Infrared Spectra and Quantum-Chemical Studies of Disilylmethanes

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New infrared data are reported for disilylmethane- d_0 and $-d_6$ in the gas phase. Quantum-chemical (QC) calculations have been carried out at HF, MP2, and B3LYP levels with a variety of basis sets. These verify the previously obtained experimental C_{2v} structure. The two possible kinds of SiH bonds are identical in strength but not in electrical properties. Reassignments are made of several vibration frequencies. The A_2 and B_2 torsional frequencies are estimated to be 78.5 and 133 cm⁻¹, respectively, from combination and difference bands. QC-based force fields scaled with 14 factors are used to predict unknown or uncertain frequencies. Scale factors for the A_2 torsion are highly variable with level and basis. By contrast, similar QC calculations of torsional frequencies in disilane, methyl-, fluoro-, chloro-, and ethyl-silanes show very little variation. Stretch/stretch interaction constants between SiH bonds in different silyl groups correlate well with the dipoledipole energies derived from the QC calculations. Electrical properties of the SiH bond are compared across a range of SiH-containing compounds. A survey of CH stretch/CH stretch interaction force constants in 17 molecules containing CH₃ or CH₂ groups indicates that MP2 calculations are unlikely to be suitable for close analyses of the stretching frequencies of these groups.

Introduction

The complete vibrational spectrum of disilylmethane $[(SiH₃)₂$ - $CH₂$], hereafter DSM, was last studied in 1970,¹ since when only the SiH stretching region has been reexamined.2 The lack of resolution in the earlier work means that a number of aspects of the spectra remain to be clarified, for which purpose a reliable general force field is essential. This can be provided by a good quality ab initio or DFT treatment. The structural interest in DSM in the past has tended to focus on the SiCSi angle which by electron diffraction $(ED)^3$ or microwave spectroscopy $(MW)^4$ is somewhat wider than tetrahedral (ED, 114.4(2)°; MW, 114.1- (6)[°]) in a C_{2v} structure, Figure 1, which in other respects resembles that of propane. There is no indication in the MW investigation of a twisting of the silyl groups away from a staggered configuration which characterizes the structures of tri- and tetrasilylmethanes,^{5,6} for which a companion QC study has been made.⁷ Technical interest in the silylmethanes generally has been stirred by their possible use in the preparation of thin films by chemical vapor deposition.^{5,6}

Another structural aspect is the extent to which the two kinds of SiH bonds present can be said to be equivalent. The simplest source of such information is the spectrum of the partially deuterated species $(SiHD_2)(SiD_3)CH_2$, where the solitary SiH bond lies in either the skeletal plane (H^s) or above/below this plane (H^a), leading to the possibility of two isolated SiH stretching frequencies $ν$ ^{is}SiH^s and $ν$ ^{is}SiH^a. Observation of only a single *ν*isSiH band suggested that the two kinds of bonds were identical in strength.² In attempts to interpret both the ν ^{is}SiH value and the SiH stretching frequencies of the parent molecule, long-range dipole-dipole forces acting between the stretching motions of SiH bonds in the two silyl groups were invoked.² The presence of such forces acting between SiH bonds similarly disposed has been predicted by quantum-chemical (QC) calcula-

Figure 1. C_{2v} structure of disilylmethane, showing atom numbering.

tions in several HSiNSiH systems, $8-10$ where it has been associated with the substantial negative charge associated with the moving hydrogen atom in an SiH bond. This moving charge is the source of the high SiH stretching infrared intensities typical of SiH compounds, whose variations determined experimentally in methylsilanes correlate very well with those computed in QC calculations.¹¹

The latter also show a close link between the static charge on the hydrogen atom as measured by the Mulliken charge and the charges associated with stretching and bending motions of the SiH bonds.11 The possibility of such a connection in DSM has the added interest that in principle two chemically distinct kinds of SiH bond are present.

Experimental Section

Samples of $(SiH₃)₂CH₂ (DSM-d₀)$ and $(SiD₃)₂CH₂ (DSM-d₁)$ d_6) were available from earlier work² and were refractionated before use. Infrared intensities indicated that the concentration of H in d_6 was about 2 atom %.

Mid-infrared spectra were recorded on a Nicolet 7199 FTIR instrument at a resolution of 0.25 cm^{-1} . Approximate intensity * To whom correspondence should be addressed. E-mail: dcmck@
lyrood.ed.ac.uk. data were obtained which will be reported elsewhere.¹²

Theoretical Section

Calculations were performed with the program Gaussian 9813 under the conditions of previous studies¹¹ ("tight" condition for convergence, density functional calculations using 99 shells, 302 angular points per shell). At the HF, MP2, and B3LYP levels, the bases 6-31G* (hereafter "*s*V") and 6-311G** (hereafter "*tz*") were employed. The basis $6-311++G^{**}$ (hereafter " $tz+$ "), which includes diffuse functions, was used at the MP2 and B3LYP levels, whereas a single B3LYP/cc-pVTZ14 (hereafter "B3LYP/*cc*") calculation was also carried out.

The Gaussian 98 output of geometry and Cartesian force constants was input into the program ASYM4015 for the calculation of symmetry internal force constants and their subsequent scaling to observed frequencies according to the procedure recommended by Pulay et al.16

In this type of program, the off-diagonal symmetry force constants are scaled by the geometric mean of the corresponding diagonal constants, an assumption which is hard to justify on theoretical grounds. Instances of a breakdown in this assumption are not uncommon. $17-19$ However, the existence of such a breakdown can only be detected by showing that scaling with a maximum number of diagonal scale factors fails to give a satisfactory fit to the range of frequency data available.

Although a "broad brush" approach to scaling, using a limited number of scale factors, is often employed,²⁰ such an approach is not suited to the detailed reproduction of the experimental frequencies of a particular molecule. The latter can be a valuable source of information regarding small Fermi resonances.

The failure of a simple approach to scaling may be seen in different ways. The use of a single factor for the stretching constant of a particular type of CH or SiH bond rarely succeeds in reproducing accurately the small but precise differences in *ν*isCH or *ν*isSiH values found for these bonds, either in the same molecule or in related ones.18-²² Differing scale factors are then needed. More dramatic instances are the markedly different scale factors needed for the symmetric and antisymmetric SiN stretching force constants in $N(SiH₃)₃¹⁹$ and $MeN(SiH₃)₂²³$

Similar considerations apply to the bending modes of CH3, CH₂, and SiH₃ groups.^{18,24,25}

It has been argued that the merit of a particular scaled QC force field is as much to be judged by the range of scale factors it requires as by the extent of their departure from unity.^{11,18,23,25,26} This can only be properly assessed if a large number of factors is determined independently for each molecule. The transfer of factors from one molecule to the next should be regarded with suspicion.

Experimental Results

A complete listing of the new infrared frequencies obtained in this work is given in the Supplementary Tables S1 and S2 for the d_0 and d_6 isotopomers, respectively. Two weak bands at 3103 and 1262 cm⁻¹ in the d_0 spectrum, both seen before,¹ had unusual contours and may be due to impurity. The d_6 spectrum contained the usual d_5 impurity bands. Table 1 shows frequencies obtained in the *ν*SiH and *ν*SiD regions from these species, including those assigned as combination and differences involving *ν*SiH and SiH₃ torsional levels, shown in Figure 2. Other fundamental frequencies used in the scaling of the force fields are included in Table 2. The silyl deformation regions are shown in Figures 3 and 4 for d_0 and d_6 , respectively.

Problems of assignment arise throughout the spectra, discussion of which depends critically upon assistance from the QC calculations, whose results must first be presented.

