

Preliminary Communication

η^3 -Ligation of tetrathiamacrocycles to Mo and W: structures of $[\text{W}(\text{CO})_3\{\eta^3\text{-(Me}_8\text{[16]aneS}_4)\}]\text{[WI}_3(\text{CO})_4]$ and $[\text{W}(\text{CO})_2\{\eta^4\text{-(Me}_8\text{[16]aneS}_4)\}]\text{[WI}_3(\text{CO})_4]$ ($\text{Me}_8\text{[16]aneS}_4 = 3,3,7,7,11,11,15,15\text{-octamethyl-1,5,9,13-tetrathia-cyclohexadecane}$)

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Abstract

Reaction of $[\text{W}_2(\text{CO})_3(\text{MeCN})_2]$ with $\text{Me}_8\text{[16]aneS}_4$ gives $[\text{W}(\text{CO})_3\{\eta^3\text{-(Me}_8\text{[16]aneS}_4)\}]\text{[WI}_3(\text{CO})_4]$ and $[\text{W}(\text{CO})_2\{\eta^4\text{-(Me}_8\text{[16]aneS}_4)\}]\text{[WI}_3(\text{CO})_4]$, whose crystal structures have been determined; $[\text{Mo}(\text{CO})_3(\text{MeCN})_3]$ reacts with $[\text{16]aneS}_4$ to give $[\text{Mo}(\text{CO})_3\{\eta^3\text{-(16]aneS}_4)\}]$.

Key words: Molybdenum; Tungsten; Thiamacrocycles; Crystal structure; Iodide; Carbonyl

The use of sulfur-donor macrocyclic ligands has increased greatly in recent times because they confer the metal–sulfur environment essential in model studies of metal-catalysed processes such as biological nitrogen fixation [1,2]. Moreover, a range of coordination geometries at the metal centre is being delineated [3]. We have recently demonstrated [4] a novel “piano-stool” geometry for a tetrathiamacrocyclic ligand in the complexes $[\text{MX}(\text{CO})_2\{\eta^4\text{-(Me}_8\text{[16]aneS}_4)\}]\text{[MX}_3(\text{CO})_4]$ [4] ($\text{M} = \text{Mo}$, $\text{X} = \text{Br}$; $\text{M} = \text{W}$, $\text{X} = \text{I}$), and here we report the first X-ray structural characterisation of η^3 -ligation for these ligands.

Treatment of two equivalents of $[\text{W}_2(\text{CO})_3(\text{MeCN})_2]$ with one equivalent of 3,3,7,7,11,11,15,15-octamethyl-1,5,9,13-tetrathiacyclohexadecane ($\text{Me}_8\text{[16]aneS}_4$) in CH_2Cl_2 affords two complexes, in moderate

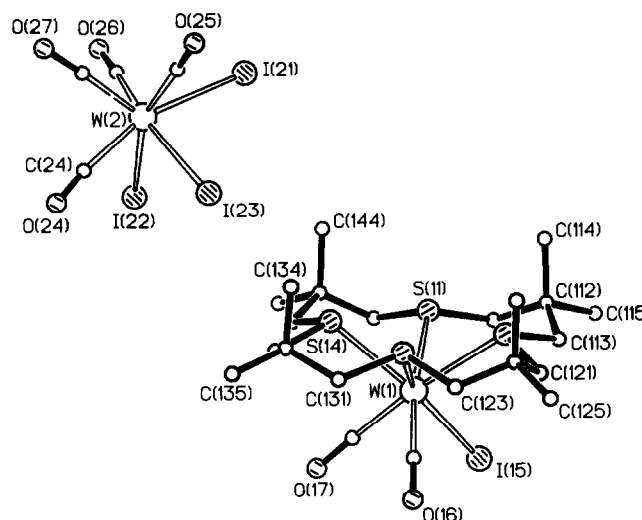


Fig. 1. Molecular structure of $[\text{W}(\text{CO})_2\{\eta^4\text{-(Me}_8\text{[16]aneS}_4)\}]\text{[WI}_3(\text{CO})_4]$, compound 2.

