

# [3 + 2] Cycloaddition reactions of the phospho-alkyne, Bu'CP, with a transition metal trisulphide containing terminal thio ligands: syntheses, crystal and molecular structures of $[W(\eta^5\text{-C}_5\text{Me}_5)(S)(S_2\text{PC}'\text{Bu})]_2$ and $[W(\eta^5\text{-C}_5\text{Me}_5)(S)(S_2\text{C}_2\text{Ph}_2)]_2$

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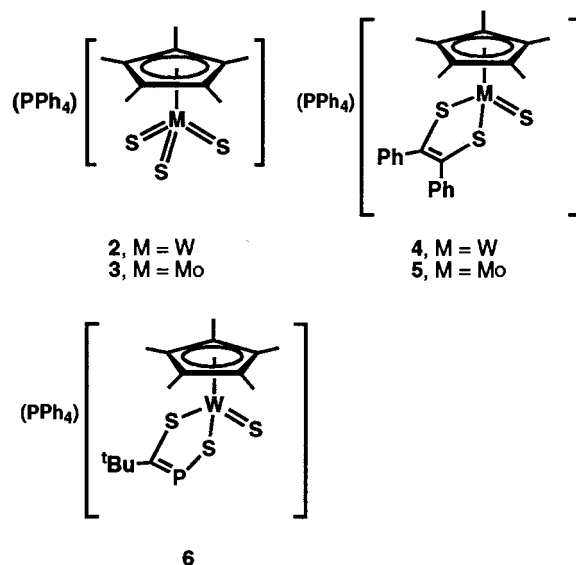
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## Abstract

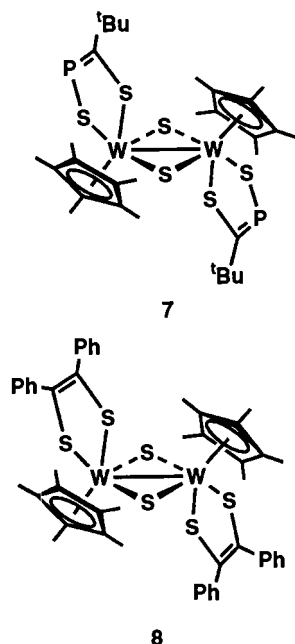
The trisulphido complex  $[\text{PPh}_4][\text{W}(\eta^5\text{-C}_5\text{Me}_5)(\text{S})_3]$ , reacts with the phospho-alkyne Bu'CP and O<sub>2</sub> to afford the dimeric complex  $[\text{W}(\eta^5\text{-C}_5\text{Me}_5)(\text{S})(\text{S}_2\text{PC}'\text{Bu})]_2$ . Analogous reactions occur with the alkyne Ph<sub>2</sub>C<sub>2</sub> to give  $[\text{W}(\eta^5\text{-C}_5\text{Me}_5)(\text{S})(\text{S}_2\text{C}_2\text{Ph}_2)]_2$ . The molecular structures of both complexes are presented and discussed. © 1998 Elsevier Science S.A. All rights reserved.

**Keywords:** Cycloaddition; Thio ligands; Trisulfide

There is considerable current interest in the study of metal sulphides, in view of their biological importance and their potential in the design of new materials [1,2]. The reactivity of certain transition metal sulphides towards alkenes [3] and alkynes [4,5] has been discussed, but a very recent report by Goodman and Rauchfuss [6], which describes the addition of nitriles to the tetrasulphido anion  $[\text{ReS}_4]^-$ , prompts us to report the first example of the ready addition of the phospho-alkyne, Bu'CP, **1**, to the anionic mononuclear trisulphido complex  $[\text{PPh}_4][\text{W}(\eta^5\text{-C}_5\text{Me}_5)(\text{S})_3]$  **2**, containing three terminal W=S bonds.



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Previously [7,8], we showed that the addition of the alkyne  $\text{Ph}_2\text{C}_2$  to complex **2** and its molybdenum analogue **3**, afforded mononuclear complexes **4** ( $M = W$ ) and **5** ( $M = \text{Mo}$ ), respectively, which were both fully characterised.

In view of the well-known similarity in behaviour of alkynes and phospho-alkynes towards transition metals [9–12] and the previously reported synthesis of several five-membered heterocyclic compounds by [3 + 2] cycloaddition reactions involving phospho-alkynes, it was of interest to study the reaction of the anionic trisulphide complex **2** with the phospho-alkyne  $\text{Bu}'\text{CP}$ . Monitoring the r.t. reaction between **2** and one equivalent of  $\text{Bu}'\text{CP}$  in acetonitrile by  $^{31}\text{P}$ -NMR spectroscopy, showed the rapid development of a new signal ( $\delta_{\text{P}} = 210.3$  ppm) assigned to complex **6**, which has not yet been possible to isolate. The same complex is obtained in the presence of five equivalents of  $\text{Bu}'\text{CP}$ .

