A Phosphoniobate with an Intersecting Tunnel Structure Related to Pyrochlore: Rb₃Nb₅P₂O₁₉

A. Leclaire, M. M. Borel, J. Chardon, and B. Raveau

Laboratoire CRISMAT, Associé au CNRS, ISMRA, Université de Caen, Boulevard du Maréchal Juin, 14050 Caen Cedex, France

Received August 2, 1993; accepted January 12, 1994 IN HONOR OF C. N. R. RAO ON HIS 60TH BIRTHDAY

A new phosphoniobate $Rb_3Nb_5P_2O_{19}$ with an intersecting tunnel structure has been synthesized. It crystallizes in the $R\bar{3}c$ space group with $a_H=12.989(2)$ Å and $c_H=53.912(6)$ Å. The host lattice, related to the pyrochlore framework, is built up from " Nb_6O_{27} " and " $Nb_3P_3O_{24}$ " units which share their corners in the (001) planes, forming $|Nb_2PO_8|_\infty$ layers. These layers, connected through NbO_6 octahedra, delimit large cages along c which contain the rubidium ions. The cages are connected through hexagonal tunnels running along $(2\ 1\ 0)_H$. The relationships with pyrochlore and $Rb_{10}Nb_{29.2}O_{78}$ are studied. © 1994 Academic Press, Inc.

INTRODUCTION

Recent studies of the niobium phosphates have shown the close relationships between these mixed frameworks and the pure octahedral host lattices. The niobium phosphate bronzes form a large family whose structures are characterized either by hexagonal tunnels or by pentagonal tunnels. This is the case for the monophosphate bronzes $ANb_3P_3O_{15}$ with A = K, Rb, Cs (1-3) that are closely related to the tetragonal tungsten bronze (4); for the phosphate niobium bronzes with hexagonal tunnels $(A_3Nb_6P_4O_{26})_n \cdot ANb_2PO_8$ with A = K, Na (5) that are closely related to the hexagonal tungsten bronzes (HTB) (6), and for the intergrowth tungsten bronzes (ITB) (7). This similarity to pure octahedral structures has also been observed very recently for a phosphoniobate RbNb₂PO₈ (8) whose structure corresponds to an ordered substitution of PO₄ tetrahedra for NbO₆ octahedra in the pure octahedral HTB framework. Curiously, no mixed framework derived from the pure octahedral pyrochlore host lattice has been obtained until now, in spite of the close relationships between HTB and pyrochlore structures. For this reason, we have more extensively investigated the Rb-Nb(V)-P-O system, especially in the region with higher Rb content. We report here the synthesis and structure of a new Nb(V) phosphate, Rb₃Nb₅P₂O₁₉, whose structure exhibits strong similarities with pyrochlore. The isotypic thallium phase was also synthesized.

SYNTHESIS AND CRYSTAL GROWTH

Single crystals of $A_3 \text{Nb}_5 P_2 O_{19}$ (A = Rb, Tl) were extracted from mixtures of nominal composition $A \text{Nb}_2 P O_8$ (A = Rb, Tl). The latter were prepared from $A_2 C O_3$, $H(NH_4)_2 P O_4$, and $Nb_2 O_5$ in appropriate proportions. The mixtures were mixed and heated to 873 K in a platinum crucible for 6 hr to decompose $H(NH_4)_2 P O_4$ and $A_2 C O_3$. The samples were heated in an evacuated silica ampoule at 1298 K for 24 hr; the ampoules were then cooled slowly (3.5 K hr⁻¹) to 1023 K. The samples were finally quenched to room temperature. From the resulting mixtures, some colorless hexagonal plates were extracted. Microprobe analysis of these crystals confirmed the composition $A_3 \text{Nb}_5 P_2 O_{19}$, in agreement with the structure determination.

Single phase powder samples of $A_3Nb_5P_2O_{19}$ with A = Rb, TI were prepared by solid state reaction in air from A_2CO_3 , $H(NH_4)_2PO_4$, and Nb_2O_5 , mixed in appropriate proportions. The mixtures were first heated in air in a platinum crucible to 653 K. After regrinding, the products were finally heated to 1273 K for 12 hr and quenched at room temperature.

