

## Preliminary Communication

### A convenient stereoselective synthesis of disubstituted alk-1-enyl phosphonates

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#### Abstract

A stereoselective synthesis of disubstituted diethyl alk-1-enylphosphonates has been developed via *syn* addition of RMgX/CuCl to alk-1-ynylphosphonates.

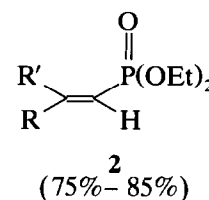
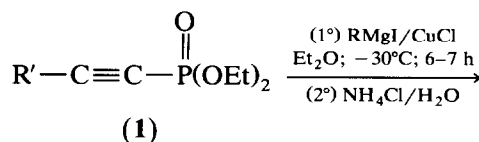
**Key words:** Phosphonate; Stereoselective synthesis; Cuprate addition; Phosphorus

Vinyl phosphonates are useful intermediates in the synthesis of biologically active compounds; for example, in the preparation of phosphomycin *via* stereospecific epoxidation of an unsaturated phosphonate [1]. Furthermore they may undergo useful transformations as reagents for organic synthesis [2].

Few methods are known for the preparation of vinyl phosphonates; they are not easily obtained using the classical Arbuzov–Michaelis reaction [3], and so their efficient synthesis is of interest. Alk-1-enyl phosphonates are generally prepared by the Wittig reaction of

acylphosphonates with methylene bisphosphonate [2,4], dehydration of  $\beta$ -hydroxyethyl phosphonate [2], oxidative elimination of organosulfonyl and organoselenyl moieties from  $\alpha$ -sulfonyl and  $\alpha$ -selenylalkyl phosphonates [2], via catalytic substitution on alk-1-enylbromide [2,5], and finally by  $\beta$ -elimination of nitrous acid from  $\alpha$ -nitroalkyl phosphonate [6].

Here we report a new and easy synthesis of disubstituted vinyl phosphonates starting from readily available alk-1-ynyl phosphonates, **1** [7] and an excess of alkyl- or aryl-magnesium iodide with cuprous salt at  $-30^\circ\text{C}$  in ether. This affords the corresponding vinyl phosphonates **2** in good yields (Table 1).



R = Oct<sup>n</sup>, Ph or 4-Tol, R' = Pr<sup>n</sup> or Ph

This addition is a typically regioselective and *syn*-addition process like the addition of organo-copper species to disubstituted alkynes [8]. The *syn*-stereoselective addition of organo-copper to  $\alpha,\beta$ -alkynyl phosphonate **1** has been established unambiguously by  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopy for compounds **2a–c**; the  $^3J$  coupling constants between  $^{13}\text{C}$  and  $^{31}\text{P}$  is typical of a *trans* value, between 21 and 24 Hz [9] (Table 2).

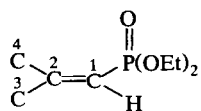
The addition of lithium organocuprate to diethyl

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TABLE 1. Synthesis of diethyl alk-1-enyl phosphonates, **2**

Compound	R	R'	Reaction time (h)	Yield <sup>a</sup> (%)	IR cm <sup>-1</sup>		$\delta^{31}\text{P}$ (CDCl <sub>3</sub> , ppm)
					$\nu_{\text{C}=\text{C}}$	$\nu_{\text{P}=\text{O}}$	
<b>2a</b>	Oct <sup>n</sup>	Ph	6	77	1615	1240	17.1
<b>2b</b>	Oct <sup>n</sup>	Pr <sup>n</sup>	7	85	1620	1250	19.2
<b>2c</b>	Ph	Pr <sup>n</sup>	6	88	1610	1250	18.5
<b>2d</b>	4-Tol	Ph	7	75	1590	1250	17.6

<sup>a</sup> Isolated pure compounds.

