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## Erratum

### Erratum to “Nickel-catalysed electrochemical coupling between mono- or di-chlorophenylphosphines and aryl or heteroaryl halides”

[Journal of Organometallic Chemistry 575 (1999) 63–66]<sup>☆</sup>

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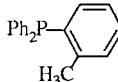
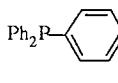
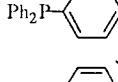
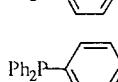
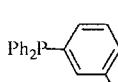
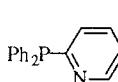
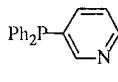
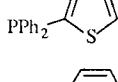
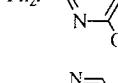
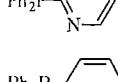
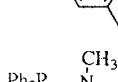
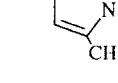
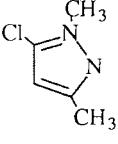
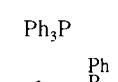
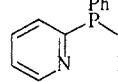
The publisher sincerely regrets that Table 1 on p. 65 of the above-mentioned paper was printed without the chemical structures. The complete Table 1 is given overleaf.

<sup>☆</sup> PII of original article S0022-328X(98)00963-2.

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**Table 1**  
Nickel-catalyzed electrochemical coupling of aryl or heteroaryl halides and chlorophosphines

entry	Ph <sub>n</sub> PCl <sub>3-n</sub>	ArX	Product	yield % <sup>(a)</sup>
1	Ph <sub>2</sub> PCl	PhBr	Ph <sub>3</sub> P	1 87
2		2-BrC <sub>6</sub> H <sub>4</sub> CH <sub>3</sub>	Ph <sub>2</sub> P- 	2 54
3		4-BrC <sub>6</sub> H <sub>4</sub> CH <sub>3</sub>	Ph <sub>2</sub> P- 	3 56
4		4-CH <sub>3</sub> O-C <sub>4</sub> H <sub>5</sub> Br	Ph <sub>2</sub> P- 	4 52
5		4-Me <sub>2</sub> N-C <sub>6</sub> H <sub>4</sub> Br	Ph <sub>2</sub> P- 	5 70
6		4-CO <sub>2</sub> Et-C <sub>6</sub> H <sub>4</sub> Br	Ph <sub>2</sub> P- 	6 63 <sup>(b)</sup>
7		3-CO <sub>2</sub> Et-C <sub>6</sub> H <sub>4</sub> Br	Ph <sub>2</sub> P- 	7 66 <sup>(b)</sup>
8		2-Cl-C <sub>5</sub> H <sub>4</sub> N	Ph <sub>2</sub> P- 	8 80
9		3-Cl-C <sub>5</sub> H <sub>4</sub> N	Ph <sub>2</sub> P- 	9 63
10		2-Cl-C <sub>4</sub> H <sub>3</sub> S	Ph <sub>2</sub> P- 	10 65
11		2-Cl-6-CH <sub>3</sub> O-C <sub>5</sub> H <sub>3</sub> N	Ph <sub>2</sub> P- 	11 66
12		2-Cl-C <sub>4</sub> H <sub>3</sub> N <sub>2</sub>	Ph <sub>2</sub> P- 	12 50
13		3-CN-C <sub>6</sub> H <sub>4</sub> Br	Ph <sub>2</sub> P- 	13 45 <sup>(b)</sup>
14			Ph <sub>2</sub> P- 	14 25
15	PhPCl <sub>2</sub>	PhBr	Ph <sub>3</sub> P	1 72
16		2-Cl-C <sub>5</sub> H <sub>4</sub> N	Ph <sub>2</sub> P- 	15 68

<sup>(a)</sup> experimental conditions: see text; isolated yields % vs chlorophosphine; <sup>(b)</sup> Anode of Zn