Powder X-ray diffractograms registered with a PW 3710 Philips diffractometer were indexed in a hexagonal cell (Table 1) with parameters obtained from the single crystal study (Table 2).

STRUCTURE DETERMINATION

Two colorless hexagonal plates were selected for structure determination, with dimensions $0.116 \times 0.090 \times 0.026$ mm and $0.064 \times 0.038 \times 0.019$ mm for the rubidium and thallium phases, respectively. The cell parameters were determined and refined by diffractometric techniques at 294 K with a least-squares refinement based upon 25 reflections with $18^{\circ} < \theta < 22^{\circ}$ (Table 2).

The data were collected on a CAD 4 Enraf-Nonius diffractometer as reported in Table 3. Unfortunately, for $Tl_3Nb_5P_2O_{19}$, the small number of reflections with $I \ge 3$

	TABLE	1	
Interreticular	Distances	for	Rb ₃ Nb ₅ P ₂ O ₁₉

h	k	1	d_{calc} (Å)	d_{obs} (Å)	I_0	h	k	1	d_{calc} (Å)	d_{obs} (Å)	I_0
1	0	4	8.636	8.631	27	1	1	21	2.387	2.388	3.
1	1	0	6.495	6.490	15.5	0	3	18	2.340	2.342	3.
1	1	3	6.108	6.103	8.5	2	3	11	2.283	2.285	3.
0	1	8	5.781	5.779	20.	0	2	22	2.246	2.246	5.
1	1	6	5.264	5.263	7.	0	0	24	2.246		
1	0	10	4.861	4.857	8.	5	0	2	2.242	2.241	1.
0	0	12	4.493	4.500	10.	3	3	6	2.105	2.105	3.
1	1	9	4.403	4.405	1.	1	2	23	2.053	2.050	5.
2	1	1	4.238	4.243	2.	2	4	7	2.049		
0	3	0	3.750	3.751	9.5	3 `	2	16	2.049		
0	1	14	3.643	3.645	39.5	3	1	20	2.040	2.040	13
0	3	6	3.460	3.461	31.	0	1	26	2.039		
2	2	0	3.247	3.250	17.5	4	2	8	2.027	2.028	16.5
2	2	3	3.196	3.200	79.5	1	4	15	2.027		
2	0	14	3.177	3,170	22.	1	5	2	2.015	2.014	8.
1	l	15	3.145	3.144	47.5	2	2	21	2.014		
1	3	1	3.115	3.114	100.	2	4	10	1.977	1.974	17.
2	2	6	3.054	3.052	15.	5	1	7	1.954	1.953	19
3	1	5	2.997	2.999	80.	4	2	11	1.950	1.949	24.
0	0	18	2.995			3	3	12	1.950		
2	1	13	2.969	2.976	31.5	1	3	22	1.927	1.927	4.5
0	2	16	2.890	2.884	21.6	0	3	24	1.927		
0	3	12	2.879	2.873	18.	3	2	19	1.909	1.909	3.
2	2	9	2.854	2.851	8.	1	4	18	1.899	1.900	3.
1	2	14	2.854			1	0	28	1.898		
3	1	8	2.831	2.829	8.	5	1	10	1.892	1.894	3.
4	Ô	4	2.753	2.749	12.5	2	4	13	1.892		
1	1	18	2.720	2.720	9.5	6	0	0	1.875	1.875	18.5
3	1	11	2.632	2.635	7.5	3	1	23	1.874		
2	2	12	2.632			0	5	16	1.871	1.870	10.
3	2	1	2.578	2.580	4.	2	3	20	1.864	1.864	4.5
2	3	2	2.569	2.570	4.5	1	2	26	1.864	11001	

 σ (I) (only 799) that could be collected did not allow the structure to be refined. The reflections of the rubidium phase were corrected for Lorentz, polarization, absorption (Gaussian method), and secondary extinction effects.