TABLE 2.  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra of diethyl alk-1-enyl phosphonates, **2**

Compound	R	R'	$^1\text{H}$ (CDCl <sub>3</sub> /TMS) (ppm, Hz)	$^{13}\text{C}$ (CDCl <sub>3</sub> ) (ppm, Hz)
<b>2a</b>	Oct <sup>n</sup>	Ph	5.65 (d, 1H, $^2J_{\text{P-H}} = 17.8$ , C=CH-P-); 2.45 (t, 2H, $^2J_{\text{H-H}} = 6.74$ , CH <sub>2</sub> -C=C)	162.9 (C <sub>2</sub> , $^2J_{\text{P-C}} = 3.9$ ) 139.8 (C <sub>4</sub> , $^3J_{\text{P-C}} = 7.8$ , <i>cis</i> ) 113.4 (C <sub>1</sub> , $^1J_{\text{P-C}} = 191.3$ ) 41.2 (C <sub>3</sub> , $^3J_{\text{P-C}} = 21.1$ , <i>trans</i> )
<b>2b</b>	Oct <sup>n</sup>	Pr <sup>n</sup>	5.28 (d, 1H, $^2J_{\text{P-H}} = 18.7$ , C=CH-P); 2.4 (dt, 2H, $^4J_{\text{P-H}} = 2.2$ , $^3J_{\text{H-H}} = 5.7$ ( <i>cis</i> ) C <sub>2</sub> H <sub>5</sub> CH <sub>2</sub> -C=C-); 2.1 (t, 2H, $^3J_{\text{H-H}} = 7$ , C <sub>3</sub> H <sub>7</sub> -C=C-)	167.1 (C <sub>2</sub> , $^2J_{\text{P-C}} = 6.8$ ) 110.8 (C <sub>1</sub> , $^1J_{\text{P-C}} = 189$ ) 37.8 (C <sub>3</sub> , $^3J_{\text{P-C}} = 22.5$ , <i>trans</i> ) 35.3 (C <sub>4</sub> , $^3J_{\text{P-C}} = 7.0$ , <i>cis</i> )
<b>2c</b>	Ph	Pr <sup>n</sup>	5.73 (d, 1H, $^2J_{\text{P-H}} = 17.2$ , C=CH-P); 2.95 (dt, 2H, $^4J_{\text{P-H}} = 2.4$ , $^3J_{\text{H-H}} = 7.8$ CH <sub>2</sub> -C=C)	163.7 (C <sub>2</sub> , $^2J_{\text{P-C}} = 8.7$ ) 140.6 (C <sub>3</sub> , $^3J_{\text{P-C}} = 23.7$ , <i>trans</i> ) 113.8 (C <sub>1</sub> , $^1J_{\text{P-C}} = 189$ ) 34.2 (C <sub>4</sub> , $^3J_{\text{P-C}} = 6.5$ , <i>cis</i> )
<b>2d</b>	4-Tol	Ph	6.1 (d, 1H, $^2J_{\text{P-H}} = 15.6$ , C=CH-P)	159.8 (C <sub>2</sub> , $^2J_{\text{P-C}} = 6.1$ ) 138.7 (C <sub>4</sub> , $^3J_{\text{P-C}} = 7.5$ , <i>cis</i> ) 138.4 (C <sub>3</sub> , $^3J_{\text{P-C}} = 22.3$ , <i>trans</i> ) 113.4 (C <sub>1</sub> , $^1J_{\text{P-C}} = 194$ )

alk-1-enyl phosphonates takes place in a similar way, but our first result suggest poorer selectivities and yields (they are under further investigations).

In a typical experiment, one equivalent of CuCl is introduced into a solution of five equivalents of Grignard reagent in anhydrous ether (with less than five equivalents, the reaction is slower and produces more by-products). The temperature is then lowered to  $-30^\circ\text{C}$  and one equivalent of alk-1-enyl phosphonate is slowly added and the mixture is stirred at  $-30^\circ\text{C}$  for 6 to 7 h. After hydrolysis by an aqueous solution of NH<sub>4</sub>Cl the aqueous phase is extracted with ether and the organic phases are dried with Na<sub>2</sub>SO<sub>4</sub> before concentration. The resulting crude oil is purified by column chromatography on silica gel, with ether as eluent.

In summary, the protocol represents a short regio- and stereoselective way to new disubstituted vinyl phosphonates in high yields.

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