However, when dry  $\text{O}_2$  was passed through a solution of **6**, the green–brown complex  $[\text{W}(\eta^5\text{-C}_5\text{Me}_5)(\text{S})(\text{S}_2\text{PC}'\text{Bu})]_2$  **7** was readily formed and isolated, ( $\delta_{\text{P}} = 101.4$  ppm,  $^2J(\text{WP}) = 26.8$  Hz) (52%) [13]. A single crystal X-ray diffraction study of the  $\text{CH}_2\text{Cl}_2$  solvate, revealed the dimeric structure shown in Fig. 1 [14].

In a similar way, the green solution of **4** reacted with dioxygen, instantly turning purple. The purple intermediate has not yet been characterised but it seems to be paramagnetic. After 1 day at r.t., the colour of the solution gradually changed to green and deposited crystals (41%) of the dark-green dimeric complex **8** [15], whose structure, also determined by a single crystal X-ray diffraction study, is shown in Fig. 2 [16].

The close similarity in the structures of **7** and **8** is self-evident from an inspection of the bond length and bond angle data summarised in Figs. 1 and 2. The  $\text{W}\cdots\text{W}$ ,  $\text{W-S}$  and  $\text{W=S}$  distances in the two complexes are identical within experimental error. The  $\text{WS}_2\text{PC}$  and  $\text{WS}_2\text{C}_2$  ring systems are planar in **7** and **8**, respectively. This could be interpreted as involving some  $\pi$ -delocalisation as expected for 1,2-phospho-ene thiolato and 1,2-enethiolato coordination; however the  $\text{C}(1)\text{-P}$  distance, 1.66(2) Å in **7**, and the  $\text{C}(11)\text{-C}(18)$  distance, 1.358(5) Å in **8**, are consistent with localised  $\text{P=C}$  and  $\text{C=C}$  double bonds, respectively.

The ready [3 + 2] cycloaddition reaction of these anionic trithio complexes with phospho-alkynes, suggests that other metal sulphides containing terminal  $\text{M=S}$  bonds may also form this type of compound. Complex **6** also readily reacts with metal carbonyl compounds, in the presence of an excess of  $\text{Bu}'\text{CP}$ , to give the cage compound  $\text{P}_2\text{C}_2\text{Bu}'_2\text{CO}$ , (previously obtained by Cowley et al. [17] from the reaction of  $\text{Bu}'\text{CP}$  and  $[\text{Ti}(\eta^5\text{-C}_5\text{H}_5)(\text{CO})_2]$ ), and a mixture of the 1,2,4- and 1,3,4-thiadiphospholes  $\text{P}_2\text{SC}_2\text{Bu}'_2$ . The 1,2,4-thiadiphosphole was first reported by Lindner et al. [18] in an unusual reaction involving  $\text{Bu}'\text{CP}$  and  $\eta^2$ -thiophosphinato cobalt and manganese complexes, but the 1,3,4-isomer has not been previously reported, and it will be the subject of a separate publication [19].

