# Generalized Sturmian Solutions for Many-Particle Schrödinger Equations ${ }^{\dagger}$ 

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#### Abstract

The generalized Sturmian method for obtaining solutions to the many-particle Schrödinger equation is reviewed. The method makes use of basis functions that are solutions of an approximate Schrödinger equation with a weighted zeroth-order potential. The weighting factors are especially chosen so that all the basis functions are isoenergetic with the state they are used to represent. This means that all the basis functions have turning points located in such a position that they can contribute usefully to the synthesis of the wave function. The method is illustrated by a simple example-the calculation of atomic spectra.


## Introduction

The Schrödinger equation for an $N$-particle system can be written in the form

$$
\begin{equation*}
\left[-\frac{1}{2} \Delta+V(\mathbf{x})-E_{\kappa}\right] \Psi_{\kappa}(\mathbf{x})=0 \tag{1}
\end{equation*}
$$

where

$$
\begin{gather*}
\mathbf{x}=\left\{\mathbf{x}_{1}, \mathbf{x}_{2}, \ldots, \mathbf{x}_{N}\right\}  \tag{2}\\
\mathbf{x}_{j} \equiv\left\{x_{j}, y_{j}, z_{j}\right\} \quad j=1,2, \ldots, N \tag{3}
\end{gather*}
$$

and

$$
\begin{equation*}
\Delta \equiv \sum_{j=1}^{N} \frac{1}{m_{j}} \nabla_{j}^{2} \tag{4}
\end{equation*}
$$

It is usual to build up the solutions to (1) in terms of a set of basis functions, $\Phi_{\nu}(\mathbf{x})$ :

$$
\begin{equation*}
\Psi_{\kappa}(\mathbf{x})=\sum_{\nu} \Phi_{\nu}(\mathbf{x}) C_{\nu, \kappa} \tag{5}
\end{equation*}
$$

In the generalized Sturmian method, ${ }^{1-26}$ these basis functions are chosen to be solutions to an approximate N -particle Schrödinger equation of the form

$$
\begin{equation*}
\left[-\frac{1}{2} \Delta+\beta_{\nu} V_{0}(\mathbf{x})-E_{\kappa}\right] \Phi_{\nu}(\mathbf{x})=0 \tag{6}
\end{equation*}
$$

Here $V_{0}$ resembles $V$ as closely as possible (remembering, of course, that one must be able to solve (6)), and the constants $\beta_{v}$ are weighting factors especially chosen in such a way as to make the energy eigenvalue $E_{\kappa}$ in (6) equal to the energy of the desired solution to (1). The advantage of this type of basis set is that in the asymptotic limit where both $V$ and $V_{0}$ become

[^0]small, all of the basis functions obey the same Schrödinger equation as the desired solution, since in this region
\[

$$
\begin{equation*}
\left[-\frac{1}{2} \Delta+V(\mathbf{x})-E_{\kappa}\right] \Psi_{\kappa}(\mathbf{x})=0 \rightarrow\left[-\frac{1}{2} \Delta-E_{\kappa}\right] \Psi_{\kappa}(\mathbf{x})=0 \tag{7}
\end{equation*}
$$

\]

and

$$
\begin{equation*}
\left[-\frac{1}{2} \Delta+\beta_{\nu} V_{0}(\mathbf{x})-E_{\kappa}\right] \Phi_{\nu}(\mathbf{x})=0 \rightarrow\left[-\frac{1}{2} \Delta-E_{\kappa}\right] \Phi_{\nu}(\mathbf{x})=0 \tag{8}
\end{equation*}
$$

Thus all the basis functions have the right asymptotic behavior, and their turning points occur in the right place, allowing them to contribute usefully to the synthesis of $\Psi_{\kappa}$. By contrast, conventional methods often represent $\Psi_{\kappa}$ by superpositions of basis functions corresponding to various energies, and these may have the wrong behavior in the asymptotic region,

$$
\begin{equation*}
\left[-\frac{1}{2} \Delta+V_{0}(\mathbf{x})-E_{v}\right] \Phi_{v}(\mathbf{x})=0 \rightarrow\left[-\frac{1}{2} \Delta-E_{v}\right] \Phi_{v}(\mathbf{x})=0 \tag{9}
\end{equation*}
$$

as well as inappropriate turning points.

