

**Preliminary communication**

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**Functionally substituted organotins:  
formation of organotin-containing phosphines**

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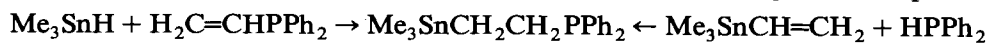
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**Abstract**

Additions of P–H bonds to vinylstannes, of Sn–H bonds to alkynylphosphines, and of Sn–P bonds to allenes and alkynes are described.

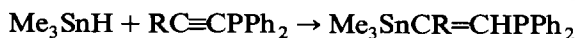
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There are few references in the literature to the functionalisation of organotin compounds by introduction of a phosphine residue. In 1980 Weichmann [1] described the addition of trimethyltin hydride to vinyl-diphenylphosphine and of diphenylphosphine to trimethylvinyltin, both reactions yielding the same product:



We decided to attempt to extend this work to include substituted vinyl residues and found that only the UV light-catalysed phosphine addition to substituted vinyltins provides a suitable route to compounds  $\text{Me}_3\text{SnCHRCH}_2\text{PPh}_2$  (R = Me, Ph,  $\text{Me}_3\text{Si}$ ,  $\text{Me}_3\text{SN}$ , OEt; yields 75–90%), attempted hydrostannation leading to polymerisation or formation of complex reaction mixtures. The stannyl-substituted phosphines can be readily oxidised by  $\text{KMnO}_4$  in acetone to give the corresponding phosphine oxides in 61–73% yields.

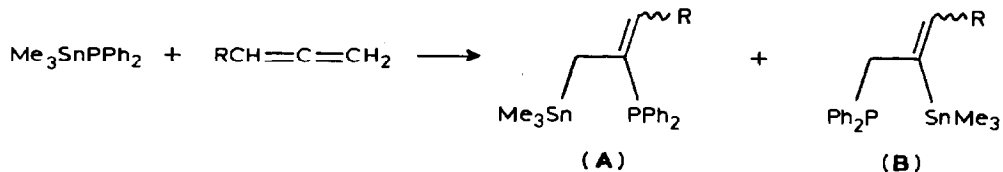
Alkynyl-diphenylphosphines also undergo hydrostannation, e.g.:



(R = H,  $\text{Me}_3\text{Sn}$ )

An alternative approach to organotin-substituted phosphines involves the addition of the Sn–P bond of a stannyl phosphine to a multiple bond. Stannyl phosphines have been known for nearly 30 years [2–8], but have remained somewhat of a laboratory curiosity. Schumann [10] described some reactions carried out with  $\text{Ph}_3\text{SnPPh}_2$ , including its additions to styrene and allyl chloride, and also the addition of  $\text{Me}_3\text{SnPPh}_2$  to acrylonitrile. Very recently, Stille [11] has reported the palladium-catalysed coupling of  $\text{Me}_3\text{SnPPh}_2$  and aryl halides to give aryl-diphenylphosphines, a reaction that we find to occur under photochemical conditions, although not so cleanly as when the palladium catalyst is present.

The addition of  $\text{Me}_3\text{SnPPh}_2$  to alkenes is probably not a particularly useful reaction (styrene polymerises and allyl chloride undergoes substitution), but its addition to both allenes and alkynes under photolytic conditions is much more promising. In the former case a mixture of two regioisomeric products is observed, with that having the phosphine residue attached to the central carbon atom predominating:

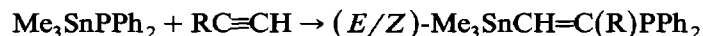


The isolated yields and proportions of **A** and **B** are as follows:

R	Yield (%)	A/B ratio
H	78	89/11
Me	67	73/27
Bu	58	88/12
Ph	83	100/0

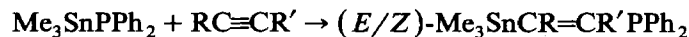
The last reaction was carried out under thermal conditions (80 °C).

The addition to both terminal and non-terminal alkynes occurs under thermal or photolytic conditions. Schumann [10] reported that  $\text{Ph}_3\text{SnPPh}_2$  adds to phenylacetylene to give a mixture of the two possible regioisomers, but we find that both this and other terminal alkynes react to give two stereoisomers and that the regiochemistry is uniform:



(R = Bu, Ph,  $\text{Et}_2\text{NCH}_2$ )

Isolated yields are between 60 and 80%: the *E*-isomer is formed preferentially (60–90%). Stannylalkyne and diphenylphosphine are formed as by-products. In the case of non-terminal alkynes, mixtures of (*E*)- and (*Z*)-isomers are again obtained, but regioisomers are absent.



(R, R' = Ph, Ph; Ph, Et;  $\text{MeOCH}_2$ ,  $\text{MeOCH}_2$ )

Isolated yields are in this case 50–90%. The (*E*)-isomer is again formed preferentially, but the *E/Z* ratio depends on the reaction conditions used.

All new compounds have been fully characterised by multinuclear NMR spectroscopy ( $^1\text{H}$ ,  $^{13}\text{C}$ ,  $^{31}\text{P}$ ,  $^{119}\text{Sn}$ ), structural assignments following from these data.

We are continuing to study these interesting and potentially valuable addition reactions of stannyl phosphines, which appear likely to join the ever-increasing list of important organotin reagents for organic synthesis.

## References

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