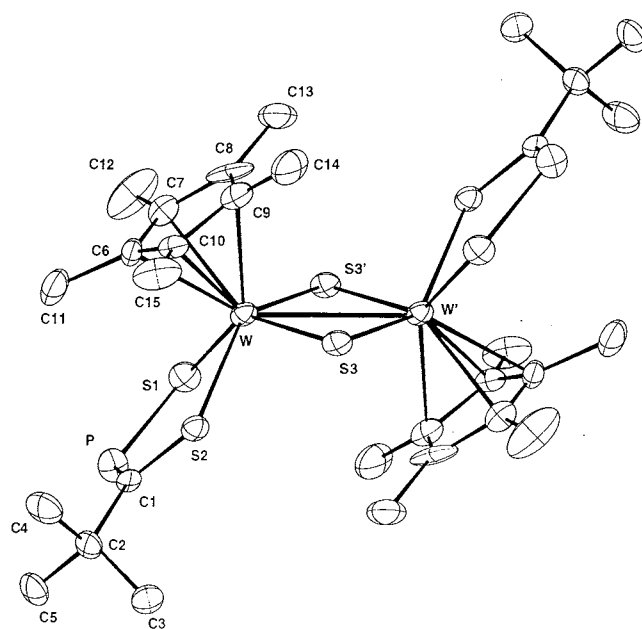


Fig. 1. Molecular structure of **7** together with selected bond length (Å) and bond angle ( $^\circ$ ) data:  $\text{W}\cdots\text{W}$  3.068(2),  $\text{W-S}(1)$  2.404(6),  $\text{W-S}(2)$  2.403(4),  $\text{W-S}(3)$  2.337(5),  $\text{W-S}(3')$  2.324(5),  $\text{S}(1)\text{-P}$  2.039(7),  $\text{S}(2)\text{-C}(1)$  1.74(2),  $\text{C}(1)\text{-P}$  1.66(2);  $\text{W-S}(3)\text{-W}$  82.3(2),  $\text{S}(2)\text{-W-S}(1)$  79.9(2),  $\text{W-S}(2)\text{-C}(1)$  116.1(6),  $\text{W-S}(1)\text{-P}$  116.3(3),  $\text{S}(2)\text{-C}(1)\text{-P}$  122.4(9),  $\text{S}(1)\text{-P-C}(1)$  102.8(7).

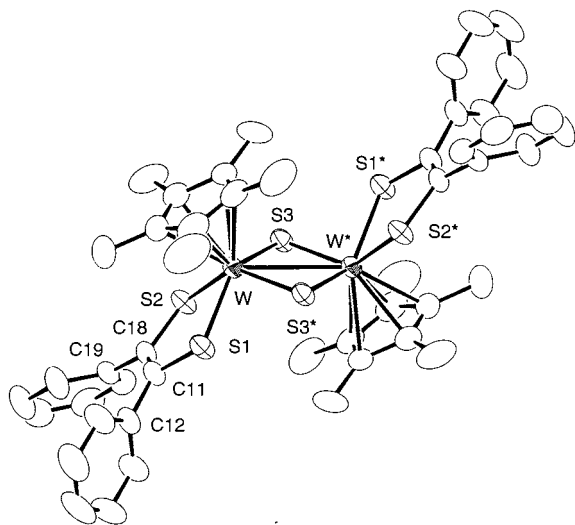


Fig. 2. Molecular structure of **8** together with selected bond length (Å) and bond angle ( $^{\circ}$ ) data: W...W 3.0652(3), W–S(1) 2.402(1), W–S(2) 2.401(1), W–S(3) 2.348(1), W–S(3)\* 2.344(1), S(1)–C(11) 1.736(4), S(2)–C(18) 1.740(4), C(11)–C(18) 1.358(5); W–S(3)–W\* 81.58(3), S(2)–W–S(1) 78.07(3), W–S(2)–C(18) 111.3(1), W–S(1)–C(11) 111.1(1), S(2)–C(18)–C(11) 118.3(3), S(1)–C(11)–C(18) 118.9(3).