Figure 2. Infrared spectrum of DSM-*d*⁰ in the *ν*SiH region, showing combination and difference bands involving torsional frequencies. Partial pressure of sample, 70 Torr in a 12 cm cell.

TABLE 1: Infrared Frequencies Observed in the *ν***SiH and** *ν***SiD Regions of DSM-***d***⁰ and -***d***⁶ in the Gas Phase**

$v_{\rm obs}^{a}$	assignment
	d_0
2303? sh	$\nu_2 + \nu_{27} + \nu_{14} - \nu_{14}$
2298 sh, C	$\nu_2 + \nu_{27}$
2291? sh, C	$\nu_2 + 2\nu_{27} - \nu_{27}$
2244.5 sh. C	$v_{16} + v_{14} + v_{27} - v_{27}$
2242.8 sh, C	$v_{16} + v_{14}$
2240.8 sh. C	$v_{16} + 2v_{14} - v_{14}$
2268.1 s, C	v_{23} (b ₂)
2167.0 s, q	hot band
2164.1 s, A	v_{15} (b ₁)
2162.5 s, q	v_{16} (b ₁) or hot band
2087 sh, C	$v_{16} + v_{14} - 2v_{14}$
2084 sh, C	$v_{16} - v_{14}$
2082 sh. C	$v_{16} + v_{27} - v_{14} - v_{27}$
2040 sh, C	$\nu_2 + \nu_{27} - 2\nu_{27}$
2033.7 sh, C	$\nu_{2} - \nu_{27}$
2029 sh, C	$\nu_2 + \nu_{14} - \nu_{27} - \nu_{14}$
	d_{6}
2164.6 s, A/C	ν ^{is} SiH in (SiHD ₂)(SiD ₃)CH ₂
1583.2 s, C	v_{23} (b ₂)
1580.2 s, A	v_{15} (b ₁)
1556.8 ms, A	v_{16} (b ₁)

^a Intensity indications refer to partial pressures in a 12 cm cell of $∼1.7$ and $∼3.7$ Torr respectively for d_0 and d_6 .

Theoretical Results

Optimization of the structure at each level with the *^s*V and *tz* basis sets, with constraint only to a C_2 point group, gave the C_{2v} conformation of the microwave investigation, as in Figure 1.⁴ Thereafter, a C_{2v} constraint was applied in all calculations. Table 3 compares the geometrical parameters obtained in all of the QC treatments with the MW and ED experimental values. The significantly larger than tetrahedral SiCSi angle observed (MW, 114.1°) is reproduced fairly well at all levels, with a

a Unscaled QC frequency (cm⁻¹). *b* QC infrared intensity (km mol⁻¹). *c* QC Raman scattering activity (\AA ⁴ amu⁻¹). *d* Frequency observed in this work, except where otherwise indicated. *^e* Frequency fit: $\epsilon_y = v_{obs} - v_{calc}$, or in parentheses, the calculated frequency. *f* Frequency omitted from the refinement *8* Raman liquid value from ref 1 ^{*i*} Solid solution v refinement. ^{*g*} Raman liquid value from ref 1. *h* Solid solution value from ref 1. *i* Refinement is only possible if 2164.1 cm⁻¹ is the higher frequency in the MP2 and B3LYP refinements, but the lower one in the HF calculations. *j* Σ WSE = sum of weighted squares of errors, calculated as described in ref 15. *k* Calculated frequency multiplied by 1.01, to offset effect of anharmonicity.²⁷ *l* Fermi resonance correction applied.

tendency to overestimate with the *tz* basis and at the HF and B3LYP levels. Of particular relevance to the interpretation of the *ν*SiH region of the spectrum are the very small differences in length between the SiH^s and SiH^a bonds. These lengths are quoted to five decimal places to avoid rounding error. (Reproducibility in the optimization procedure is about 0.00002 Å). Unlike the case of the similarly disposed CH bonds in $Me₂SiH₂$, where there is good agreement in all calculations on a small positive CHa -CHs difference, the variation in sign of the minute differences seen between the SiH^s and SiH^a lengths means that these bonds must be considered to be indistinguishable. The tilt of the silyl group, as measured by the difference between

Figure 3. Infrared spectrum of DSM- d_0 in the region $1200-600$ cm⁻¹.
Partial pressure of sample 1.8 Torr in a 12 cm cell Partial pressure of sample, 1.8 Torr in a 12 cm cell.

Figure 4. Infrared spectrum of DSM- d_6 in the region $1200-600$ cm⁻¹.
Partial pressure of sample 3.8 Torr in a 12 cm cell Partial pressure of sample, 3.8 Torr in a 12 cm cell.

the H^aSiC and H^sSiC angles, is small or negligible at all levels. A further illustration of this equivalence is provided by the unscaled *ν*^{is}SiH values calculated in the species (SiHD₂)(SiD₃)-CH2, shown in Table 4. There is a modest sensitivity to basis set, but in all cases the SiH^s bond is only marginally weaker than the SiH^a one.

By contrast the associated intensities are markedly different. The MP2 and B3LYP calculations agree well in predicting a fall of $22-25\%$ in the IR intensity *A* from SiH^s to SiH^a. There is a similar fall in the Raman scattering activity from the two available calculations, at the HF level.

Table 2 includes the unscaled vibrational data produced by the HF and B3LYP calculations with the *tz* basis, the observed frequencies, and the frequency fit obtained after scaling the HF/ *tz*, MP2/*tz*, and B3LYP/*tz* force fields. Only those frequencies considered to be reliable were used in the scaling of the force fields, as detailed below.

Also given in Table 2 are the frequency shifts resulting from ¹³C substitution in d_0 calculated from the scaled B3LYP/tz force field.

Assignments. Attention is focused on aspects of the spectra unclear in the previous work.

(1) ν *CH Region*. The type C bands at 2946.2 (d_0) and 2944 cm^{-1} (d_6) identify v_{22} (b_2) in the two isotopomers, whereas the type B band at 2907 cm⁻¹ due to $v_1(a_1)$ is well defined in d_0 , less well so in d_6 . The upper level of the latter is likely to be in weak Fermi resonance with $2 \times \nu_4$ at 2822 cm⁻¹, but the extent of the resonance shift appears to be small, perhaps $6-7$ cm⁻¹, judging by the results in Table 2 of scaling both CH stretching symmetry force constants to the value of *ν*₂₂.

(2) νSiH Region. The large mass of the silicon atom, as compared to that of carbon, means that the composition of stretching modes of the silyl group is the consequence of a delicate balance between opposing G and F matrix terms, such that in-phase stretches may lie above or below out-of-phase ones, depending on the exact size of the valence interaction force constants. (The terms "symmetric" and "antisymmetric" are usually inappropriate in such circumstances). In addition the coupling is markedly affected by small differences in the diagonal valence constants. It is therefore not surprising to find markedly different descriptions of the unscaled frequencies given in Table 5 from the different calculations. In part these differences stem from the differences in *ν*isSiH value seen in Table 4.

