

Compositional Dependence of Upconversion Luminescence of Er^{3+} in ZrF_4 -Based Glasses

III. Anion Substitution Effect

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As one of the compositional designs for enhancing the upconversion luminescence efficiency of Er^{3+} in ZrF_4 -based glasses, the anion substitution effect has been examined in glasses $58\text{ZrF}_4 \cdot (39-x)\text{BaF}_2 \cdot x\text{BaX}_2 \cdot 3\text{Er}$ (or Eu) F_3 ($x = 0-29$ for $X = \text{Cl}$ and $x = 0-10$ for $X = \text{Br}$ in batch compositions). Prominent enhancement of upconversion emission intensity was observed on the substitution of Cl for F. The Cl substitution brought about two effects, of which one was an increase in the deviation from inversion symmetry in the F^- coordination around Er^{3+} and the other was an increase in both phonon energy and electron-phonon coupling strength. The former raises the transition probability in the upconversion process, leading to an enhancement of the emission intensity, while the latter increases the multiphonon relaxation in the upconversion process, leading to a reduction of the emission intensity. Due to these antithetical effects on the upconversion process, the upconversion emission intensity of the Cl substituted glasses exhibits a maximum value around the $58\text{ZrF}_4 \cdot 20\text{BaF}_2 \cdot 19\text{BaCl}_2 \cdot 3\text{ErF}_3$ composition. The substitution of small amounts of Cl for F is concluded to be quite effective for enhancing the upconversion efficiency of ZrF_4 -based glasses. © 1993 Academic Press, Inc.

I. Introduction

The conversion of infrared radiation to shorter wavelengths, i.e., frequency upconversion luminescence, by rare earth-

doped glasses presently receives much interest since the development of near-infrared semiconductor diode lasers has made the upconversion pumping of rare-earth laser systems possible (1). Among various glass-forming systems, heavy metal fluoride glasses are considered to be the best candi-

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dates for producing upconversion luminescence with high emission efficiency. This is because the maximum phonon frequencies of the glasses are so small that the nonradiative loss due to multiphonon relaxation can be substantially reduced (2). Until now, therefore, upconversion luminescence phenomena have been intensively examined in heavy metal fluoride glasses based on ThF₄ (3), HfF₄ (4), and so on. Accordingly, compositional designs for upconversion laser glasses also are an important subject in heavy metal fluoride-based systems. However, there are few reports thus far on the compositional dependence of upconversion luminescence in these glasses.

The present authors are advancing a series of experiments as to how the upconversion luminescence of Er³⁺ in fluoride glasses is affected by the kinds and quantities of glass-constituting ions. So far the effect of glass-modifying cations on Er³⁺ upconversion luminescence has been examined in the 21.25MF · 21.25LiF · 52.5ZrF₄ · 5ErF₃ (M: Li, Na, K, Rb or Cs) and 5M'F₂ · 25BaF₂ · 65ZrF₄ · 5ErF₃ (M': Mg, Ca, Sr or Ba) glasses (5, 6). Also the difference in Er³⁺ upconversion luminescence with the kinds of glass-forming cations has been examined on the MF_n-BaF₂-YF₃ (M: Zr, Hf, Al, Ga, In, Sc, or Zn) glasses (7).

This paper reports the effect of anion substitution on the upconversion properties of Er³⁺ in fluoride-based glasses. The glass composition chosen was a ZrF₄-based one in which part of the F⁻ ions was replaced by other anions such as Cl⁻, Br⁻, I⁻, or O²⁻ ions.

Experimental Procedure

(a) Sample Preparation

The glass composition employed is 58ZrF₄ · (39-x)BaF₂ · xBaX₂ · 3LnF₃ (Ln: Er or Eu, X₂: Cl₂, Br₂, I₂ or O) in mol%. The x values for the respective Xs are given in Table I. High purity reagents ZrF₄, BaF₂,

BaCl₂, BaBr₂, BaI₂, BaCO₃, ErF₃, and EuF₃ were used as raw materials in preparing the glasses. About 4 g batches of the raw materials were melted at 700°C for about 20 min in Pt crucibles under an Ar gas stream. Then the melts were poured into brass molds kept at a temperature of about 100°C. The glasses obtained were annealed at the respective glass transition temperatures determined by a differential thermal analysis. In the samples of O substitution (i.e., BaX₂ = BaO), small but numerous crystals precipitated in the glass matrices, giving no good quality glasses with high transparency.