The structure was solved by the heavy atom method and refined by the full matrix least-squares method. The refinement of the atomic coordinates and their thermal anisotropic factors led to R=0.031 and $R_{\rm W}=0.030$ and to the atomic parameters summarized in Table 4.

DESCRIPTION OF THE STRUCTURE AND DISCUSSION

The projection of the structure of $Rb_3Nb_5P_2O_{19}$ along [210] (Fig. 1a) shows that the host lattice $[Nb_5P_2O_{19}]_{\infty}$

TABLE 2 Crystallographic Parameters

Compound	a (Å)	b (Å)	c (Å)	γ(°)	V (Å ³)
$Rb_3Nb_5P_2O_{19} \\ Tl_3Nb_5P_2O_{19}$	12.989(2)	12.989(2)	53.912(6)	120	7877(40)
	13.027(3)	13.027(3)	53.616(6)	120	7878(70)

consists of corner-sharing NbO_6 octahedra and PO_4 tetrahedra. In this framework, each PO_4 tetrahedron shares its four apices with four octahedra, whereas there exist four kinds of NbO_6 octahedra. The $Nb(1)O_6$ octahedra share their six apices with six NbO_6 octahedra, the $Nb(2)O_6$ octahedra are linked to three NbO_6 octahedra and three PO_4 tetrahedra, the $Nb(3)O_6$ octahedra share four apices with four NbO_6 octahedra and two PO_4 tetrahedra, and the $Nb(4)O_6$ octahedra are linked to five NbO_6 octahedra and one PO_4 tetrahedron.

The PO₄ tetrahedron has the classical geometry of monophosphate groups, as shown in Table 5. The "O₆" octahedra that ensure the coordination of Nb(V) are not strongly distorted. The mean Nb-O distance does not vary dramatically from one octahedron to the other (1.959 to 2.001 Å). On the other hand, a significant variation of this distance is observed by comparing one Nb-O-Nb bond with one Nb-O-P bond. For Nb(1), located on a 3 axis, one observes, of course, six identical Nb-O distances corresponding to Nb-O-Nb bonds. On the opposite, the other Nb-O-P bonds of Nb(2) are significantly larger (2.11 Å) than the three Nb-O-Nb bonds (1.888 Å)

28 LECLAIRE ET AL.

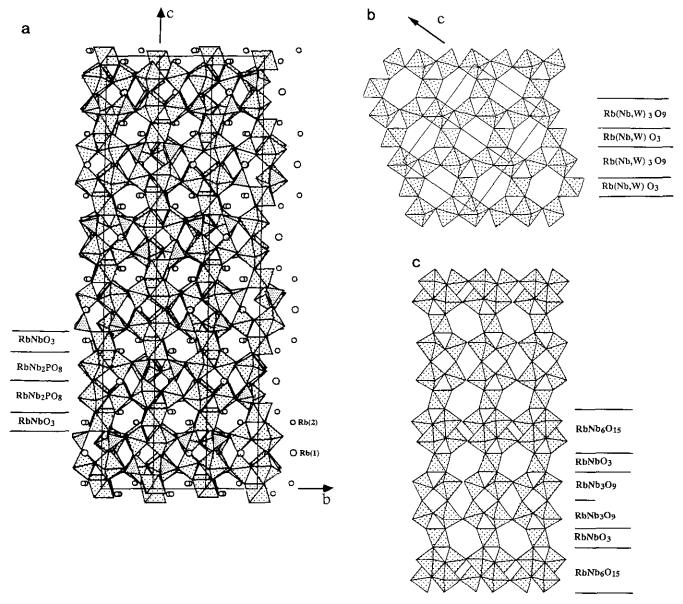


FIG. 1. The six-sided tunnels running along (a) [210]_H in Rb₃Nb₅P₂O₁₉, (b) [110]_H in pyrochlore, and (c) [110]_H in Rb₁₀Nb_{29,2}O₇₈

Similar behavior is observed for Nb(3) and Nb(4), which exhibit Nb-O-P bonds significantly larger than the Nb-O-Nb bonds (Table 5).