## Application to the Calculation of Atomic Spectra

The generalized Sturmian method can be applied to a wide variety of problems in atomic and molecular physics. We can illustrate it with a particularly simple example-calculation of the spectra of few-electron atoms, in the nonrelativistic approximation, neglecting the motion of the nucleus as well as spin-orbit coupling. This case is especially simple because if $V_{0}$ is chosen to be the Coulomb attraction of the nucleus

$$
\begin{equation*}
V_{0}(\mathbf{x})=-\sum_{j=1}^{N} \frac{Z}{r_{j}} \tag{10}
\end{equation*}
$$

then (6) can be solved exactly ${ }^{1}$ by Slater determinants of the form

$$
\begin{equation*}
\Phi_{v}=\left|\chi_{n, l, m, m_{s}} \chi_{n^{\prime}, l^{\prime}, m^{\prime}, m_{s}^{\prime}} \chi_{n^{\prime \prime}, l^{\prime \prime}, m^{\prime \prime}, m_{s}^{\prime \prime} \ldots}\right| \tag{11}
\end{equation*}
$$

In eq 11, the one-electron functions $\chi_{n, l, m, m_{s}}$ are hydrogen-like
spin-orbitals corresponding to an effective charge $Q_{\nu}$, which is related to $E_{\kappa}$ and to the principal quantum numbers in the configuration by

$$
\begin{equation*}
Q_{v}=\beta_{v} Z=\left(\frac{-2 E_{\kappa}}{\frac{1}{n^{2}}+\frac{1}{n^{\prime 2}}+\ldots}\right)^{1 / 2} \tag{12}
\end{equation*}
$$

To see that (11) will then be an exact solution of (6), we notice that each of the hydrogenlike spin-orbitals in (11) obeys a oneelectron Schrödinger equation of the form

$$
\begin{equation*}
\left[-\frac{1}{2} \nabla_{j}^{2}+\frac{Q_{v}{ }^{2}}{2 n^{2}}-\frac{Q_{v}}{r_{j}}\right] \chi_{n, l, m, m_{s}}\left(\mathbf{x}_{j}\right)=0 \tag{13}
\end{equation*}
$$

Thus the kinetic energy operator acting on (11) yields
$\left[-\frac{1}{2} \Delta\right] \Phi_{\nu}(\mathbf{x})=$
$\left[-\frac{Q_{v}{ }^{2}}{2}\left(\frac{1}{n^{2}}+\frac{1}{n^{\prime 2}}+\ldots\right)+\frac{Q_{v}}{r_{1}}+\frac{Q_{v}}{r_{2}}+\ldots\right] \Phi_{v}(\mathbf{x})$

$$
\begin{equation*}
=\left[E_{\kappa}-\beta_{\nu} V_{0}(\mathbf{x})\right] \Phi_{v}(\mathbf{x}) \tag{14}
\end{equation*}
$$

Comparing this with eq 6 , we can see that if $\Phi_{v}$ has the form indicated by (11)-(13), it will be an exact solution of (6). One can show ${ }^{16}$ that whenever $\beta_{v} \neq \beta_{v^{\prime}}$ the configurations $\Phi_{v}$ and $\Phi_{\nu^{\prime}}$ will then obey a potential-weighted orthonormality relation of the form

$$
\begin{equation*}
\int \mathrm{d} x \Phi_{v^{\prime}}^{*}(\mathbf{x}) V_{0}(\mathbf{x}) \Phi_{\nu}(\mathbf{x})=\delta_{\nu^{\prime}, v} \frac{2 E_{\kappa}}{\beta_{v}} \tag{15}
\end{equation*}
$$

When two different configurations have the same value of $\beta$, a basis set transformation sometimes needs to be made to ensure that (15) holds.