Glasses obtained by Cl, Br, or I substitution were subjected to X-ray fluorescence analysis to determine the substituted halogen contents. The analyzed values of Cl, Br, and I contents are given in Table I, where glass-transition temperature, density and Er³⁺ concentration are also given for the Cl or Br substituted glasses. Since no significant amounts of I were found to remain in the I substituted glasses, no experiments were performed on these glasses.

Since the Cl or Br substituted glasses were prepared without the addition of a fluoridizing agent, NH₄F · HF, to the batches, appreciable amounts of oxygen might be included as an impurity. Thus, the oxygen content was analyzed in some of the Cl substituted glasses by using an oxygen and nitrogen analyzer for metals (HORIBA, EMGA-2200). The analyzed oxygen contents in glasses of x = 0, 15, and 24 were 0.469(18), 0.223(8) and 0.357(16)ppm, respectively. Therefore the oxygen amounts of the Cl substituted glasses are concluded to be negligibly small regardless of the Cl content.

(b) Upconversion Luminescence Spectra of Er³⁺

The upconversion luminescence spectra of Er³⁺ in the 58ZrF₄ · (39-x)BaF₂ · xBaX₂ · 3ErF₃ (X = Cl or Br) glasses were measured in the wavelength range 380–780 nm

TABLE I
COMPOSITIONS AND SOME PROPERTIES OF GLASSES $58\text{ZrF}_4 \cdot (39-x)\text{BaF}_2 \cdot x\text{BaX}_2 \cdot 3\text{ErF}_3$
($X_2 = \text{Cl}_2, \text{Br}_2, \text{I}_2$ or O)

X	x (Mol%) batch	Substituted F (Anion%)		T_g (°C)	Density ($\text{g} \cdot \text{cm}^{-3}$)	Er^{3+} concentration ($10^{-4} \text{ mol} \cdot \text{cm}^{-3}$)
		Batch	Analyzed			
Cl	0	0	~0.00	309	4.909	8.193
	5	3	1.71	287	4.653	8.029
	10	6	3.20	285	4.603	7.868
	15	9	5.09	276	4.492	7.609
	19	12	6.23	271	4.453	7.485
	24	15	7.50	262	4.550	7.400
	29	18	8.95	249	4.394	7.251
Br	5	3	2.7	297	4.655	7.863
	10	6	5.4	287	3.463	7.538
I	5	3	~0.0			
	10	6	~0.0			
O	2.5	1.5				
	5	3				

with a fluorescence spectrophotometer (HITACHI, 850) at room temperature, using 800.0 nm radiation from an AlGaAs diode laser (SONY, SLD303XT-22) with a 250 mW power excitation source. Glass plates of size $3.0 \times 4.0 \times 1.0$ mm with optically flat surfaces were used.

(c) Emission Spectra of Eu^{3+}

The emission spectra of the ${}^5D_0 \rightarrow {}^7F_J$ ($J = 0, 1$ and 2) transitions of Eu^{3+} in the $58\text{ZrF}_4 \cdot (39-x)\text{BaF}_2 \cdot x\text{BaCl}_2 \cdot 3\text{ErF}_3$ glasses were measured in the wavelength range 550–650 nm with a fluorescence spectrophotometer (HITACHI, 850) at room temperature, using 395 nm light of a Xe lamp as the excitation source.

(d) Phonon Sideband Spectra of Eu^{3+}

The excitation spectra of Eu^{3+} in the $58\text{ZrF}_4 \cdot (39-x)\text{BaF}_2 \cdot x\text{BaCl}_2 \cdot 3\text{ErF}_3$ glasses were measured by monitoring the ${}^5D_0 \rightarrow {}^7F_2$ emission at 616 nm. The measurement was carried out in the wavelength range 420–475 nm at room temperature with the same appa-

ratus as that in (c). The phonon sidebands associated with the ${}^5D_2 \leftarrow {}^7F_0$ transition were observed for all the glasses at high energy sides of the electronic bands.