The $[{\rm Nb_5P_2O_{19}}]_{\infty}$ framework can be described from two sorts of structural units: the octahedral blocks "Nb₆O₂₇" previously encountered in the intersecting tunnel structure Rb₁₀Nb_{29.2}O₇₈ (9) and the mixed octahedral-tetrahedral units "Nb₃P₃O₂₄" built up from three Nb(3)O₆ octahedra and three PO₄ tetrahedra. Along $\bf c$ these two kinds of units are linked through an Nb(1)O₆ or an Nb(2)O₆ octahedron whose ternary axis is parallel to $\bf c$, forming columns (Fig. 2a) according to the sequence

$$\begin{split} [-Nb(1)-Nb_3P_3O_{24}-Nb(2)-Nb_6O_{27} \\ -Nb(2)-Nb_3P_3O_{24}--Nb(1)-]_2. \end{split}$$

In each column the successive Nb_6O_{27} (or $Nb_3P_3O_{24}$) blocks are related one to the other through the $\overline{3}$ axis that coincides with the axis of the column. Laterally, i.e., in the $(001)_H$ plane, three columns share the corners of their polyhedra, forming $[Nb_5P_2O_{23}]_{\mathbb{Z}}$ layers parallel to $(110)_H$ (Fig. 3) that exhibit two kinds of hexagonal windows, built up from four octahedra and two tetrahedra and from five octahedra and one tetrahedron, respectively. The stacking of the $[Nb_5P_2O_{23}]_{\mathbb{Z}}$ layers along $[110]_H$ leads to the tridimensional framework $[Nb_5P_2O_{19}]_{\mathbb{Z}}$; in this the hexagonal windows delimit tunnels running along $[210]_H$ (Fig. 1) and zig-zag tunnels parallel to \mathbf{c} (Fig. 4).

In fact, this framework exhibits strong similarities with those of the pyrochlore RbNbWO₆ and of the niobate Rb₁₀Nb_{29,2}O₇₈ (9), as shown by the comparison of the

TABLE 3
Summary of Crystal Data, Intensity Measurements, and Structure Refinement Parameters for Rb₃Nb₅P₂O₁₉

i. Cr	ystal data
Space group	$R\overline{3}c$
Cell dimensions	a = 12.989 (2) Å
	$b = 12.989 (2) \text{ Å } \gamma = 120^{\circ}$
	c = 53.912 (6) Å
Volume	7877(3) Å ³
Z	18
$d_{\mathrm{calc.}}$	4.12
$d_{\text{exp.}}$	4.07
	measurements
$\lambda (MoK\alpha)$	0.71 07 3 Å
Scan mode	$\omega - \frac{4}{3}\theta$
Scan width (°)	$1 + 0.35 \tan \theta$
Slit aperture (mm)	$1.06 + \tan \theta$
max θ (°)	38°
Standard reflections	3 (every 3000 sec)
Reflections with $I > 3\sigma$	2337
Measured reflections	10,064
$\mu \text{ (mm}^{-1})$	11.38
 Structure sol 	ution and refinement
Parameters refined	134
Agreement factors	$R = 0.031, R_{\rm w} = 0.030$
Weighting scheme	$w = f(\sin \theta/\lambda)$
Δ/σ max	< 0.004

[210]_H projection (Fig. 1a) with those of these structure along [110]_H (Figs. 1b,1c). The three structures are characterized by a similar disposition of the hexagonal tunnels running along [210]_H in the phosphoniobate and along [110]_H in the two other phases. The three structures con-