yield, which can be separated by fractional crystallisation. These are $[\text{W}(\text{CO})_3\{\eta^3\text{-(Me}_8\text{[16]aneS}_4)\}]\text{[WI}_3(\text{CO})_4]$ (1) and $[\text{W}(\text{CO})_2\{\eta^4\text{-(Me}_8\text{[16]aneS}_4)\}]\text{[WI}_3(\text{CO})_4]$ (2). Compound 2, which is the major product, is a further example of piano-stool η^4 -coordination by $\text{Me}_8\text{[16]aneS}_4$ (we have briefly described the structure of a molybdenum bromide analogue [4]), and its molecular structure [5*] is shown in Fig. 1. Compound 1, on the other hand, exhibits $\eta^3\text{-Me}_8\text{[16]aneS}_4$ ligation (molecular structure [6*] in Fig. 2) and to our knowledge is the first structurally-characterised example of this binding mode in a mononuclear complex. A single example of η^3 -ligation has been structurally demonstrated for the related macrocycle 1,4,8,11-tetrathiacyclotetradecane ($[\text{14]aneS}_4$) in the polymeric compound $[\{\text{Cu}(\mu\text{-}\eta^1\text{-}\eta^3\text{-[14]aneS}_4)\}_n]$ (the fourth sulfur binds a second metal in this complex) [7]. Compounds 1 and 2 show the expected spectroscopic properties [8*] and are clearly related in that conceptually 1 can be converted into 2 by intramolecular displacement of CO from the metal by the unligated sulfur [S(14) in Fig. 2]. These complexes therefore represent intermediates on the reaction pathway in the synthesis of $[\text{MX}_2\{\eta^4\text{-(Me}_8\text{[16]aneS}_4)\}]$ by displacement of MeCN and/or CO from $[\text{MX}_2(\text{CO})_3(\text{MeCN})_2]$ or $[\{\text{MX}_2\text{-}$

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* Reference number with asterisk indicates a note in the list of references.

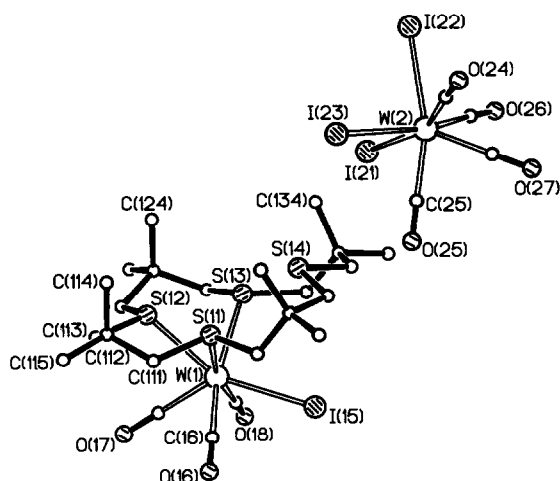


Fig. 2. Molecular structure of $[\text{W}(\text{CO})_3(\eta^3\text{-(Me}_8\text{[16]aneS}_4\text{))}] [\text{W}(\text{CO})_4]$, compound 1.

$(\text{CO})_4)_2]$ with $\text{Me}_8\text{[16]aneS}_4$ [2,4]. It is interesting to note that reaction of $[\text{W}_2(\text{CO})_3(\text{MeCN})_2]$ with the acyclic trithioether 2,5,8-trithianonane (ttn) does not give the analogous $[\text{W}(\text{CO})_3(\eta^3\text{-ttn})]^+$ cation, but rather a neutral complex identified by X-ray crystallography as $[\text{W}_2(\text{CO})_2(\eta^3\text{-ttn})]$, in which CO rather than I^- has been displaced [9]. Clearly there are subtle factors controlling product formation in these reactions.

A further example of an η^3 -tetrathiamacrocyclic ligand results when *fac*- $[\text{Mo}(\text{CO})_3(\text{MeCN})_3]$ is treated with 1,5,9,13-tetrathiacyclohexadecane ([16]aneS₄) in boiling MeCN. This reaction gives the brown, poorly soluble, diamagnetic complex *fac*- $[\text{Mo}(\text{CO})_3(\eta^3\text{-[16]aneS}_4)]$ (3), which has been characterised by analysis and spectroscopy [10*].

Clearly the flexibility of these versatile ligands allows a range of bonding modes to be adopted, depending upon the demands of the metal in terms of its available geometries and potential co-ligands [3]. We are extending this work to a range of other macrocyclic sulfur-donor ligands.