Results and Discussion

(a) Change in Upconversion Luminescence Spectra with Anion Substitution

The upconversion luminescence spectra of the $58\text{ZrF}_4 \cdot (39-x)\text{BaF}_2 \cdot x\text{BaCl}_2 \cdot 3\text{ErF}_3$ glasses are shown in Fig. 1. The upconver-

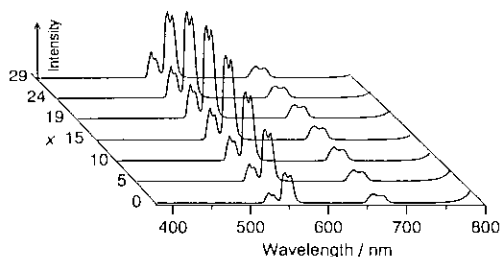


FIG. 1. Upconversion luminescence spectra of Er^{3+} in glasses $58\text{ZrF}_4 \cdot (39-x)\text{BaF}_2 \cdot x\text{BaCl}_2 \cdot 3\text{ErF}_3$.

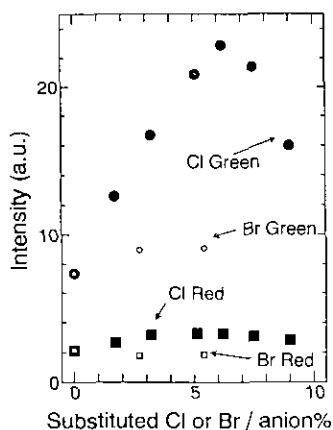


FIG. 2. Anion substitution dependences of upconversion emission intensities in glasses $58\text{ZrF}_4 \cdot (39-x)\text{BaF}_2 \cdot x\text{BaX}_2 \cdot 3\text{ErF}_3$ ($X = \text{Cl}$ or Br).

sion luminescences of Er^{3+} were observed at about 525 and 550 nm, due to the ${}^2\text{H}_{11/2} \rightarrow {}^4\text{I}_{15/2}$ and ${}^4\text{S}_{3/2} \rightarrow {}^4\text{I}_{15/2}$ transitions, respectively, and at about 660 nm due to the ${}^4\text{F}_{9/2} \rightarrow {}^4\text{I}_{15/2}$ transition. A faint emission band was observed at about 410 nm, corresponding to the ${}^2\text{H}_{9/2} \rightarrow {}^4\text{I}_{15/2}$ transition. Peak wavelengths of these emission bands were almost the same in all of the Cl or Br substituted glasses.

Figure 2 shows the Cl- and Br-content dependences of the emission intensities of the green color (525 and 550 nm) and red color (660 nm) bands. The following three features can be seen from the figure: (1) The magnitude of the change in the emission intensities with anion substitution is much more remarkable for the Cl substitution than for the Br substitution. (2) In the Cl substituted glasses the emission intensities of both green and red colors become larger with substitution, although substitution beyond about 6 anion% leads to a reduction of the emission intensities. (3) In the Br substituted glasses the emission intensity of the green color slightly increases with substitution, while that of the red color remains almost unchanged.

(b) F^- Coordination Environment of Eu^{3+}

The observed emission spectra of Eu^{3+} in the $58\text{ZrF}_4 \cdot (39-x)\text{BaF}_2 \cdot x\text{BaCl}_2 \cdot 3\text{EuF}_3$ glasses are shown in Fig. 3. In the luminescence spectral bands, the ${}^5\text{D}_0 \rightarrow {}^7\text{F}_2$ emission band is due to the electric-dipole transition, depending largely on a local symmetry of the coordination environment around Eu^{3+} , while the ${}^5\text{D}_0 \rightarrow {}^7\text{F}_1$ emission band is due to the magnetic-dipole transition, which is independent of a local symmetry (8). Therefore the ratio of the ${}^5\text{D}_0 \rightarrow {}^7\text{F}_2$ to ${}^5\text{D}_0 \rightarrow {}^7\text{F}_1$ emission intensities gives information about the deviation from inversion symmetry in the coordination environment of Eu^{3+} (9).