## Generalized Sturmian Secular Equations

It is convenient to introduce the variables

$$
\begin{equation*}
p_{\kappa} \equiv \sqrt{-2 E_{\kappa}} \tag{16}
\end{equation*}
$$

and

$$
\begin{equation*}
\mathscr{R}_{v} \equiv \sqrt{\frac{1}{n^{2}}+\frac{1}{n^{\prime 2}}+\ldots} \tag{17}
\end{equation*}
$$

In terms of them, (12) can be rewritten in the form

$$
\begin{equation*}
Q_{v}=\beta_{v} Z=\frac{p_{\kappa}}{\mathscr{R}_{v}} \tag{18}
\end{equation*}
$$

while the energy becomes

$$
\begin{equation*}
E_{\kappa}=-\frac{p_{\kappa}{ }^{2}}{2} \tag{19}
\end{equation*}
$$

The potential $V$ that enters the $N$-electron Schrödinger eq 1 includes both the nuclear attraction term $V_{0}$ and an interelectron repulsion term $V^{\prime}$ :

$$
\begin{equation*}
V(\mathbf{x})=V_{0}(\mathbf{x})+V^{\prime}(\mathbf{x})=-\sum_{j=1}^{N} \frac{Z}{r_{j}}+\sum_{i>i}^{N} \sum_{i=1}^{N} \frac{1}{r_{i j}} \tag{20}
\end{equation*}
$$

TABLE 1: ${ }^{1}$ S Excited State Energies (in Hartrees) for the 2-Electron Isoelectronic Series ${ }^{a}$

|  | He | $\mathrm{Li}^{+}$ | $\mathrm{Be}^{2+}$ | $\mathrm{B}^{3+}$ | $\mathrm{C}^{4+}$ | $\mathrm{N}^{5+}$ |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| 1s2s ${ }^{1} \mathrm{~S}$ | -2.1429 | -5.0329 | -9.1730 | -14.564 | -21.206 | -29.098 |
| basis2 | -2.1442 | -5.0348 | -9.1759 | -14.567 | -21.209 | -29.102 |
| exp | -2.1458 | -5.0410 | -9.1860 | -14.582 | -21.230 | -29.131 |
| 1s3s ${ }^{1} \mathrm{~S}$ | -2.0603 | -4.7297 | -8.5099 | -13.402 | -19.406 | -26.521 |
| basis2 | -2.0606 | -4.7301 | -8.5107 | -13.403 | -19.407 | -26.522 |
| exp | -2.0611 | -4.7339 | -8.5183 | -13.415 | -19.425 | -26.548 |
| 1s4s ${ }^{1} \mathrm{~S}$ | -2.0332 | -4.6276 | -8.2837 | -13.003 | -18.785 | -25.629 |
| basis2 | -2.0333 | -4.6277 | -8.2840 | -13.003 | -18.785 | -25.630 |
| exp | -2.0334 | -4.6299 | -8.2891 |  |  | -25.654 |
| 1s5s ${ }^{1} \mathrm{~S}$ | -2.0210 | -4.5811 | -8.1806 | -12.820 | -18.500 | -25.220 |
| exp | -2.0210 | -4.5825 |  |  |  | -25.241 |
| 1s6s ${ }^{1} \mathrm{~S}$ | -2.0144 | -4.5562 | -8.1250 | -12.721 | -18.346 | -24.998 |
| exp | -2.0144 | -4.5571 |  |  |  |  |
| 1s7s ${ }^{1} \mathrm{~S}$ | -2.0105 | -4.5412 | -8.0917 | -12.662 | -18.253 | -24.865 |
| exp | -2.0104 | -4.5418 |  |  |  |  |
| 1s8s ${ }^{1} \mathrm{~S}$ | -2.0080 | -4.5315 | -8.0701 | -12.624 | -18.194 | -24.779 |
| exp | -2.0079 |  |  |  |  |  |
| 1s9s ${ }^{1} \mathrm{~S}$ | -2.0063 | -4.5248 | -8.0554 | -12.598 | -18.153 | -24.720 |
| exp | -2.0062 |  |  |  |  |  |
| 1s10s ${ }^{1} \mathrm{~S}$ | -2.0051 | -4.5201 | -8.0449 | -12.579 | -18.124 | -24.678 |
| exp | -2.0050 |  |  |  |  |  |
| 1s11s ${ }^{1} \mathrm{~S}$ | -2.0042 | -4.5166 | -8.0371 | -12.566 | -18.102 | -24.647 |
| exp | -2.0041 |  |  |  |  |  |
| 1s12s ${ }^{1} \mathrm{~S}$ | -2.0034 | -4.5140 | -8.0312 | -12.555 | -18.086 | -24.624 |
| exp | -2.0034 |  |  |  |  |  |