The IR spectrum of d_0 (Figure 4, ref 2) shows clearly identifiable type A and type C bands at 2164.1 and 2168.1 cm-¹ which must arise from B_1 and B_2 modes, respectively. One of two subsidiary Q branches at 2162.5 and 2167.0 cm^{-1} may represent the other B_1 band expected here or else both of these may be hot bands. In d_6 , a single A/C hybrid band at 2164.6 cm^{-1} has the appearance expected from a solitary SiH^a bond in d_5 impurity, but the absence of any sign of a second Q branch suggests that the *ν*isSiHs and *ν*isSiHa bands in fact coincide. The B_2 fundamental at 2168.1 cm⁻¹ has to be ν_{23} . Because its normal coordinate involves the stretching of the SiH^a bonds only, this frequency provides an unambiguous datum for scaling the four symmetry force constants involving SiH^a stretching. However, the B₁ 2164.1 cm⁻¹ band may represent either v_{15} or v_{16} , each of which should involve stretching of both SiH^a and SiH^s bonds. It proved possible to scale the SiH^s stretching symmetry force constants using this second datum but on a different basis for the HF as compared with the MP2 and B3LYP treatments. The HF data only allowed refinement when the B_1 band was taken to be the lower mode, *ν*16. By contrast, the MP2 and B3LYP treatments required the B_1 band to be ν_{15} . Table 5 shows the resulting predictions for the unassigned *ν*SiH bands, the associated scaled *ν*isSiH frequencies, and all six of the scaled ν SiD frequencies in d_6 . The latter have all been multiplied by the factor 1.01 to offset the differing effects of anharmonicity on SiH and SiD stretching frequencies, for a more meaningful comparison with the observed spectrum.27,28

A clear choice between the alternative assignments for the 2164.1 cm^{-1} band can then be made by examining the associated

TABLE 3: Comparison of Observed and Calculated Structural Parameters for DSM

calculated*^b*

^a Bond lengths are in angstroms, angles are in degrees, and dipole moments *µ* are in Debye. Atom Hs lies in the skeletal plane of the molecule, and H^a lies above or below this plane. ^b This work. ^c Reference 3. ^d Reference 4. ^{*e*} Assumed. *f* C_{3v} local symmetry assumed.

TABLE 4: Unscaled QC Frequencies (cm-**1) and Intensities for** *ν***isSiHs and** *ν***isSiHa Bands in DSM**

		SiH ^s		SiH ^a		
level/basis	ν ¹⁸	$A(R)^a$	ν ¹⁸	$A(R)^a$	$\Delta \nu$ ^{is} (H ^s -H ^a)	$\Delta A(R)$ (H ^s -H ^a)
HF/sv	2367.5	227.8(141.3)	2368.8	186.2(95.8)	-1.3	41.6(45.5)
HF/tz	2321.9	176.4(176.3)	2322.5	142.3(120.7)	-0.6	34.1(55.6)
MP2/sv	2304.8	172.0	2309.8	132.1	-5.0	39.9
MP2/tz	2308.7	142.4	2310.0	110.3	-1.3	32.1
$MP2/tz+$	2306.2	144.2	2307.0	111.0	-0.8	33.2
$B3LYP$ /sv	2238.6	169.4	2246.0	129.6	-7.4	39.8
B3LYP/tz	2221.2	136.9	2226.2	103.7	-5.0	33.2
$B3LYP/tz+$	2220.3	139.7	2225.1	105.0	-4.8	34.7
B3LYP/cc	2213.3	133.3	2218.6	98.1	-5.3	35.2 \sim

a Where available from Gaussian 98. *A* is the infrared intensity in km mol⁻¹, and *R* is the Raman scattering activity in \AA^4 amu⁻¹.

TABLE 5: QC-Based SiH and SiD Stretching Frequencies (cm⁻¹) after Scaling to v_{15} or v_{16} (b₁) = 2164.1, v_{23} (b₂) = 2168.1 **cm**-**¹**

					A. DSM- d_0								
level/basis	$v_2(a_1)$	$v_3(a_1)$	$v_{10}(a_2)$	ν_{16} or $\nu_{15}(b_1)^a$	ν ^{is} SiH ^s	ν ^{is} SiH ^a	Δv ^{is} (H ^s -H ^a)						
HF/sv	2185.4	2171.7	2159.5	2178.1	2179.2	2166.8	12.4						
HF/tz	2184.6	2168.3	2160.7	2175.9	2172.6	2168.8	3.8						
MP2/sv	2167.1	2162.5	2161.3	2154.2	2164.0	2162.3	1.7						
MP2/tz	2168.5	2166.5	2162.1	2160.8	2165.1	2164.7	0.4						
$MP2/tz+$	2169.2	2166.5	2162.1	2161.5	2165.3	2165.0	0.3						
$B3LYP$ /sv	2166.9	2164.3	2160.7	2154.7	2164.5	2162.3	2.2						
B3LYP/tz	2167.7	2166.2	2161.9	2159.2	2164.9	2164.1	0.8						
$B3LYP/tz+$	2168.0	2166.3	2161.8	2159.5	2164.9	2164.2	0.7						
B3LYP/cc	2166.7	2166.2	2161.7	2158.1	2164.6	2163.7	0.9						
obs					?	2164.6	$\overline{\cdot}$						
				B. DSM- d_6 (scaled frequencies \times 1.01)									
level/basis	$v_2(a_1)$		$v_3(a_1)$	$v_{10}(a_2)$	$v_{15}(b_1)$	$v_{16}(b_1)$	$v_{23}(b_2)$						
HF/sv	1587.5		1569.4	1575.3	1584.3	1559.8	1582.3						
HF/tz	1582.6		1570.8	1576.1	1579.7	1562.4	1582.2						
MP2/sv	1582.1		1555.4	1577.2	1579.3	1548.1	1582.8						
MP2/tz	1581.6		1559.2	1577.5	1579.1	1552.6	1582.5						
$MP2/tz+$	1581.5		1559.7	1577.4	1579.0	1553.1	1582.5						
$B3LYP$ /sv	1581.8		1557.1	1576.6	1579.0	1548.9	1582.7						
B3LYP/tz	1581.2		1559.0	1577.3	1579.0	1551.7	1582.4						
$B3LYP/tz+$	1581.2		1559.2	1577.2	1579.0	1551.9	1582.4						
B3LYP/cc	1581.2		1558.2	1577.2	1578.9	1551.0	1582.4						
obs	$\overline{\mathcal{L}}$		$\overline{\cdot}$	$\overline{\cdot}$	1580.2	1556.8	1583.2						

 $a \nu_{15}$ for the HF calculations, ν_{16} for all of the others.

*ν*isSiH values. The MP2 and B3LYP scaled *ν*isSiH values lie close to each other and also to the Q branch observed at 2164.6 cm-1, whereas both HF *ν*isSiH values lie unacceptably high, at 2172.6 (H^s) and 2168.8 (H^a) (*tz* basis) or 2178.1 and 2166.8 cm^{-1} (*sv* basis).

