

## ADDITIONS AND CORRECTIONS

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2003, Volume 107A

**E. K. L. Yeow, L. D. Slep, A. K. Chibisov, and S. E. Braslavsky:** Photoinduced Electron-Transfer Reaction between the Erythrosin Dianion and  $\text{Mo}(\text{CN})_8^{4-}$  in the Presence of Various Cations. The First Example of Enthalpy–Entropy Compensation in Electron Transfer between Anions

Page 442, Table 3. In the Table title and column headings,  $\text{E}^{3-}$  was inadvertently used instead of  $\text{Er}^{3-}$ , which is used elsewhere in the text. Footnote b applies to column  $\text{Er}^{3-}$  production (LIOAS), and footnote c applies to column  $\text{Er}^{3-}$  production (TA) and column  $\text{Er}^{3-}$  decay (TA).

Page 443, Scheme 1. The values of  $\lambda_{\text{max}}$  have been inadvertently exchanged in the Scheme. They should be  $\lambda_{\text{max}} = 620$  nm for  ${}^3\text{Er}^{2-}$  and  $\lambda_{\text{max}} = 410$  nm for  $\text{Er}^{3-}$ , as quoted several times in the text.

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2002, Volume 106A

**Pradyot K. Chowdhury:** ArF Laser Photodissociation Dynamics of 1,4-Pentadien-3-ol: Laser-Induced Fluorescence Observation of OH Rovibrational States

Page 6226. The correct form of eq 3 in section 3.3 is

$$I(t) = [A/(B + Ct^2)][1 - \exp(-kt)]$$

which has been used to derive the dissociation rate constant  $k$  by fitting the time-dependent data of Figure 5.

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