${ }^{a}$ In making Table 1, the basis set used consisted of 63 generalized Sturmians. Singlet and triplet states were calculated simultaneously, 0.5 S of 499 MHz Intel Pentium III time being required for the calculation of 154 states. Basis 2 consisted of 245 symmetry-adapted configurations ${ }^{20}$ including angular correlation up to $l_{\text {max }}=4$. Having constructed the interelectron repulsion matrix, eq 26, one can use the same matrix for any value of nuclear charge. Experimental values are taken from the NIST tables. ${ }^{27}$

TABLE 2: Doubly Excited Singlet S States of the 2-Electron Isoelectronic Series ${ }^{a}$

|  | $\mathrm{C}^{4+}$ | $\mathrm{N}^{5+}$ | $\mathrm{O}^{6+}$ | $\mathrm{F}^{7+}$ | $\mathrm{Ne}^{8+}$ | $\mathrm{Na}^{9+}$ |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: |
| 2s2s ${ }^{1} \mathrm{~S}$ | -8.2861 | -11.4133 | -15.0406 | -19.1678 | -23.7949 | -28.9221 |
| 2p2p ${ }^{1} \mathrm{~S}$ | -7.6460 | -10.6520 | -14.1580 | -18.1639 | -22.6698 | -27.6757 |
| exp | -7.6531 | -10.6604 | -14.1699 | -18.1777 | -22.6908 | -27.7075 |
| 2s3s ${ }^{1} \mathrm{~S}$ | -6.0563 | -8.3257 | -10.9563 | -13.9479 | -17.3007 | -21.0145 |
| exp | -6.0618 | -8.3317 | -10.9642 | -13.9618 | -17.3220 | -21.0470 |
| 2p3p ${ }^{1} \mathrm{~S}$ | -5.8049 | -8.0200 | -10.5961 | -13.5333 | -16.8315 | -20.4907 |
| exp | -5.8105 | -8.0252 | -10.6018 | -13.5441 | -16.8478 | -20.5122 |
| 2s4s ${ }^{1} \mathrm{~S}$ | -5.3537 | -7.3373 | -9.6334 | -12.2420 | -15.1631 | -18.3967 |
| 2p4p ${ }^{1} \mathrm{~S}$ | -5.2453 | -7.2056 | -9.4786 | -12.0641 | -14.9622 | -18.1729 |
| 2s5s ${ }^{1} \mathrm{~S}$ | -5.0370 | -6.8897 | -9.0324 | -11.4651 | -14.1878 | -17.2006 |
| 2p5pp ${ }^{1} \mathrm{~S}$ | -4.9810 | -6.8216 | -8.9522 | -11.3730 | -14.0838 | -17.0846 |

${ }^{a}$ These are autoionizing states embedded in a continuum. The few available experimental energies ${ }^{28-30}$ agree well with the calculated energies (in hartrees) shown in the table. Vainshtein and Safronova ${ }^{29,30}$ give wavelengths for transitions to states that are listed in the NIST tables, ${ }^{27}$ and from these the experimental energies were deduced. Solution of the Sturmian secular equations with basis 2 (Table 1) yielded the energies of all the ${ }^{1} \mathrm{~S}$ states shown in Tables 1-5.

Substituting the expansion of the wave function (5) into (1), we obtain

$$
\begin{equation*}
\sum_{\nu}\left[-\frac{1}{2} \Delta+V(\mathbf{x})-E_{\kappa}\right] \Phi_{\nu}(\mathbf{x}) C_{v, \kappa}=0 \tag{21}
\end{equation*}
$$

We now make use of the fact that each of the basis configurations $\Phi_{\nu}$ obeys eq 6. This allows us to rewrite (21) in the form

$$
\begin{equation*}
\sum_{v}\left[V(\mathbf{x})-\beta_{\nu} V_{0}(\mathbf{x})\right] \Phi_{\nu}(\mathbf{x}) C_{v, \kappa}=0 \tag{22}
\end{equation*}
$$