A further argument for assigning the 2164.1 cm⁻¹ Q branch to *ν*¹⁵ comes from using a bond dipole model to predict the relative intensities of the two modes. Both MP2 and B3LYP approaches indicate that v_{15} is primarily a motion of the two SiH^s bonds, whereas $ν_{16}$ derives largely from $ν_s$ SiH^a₂ stretching

TABLE 6: Comparison of Calculated and Observed Torsional Frequencies in Some Compounds Containing Silyl Groups*^a*

		DSM	DS	MS	FMS	CMS	ETS	
level/basis	$v_{14}(a_2)$	$v_{27}(b_2)$	$v_{12}(a_{1u})$	$v_6(a_2)$	$v_{18}(a'')$	$v_{18}(a'')$	$v_{27}(a'')$	$v_{26}(a'')$
HF/sv	69.1	144.0	133.0	193.6	157.6	157.2	137.9	235.2
HF/tz	75.7	142.6	134.7	197.9	154.5	158.9	141.2	234.3
MP2/sv	47.8	145.7	140.4	201.7	169.4	161.8	142.1	235.8
MP2/tz	56.7	139.0	147.6	196.0	158.1	159.9	136.0	232.4
$MP2/tz+$	44.9	141.1	146.6	199.6	148.9	156.1	135.3	225.0
$B3LYP$ /sv	73.1	136.1	129.7	192.3	161.3	153.7	137.3	226.0
B3LYP/tz	81.0	136.3	135.4	193.9	154.4	155.5	139.9	220.8
$B3LYP/tz+$	77.9	135.2	131.5	192.4	151.0	154.0	139.0	220.0
B3LYP/cc	85.8	139.6						
obs	78.5	133	125.5^{b}	190^c	149 ^d	161 ^e	132 ^f	207 ^{f,g}

a DSM = disilylmethane; DS = disilane; MS = methylsilane; FMS = FCH₂SiH₃; CMS = ClCH₂SiH₃; ETS = ethylsilane, where $\nu_{27} = \tau S$ iH₃, $\nu_{26} = \tau CH_3$. This work, except where otherwise indicated. *b* Reference 29. This value is exactly one-half of the frequency of the $2 \leftarrow 0$ transition quoted earlier,³⁰ which is surprising for a strongly anharmonic situation. *c* Reference 31. *d* Reference 32. *e* Reference 33. *f* Reference 34. *g* Value from solid-state spectra.

motion. The bond dipole model then predicts that v_{15} will produce a more intense band than v_{16} . In the latter, the change of moment in each silyl group lies almost exactly along the *C*² symmetry axis, with only a small component along the *A* inertial axis. The resultant dipole change is therefore small. The SiHs bond, by contrast, is directed within 33° of this axis, so that the resultant dipole change is large. We conclude that *ν*¹⁶ lies below 2164.1 cm^{-1} , the best predictions being those from the MP2/tz and MP2/ tz + calculations, 2160.8 and 2161.5 cm⁻¹, respectively. These are sufficiently close to the Q branch at 2962.5 cm-¹ for the latter to be a strong candidate for *ν*16. As such, it provides a reasonable interpretation of the torsional combination and difference bands, as detailed below. At all events, it can safely be concluded that the fourth Q branch at 2167.0 cm^{-1} derives from a hot band.

Assignment of the combination and difference bands involving $v\sinh$ and $\tau\sinh$ ₃ levels listed in Table 1 must now be considered. These bands are seen as very weak sharp shoulders on each side of the intense *ν*SiH absorption in Figure 2, which indicate the presence of type C bands. Table 6 lists the unscaled predictions for the torsional frequencies v_{14} and v_{27} , together with similar predictions for some other molecules containing silyl groups. In DSM, there is quite good agreement among the nine calculations on the magnitude of v_{27} , which is predicted in the range $135-146$ cm⁻¹, but a surprising variation in the value of v_{14} , which ranges from 45 to 86 cm⁻¹. A similar variability in the prediction of silyl torsional frequencies is found in trisilylmethane.7 By contrast, calculated values of torsional frequencies in disilane (DS), methylsilane (MS), fluoro- and chloromethylsilanes (FMS, CMS), and ethylsilane (ETS), also listed in Table 6, show in general good agreement both between levels of theory and also between theory and experiment.29-³⁴

Focusing on the spectrum of DSM, it is clear that the pairs of Q branches 2242.8, 2240.8, and 2087, 2084 cm⁻¹ must be associated with the lower torsional frequency $v_{14}(a_2)$. Type C bands will result when a quantum of this vibration is associated with a *ν*SiH one of B₁ symmetry. Of the two B₁ candidates, $ν_{15}$ and v_{16} , v_{16} is largely an asymmetric stretch of the SiH^a bonds and, as such, seems more likely to interact with torsional motion than the stretching of SiHs . Assuming *ν*¹⁶ to be responsible for the Q branch at 2162.5 cm^{-1} , the tentative assignment is made (Table 1) of 2242.8 and 2084 cm^{-1} , respectively, to the combination and difference bands involving v_{16} with v_{14} . This places v_{14} at 78.5 cm⁻¹. The other Q branches are then assigned to hot bands in v_{14} and v_{27} , rather than to levels involving v_{15} .

Type C combination and difference bands arising from *ν*₂₇(b₂) can arise from either of the A₁ modes ν_2 and ν_3 , the best predictions for which are about 2169 and 2166.5 cm^{-1} (Table 5). The closest match in appearance lies between the Q branches at 2298 and 2033.7 cm^{-1} , which together fit quite well as combination and differences between ν_3 at about 2166.5 and v_{27} at about 133 cm⁻¹. The remaining difference bands at 2040 and 2029 cm-¹ and the very doubtful combinations at 2303 and 2291 cm⁻¹ are tentatively assigned as hot bands, as in the discussion of *ν*₁₄ above.

This analysis depends markedly on the reliability of the QC calculations of values of the sretch-stretch interaction force constants *f* ′ connecting SiH bonds in the same and in differing silyl groups. This is hard to assess. A final judgment on the assignments of these Q branches must await a more careful investigation at higher resolution in a spectrum which is free of the interference from atmospheric $CO₂$ absorption visible in Figure 2. However, the present analysis and prior examples such as that of ethylsilane³⁴ together suggest that torsional modes should normally be identifiable from their combination and difference bands involving SiH stretching quanta. The problem is to discover which *ν*SiH fundamentals are involved in each transition.

(3) νSiD Region. The change here in the relative masses of the silicon and terminal atoms means that in contrast to the SiH region there is a well-defined distinction between higher-lying v_{as} SiD₃ and lower-lying v_s SiD₃ modes. The Q branch of the B₂ v_{as} SiD₃ band v_{23} is plain at 1583.2 cm⁻¹, whereas type A bands centered at 1580.2 and 1556.8 cm⁻¹ identify the B_1 bands due to v_{15} and v_{16} . As seen in Table 2, the two bands assigned to v_{15} and v_{23} are fitted to within 1 cm⁻¹ using the force constants scaled on the d_0 values of ν_{15} and ν_{23} above, plus the usual anharmonicity correction factor of 1.01.²⁷ However, *ν*₁₆ is predicted not higher than 1553 cm^{-1} (ignoring the HF results, rejected above), which is ∼4 cm-¹ lower than observed. This may reflect the need for a variation in the above anharmonicity correction.

(4) A₂ Modes. In the previous study,¹ the A₂ τ CH₂ mode ν_{11} was identified as the weak Raman line at 1101 cm^{-1} in the liquid phase, above the wCH₂ band (v_{17}) at 1058 cm⁻¹. However, all of the QC unscaled calculations place v_{11} substantially below ν_{17} , making the above assignment untenable. A plausible position for v_{11} is the hitherto unassigned sharp feature at 1048.3 cm^{-1} in the solid solution spectrum,¹ its infrared activity being ascribed to a lowering of the symmetry in the crystal phase. The assignment of the weak sharp band in the solid spectrum just below at 1038.4 cm^{-1} as a 13 C counterpart of v_{17} is confirmed by the calculated ¹³C frequency shifts in Table 2.

Assigning v_{11} near 1048 cm⁻¹ would be likely to render the band indistinguishable from v_{17} in the liquid-phase Raman spectrum.

TABLE 7: Scale Factors for QC-Based Force Fields for DSM

^a As defined in Reference 11, with the interchange of silicon and carbon atoms. *b,c* Constrained equal. *^d* Excluding *τ*SiH3. *^e* Sample standard deviation. f Sum of weighted squares of errors for d_0 and d_6 frequencies.

