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Mini-review

Recent advances on the synthetic applications of the dithioacetal functionality

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Dedicated to Professor Robert J.H. Corriu, Professor Jay K. Kochi, Makoto Kumada, and Akio Yamamoto for their pioneering contributions to the cross coupling reactions

Abstract

Recent advances on the synthetic applications of the dithioacetal functionality are briefly reviewed. Nickel-catalyzed silylolefination reaction has led to the synthesis of a range of silyl-substituted olefins for optoelectronic interests. The reactions of propargylic dithioacetals with organocopper or lithium reagent followed by treatment with electrophiles yield the corresponding sulfur-substituted allenes. Further cross coupling with the Grignard reagent in the presence of a nickel catalyst affords highly substituted allenes. Acid-catalyzed cyclization of the sulfur-substituted allenyl alcohols furnishes a useful route for the synthesis of oligoaryls having highly substituted furan or pyrrole moieties. © 2002 Elsevier Science B.V. All rights reserved.

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1. Introduction

The discovery of the cross coupling reactions have led to a great leap forward in organic synthesis [1]. Carbon-carbon bond, as well as carbon-heteroatom bond, involving a C_{sp2} center can easily be formed by these protocols. An extension of this reaction to convert a carbon-sulfur bond into a carbon-carbon bond was achieved in 1979 by Takei and Wenkert (Eq. (1) and (2)) [2]. Whereas the benzylic thioether substrate is quite unreactive under these conditions [3], the discovery of the nickel-catalyzed cross coupling reaction of benzylic or allylic dithioacetals with Grignard reagents has paved a new route for the olefination of a carbonyl equivalent [4]. Representative examples are shown in Scheme 1 [4-8]. In these reactions, one of the carbon-sulfur bonds is replaced by a carbon-carbon bond, and the second carbon-sulfur bond undergoes formal

$$= \stackrel{\mathsf{SPh}}{\longrightarrow} \frac{\mathsf{PhMgBr}}{\mathsf{NiCl}_2(\mathsf{PPh_3})_2} = \stackrel{\mathsf{Ph}}{\longrightarrow} (1)$$

$$\begin{array}{c}
\text{SMe} & \underline{\text{BuMgBr}} \\
\hline
\text{NiCl}_2(\text{dppe})
\end{array}$$

This brief account summarizes our recent results based on this olefination reactions and related reactions on the synthetic applications of the dithioacetal functionality.

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elimination to give an alkene. The reaction is particularly useful for the synthesis of vinylsilanes [5]. When the cyclopropyl Grignard reagent is employed in these cross coupling reactions, the corresponding butadienes are obtained in good yield [6]. Accordingly, the dithioacetal functionality can be considered as a germinal dication synthon. Indeed, the reaction of a dimetallic species with benzylic dithioacetals furnishes an alternative procedure for the olefination of dithioacetals [7]. The key to the success of this reaction relies on the involvement of a chelation intermediate 1, which activates the second carbon–sulfur bond [8].

2. Silylolefination

Organic conjugated chromophores can exhibit a variety of fascinating optoelectronic properties [9]. Introduction of a bulky substituent to the substrates has been occasionally employed to prevent aggregation of the molecules in an electronic device. In addition, the thermal stability of organic substrates plays a pivotal role on the lifetime of devices. It is noteworthy that the benzylic carbon-carbon bond is somewhat weaker than the silicon-carbon bond in the corresponding silicon analogues. Consequently, there has been an ever burgeon use of organosilicon compounds in the optoelectronic applications [10]. As shown in Scheme 1, the olefination of dithioacetals is particularly useful for the synthesis of vinylsilanes [5]. The substituent(s) on the silicon atom can readily be tuned, and organosilicon compounds of different structural variety can be conveniently synthesized. With this strategy in mind, we have achieved in synthesizing several vinylsilanes that exhibit interesting potential for electroluminescent applications.

Reaction of bisdithioacetal **2** with Ph₂MeSiCH₂MgCl in the presence of 3 mol% of NiCl₂(PPh₃)₂ gives the corresponding bisolefination product **3** in 80% yield (Eq. (3)). It is interesting to note that **3** can serve as a hole-blocking material for a device constituted of ITO–**4**–**3**–Alq₃–Mg–Ag [11].

Scheme 1.

Alkoxysilyl-substituted styrene derivatives can be conveniently obtained from this silylolefination protocol. Bisvinylsilane 6 is synthesized from the corresponding bisdithioacetal 5 [12]. Base or TBAF-induced dimerization of 6 yields the corresponding siloxanetethered paracyclophane 7 (Eq. (4)) [13]. Treatment of 8 with base in the presence of (EtO)₄Si affords the corresponding sol gel 9 having divinylarene chromophore (Eq. (5)) [14].