Multiplying (22) on the left by a conjugate configuration from

TABLE 3

|  | $\mathrm{C}^{4+}$ | $\mathrm{N}^{5+}$ | $\mathrm{O}^{6+}$ | $\mathrm{F}^{7+}$ | $\mathrm{Ne}^{8+}$ | $\mathrm{Na}^{9+}$ |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| 3s3s ${ }^{1} \mathrm{~S}$ | -3.7116 | -5.1069 | -6.7244 | -8.5641 | -10.6261 | -12.9102 |
| 3p3p ${ }^{1} \mathrm{~S}$ | -3.5578 | -4.9241 | -6.5126 | -8.3231 | -10.3559 | -12.6108 |
| 3d3d ${ }^{1} \mathrm{~S}$ | -3.2860 | -4.5981 | -6.1323 | -7.8887 | -9.8674 | -12.0683 |
| 3s4s ${ }^{1} \mathrm{~S}$ | -2.9098 | -4.0011 | -5.2661 | -6.7046 | -8.3168 | -10.1025 |
| 3p4p ${ }^{1} \mathrm{~S}$ | -2.8134 | -3.8850 | -5.1302 | -6.5489 | -8.1411 | -9.9070 |
| 3d4d ${ }^{1} \mathrm{~S}$ | -2.6940 | -3.7379 | -4.9554 | -6.3463 | -7.9107 | -9.6488 |
| 3s5s ${ }^{1} \mathrm{~S}$ | -2.5687 | -3.5248 | -4.6319 | -5.8902 | -7.2996 | -8.8601 |
| 3p5p ${ }^{1} \mathrm{~S}$ | -2.5165 | -3.4620 | -4.5586 | -5.8062 | -7.2050 | -8.7549 |
| 3d5d ${ }^{1} \mathrm{~S}$ | -2.4563 | -3.3885 | -4.4719 | -5.7064 | -7.0921 | -8.6288 |

TABLE 4

|  | $\mathrm{C}^{4+}$ | $\mathrm{N}^{5+}$ | $\mathrm{O}^{6+}$ | $\mathrm{F}^{7+}$ | $\mathrm{Ne}^{8+}$ | $\mathrm{Na}^{9+}$ |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| 4s4s ${ }^{1} \mathrm{~S}$ | -1.7189 | -2.3635 | -3.1107 | -3.9603 | -4.9125 | -5.9671 |
| 4p4p ${ }^{1} \mathrm{~S}$ | -1.6754 | -2.3116 | -3.0503 | -3.8914 | -4.8350 | -5.8811 |
| 4d4d ${ }^{1} \mathrm{~S}$ | -1.6169 | -2.2405 | -2.9665 | -3.7949 | -4.7258 | -5.7592 |
| 4f4f ${ }^{1} \mathrm{~S}$ | -1.5517 | -2.1588 | -2.8682 | -3.6801 | -4.5943 | -5.6111 |
| 4s5s ${ }^{1} \mathrm{~S}$ | -1.5293 | -2.0997 | -2.7604 | -3.5113 | -4.3525 | -5.2840 |
| 4p5p ${ }^{1} \mathrm{~S}$ | -1.5031 | -2.0683 | -2.7238 | -3.4695 | -4.3055 | -5.2318 |
| 4d5d ${ }^{1} \mathrm{~S}$ | -1.4682 | -2.0263 | -2.6746 | -3.4132 | -4.2420 | -5.1611 |
| 4f5f ${ }^{1} \mathrm{~S}$ | -1.4308 | -1.9801 | -2.6198 | -3.3497 | -4.1700 | -5.0806 |
| 4s6s ${ }^{1} \mathrm{~S}$ | -1.4170 | -1.9429 | -2.5517 | -3.2434 | -4.0180 | -4.8755 |
| 4p6p ${ }^{1} \mathrm{~S}$ | -1.4004 | -1.9230 | -2.5285 | -3.2169 | -3.9881 | -4.8423 |
| 4d6d ${ }^{1} \mathrm{~S}$ | -1.3790 | -1.8974 | -2.4986 | -3.1827 | -3.9496 | -4.7994 |
| 4f6f ${ }^{1} \mathrm{~S}$ | -1.3553 | -1.8684 | -2.4646 | -3.1436 | -3.9056 | -4.7504 |