Apart from a dubious feature at 1113 cm^{-1} in the Raman spectrum of d_6 , whose previous asignment to v_{11} is untenable, the only other Raman feature previously assigned to an A_2 mode is the weak band in d_6 at 417 cm⁻¹. This fits well with the prediction of \sim 426 cm⁻¹ for the ρ SiD₃ mode ν_{13} which results from the assumption of equal scale factors for the ρ SiD₃ constants in the A_2 and B_2 species. An independent datum for the scaling of the A_2 ρ SiD₃ constant is therefore available.

(5) Silyl Deformations. In d_0 , these are predicted to lie close to each other near 950 cm⁻¹, except for the very intense B_1 band due to the δ_s SiH₃ mode ν_{19} , expected some 30 cm⁻¹ lower. The latter may be confidently assigned to the very strong band with a minimum at 921.5 cm⁻¹ and likely center near the maximum at 918.4 cm⁻¹, although no discrete Q branch was discernible. Q branches at 952.8 and 947.9 cm^{-1} are likely to arise from $v_{24}(b_2)$ and $v_{18}(b_1)$, but the contours here are confusing, as Figure 3 shows. In the d_6 spectrum, Figure 4, apart from the well-defined type A band at 765.5 cm⁻¹ due to $v_{18}(b_1, b_2)$ $v_{\text{as}}\text{CSi}_2$), the contours are also confusing. The A₁ $\delta_{\text{s}}\text{SiD}_3$ mode is displaced above all of the δ_{as} SiD₃ modes by coupling to ν_s - $CSi₂$ motion and must contribute most of the IR intensity to the band with a maximum at 733 cm⁻¹. However, $v_{24}(b_2)$, primarily a ρ CH₂ mode, is also expected here. A likely A type band has a Q branch at 670.4 cm^{-1} , and this is confidently assigned to $v_{20}(b_1)$. A weak Q branch at 687.8 cm⁻¹ may arise from $v_{19}(b_1)$ or $v_{25}(b_2)$. With only two reliable δ SiH₃ or δ SiD₃ assignments available, both associated with symmetric deformation, in the subsequent scaling of the force fields, it was necessary to assume identical scale factors for the δ_{as} and δ_{s} force constants.

(6) Silyl Rocking Motions. Figure 3 illustrates the imperfect nature of the d_0 contour associated with the $B_2(\nu_{25})$ band at 812 cm⁻¹. The QC calculations make it clear that the A₁ mode ν_7 cannot be at 595 cm^{-1} as previously assigned from the Raman spectrum but must instead underlie the distorted type A band at 763.4 cm⁻¹ due to v_{20} (b₁, v_{as} CSi₂). In d_6 , there is slight evidence to suggest that the center of v_{21} is a little lower than the Raman liquid values of 438 cm⁻¹. The band at 390 cm⁻¹ previously assigned¹ to ν_{26} was not rerecorded.

(7) Fermi Resonance Correction. A small resonance is likely in the spectrum of d_6 between v_4 at 1363 cm⁻¹ and the combination 738(a₁) + 664(a₁), expected at 1402 cm⁻¹, but observed at 1407 cm⁻¹. Correcting v_4 to 1368 cm⁻¹ (Table 2) then increases the compatibility between the scaled d_0 and d_6 values of *ν*4.

Force Field Scaling. The symmetry coordinates employed were identical to those employed for dimethylsilane, with exchange of silicon for carbon.¹¹ It was found possible to define 14 independent scale factors, as listed in Table 7 for each of the levels and basis sets employed. A high correlation coefficient of 0.92 was found between the factors for the strongly coupled ρ CH₂ and ρ SiH₃ motions in the B₂ symmetry species, and it was thought advisable to constrain these to be equal. There were slight indications in the case of the HF and MP2 force fields that this assumption was imperfect. However, with or without this constraint, the fit to the d_0 and d_6 B₂ frequencies remained poor, the cause of which was identified as incorrect values of the relatively large interaction force constant $F_{25,27}$ in all of the calculations. This was remedied by a manual adjustment of this constant prior to scaling, reducing it by between 8 and 15% according to the level of theory.

The 14 scale factors included independent ones for the two kinds of SiH bond stretching force constants, three such for silyl rocking constants, and two for the two silyl torsions. The A_2 and $B_2 \rho''$ SiH₃ factors would diverge further from each other if, as seems likely, the gas-phase value of v_{13} (a₂) in d_6 lies below the liquid value of 417 cm^{-1} .

Table 7 includes for each set of factors for a particular calculation the average value, the standard deviation, and the range. However, the two torsional factors were excluded from these comparisons for two reasons. The first reason was the present uncertainty in the assignment of the torsional frequencies, and the second, the highly anomalous values which these factors exhibit in most cases. The size of the factor for the A_2 constant from the MP2/*tz*+ calculation, 3.11, is unprecedented in our experience. Anomalies such as these will remain even if a drastic revision in the frequencies becomes necessary. Thus, were an MP2 calculation to give more reasonable torsional factors as the result of reassignment, the B3LYP ones would then become unacceptable.

Without the torsions, the ranges and standard deviations for the various calculations are similar, with a slight preference for

TABLE 8: Potential Energy Distributions (PED) in DSM-*d***⁰ and -***d***⁶ from the Scaled B3LYP/***tz* **Calculation**

			$DMS-d_0$		$DMS-d_6$			
		$\nu_{\rm calc}$	P. E. D^a	main component	$\nu_{\rm calc}$	P. E. D^a	main component	
A ₁	v_1	2903.7	$101S_4$	v_sCH_2	2903.7	$101S_4$	v_sCH_2	
	v_2	2167.7	$17S_1$, $82S_2$	ν SiH ^a	1565.5	$67S_1$, $35S_2$	$v_{\rm as}$ SiD ₃	
	v_3	2166.2	$83S_1$, $18S_2$	$v_{s}SiH^{s}$	1543.5	$34S_1, 65S_2$	v_s SiD ₃	
	v_4	1368.7	103S ₉	$\delta_{\rm s}CH_2$	1368.3	103S ₉	$\delta_{\rm s}CH_2$	
	v ₅	959.8	$11S_5$, 89 S_6	δ_s SiH ₃	737.6	$35S_3, 61S_6$	δ _s SiD ₃	
	v_6	955.5	$91S_5$, $11S_6$	δ_{as} SiH ₃	689.3	$93S_5$	δ_{as} SiD ₃	
	v ₇	778.6	$19S_3, 66S_7$	ρ' SiH ₃	662.1	$17S_3$, $9S_5$, $33S_6$, $34S_7$	coupled	
	ν_8	548.9	$78S_3$, $23S_7$	v_sCSi_2	467.8	$48S_3, 46S_7$	coupled	
	v_9	175.4	$19S_7, 108S_8$	δ CSi ₂	156.2	$25S_7, 106S_8$	δ CS ₁₂	
A ₂	v_{10}	2161.9	$100S_{10}$	ν SiH ^a ₂	1561.7	$101S_{10}$	ν SiD ^a ₂	
	v_{11}	1048.3	$85S_{14}$	τ CH ₂	1031.2	$92S_{14}$	τ CH ₂	
	v_{12}	953.1	$101S_{11}$	$\delta_{as}SiH_3$	683.7	$102S_{11}$	δ_{as} SiD ₃	
	v_{13}	555.1	$106S_{12}$, $26S_{14}$	ρ "SiH ₃	417.0	$110S_{12}$, $19S_{16}$	ρ "SiD ₃	
	v_{14}	78.5	$104S_{13}$	τ SiH ₃	55.8	$104S_{13}$	τ SiD ₃	
B_1	v_{15}	2164.1	$90S_{15}$, $11S_{16}$	ν SiH ^s	1563.3	$72S_{15}$, $30S_{16}$	v_{as} SiD ₃	
	v_{16}	2159.2	$10S_{15}$, 89 S_{16}	ν SiH ^a	1536.4	$29S_{15}$, $70S_{16}$	ν _s SiD ₃	
	v_{17}	1057.8	$23S_{17}$, $80S_{21}$	WCH ₂	1043.4	$24S_{17}$, 87 S_{21}	WCH ₂	
	v_{18}	950.1	$100S_{18}$	$\delta_{as}SiH_3$	767.8	$78S_{17}$, $11S_{21}$	$v_{\rm as}$ CSi ₂	
	v_{19}	916.8	$101S_{19}$	$\delta_{\rm s}$ SiH ₃	682.6	$99S_{18}$	δ_{as} SiD ₃	
	v_{20}	766.4	$79S_{17}$, $11S_{21}$	$\nu_{\rm as}$ CSi ₂	666.7	$92S_{19}$	δ _s SiD ₃	
	v_{21}	579.8	$97S_{20}$, $18S_{21}$	ρ' SiH ₃	434.8	$102S_{20}$, $13S_{21}$	$\rho' \text{SiD}_3$	
B_2	v_{22}	2946.2	$101S_{23}$	$v_{\rm as}CH_2$	2946.1	$101S_{23}$	$v_{as}CH_2$	
	v_{23}	2168.1	$100S_{22}$	ν SiH ^a	1566.8	$100S_{22}$	ν SiD ^a ₂	
	v_{24}	958.0	$102S_{24}$	$\delta_{as}SiH_3$	736.4	$24S_{25}$, $50S_{27}$	ρ CH ₂	
	v_{25}	810.7	$42S_{25}$, $33S_{27}$	ρ "SiH ₃	686.9	$101S_{24}$	δ_{as} SiD ₃	
	v_{26}	463.7	$65S_{25}$, 74 S_{27}	ρ CH ₂	390.0	$82S_{25}$, 55 S_{27}	ρ "SiD ₃	
	v_{27}	133.0	$93S_{26}$	τ SiH ₃	104.0	$92S_{26}$	τ SiD ₃	