In the emission spectra of Eu^{3+} ions observed for the $58\text{ZrF}_4 \cdot (39-x)\text{BaF}_2 \cdot x\text{BaCl}_2 \cdot 3\text{EuF}_3$ glasses, the ${}^5\text{D}_0 \rightarrow {}^7\text{F}_0$, ${}^5\text{D}_1 \rightarrow {}^7\text{F}_3$ and ${}^5\text{D}_0 \rightarrow {}^7\text{F}_1$ emission bands overlap one another, as can be seen from the assignment given in Fig. 3. Thus the deconvolution of each band was necessary in order to calculate the ratio of the ${}^5\text{D}_0 \rightarrow {}^7\text{F}_2$ to ${}^5\text{D}_0 \rightarrow {}^7\text{F}_1$ emission intensities. The band deconvolu-

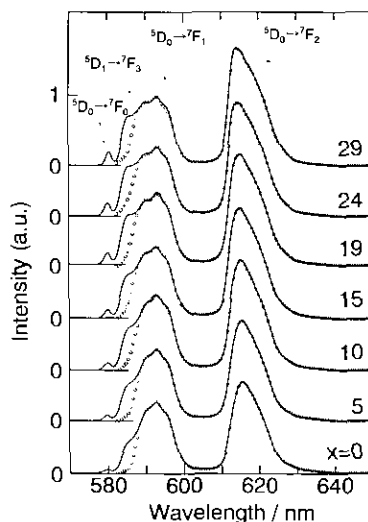


FIG. 3. Emission spectra of Eu^{3+} in glasses $58\text{ZrF}_4 \cdot (39-x)\text{BaF}_2 \cdot x\text{BaCl}_2 \cdot 3\text{EuF}_3$. (Dotted lines of open circles are ${}^5\text{D}_0 \rightarrow {}^7\text{F}_1$ and ${}^5\text{D}_0 \rightarrow {}^7\text{F}_2$ emission spectra obtained by convolution fitting.)

tion was performed using a curve-fitting program of the least squares method. An example of the convolution fitting is shown for the $x = 5$ glass in Fig. 4. The ${}^5D_0 \rightarrow {}^7F_1$ bands obtained for all the Cl substituted glasses by convolution fitting are shown by dotted lines of open circles in Fig. 3, together with the ${}^5D_0 \rightarrow {}^7F_2$ bands.

Figure 5 shows the ratios of the ${}^5D_0 \rightarrow {}^7F_2$ to ${}^5D_0 \rightarrow {}^7F_1$ emission intensities in the $58ZrF_4 \cdot (39-x)BaF_2 \cdot xBaCl_2 \cdot 3EuF_3$ glasses, which are plotted against the substituted Cl content in anion%. The compositional dependence of the ${}^5D_0 \rightarrow {}^7F_2$ to ${}^5D_0 \rightarrow {}^7F_1$ ratios clearly indicates that a deviation from inversion symmetry in the F^- coordination of Eu^{3+} increases with an increase of the substituted Cl content. Since the Er^{3+} and Eu^{3+} ions can be considered to be chemically equivalent and thus to play almost the same role in the glasses, the F^- coordination environments of Er^{3+} are presumed to be the same as those of Eu^{3+} .

Therefore, the increasing tendency of upconversion intensities in the $58ZrF_4 \cdot (39-x)BaF_2 \cdot xBaCl_2 \cdot 3ErF_3$ glasses is ascribed to the increased transition probability in the upconversion process with successive substitution of the Cl^- ions for the F^- ions around the Er^{3+} ions. In actual, however, the upconversion luminescence intensities were reduced in the Cl substitution beyond

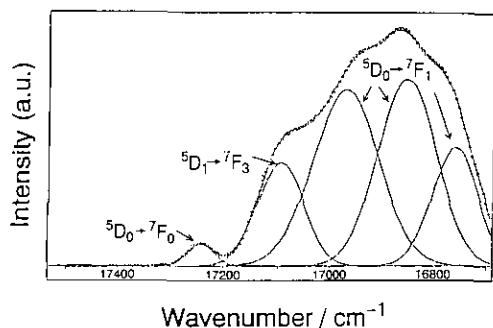


FIG. 4. Convolution fitting for emission spectrum of Eu^{3+} in glass $58ZrF_4 \cdot 34BaF_2 \cdot 5BaCl_2 \cdot 3EuF_3$.