TABLE 4
Positional Parameters and Their Estimated Standard Deviations

Atom	х	у	z	$B(\mathring{\mathbf{A}}^2)$
Nb(1)	0.	0.	0.	0.83(2)
Nb(2)	0.	0.	0.15913(2)	0.54(1)
Nb(3)	0.16987(4)	0.14913(4)	0.05416(1)	0.530(8)
Nb(4)	0.01723(4)	0.16947(5)	0.21506(1)	0.784(9)
Rb(1)	0.6723(1)	0.	0.25	3.47(3)
Rb(2)	0.6529(1)	0.0031(1)	0.32112(2)	4.87(3)
P	0.2187(1)	0.2165(1)	0.11778(3)	0.46(2)
O(1)	0.1263(3)	0.1223(3)	0.02055(7)	0.79(8)
O(2)	0.1189(3)	0.1319(4)	0.13426(8)	1.00(8)
O(3)	0.0063(3)	0.1255(3)	0.17797(7)	0.73(7)
O(4)	0.1744(4)	0.2056(4)	0.40892(8)	1.35(9)
O(5)	0.1230(3)	-0.0095(3)	0.06028(8)	0.81(8)
O(6)	0.3314(3)	0.1957(3)	0.04485(8)	0.72(8)
O(7)	0.2189(4)	0.3299(4)	0.04840(9)	1.42(9)
O(8)	0.1325(3)	0.1256(3)	0.21878(8)	0.91(8)
O(9)	0.	0.1901(4)	0.25	0.9(1)
O(10)	-0.1041(3)	0.2272(3)	0.20524(8)	0.89(8)

Note. Anisotropically refined atoms are given in the isotropic equivalent displacement parameter, defined as $B = \frac{4}{3} \left[\beta_{11} a^2 + \beta_{22} b^2 + \beta_{33} c^2 + \beta_{12} ab \cos \gamma + \beta_{13} ac \cos \beta + \beta_{23} bc \cos \alpha \right]$.

tain identical [RbNbO₃]_∞ layers involving the isolated Nb(1) and Nb(2) octahedra whose ternary axes are parallel to c_H . They differ by the nature of the layers that are sandwiched between the [RbNbO₁]_∞ layers. In the phosphoniobate one observes double [RbNb₂PO₈]_x layers stacked with the [RbNbO₃]_x layers according to the sequence (RbNb₂PO₈)₂ · RbNbO₃ (Fig. 1a). In the pyrochlore, the double phosphoniobate layers are replaced by single HTB layers [RbNb₃O₉]_∞ (Fig. 1b), so that the sequence (RbNb₃O₉). RbNbO₃ is observed. In the niobate Rb₁₀Nb_{29,2}O₇₈ there exist two kinds of layers between the RbNbO₃ layers (Fig. 1c), double HTB [RbNb₃O₉]_∞ layers and "sheared" HTB layers [RbNb₆O₁₅]_∞, leading to the sequence $(RbNb_3O_9)_2 \cdot RbNbO_3 \cdot (RbNb_6O_{15}) \cdot RbNbO_3$. The similarity between the double HTB layers [RbNb₃O₉]₂ and the double phosphate layers [RbNb₂PO₈], is also seen by considering the columns of polyhedra running along c

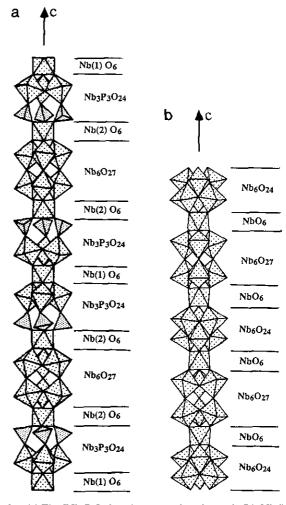


FIG. 2. (a) The $[Nb_5P_2O_{25}]_x$ column running along e in $Rb_3Nb_3P_2O_{19}$, built up from Nb_6O_{27} and $Nb_3P_3O_{24}$ blocks linked through NbO_6 octahedra; (b) the $[Nb_{14}O_{51}]_x$ column running along e in $Rb_{10}Nb_{29,2}O_{78}$, built up from Nb_6O_{27} and Nb_6O_{24} octahedral blocks linked through NbO_6 octahedra.