a Diagonal terms \geq 9%.

the B3LYP force fields. The sums of weighted squares of errors for the fit to the frequency data are also similar, reflecting the comparable fit data of Table 2. The apparent slight superiority of the fit with the HF calculations is offset by their failure outlined above in the prediction of *ν*SiH values.

The need for a range of independent scale factors is present in all cases, even for motions as apparently similar as the two kinds of SiH stretching or the three involving silyl rocking. Substantial differences are also needed in the factors for the deformation, rocking, and wagging constants associated with the $CH₂$ group. In all of the treatments, the minimum value for any factor is associated with the $CH₂$ wag, wCH₂, which means that all are tending to calculate this frequency too high.

For a quantitative description of the vibrational modes in terms of the potential energy distributions (PED), results of the B3LYP/*tz* calculation are cited in Table 8. The scaled B3LYP/ *tz* force field is given in Table 3S of the Supporting Information. The MP2/*tz*-based PED has slightly more coupling between wCH₂ and ν_{as} CSi₂ motion in the B₁ modes ν_{17} and ν_{20} (*d*₀) and their counterparts v_{17} and v_{18} in d_6 and also between δ_s SiD₃ and δ_{as} SiD₃ in d_6 .

Prediction of Unknown or Uncertain Frequencies. Apart from the variations in prediction of *ν*SiH and silyl torsional frequencies discussed above, the MP2/*tz* calculation is notable for a low value of ν_1 and a high value of ν_7 (d_0). The former is associated with a lower valence interaction force constant involving the stretchings of two CH bonds sharing the same carbon atom (see further below).The B3LYP or HF predictions are preferred because they yield more acceptable Fermi resonance shifts on *ν*1.

Elsewhere, there is broad agreement between the predictions in Table 2 from the three levels of theory for the silyl deformations ν_5 , ν_6 , and ν_{12} . It is interesting to note the inversion of the symmetric and antisymmetric deformations δ_s SiH₃ and δ _{as}SiH₃ that occurs in *d*₀ from the A₁ to the B₁ species, which is enhanced in d_6 due to the greater coupling there between the δ _sSiD₃ and ν _sSiC₂ motions.

Prediction of the position of the silyl rocking mode ν_7 in d_0 is more variable, the MP2 value of about 786 cm^{-1} lying 7 cm^{-1} higher than the B3LYP value of 778 cm^{-1} . The value of ∼555 cm-¹ predicted for the A2 silyl rock *ν*¹³ is based on the liquid-phase Raman frequency seen in d_6 and may differ significantly from its actual position in the gas. Variations due to basis set are negligible in all of these predictions. The main source of error will lie in the constraints applied to the scale factors.

Unscaled Valence Stretching Interaction Force Constants. The full set of unscaled valence stretching force constants is given in Table 4S. Quoted here in Table 9 are the interaction constants f' which fall into three categories, α or gem constants, involving a central atom common to both bonds, β constants which relate bonds either gauche or trans (antiperiplanar) with respect to each other, and longer range *γ* constants between bonds separated by a motionless atom, in this case the SiH bonds in different silyl groups.

Among the α constants, there is a contrast between a reasonable agreement between MP2 and B3LYP values for SiH bonds, with the marked fall in value for the CH bond when an MP2 value is compared with an HF or B3LYP one. This is the basis for the larger $v_{as}CH_2/v_sCH_2$ splitting predicted from an MP2 calculation, which in turn yields an improbable Fermi resonance correction on ν_s CH₂. The same trend is found with the CH/CSi and CSi/CSi interactions, but not the SiH/SiH one.

This characteristic of MP2 calculations has been noted previously on a number of occasions.11,17,18,23,26 It seemed profitable to explore how widespread the phenomenon might be. Table 10 shows a comparison of gem *f* ′ values from B3LYP and MP2 calculations for 17 molecules containing $CH₃$ and/or $CH₂$ groups. In almost every case the MP2 value is substantially below the B3LYP one. The apparent universality of this effect

TABLE 9: Unscaled Valence Stretching Interaction Force Constants *f*′ **(aJ Å**-**2) in DSM**

a a', a1', and a2' denote hydrogen atoms in SiH bonds in the second silyl group; a1 and a1' lie on the same side of the skeletal plane, and a1 and a2' lie on opposite sides. ${}^b f_{\beta g}$ ' involves SiH and SiC or CH bonds oriented gauche toward each other; $f_{\beta t}$ ' involves similar bonds trans (antiperiplanar) related.

