(AB))	³¹ P NMR (32.38 MHz) δ(PR ₃)(s) ^b (ppm)	{ ¹ H} ¹³ C NMR (75.47 MHz) δ(Ru=C)(ppm)(² J(PC))
2a	5.04; 4.50 (13 Hz)	6.5	323.0 (20.6 Hz)
2b	5.49; 4.09 (12 Hz)	12.6	319.3 (19.8 Hz)
3a	3.70; 3.10 (20.6 Hz)	0.7	330.5 (16.1 Hz)
4a	2.98 (=C-CH ₃)	10.5	330.9 (21.2 Hz)
5a	3.70; 3.30 (8 Hz) 5.18; 5.16 (23 Hz) ^c	8.6	317.4 (22.1 Hz)

^a In CD₂Cl₂ at 300 K. ^b In addition, a septuplet was observed at ca δ -144 ppm, with respect to H₃PO₄ as external reference, for the PF₆⁻ anion. ^c O-CH₂.

1a with 2,2-dimethylbutyne-1. Complex 4a was obtained in 60% yield by treating 1a with trimethylsilylacetylene at room temperature (Scheme 2); its formation may involve a protonolysis of the C-Si bond by methanol as it was observed in carbene-platinum [5] and -tungsten complexes [6]. Reaction of complex 1a with 4-hydroxybutyne-1 in methanol gave exclusively complex 5a, isolated in 61% yield, indicating that the intramolecular addition of the alcohol function is favoured over addition of the solvent methanol. The complexes 2-5 gave satisfactory analyses, showed in the infrared absorptions at ca. 1280 cm⁻¹ (C-OMe) and at 840 cm⁻¹ (PF₆⁻) and were characterized by NMR (Table 1).

The addition of an alcohol to an electrophilic vinylidene ligand is now well-established [2,7]. Thus the formation of the carbeneruthenium(II) complexes 2-5 probably proceeds via the addition of alcohol to the vinylidene ligand in the arene ruthenium(II) intermediates of type A (Scheme 1). The intermediates A may also be the active species in the additions to terminal alkynes catalyzed by precursors of type 1.

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