

Erratum

Erratum to “Fundamental metal carbonyl equilibria, V:  
Reinvestigation of the equilibrium between dicobalt octacarbonyl and  
cobalt tetracarbonyl hydride under hydrogen pressure”  
[J. Organomet. Chem. 570 (1998) 39–47]<sup>☆</sup>

Rina Tannenbaum<sup>a,\*</sup>, Urs K. Dietler<sup>b</sup>, György Bor<sup>c</sup>, Ferenc Ungváry<sup>c</sup>

<sup>a</sup> Department of Chemical Engineering and Materials Science, University of Minnesota, Minneapolis, MN 55455, USA

<sup>b</sup> Department of Industrial and Engineering Chemistry, Swiss Federal Institute of Technology, ETH-Zentrum CH-8092 Zürich, Switzerland

<sup>c</sup> Department of Organic Chemistry, University of Veszprém, P.O. Box 158, H-8201 Veszprém, Hungary

Received 15 April 1998; received in revised form 5 June 1998

The heading to the third column in Table 2 should be  $\Delta S$  (cal mol<sup>-1</sup> K<sup>-1</sup>), and not  $\Delta H$  (kcal mol<sup>-1</sup>). The correct Table 2 is shown below.

Table 2  
Thermochemical parameters for the hydrogenation of dicobalt octacarbonyl

Reaction medium	$\Delta H$ (kcal mol <sup>-1</sup> )	$\Delta S$ (cal mol <sup>-1</sup> K <sup>-1</sup> )
<i>n</i> -hexane <sup>a</sup>	4.1	-3.1
<i>n</i> -heptane <sup>b</sup>	4.3	-2.6
CO <sub>2</sub> <sup>c</sup>	4.0	-4.2

<sup>a</sup> This work.  $\Delta H = 4.054 \pm 0.175$  (kcal mol<sup>-1</sup>) and  $\Delta S = -3.067 \pm 0.488$  (cal mol<sup>-1</sup> K<sup>-1</sup>).

<sup>b</sup> Ungváry, 1972 [24].

<sup>c</sup> Rathke et al., 1992 [45].

\* Corresponding author.

<sup>☆</sup> PII of original article S0022-328X(98)00798-0.