TABLE~5 Distances (Å) and Angles (°) in the Octahedra and Tetrahedra for $Rb_3Nb_5P_2O_{19}$

	Distances (A) a	` _ ` _ ` _			, ,	
Nb(1)	O(1 ⁱ)	O(1)	O(1ii)	O(1 ⁱⁱⁱ)	O(1 ^{iv})	O(1 ^v)
O(1i)	1.959(1)	3.917(9)	2.742(9)	2.798(9)	2.742(9)	2.798(9)
O(1)	180.	1.959(1)	2.798(9)	2.742(9)	2.798(9)	2.742(9)
$O(1^{ii})$	91.2(2)	88.8(2)	1.959(1)	3.917(9)	2.798(9)	2.742(9)
O(1 ⁱⁱⁱ)	88.8(2)	91.2(2)	180.	1.959(1)	2.742(9)	2.798(9)
O(1iv)	91.2(2)	88.8(2)	91.2(2)	88.8(2)	1.959(1)	3.917(9)
O(1 ^v)	88.8(2)	91.2(2)	88.8(2)	91.2(2)	180.	1.959(1)
Nb(2)	O(2)	$O(2^{ii})$	O(2iv)	O(3)	O(3 ⁱⁱ)	O(3iv)
O(2)	2.115(5)	2.833(8)	2.833(8)	2.754(8)	3.990(8)	2.964(8)
O(2 ⁱⁱ)	84.1(2)	2.115(5)	2.833(8)	2.964(8)	2.754(8)	3.990(8)
$O(2^{iv})$	84.1(2)	84.1(2)	2.115(5)	3.990(8)	2.964(8)	2.754(8)
O(3)	86.7(2)	95.4(2)	170.8(2)	1.888(5)	2.756(8)	2.756(8)
O(3ii)	170.8(2)	86.7(2)	95.4(2)	93.8(2)	1.888(5)	2.756(8)
O(iv)	95.4(2)	170.8(2)	86.7(2)	93.8(2)	93.8(2)	1.888(5)
Nb(3)	O(1)	O(4vi)	O(5)	O(5 ⁱⁱ)	O(6)	O(7)
O(1)	1.879(5)	3.909(8)	2.729(8)	2.666(8)	2.681(8)	2.780(8)
O(4 ^{iv})	176.2(2)	2.035(5)	2.655(8)	2.855(8)	2.917(8)	3.009(8)
O(5)	93.7(2)	85.8(2)	1.863(5)	2.879(8)	2.812(8)	3.987(8)
O(5 ⁱⁱ)	86.5(2)	89.8(2)	95.9(3)	2.011(5)	3.927(8)	2.722(8)
O(6)	89.3(2)	94.6(2)	95.5(2)	168.1(2)	1.937(5)	2.786(5)
O(7)	87.8(2)	92.7(2)	177.6(2)	82.2(2)	86.4(2)	
						2.125(5)
Nb(4)	O(3)	O(6 ^{vii})	O(8)	O(8 ⁱⁱ)	O(9)	O(10)
O(3)	2.065(5)	2.992(9)	2.743(9)	2.740(9)	3.982(9)	2.802(9)
O(6 ⁱⁱ)	96.8(2)	1.937(5)	2.718(9)	3.931(9)	2.772(9)	2.745(9)
O(8)	88.5(2)	91.4(2)	1.866(5)	2.907(9)	2.817(9)	3.972(9)
O(iii)	84.6(2)	171.1(2)	97.5(3)	2.006(5)	2.697(9)	2.828(9)
O(9)	170.4(2)	91.6(2)	96.0(2)	86.4(2)	1.931(2)	2.922(9)
O(10)	84.0(2)	85.0(2)	171.2(2)	86.4(2)	92.1(2)	2.123(5)
	P	O(2)	O(4 ^{vi})	O(7 ^{viii})	O(10 ⁱ	x)
	O(2)	1.501(5)	2.624/65	2.461(6)	2.4610	
	U(2)	1.301(3)	2.324(0)	2.461(6)		
		1.501(5) 113.4(3)	2.524(6) 1.519(5)	2.461(6) 2.487(6)		6)
	O(4 ^{vi})	113.4(3)	1.519(5)	2.487(6)	2.4760	
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3)	1.519(5) 108.1(3)	2.487(6) 1.504(5)	2.476(2.538(6)
	O(4 ^{vi})	113.4(3)	1.519(5) 108.1(3) 107.9(3)	2.487(6) 1.504(5) 110.8(3)	2.4760 2.5380 1.5440	6) 5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3)	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x)	2.487(6) 1.504(5) 110.8(3) = 3.432(5)	2.476(2.538(1.544(Rb(2)-O(1 ^{xviii}	(6) (5) (5) = 3.133(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3)	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(4 ^{xi})	2.487(6) 1.504(5) 110.8(3) = 3.432(5) = 3.432(5)	2.476(2.538(1.544(Rb(2)=O(1 ^{xviii} -O(3 ^{xv}) -O(6 ^{xii})	(6) (5) (5) (6) (6) (7) (8) (8) (8) (9) (9) (9) (9) (9) (9) (9) (9) (9) (9