TABLE 5

|  | $\mathrm{C}^{4+}$ | $\mathrm{N}^{5+}$ | $\mathrm{O}^{6+}$ | $\mathrm{F}^{7+}$ | $\mathrm{Ne}^{8+}$ | $\mathrm{Na}^{9+}$ |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| 5s5s ${ }^{1} \mathrm{~S}$ | -1.0065 | -1.3957 | -1.8448 | -2.3470 | -2.9096 | -3.5327 |
| 5p5p ${ }^{1} \mathrm{~S}$ | -0.9995 | -1.3851 | -1.8241 | -2.3234 | -2.8832 | -3.5033 |
| 5d5d ${ }^{1} \mathrm{~S}$ | -0.9858 | -1.3606 | -1.7957 | -2.2911 | -2.8470 | -3.4632 |
| 5f5f ${ }^{1} \mathrm{~S}$ | -0.9577 | -1.3268 | -1.7564 | -2.2466 | -2.7972 | -3.4083 |
| 5g5g ${ }^{1} \mathrm{~S}$ | -0.9479 | -1.3003 | -1.7196 | -2.2039 | -2.7486 | -3.3537 |
| 5s6 ${ }^{1} \mathrm{~S}$ | -0.9387 | -1.2959 | -1.7084 | -2.1721 | -2.6914 | -3.2663 |
| 5p6p ${ }^{1} \mathrm{~S}$ | -0.9336 | -1.2894 | -1.6958 | -2.1578 | -2.6754 | -3.2486 |
| 5d6d ${ }^{1} \mathrm{~S}$ | -0.9327 | -1.2834 | -1.6887 | -2.1497 | -2.6662 | -3.2382 |
| 5f6f ${ }^{1} \mathrm{~S}$ | -0.9261 | -1.2746 | -1.6787 | -2.1385 | -2.6538 | -3.2247 |
| 5g6g ${ }^{1} \mathrm{~S}$ | -0.9186 | -1.2653 | -1.6676 | -2.1256 | -2.6391 | -3.2082 |
| 5s7s ${ }^{1} \mathrm{~S}$ | -0.9056 | -1.2504 | -1.6510 | -2.1073 | -2.6192 | -3.1867 |
| 5p7p ${ }^{1} \mathrm{~S}$ | -0.8989 | -1.2408 | -1.6385 | -2.0919 | -2.6010 | -3.1657 |
| 5d7d ${ }^{1} \mathrm{~S}$ | -0.8967 | -1.2296 | -1.6263 | -2.0786 | -2.5865 | -3.1501 |
| 5f7f ${ }^{1} \mathrm{~S}$ | -0.8888 | -1.2292 | -1.6140 | -2.0512 | -2.5407 | -3.0944 |
| 5g7g ${ }^{1} \mathrm{~S}$ | -0.8883 | -1.2195 | -1.6029 | -2.0391 | -2.5387 | -3.0825 |

TABLE 6: Energies for Doubly Excited States of C ${ }^{4+}$ with Varying Angular Correlation ${ }^{a}$

| $\mathrm{C}^{4+}$ | $l_{\max }=0$ | $l_{\max }=1$ | $l_{\max }=2$ | $l_{\max }=3$ | $l_{\max }=4$ |
| :--- | :--- | :--- | :--- | :--- | :--- |
| 2s2s ${ }^{1} \mathrm{~S}$ | -8.1172 | -8.2854 | -8.2860 | -8.2861 | -8.2861 |
| 2p2p ${ }^{1} \mathrm{~S}$ | -5.9810 | -7.6058 | -7.6400 | -7.6449 | -7.6460 |
| 2s3s ${ }^{1} \mathrm{~S}$ | -5.3211 | -6.0557 | -6.0562 | -6.0563 | -6.0563 |
| 2p3p ${ }^{1} \mathrm{~S}$ | -5.0202 | -5.7949 | -5.8038 | -5.8048 | -5.8049 |
| 2s4s ${ }^{1} \mathrm{~S}$ | -4.8586 | -5.3534 | -5.3537 | -5.3537 | -5.3537 |
| 2p4p ${ }^{1} \mathrm{~S}$ | -4.7620 | -5.2411 | -5.2448 | -5.2452 | -5.2453 |
| 2s5s ${ }^{1} \mathrm{~S}$ | -4.6997 | -5.0369 | -5.0370 | -5.0370 | -5.0370 |
| 2p5p ${ }^{1} \mathrm{~S}$ | -4.6572 | -4.9789 | -4.9808 | -4.9810 | -4.9810 |