TABLE 10: Comparison of Unscaled CH Stretch/CH Stretch Interaction Constants *f*′**(aJ Å**-**2) from B3LYP and MP2 Treatments***^a*

molecule	basis	group ^b	B3LYP	MP ₂
CH_4	tz.		0.036	0.013
CH ₃ F	tz.	CH ₃	0.054	0.033
CH ₃ Cl	tz.	CH ₃	0.037	0.022
C_2H_6	tz.	CH ₃	0.051	0.024
CH ₃ CH ₂ F	tz.	CH ₃	0.040, 0.044	0.013, 0.018
		CH ₂	0.065	0.042
$CH_3CH_2CH_3$	$tz +$	CH ₃	0.048, 0.049	0.021, 0.022
		CH ₂	0.064	0.033
$CH3CH2SiH3$	$t\overline{z}$	CH ₃	0.052, 0.048	0.026, 0.022
		CH ₂	0.048	0.021
CH_3SiH_3	tz.	CH ₃	0.035	0.011
$(CH_3)_2SH_2$	s v	CH ₃	0.033, 0.033	0.012, 0.013
$(CH_3)_3SH$	$s\upsilon$	CH ₃	0.033, 0.033	0.012, 0.012
CH ₂ F ₂	tz.	CH ₂	0.059	0.042
CH_2Cl_2	tz.	CH ₂	0.039	0.034
$CH2(SiH3)2$	$t\overline{z}$	CH ₂	0.033	0.009
FCH ₂ SiH ₃	$t\overline{z}$	CH ₂	0.052	0.031
ClCH ₂ SiH ₃	tz.	CH ₂	0.039	0.023
$CH2=CH2$	$tz +$	CH ₂	0.034c	0.013
$CH2=CF2$	$tz+$	CH ₂	0.014	-0.007

^a This work. *^b* In asymmetric methyl groups, the two *f* ′ values are listed in the order CHa/CHs and CHa/CHa. *^c* CCSD(T)/cc-pVTZ calculations yield a slightly lower value (0.028 aJ \AA^{-2}),³⁵ whereas local mode studies place it higher (\sim 0.044 aJ Å⁻²).³⁶

suggests that MP2 calculations should be avoided if a close analysis of CH_3 or CH_2 stretching vibrations is to be attempted, a fact not commonly appreciated.

The β interactions fall into two categories. Those between the CH and SiH bonds exhibit the familiar pattern of positive gauche interactions and smaller, negative trans ones.18 The two gauche interactions are both smaller from an MP2 calculation.

However, the β interactions between SiC and SiH bonds are negative and very similar in size for both gauche- and transrelated pairs, with little variation with level and basis set.

The largest of the *γ* interactions, f_{alal} ^{*'*}, operates between SiH bonds roughly parallel to each other in the two silyl groups and is larger than either of the β constants. The other γ constants are either comparable in size to the β ones or much smaller. However, all are calculated consistently positive throughout. Further light on the way these *γ* constants vary with level of theory and basis set and also with bond orientation is thrown by the calculations of dipole/dipole interactions described below.

Centrifugal Distortion Constants. Results for these from MP2 and B3LYP calculations are listed in Table 11. They show a significant dependence on both basis and level of theory, but may be found of use in guiding future microwave investigations. No experimental values are currently available.

Electrical Properties of the SiH Bonds. Table 12 lists the atomic polar tensors for DSM from the B3LYP/*tz* calculation. For the silicon and hydrogen atoms, the reference axes have been rotated so that the *z* axis lies along the direction of the bond containing the atom. In such a case, in the zero order optical parameter approximation, $\frac{\partial p_x}{\partial x}$ (= p_{xx}) and $\frac{\partial p_y}{\partial y}$ (= p*yy*), which relate to bending of the bond concerned, would both equal μ/r , where μ is the bond dipole moment and *r* the bond length, whereas $\partial p_z / \partial z$ (= p_{zz}) would be the dipole derivative with respect to stretching of the bond, *∂µ*/*∂r*. ³⁷ Nonzero values of *∂*p*x*/*∂z* and *∂*p*y*/*∂z* indicate the extent to which the vector *∂µ*/ *∂r* in fact is directed away from the bond direction. As usual

TABLE 12: Atomic Polar Tensors (e) in DSM, from B3LYP/*tz* **Calculations**

		$C(1)^{a,b}$			$Si(2)^{c,d}$				
	∂x	∂y	∂z	∂x	∂y	∂z	∂x	∂y	∂z
∂p_x	-0.880	0.000	0.000	0.909	0.000	0.186	0.041	0.000	0.000
∂p_y	0.000	-0.412	0.000	0.000	0.953	0.000	0.000	0.058	-0.013
∂p_z	0.000 H ^s (6) ^{c,d}	0.000	-0.484	-0.035 $H^a(7)^{c,e}$	0.000	1.429	0.000	-0.041	-0.088
∂p_x	-0.295	0.000	0.052	-0.216	0.011	0.004			
∂p_y	0.000	-0.212	0.000	0.020	-0.280	-0.046			
∂p_z	0.077	0.000	-0.349	-0.003	-0.002	-0.296			

a z axis along C_2 symmetry axis. *b x* axis in the CSi₂ plane. *c z* axis along the C-Si(2), C-H(4), Si-H(6), or Si-H(7) bond. *d y* axis in the CH₂ plane. *^e x* and *y* axes in arbitrary orientation.

TABLE 13: Infrared Stretching Intensities and Electrical Properties Associated with Some SiH Bonds, from B3LYP/*tz* **Calculations***^a*

	SiH_4^b	MeSiH ₃	$Me2SiH2b$	Me ₃ SiH ^b	Si ₂ H ₆	DSM(H _s)	$DSM(H^a)$	DSM average	SiH_2Cl_2
A_{av}^c obs	$70(1)^d$	$102(5)^e$	$127(6)^e$		$72(4)^e$			$90(3)^e$	$67(2)^e$
A_{av}^c calc	91.0	120.0	144.9	167.3	89.8			114.8	74.8
A^{is}	94.9	122.7	146.3	167.3	93.0	136.9	103.7		75.7
$\partial \mu / \partial r^g$	-0.287	-0.329	-0.361	-0.387	-0.293	-0.353	-0.300		-0.242
θ^h	0.0	4.8	4.1	0.0	9.1	8.4	0.8, 8.8		7.2
q_H	-0.126	-0.139	-0.150	-0.154	-0.114	-0.133	-0.132		-0.088
	0.248	0.280	0.308	0.330	0.262	0.296	0.268		0.272
$\frac{\chi_{\textrm{H}}^{\prime}}{\bar{p}_{\textrm{H}}^{\prime}}$	-0.246	-0.277	-0.304	-0.327	-0.253	-0.285	-0.264		-0.269
\bar{p}_{\perp}	-0.226	-0.251	-0.276	-0.298	-0.235	-0.254	-0.248		-0.284

a This work, except where otherwise indicated. *b* Reference 11, with transcription errors corrected. *c* Average infrared intensity (km mol⁻¹) observed (obs) or calculated (calc) in the *ν*SiH region of the parent molecule, divided by the number of SiH bonds. *^d* Reference 38. *^e* Reference 12. *^f* QCcalculated infrared intensity (km mol⁻¹) for the ν^{is} SiH band. *g* Dipole derivative (e) for stretching of the Si-H bond. *h* Angle(s) (degrees) made ϕ ^{*x*} between *∂µ/∂r* and the Si-H bond direction. *i* Mulliken charge on H. *j* King atomic charge, = Tr{**P**[′]A^{*·*}**P**_A}/3.³⁹ *k* Mean atomic charge, = (p_{*xx*} + p_{*yy}*</sub> $+$ p_{zz})/3.³⁹ ^{*l*} Mean bending charge, = (p_{xx} + p_{yy})/2, when the *z* axis lies along the Si-H bond.¹⁸

TABLE 14: Dipole-**Dipole Potential Energies DPE (103 e2 Å**-**1) Involved in the Simultaneous Stretchings of SiH Bonds in Neighboring Silyl Groups, from QC Calculations**

	SiH ^s /SiH ^s '	SiH ^s /SiH ^a '	$SiH^{a1}/SiH^{a1'}$	$SiH^{a1}/SiH^{a2'}$
$HF/s\nu$	1.982	-0.423	4.362	-0.728
HF/tz	1.521	-0.318	3.218	-0.544
MP2/sv	1.491	-0.326	3.265	-0.538
MP2/tz	1.233	-0.272	2.660	-0.445
$MP2/tz+$	1.248	-0.273	2.669	-0.446
$B3LYP$ /sv	1.463	-0.301	3.067	-0.507
B3LYP/tz	1.172	-0.240	2.376	-0.397
$B3LYP/tz+$	1.197	-0.245	2.413	-0.406
B3LYP/cc	1.149	-0.234	2.294	-0.382

for CH bonds, p_{xx} and p_{yy} are both positive, but p_{zz} is negative. However, the hydrogen in an SiH bond carries a negative charge in both stretching and bending motions.