## References

- [1] E.I. Stiefel, K. Matsumoto (Eds.), *Transition Metal Sulfur Chemistry, Biological and Industrial Significance*, American Chemical Society, Washington, DC, 1996.
- [2] E.I. Stiefel, D. Coucouvanis, W.E. Newton (Eds.), *Molybdenum Enzymes, Cofactors and Model Systems*, American Chemical Society, Washington, DC, 1993.
- [3] M. Rakowski, B.R. DuBois, B.R. Jagirdar, S. Dietz, B.C. Noll, *Organometallics* 16 (1997) 294, and references therein.
- [4] A.A. Eagle, S.M. Harben, E.R.T. Tiekink, C.G. Young, *J. Am. Chem. Soc.* 116 (1994) 9749.
- [5] T. Shibahara, G. Sakane, S. Mochida, *J. Am. Chem. Soc.* 115 (1993) 10408.
- [6] J.T. Goodman, T.B. Rauchfuss, *Angew. Chem. Int. Ed. Engl.* 36 (1997) 2083.
- [7] H. Kawaguchi, K. Tatsumi, *J. Am. Chem. Soc.* 117 (1995) 3885.
- [8] H. Kawaguchi, K. Yamada, J.-P. Lang, K. Tatsumi, *J. Am. Chem. Soc.* 119 (1997) 10346.
- [9] K.B. Dillon, F. Mathey, J.F. Nixon, *Phosphorus: The Carbon Copy*, Wiley, Chichester, UK, 1998.
- [10] J. F. Nixon, *Coord. Chem. Revs.* 145 (1995) 201; *Chem. Soc. Revs.* 24 (1995) 319, and references therein.
- [11] M. Regitz, P. Binger, in M. Regitz, O.J. Scherer (Eds.), *Multiple Bonds and Low Coordination in Phosphorus Chemistry*, Georg Thieme, Stuttgart, and references therein, 1991, chapter 2.
- [12] J.F. Nixon, *Chem. Rev.* 88 (1988) 1327, and references therein.
- [13] **7**:  $^1\text{H-NMR}$  (360 MHz,  $\text{CD}_3\text{CN}$ );  $\delta$  1.82 (s,  $\text{C}_5\text{Me}_5$ , 30H), 0.61 (s,  $\text{Bu}^t$ , 18H).
- [14] Crystal data for **7**:  $\text{C}_{31}\text{H}_{50}\text{Cl}_2\text{P}_2\text{S}_6\text{W}_2$ ,  $M = 1115.6$ , monoclinic, space group  $C2/c$  (No. 15),  $a = 28.794(6)$ ,  $b = 9.435(6)$ ,  $c = 15.074(4)$  Å,  $\beta = 107.17(2)^\circ$ ,  $U = 3913(3)$  Å $^3$ ,  $Z = 4$ ,  $D_{\text{calc}} = 1.89$  g cm $^{-3}$ ,  $F(000) = 2716$ . Monochromated Mo– $K_\alpha$  radiation  $\lambda = 0.71073$  Å,  $T = 173(2)$  K. Data were collected on an Enraf-Nonius CAD 4 diffractometer using a crystal of  $0.20 \times 0.20 \times 0.05$  mm. A total of 3423 independent reflections were measured for  $2 < \theta < 25^\circ$  of which 2165 had  $I > 2\sigma I$ . The structure was solved by direct methods using SHELXS-86 and refined on  $F^2$  with all non-H atoms anisotropic using SHELXL-93. Methyl-H atoms were included in riding mode, fixed at idealised geometry, but with the torsional angle defining the H atom positions refined and  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ . The final residues were  $R_1 = 0.075$  (for  $I > 2\sigma I$ ) and  $wR_2 = 0.203$  for all data.
- [15] **8**:  $^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ );  $\delta$  2.02 (s,  $\text{C}_5\text{Me}_5$ , 30H), 7.2 (m, Ph, 20H). UV-visible ( $\lambda_{\text{max}}$ , nm THF): 409, 542, 638. Found C, 49.01, H, 4.41, S, 15.89;  $\text{C}_{48}\text{H}_{50}\text{W}_2\text{S}_6$  requires C, 48.57, H, 4.25, S, 16.21%.
- [16] Crystal data for **8**:  $\text{C}_{48}\text{H}_{50}\text{S}_6\text{W}_2$ ,  $M = 1186.98$ , triclinic, space group  $P\bar{1}$  (No. 2),  $a = 10.5450(9)$ ,  $b = 11.1888(8)$ ,  $c = 9.898(1)$  Å,  $\alpha = 96.359(8)$ ,  $\beta = 91.321(8)$ ,  $\gamma = 77.343(6)^\circ$ ,  $U = 1132.4(2)$  Å $^3$ ,  $Z = 1$ ,  $D_{\text{calc}} = 1.740$  g cm $^{-3}$ ,  $F(000) = 582.00$ . Monochromated Mo– $K_\alpha$  radiation  $\lambda = 0.71069$  Å,  $T = 296$  K. Data were collected on a Rigaku AFC7R diffractometer using a crystal of  $0.20 \times 0.25 \times 0.15$  mm. A total of 5203 independent reflections were measured for  $2 < \theta < 25^\circ$  of which 4430 had  $I > 3\sigma I$ . The structure was solved by Patterson methods and refined by full-matrix least-squares with all non-H atoms anisotropic using the TEXSAN package. The final residues were  $R = 0.023$  and  $R_w = 0.027$  (for  $I > 3\sigma I$ ).
- [17] A.R. Barron, A.H. Cowley, S.W. Hall, C.M. Nunn, *Angew. Chem. Int. Ed. Engl.* 27 (1988) 837.
- [18] (a) E. Lindner, C. Haase, H.A. Meyer, M. Kemmler, R. Fawzi, M. Steimann, *Angew. Chem. Int. Ed. Engl.* 32 (1993) 1424. (b) E. Lindner, T. Schlenker, C. Haase, *J. Organomet. Chem.* 464 (1994) C31.
- [19] S.E. d'Arbeloff, S. Krill, J.F. Nixon, L. Nyulaszi, M. Regitz, paper in preparation.