${ }^{a}$ As can be seen from Table 1, the inclusion of angular correlation has only a small effect on the singly excited states. However, Table 6 shows that angular correlation is crucial for accurate determination of the energies of doubly excited states.
the basis set, and integrating over space and spin coordinates, we obtain

$$
\begin{equation*}
\sum_{v} \int \mathrm{~d} x \Phi_{v^{\prime}}^{*}(\mathbf{x})\left[V(\mathbf{x})-\beta_{v} V_{0}(\mathbf{x})\right] \Phi_{v}(\mathbf{x}) C_{v, k}=0 \tag{23}
\end{equation*}
$$

We now introduce the definition

$$
\begin{equation*}
T_{\nu^{\prime}, \nu}^{0} \equiv-\frac{1}{p_{\kappa}} \int \mathrm{d} x \Phi_{\nu^{\prime}}^{*}(\mathbf{x}) V_{0}(\mathbf{x}) \Phi_{\nu}(\mathbf{x}) \tag{24}
\end{equation*}
$$

With the help of the potential-weighted orthonormality relation (15) and making use of (18), we have

$$
\begin{equation*}
T_{\nu^{\prime}, v}^{0}=\delta_{\nu^{\prime} \nu} \mathrm{Z} \mathscr{R}_{\nu} \tag{25}
\end{equation*}
$$

Thus we see that the matrix $T_{v^{\prime}, \nu}^{0}$ defined by (24) is independent of $p_{\kappa}$. Similarly, it can be shown ${ }^{16}$ that

$$
\begin{equation*}
T_{\nu^{\prime}, v}^{\prime} \equiv-\frac{1}{p_{\kappa}} \int \mathrm{d} x \Phi_{\nu^{\prime}}^{*}(\mathbf{x}) V^{\prime}(\mathbf{x}) \Phi_{\nu}(\mathbf{x}) \tag{26}
\end{equation*}
$$

is also independent of $p_{\kappa}$. Substituting these definitions into (23) and dividing by $p_{\kappa}$, we obtain the generalized Sturmian secular equations

$$
\begin{equation*}
\sum_{v}\left[\delta_{v^{\prime}, \nu} Z \mathscr{R}_{\nu}+T_{\nu^{\prime}, v}^{\prime}-p_{\kappa} \delta_{\nu^{\prime}, \nu}\right] C_{v, \kappa}=0 \tag{27}
\end{equation*}
$$

This set of secular equations differs greatly from the usual secular equations that are obtained when one diagonalizes the Hamiltonian of a quantum mechanical system. The kinetic energy term has disappeared, and the term representing nuclear attraction is diagonal. Furthermore, the eigenvalues are not energies but values of the parameter $p_{\kappa}$, which is related to the energy through equations (16) and (19). By solving the set of secular equations (27), one obtains a spectrum of $p_{\kappa}$ values, and hence also a spectrum of energies $E_{\kappa}=-p_{\kappa}{ }^{2} / 2$. The lowest values of $p_{\kappa}$ correspond to the most tightly bound states. From (18) we can see that this parameter can be interpreted as a "scaling parameter", since the effective charges $Q_{v}$ that characterize the configurations are proportional to $p_{\kappa}$. Thus the basis set adjusts automatically to the energy of the state. For the lowest energy states, the basis functions are localized near the nucleus, while for highly excited states, they are diffuse. By solving the set of secular equations, one obtains simultaneously the energy spectrum and the optimal basis set for each state. This characteristic of the generalized Sturmian method makes it very well suited to the rapid calculation of large numbers of excited states of few-electron atoms, as is illustrated in Tables 1-5.