Electrical properties derived from these atomic polar tensors for the SiH bonds in DSM are compared in Table 13 with those in silane, disilane, three methylsilanes, and dichlorosilane. Also included are the average SiH stretching infrared intensities, per bond, derived experimentally and from B3LYP/*tz* calculations for each parent molecule. The latter differ trivially from the computed intensities of the *ν*isSiH band, also cited. Although, as has already been noted, the computed *ν*isSiH band intensities for the SiH^s and SiH^a bonds in DSM differ markedly from each other, the computed average intensity per bond occupies a position between those for the SiH bonds in silane and methylsilane, as does the experimental average value. The rise in *ν*SiH intensity is readily attributable to the increased negative charge on the hydrogen atom expected from the inductive effect of the methyl or $CH₂$ group. However, it is interesting to note that in DSM the H^s atom is much more affected in this way than the H^a atom. The Mulliken charge q_H , also quoted in Table 13, follows the same trend from silane to methylsilane, but the difference between H^s and H^a in DSM seems less marked. An opposite type of inductive effect due to chlorine shows up in

the computed and observed intensities for the SiH bond in SiH2- $Cl₂$. By contrast, the α effect of a second silyl group, as exhibited by the disilane data, is negligible.

As expected, values of *∂µ*/*∂r* parallel very closely changes in *A*av. The direction of *∂µ*/*∂r* lies within 9° of that of the bond. Calculation of the mean bending charge¹⁸ $\bar{p}_{\perp} = (p_{xx} + p_{yy})/2$ also shows the H^s atom to carry a more negative charge in its bending motion than the H^a atom, both falling in line with variations from silane to disilane and methylsilanes. However, the same is not true in the case of SiH₂Cl₂, where \bar{p}_\perp rises markedly from silane, in contrast to the fall for *∂µ*/*∂r*.

Long-Range Effects Between Bonds in DSM. *SiH/SiH Stretching*. The QC data obtained in this work permit an exploration of a possible relation between dipole/dipole forces and the SiH sretch-stretch interactions, which were suggested earlier as being responsible for splittings of the SiH stretching modes in the parent DSM molecule.²

As described in calculations for disilylamines, $8-10$ the model employed assumes (1) that the dipole is proportional to the parameter *∂µ*/*∂r* for the bond concerned, (2) that the dipole is located at the position of the moving hydrogen atom, and (3) that the direction of the dipole is along the SiH bond. That the latter is approximately true is seen from the angle data in Table 13.

The potential energy DPE is given by the standard formula:⁴⁰

$$
DPE = \mu_i \mu_j r^{-3} (\cos \theta_i \cos \theta_j - \sin \theta_i \sin \theta_j \cos \phi_{ij})
$$

Here μ_i and μ_j are the two dipole moments concerned, equated to the values of *∂µ*/*∂r* for the two SiH bonds *i* and *j*, assuming the same unit amplitude in the stretching coordinate as for *f* ′, *r* is the distance between the two dipoles and θ_i , θ_j , and ϕ_{ij} are the angles H*i*Si*k*Si*l*, H*j*Si*l*Si*k*, and H*i*Si*k*Si*l*H*^j* (dihedral), respectively.

Table 14 lists the values of DPE obtained from the three levels of theory, for the four types of *γ* interaction possible. As might

TABLE 15: Hydrogen Interatomic Distances *R* **(Å) and Mulliken Charges** *q* **(e) at Various Levels of Theory**

Figure 5. Correlation graph of the *ν*SiH/*ν*SiH interaction force constant *f*′ (aJ Å-²) operating between SiH bonds in different silyl groups, versus the dipole/dipole interaction energy DPE (e^2 Å⁻¹), both quantities from HF, MP2, and B3LYP calculations with the *^s*V, *tz*, *tz*+, and *cc* basis sets. The line is described by the equation: 10^3 DPE = $-0.810(54)$ + 0.408(10) $10^3 f$; $R = 0.990$, $\sigma_d = 0.208$

be expected, the largest energy is associated with the SiH^{a1} and SiHa1′ bonds which lie roughly parallel to each other. The general relation of these DPE values to the QC-based *f* ′ constants is demonstrated in Figure 5, which shows a good correlation between the two quantities, independent of wave function basis or level of theory. The gradient of 0.408 is close to the value of 0.433 expected for a purely electrostatic origin for *f* ′. ¹⁰ This must be considered satisfactory in view of the three assumptions listed above. However, the influence of an additional factor is evident from the presence of a negative intercept on the DPE axis, which would be zero if only dipoledipole forces were present. Nevertheless, the suggestion is strong that the HF, MP2 and B3LYP approaches agree in finding the main source of the stretch/stretch interactions to lie in dipole/ dipole interactions.

Torsional Motion. The spread of points for a particular *f* ′ in Figure 5 is almost wholly due to the differing dipole derivatives yielded by the various calculations. It is then pertinent to enquire if differences in the calculated charge distributions on hydrogen can explain the variations in the A_2 torsional frequency seen in Table 6, particularly the low values from MP2 calculations.

Table 15 shows some of the factors which might be thought relevant. In addition to the Mulliken charges on the hydrogens in the CH bond (H_4) and SiH bonds (H_{6}^s, H_{7}^a) , the distances H_4-H_6 , H_4-H_7 , H_7-H_{10} , and H_7-H_{11} are given. (The remote hydrogens H_6 and H_9 are much further apart and considered unlikely to influence torsional motion.) Changes in the distances listed are less than 1% for H_4-H_6 and H_4-H_7 and less than 3% for H_7 - H_{10} and H_7 - H_{10} . Although there is a dip in the latter from an HF value to an MP2 one, with a consequent rise from MP2 to B3LYP, the changes are so small as to make it rather unlikely that they can produce the torsional variation seen. The Mulliken charge on H4 changes slightly with basis set but

rather less with level of theory. There is a significant fall in the negative charge on each type of hydrogen attached to silicon, H_6 and H_7 , in the sequence $HF > MP2 > B3LYP$, but this is not the order of the variation in the torsional constant. A further point is that the *A*² motion involves an in-phase rotation of both silyl groups in which changes in the repulsion between H^a atoms might be expected to be small. Repulsions between these atoms should be more noticeable in the B_2 torsional motion, but agreement on the frequency of the latter is in fact excellent between all levels of theory.

These considerations suggest that the reason for the anomalously low MP2 torsional force constant has yet to be discovered.

Acknowledgment. Professor D. W. H. Rankin and Dr S. L. Hinchley are warmly thanked for making available the Edinburgh ab initio facility for running the Gaussian 98 program. I thank also Professor H. Bürger for the data of Reference (32) and two referees for helpful comment.

Supporting Information Available: Infrared frequencies observed in $(SiH_3)_2CH_2$ (Table 1S) and $(SiD_3)_2CH_2$ (Table 2S); the B3LYP/6-311G** force field (Table 3S); unscaled valence stretching force constants for all of the QC calculations (Table 4S). This material is available free of charge via the Internet at http://pubs.acs.org.

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