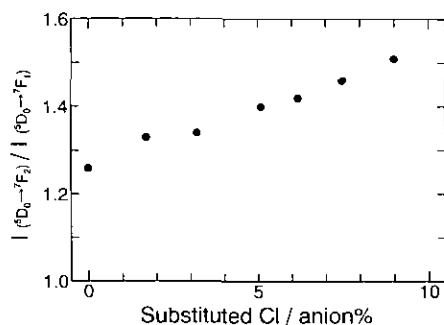


FIG. 5. Intensity ratios of emissions ${}^5D_0 \rightarrow {}^7F_2$ to ${}^5D_0 \rightarrow {}^7F_1$ in glasses $58ZrF_4 \cdot (39-x)BaF_2 \cdot xBaCl_2 \cdot 3EuF_3$.

about 6 anion%, as seen from Fig. 2. Accordingly there must be other factors by which upconversion luminescence intensities are reduced with the Cl substitution.

(c) Phonon Energy and Electron-Phonon Coupling Strength

According to the Miyakawa-Dexter equation (10), the nonradiative decay rate due to the multiphonon relaxation process is governed by the phonon energy and the electron-phonon coupling strength. The larger the phonon energy and/or the electron-phonon coupling strength, the larger becomes the decay rate. Consequently an increase in phonon energy and/or electron-phonon coupling strength reduces the lifetime and quantum efficiency of excited levels, and thus reduces the emission intensity of upconversion luminescence.

The phonon sideband (PSB) spectra observed for the $58ZrF_4 \cdot (39-x)BaF_2 \cdot xBaCl_2 \cdot 3EuF_3$ glasses are shown in Fig. 6, where all the spectra are normalized to the intensity of the pure electronic transition (PET) in order to facilitate a comparison. The phonon energy corresponds to the mean energy difference between the pure electronic transition and the phonon sideband, and the electron-phonon coupling strength is estimated as the ratio of the integrated intensity of

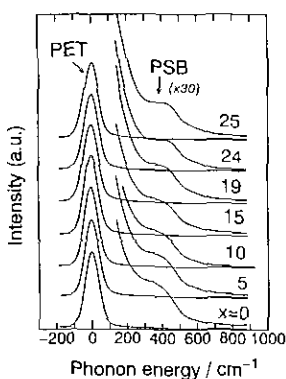


FIG. 6. Excitation spectra of ${}^3D_0 \leftarrow {}^7F_2$ emission of Eu^{3+} in glasses $58\text{ZrF}_4 \cdot (39-x)\text{BaF}_2 \cdot x\text{BaCl}_2 \cdot 3\text{EuF}_3$.

phonon sideband to that of pure electronic transition (11). It can be seen from the figure that the phonon sidebands overlap very closely with the pure electronic transition bands, making the evaluation of both phonon frequency and electron-phonon coupling strength extremely difficult. Thus, deconvolution of the phonon sidebands from the pure electronic transition bands was attempted for the Cl substituted glasses of $x = 0, 10, 19$ and 29 , using a curve-fitting program of the least squares method. The phonon energy and the electron-phonon coupling strength values obtained from the extracted phonon sidebands are given in Table II. From the table it can be seen that the phonon energy becomes considerably

TABLE II

PHONON ENERGIES, $\hbar\omega$, AND ELECTRON-PHONON COUPLING STRENGTHS, g , OF GLASSES $58\text{ZrF}_4 \cdot (39-x)\text{BaF}_2 \cdot x\text{BaCl}_2 \cdot 3\text{EuF}_3$

x (Mol%) batch	Substituted F (Anion%) batch	$\hbar\omega$ (cm^{-1})	g
0	0	375	0.0058
10	6	395	0.0072
19	12	425	0.0072
29	18	435	0.0074