The silicon-oxygen bond can readily be reduced to the corresponding silyl hydride [12]. Bis-silyl hydride 12 obtained from 10 serves as a powerful arsenal for the synthesis of a variety of alternating copolymers 13, containing the silylene moiety as the insulating spacer between chromophores (Eq. (6)) [12]. This strategy has provided a useful procedure for the synthesis of a variety of copolymers, having alternating donor and acceptor chromophores separated by

12

the silicon spacer. The photophysical properties and the optoelectronic applications of 13 have been briefly explored [15].

The silyl-substituted monomeric light emitting material 14 is synthesized in a similar manner. Bright blue emission is obtained from a device containing 14 as a dopant [16].

3. Dithioacetal as a zwitterion synthon

The electronegativities of carbon and sulfur atoms are similar. These intriguing properties lead to the rich chemistry of organosulfur compounds. The reactivity of the carbon-sulfur bond towards a nucleophile can also be altered depending on the reaction conditions. In other words, either a sulfide anion or a carbanion can serve as a leaving group in these reactions. Relatively speaking, nucleophilic substitution reactions involving the carbanionic leaving group is rare, unless this anionic species is somewhat stabilized [17,18]. The dithioacetal functionality turns out to be an ideal substrate for this purpose because the carbanion moiety generated by nucleophilic attack at one of the two sulfur atoms will be stabilized by the remaining sulfur group. Indeed, treatment of butyl lithium with a benzylic dithioacetal 15 followed by protonation yields the corresponding thioether 16 (Eq. (7)) [18]. An extension of this reaction to the propargylic system has led to the discovery of several interesting transformations.

3.1. Synthesis of substituted allenes

Treatment of propargylic dithioacetal 17 with an organocopper reagent followed by quenching with an electrophile affords either sulfur-substituted allene 20 or alkyne 21, depending on the nature of the electrophile

[19]. In the presence of a proton source, allene is obtained exclusively in 73–95% yield. (20, E = H, Eq. (8)). In a similar manner, allenyl silanes and stannanes can be prepared (20, E = Me₃Si or R₃Sn, Eq. (8)). Substituents (R¹, R²) in the starting 17 can vary from hydrogen, alkyl to aryl groups.

Bu S Bu Culi
$$\mathbb{R}^2$$

17

18

 \mathbb{R}^1
 \mathbb{R}^2
 \mathbb{R}^1
 \mathbb{R}^2
 \mathbb{R}^1
 \mathbb{R}^2
 \mathbb{R}^1
 \mathbb{R}^2
 \mathbb{R}^2
 \mathbb{R}^1
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 \mathbb{R}^4
 \mathbb{R}^2
 \mathbb{R}^4
 \mathbb{R}^2

When a soft alkyl halide electrophile is employed, selective carbon–carbon bond formation leading to an excellent yield of propargylic thioether **21** is observed.

Transmetallation of the organocopper intermediate **18** or **19** with $ZnBr_2$ followed by the palladium-catalyzed coupling reaction gives the corresponding allenes **22** in 65–78% yield (Eq. (9)).

Further reactions of the organosulfur products 20 or 21 with the Grignard reagent in the presence of a nickel catalyst yields the corresponding allenes 23 or 24, respectively (Eqs. (10) and (11)) [19]. These results demonstrate the first examples of using propargylic dithioacetals as allene zwitterion synthons 25.

18
$$\frac{R^{3}MgX}{NiCl_{2}(dppf)}$$

$$R^{1}$$

$$23$$

$$R^{2}$$

$$R^{2}$$

$$R^{2}$$

$$R^{3}$$

$$R^{2}$$

$$R^{3}$$

$$R^{3}$$

$$R^{2}$$

$$R^{3}$$

$$R^{2}$$

$$R^{4}$$

$$R^{2}$$

$$R^{4}$$

$$R^{4}$$

$$R^{2}$$

$$R^{4}$$

It is known that the propargylic ether can react with an organocopper reagent to yield an S_N2' -like product. It is intriguing to note that treatment of **26**, with Bu_2CuLi , affords the corresponding cumulene **27** in 91% yield. Apparently, the reactivity of the sulfur moiety in **26** toward the nucleophile is faster than that of the alkynyl ether group [19].

3.2. Synthesis of furans and pyrroles

It is well documented that annulation of allenyl-methanols and related compounds are known to afford the corresponding five-membered oxygen heterocycles [20].^{6–11} Thus, treatment of organocopper intermediate **18** or **19** with an aldehyde **28** or an aldimine **29** can afford the intermediate alcohol **30**, or amine **31**, respectively. Since the thioether moiety in **30** or **31** will be a good leaving group, cyclization to eliminate the sulfur moiety will lead to the substituted furan **32** or pyrrole **33** (Eq. (13)) [21].