## References and Notes

(1) Goscinski, O. Preliminary Research Report No. 217; Quantum Chemistry Group, Uppsala University, 1968; Adv. Quantum Chem. 2003, 41, 51.
(2) Shull, H.; Löwdin, P. O. J. Chem. Phys. 1959, 30, 617.
(3) Rotenberg, M. Ann. Phys. (N. Y.) 1962, 19, 62.
(4) Rotenberg, M. Adv. At. Mol. Phys. 1970, 6, 233.
(5) Weniger, E. J. J. Math. Phys. 1985, 26, 276.
(6) Aquilanti, V.; Cavalli, S.; De Fazio, D.; Grossi, G. Hyperangular Momentum: Applications to Atomic and Molecular Science in New Methods in Quantum Theory; Tsipis, C. A., Popov, V. S., Herschbach, D. R., Avery, J., Eds.; Kluwer Academic Publishers: Dordrecht, The Netherlands, 1996.
(7) Aquilanti, V.; Cavalli, S.; Coletti, C.; Grossi, G. Chem. Phys. 1996, 209, 405.
(8) Aquilanti, V.; Cavalli, S.; Coletti, C. Chem. Phys. 1997, 214, 1.
(9) Avery, J.; Herschbach, D. R. Int. J. Quantum Chem. 1992, 41, 673.
(10) Avery, J. J. Math. Chem. 1997, 21, 285.
(11) Aquilanti, V.; Avery, J. Chem. Phys. Lett. 1997, 267, 1.
(12) Avery, J.; Antonsen, F. J. Math. Chem. 1998, 24, 175.
(13) Avery, J. J. Math. Chem. 1998, 24, 169.
(14) Avery, J., J. Mol. Struct. 1999, 458, 1.
(15) Avery, J., Adv. Quantum Chem. 1999, 31, 201.
(16) Avery, J. Hyperspherical Harmonics and Generalized Sturmians; Kluwer Academic Publishers: Dordrecht, The Netherlands, 2000.
(17) Avery, J.; Shim, R. Int. J. Quantum Chem. 2000, 79, 1.
(18) Avery, J.; Shim, R. Int. J. Quantum Chem. 2000, 83, 1.
(19) Avery, J. J. Math. Chem. 2000, 4, 279.
(20) Avery, J.; Sauer, S. In Many-Electron Sturmians Applied to Molecules. Quantum Systems in Chemistry and Physics, Vol. 1; HernándezLaguna, A., Maruani, J., McWeeny, R., Wilson, S., Eds.; Kluwer Academic Publishers: Dordrecht, The Netherlands, 2000.
(21) Aquilanti, V.; Avery, J. Adv. Quantum Chem. 2001, 39, 71.
(22) Aquilanti, V.; Cavalli, S.; Coletti, C.; Di Domenico, D.; Grossi, G. Int. Rev. Phys. Chem. 2001, 20, 673.
(23) Avery, J.; Shim, R. Int. J. Quantum Chem. 2001, 83, 1.
(24) Avery, James; Avery, John. J. Math. Chem. 2003, 33, 145.
(25) Avery, John; Avery, James. Adv. Quantum Chem. 2003, 43, 185.
(26) Avery, John; Avery, James; Goscinski, Oswaldo. Adv. Quantum Chem. 2003, 43, 207.
(27) NIST Atomic Spectra Database; National Institute of Standards and Technology (NIST), http://physics.nist.gov/asd.
(28) Atomic and Molecular Databases: Data for Autoionizing States; National Institute for Fusion Science (NIFS), http://dprose.nifs.ac.jp/DB/ Auto/.
(29) Vainshtein, L. A.; Safronova, U. I. Wavelengths and Transition Probabilities of Satellites to Resonance Lines of $\mathrm{H}-$ and He-like Ions. At. Data Nucl. Data Tables 1978, 21, 49.
(30) Vainshtein, L. A.; Safronova, U. I. Dielectronic Satellite Spectra for Highy Charged H-like Ions and He-like Ions with $\mathrm{Z}=6-33$. At. Data Nucl. Data Tables 1980, 25, 311.


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