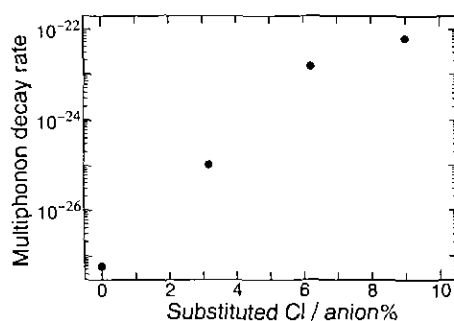


FIG. 7. Multiphonon decay rates in transition ${}^4I_{9/2} \rightarrow {}^4I_{11/2}$ obtained for glasses $58\text{ZrF}_4 \cdot (39-x)\text{BaF}_2 \cdot x\text{BaCl}_2 \cdot 3\text{ErF}_3$ by the Miyakawa-Dexter equation.

higher with the successive substitution and the electron-phonon coupling strength is slightly increased. From these values, the multiphonon decay rate of the ${}^4I_{9/2} \rightarrow {}^4I_{11/2}$ transition was calculated for the glasses of $x = 0, 10, 19$ and 29 according to the Miyakawa-Dexter equation and the results are plotted in Fig. 7 as a function of the substituted Cl content in anion%. The calculation result, though it might be a tentative estimate, suggests that the nonradiative decay rate increases with the successive Cl substitution.

Such a dependence of the nonradiative decay rate on the Cl content implies that the emission intensity of upconversion luminescence in the $58\text{ZrF}_4 \cdot (39-x)\text{BaF}_2 \cdot x\text{BaCl}_2 \cdot 3\text{ErF}_3$ glasses decreases with the successive substitution of the Cl^- ions for the F^- ions. On the other hand, as described in Section (b), the substitution of the Cl^- ions for the F^- ions gains the emission intensity of upconversion luminescence by increasing the transition probabilities in a multistep excitation and emission process. These two antithetical effects can explain the results shown in Fig. 2, where the upconversion luminescence intensity exhibits a maximum value at the Cl substitution of 6 anion% in the $58\text{ZrF}_4 \cdot (39-x)\text{BaF}_2 \cdot x\text{BaCl}_2 \cdot 3\text{ErF}_3$ glasses.

Conclusions

The effect of anion substitution on the upconversion luminescence properties of Er^{3+} in ZrF_4 -based glasses was investigated by substituting a part of the F^- ions in the $58\text{ZrF}_4 \cdot 39\text{BaF}_2 \cdot 3\text{ErF}_3$ composition by Cl^- , Br^- , I^- , or O^{2-} ions. The substitution was possible only for the Cl^- and Br^- ions and the substitution ranges for the respective ions were 0–8.95 and 0–5.4 in anion%. The upconversion luminescence of Er^{3+} of the glasses, which was measured by 800 nm excitation of a 250 mW power, gave three intense emissions around 525, 550, and 660 nm, and a faint emission around 410 nm. The substitution effect on emission intensity is much more prominent in the Cl substitution than in the Br substitution. The emission intensities of the Cl substituted glasses is largely increased by substitution, exhibiting a maximum value around 6 anion%. To investigate the local structure around Er^{3+} , the emission and the phonon sideband spectra were measured for the Cl substituted glasses containing Eu^{3+} instead of Er^{3+} . The Cl substitution was found to cause the following two changes for the glasses: An increased deviation from inversion symmetry in the F^- coordination around Er^{3+} , which raises the radiative transition probability, and an increase in both phonon energy and phonon–electron coupling strength, which increases the nonradiative decay rate due to multiphonon relaxation. The Cl substitution dependence of emission intensity was deduced to be due to the competition of these

two factors. The present study concludes that the substitution of Cl^- for a small amount of F^- in ZrF_4 -based glasses is highly effective for enhancing the upconversion luminescence of Er^{3+} .

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