18 or 19
$$\begin{array}{c}
R^{3}CH=X \\
\hline
28 X = O \\
29 X = NR^{4}
\end{array}$$

$$\begin{array}{c}
R^{3} \\
R^{1} \\
\hline
30 X = O \\
31 X = NR^{4}
\end{array}$$

$$\begin{array}{c}
R^{2} \\
R^{1} \\
\hline
X \\
R^{3}
\end{array}$$

$$\begin{array}{c}
R^{2} \\
R^{3} \\
\hline
X \\
R^{3} \\
32 X = O \\
33 X = NR^{4}
\end{array}$$

$$\begin{array}{c}
32 X = O \\
33 X = NR^{4}
\end{array}$$

Thus, 17 is allowed to react with Bu₂CuLi at -78 °C followed by treatment with an aldehyde 28. Trifluoroacetic acid-promoted annulation furnishes the corresponding furans 32 in satisfactory yield. A range of 2,3,5-trisubstituted furans 32 can be conveniently synthesized by this procedure. Pyrroles 33 are also obtained in a similar manner when imines 29 are used as electrophiles. It is noteworthy that BF₃·OEt₂ appears to be a better Lewis acid catalyst for the pyrrole synthesis. These reactions are particularly promising to introduce a substituent at C₃. It is a common strategy to have such alkyl group to increase the solubility of oligoaryls for the convenience of processing leading to devices for optoelectronic investigations.

Starting from the dialdehydes 34, oligoaryls 36 having two furan moieties are obtained in one pot (Eq. (14)). In a similar manner, when a diimine 35 is employed, bis-pyrroles 37 are prepared (Eq. (14)). A wide range of substituents having different functionalities remains intact under these reaction conditions. Accordingly, further modification of these functionalities may lead to a variety of oligoaryls and the lengths of conjugation can thus be tuned. For example, Heck reaction of 38 with PhI afforded 39 in 83% yield (Eq. (15)) [21].

Organic lithium reagents behave similarly. There is no apparent discrepancy in regioselectvities whether organocopper or lithium regent is employed. It is worthy to note that the reactivity of the ditioioacetal moiety under these conditions is faster than that of an ester functionality. Thus, 41 is prepared from the corresponding dithioacetal 40 and dialdehyde 34. Repetitive use of this strategy shown in Eq. (14) has paved the way to the synthesis of a variety of furan or pyrrolecontaining oligoaryls (e.g. 42) [22].

3.3. Reactions of allylic dithioacetals

Regioselective replacement of a carbon–heteroatom bond in an unsymmetrical allylic system by a carbon–carbon bond is important in organic synthesis. The selectivity depends on the nature of the substrates, reagents, stoichiometry and reaction conditions [23]. The reaction of allylic dithioacetals 43 with MeMgI in the presence of NiCl₂(dppe) leading to geminal dimethylation product 44 has been studied in details (Eq. (16)) [24].

No 1,3-dimethylation product has been observed at all. On the other hand, propargylic dithioacetal 17 can

serve as an allene 1,3-dication synthon leading to 1,3dimethylation product 45 under similar conditions (Eq. (17)) [25]. As mentioned earlier, propargylic dithioacetal 17 can behave as an allene 1,3-zwitterion synthon 25 and the regioselectivity of the reaction depends on the nature of the electrophile (Eq. (8)–(11)). Thus, the reaction of 46 under similar conditions would generate a substituted allylmetallic intermediate 47 which might behave differently from that of the allenyl-propargylic counterpart 18 or 19. Indeed, reaction of 46 with Bu₂CuLi or BuLi in THF at -78 °C followed by treatment with an alkyl halide affords regioselectively an E-Z mixture of the corresponding vinyl sulfide 48 in good yield (Eq. (18)). It is interesting to note that the reaction of alkyl electrophile occurs regioselectively at the position away from the sulfur substituent of the corresponding unsymmetrical allyl organometallic species 47. This selectivity is just opposite to that observed for the propargylic substrates [20]. The proton electrophile, however, gives a mixture of regioisomers.

A mixture of E-Z isomeric 48 is treated with the Grignard reagent in the presence a catalytic amount of NiCl₂(dppe) to afford the corresponding coupling product 49 in good yield (Eq. (19)) [2,26]. In general, the Grignard reagent having the same alkyl group as the R¹ group in 46 is employed so that the stereochemical problem in 48 can be lifted. The regioselectivity in this reaction just complements that of procedures shown in Eq. (16).

48
$$\xrightarrow{R^1 \text{MgX}} \xrightarrow{R^2} \xrightarrow{R^1} \text{R}^1$$
 (19)

4. Conclusions

We have demonstrated the recent advances on the synthetic applications of the dithioacetal functionality. An extension of the olefination reaction has led to the discovery of a new route towards the synthesis of a wide range of conjugated systems for optoelectronic interests. The reaction of propargylic dithioacetals with

organolithium or copper reagents furnishes a new entry for the regioselective carbon–carbon bond formation in allene synthesis as well as in five-membered heteroaromatic annulation. In a similar manner, allylic dithioacetal can be considered as a propene 1,3-zwitterion synthon. Due to the rich chemistry of organosulfur compounds, the opportunity for further development involving the dithioacetals abounds.

Acknowledgements

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