Acknowledgements

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References and notes

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- 5 Crystal data for 2: $[\text{C}_{22}\text{H}_{40}\text{IO}_2\text{S}_4\text{W}][\text{C}_4\text{I}_3\text{O}_4\text{W}]$, $M = 1452.2$, triclinic, space group $P\bar{1}$ (no. 2); $a = 11.646(1)$, $b = 13.051(3)$, $c = 14.308(2)$ Å; $\alpha = 84.345(16)^\circ$, $\beta = 102.209(10)^\circ$, $\gamma = 88.875(15)^\circ$; $V = 2113.4$ Å³; $Z = 2$; $D_{\text{calcd}} = 2.282$ g cm⁻³; $F(000) = 1336$; $\mu(\text{Mo K}\alpha) = 86.7$ cm⁻¹; $\lambda(\text{Mo K}\alpha) = 0.71069$ Å. Crystals are dark green rectangular prisms; one *ca.* 0.10 mm × 0.14 mm × 0.43 mm mounted on a glass fibre in air. Diffraction intensities measured on an Enraf-Nonius CAD4 diffractometer (with monochromated radiation): 5868 unique reflections, $\theta_{\text{max}} = 23^\circ$, 4801 with $I > 2\sigma$; corrections applied for Lorentz polarisation effects, absorption, crystal deterioration (18.7% overall), and to eliminate negative net intensities. Structure determined by automated Patterson methods (G.M. Sheldrick, SHELXS program, *Acta Cryst.*, **A46** (1990) 467). Refinement by large-block-matrix least-squares methods to $R = 0.069$ and $R_g = 0.107$ (G.M. Sheldrick, SHELX-76, *Program for crystal structure determination*, University of Cambridge, 1976; extended version, SHELXN, 1977) for 5478 reflections (with $I > \sigma$) weighted $w = (\sigma_F^2 + 0.0200F^2)^{-1}$. Hydrogen atoms of methylene groups included with idealised parameters; those in methyl groups refined with geometrical constraints; all non-hydrogen atoms refined anisotropically. Tables of atomic coordinates, bond lengths, angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre.
- 6 Crystal data for 1: $[\text{C}_{23}\text{H}_{40}\text{IO}_3\text{S}_4\text{W}][\text{C}_4\text{I}_3\text{O}_4\text{W}]$, $M = 1480.2$, orthorhombic, space group $P2_12_12_1$ (no. 19); $a = 12.449(2)$, $b = 28.875(5)$, $c = 11.825(2)$ Å; $V = 4250.6$ Å³; $Z = 4$; $D_{\text{calcd}} = 2.313$ g cm⁻³; $F(000) = 2728$; $\mu(\text{Mo K}\alpha) = 86.3$ cm⁻¹. Fragment, *ca.* 0.30 mm × 0.35 mm × 0.45 mm, of a deep red prismatic crystal mounted on a glass fibre in air. Experimental details similar to those for (2). In all, 4181 unique reflections measured, 3878 having $I > 2\sigma$; intensities corrected for absorption, and elimination of negative net intensities—no deterioration correction necessary. Refinement to $R = 0.040$ and $R_g = 0.048$ (G.M. Sheldrick, SHELX-76, *Program for crystal structure determination*, University of Cambridge, 1976; extended version, SHELXN, 1977.) for all data, weighted $w = (\sigma_F^2 + 0.00081F^2)^{-1}$. Hydrogen atoms included with geometrical constraints but with independent U_{iso} , non-hydrogen atoms all anisotropic. As in the case of 2, supplementary data have been deposited at the Cambridge Crystallographic Data Centre.
- 7 E. Dockal, L. Diaddario, M. Glick and D. Rorabacher, *J. Am. Chem. Soc.*, **99** (1977) 4532.
- 8 Satisfactory C and H elemental analyses have been obtained. Selected IR (KBr, ν/cm^{-1}): (1), 2068(s), 2036(s), 2010(sh), 1992(s), 1967(s), 1944(m) and 1924(s); (2), 2069(s), 2011(m), 1987(m), 1960(s), 1950(m) and 1865(s) (all CO stretch, s = strong, sh = shoulder, m = medium).
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- 10 Satisfactory C and H elemental analyses obtained. Selected IR (KBr, ν/cm^{-1}): 1920(s), 1785(s) (CO stretch); ¹H-NMR, ($\text{Me}_2\text{SO}-d_6$), complex second order sets of multiplets at 1.7–1.85, 2.3–2.5 and 2.75–2.95 ppm (rel. SiMe_4).