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3)	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(5 ^{xii}) -O(5 ^{xiii})	2.487(6) 1.504(5) 110.8(3) = 3.432(5) = 3.432(5) = 2.984(5) = 2.984(5)	2.476(2.538(1.544(Rb(2)-O(1 ^{xviii} -O(3 ^{xv}) -O(6 ^{xii}) -O(6 ^{xviii}	(6) (5) (5) (6) (6) (7) (8) (8) (8) (9) (9) (9) (9) (9) (9) (9) (9) (9) (9
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3)	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(5 ^{xiii}) -O(5 ^{xiii}) -O(6 ^{xiii})	2.487(6) 1.504(5) 110.8(3) = 3.432(5) = 3.432(5) = 2.984(5) = 2.984(5) = 3.263(5)	$\begin{array}{c} 2.476(\\ 2.538(\\ 1.544(\\ Rb(2)-O(1^{xviii})\\ -O(3^{xv})\\ -O(6^{xii})\\ -O(6^{xvii})\\ -O(7^{xiv}) \end{array}$	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3)	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4^x) -O(5^{xii}) -O(5^{xii}) -O(6^{xii}) -O(6^{xii})	2.487(6) 1.504(5) 110.8(3) = 3.432(5) = 3.432(5) = 2.984(5) = 2.984(5) = 3.263(5) = 3.263(5)	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3)	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(5 ^{xii}) -O(5 ^{xii}) -O(6 ^{xiii}) -O(6 ^{xiii}) -O(7 ^{xiv})	2.487(6) 1.504(5) 110.8(3) = 3.432(5) = 3.432(5) = 2.984(5) = 2.984(5) = 3.263(5) = 3.263(5) = 3.490(5)	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3)	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(5 ^{xii}) -O(5 ^{xiii}) -O(6 ^{xiii}) -O(7 ^{xiv}) -O(7 ^{xv}) -O(7 ^{xv})	2.487(6) 1.504(5) 110.8(3) = 3.432(5) = 2.984(5) = 2.984(5) = 3.263(5) = 3.263(5) = 3.490(5) = 3.490(5) = 3.082(5)	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3)	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(5 ^{xii}) -O(5 ^{xiii}) -O(6 ^{xiii}) -O(7 ^{xiv}) -O(7 ^{xv}) -O(7 ^{xv})	2.487(6) 1.504(5) 110.8(3) = 3.432(5) = 2.984(5) = 2.984(5) = 3.263(5) = 3.263(5) = 3.490(5) = 3.490(5)	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3)	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(5 ^{xii}) -O(5 ^{xiii}) -O(6 ^{xiii}) -O(7 ^{xiv}) -O(7 ^{xv}) -O(7 ^{xv})	2.487(6) 1.504(5) 110.8(3) = 3.432(5) = 3.432(5) = 2.984(5) = 2.984(5) = 3.263(5) = 3.263(5) = 3.490(5) = 3.082(5)) = 3.082(5)	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3) 107.9(3)	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(4 ^x) -O(5 ^{xii}) -O(6 ^{xii}) -O(6 ^{xiii}) -O(7 ^{xiv}) -O(8 ^{xvi}) -O(8 ^{xvi}) -O(8 ^{xvi}	2.487(6) 1.504(5) 110.8(3) = 3.432(5) = 2.984(5) = 2.984(5) = 3.263(5) = 3.263(5) = 3.490(5) = 3.490(5) = 3.082(5) 3.082(5)	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	i: - ii: - ii: -	1.519(5) 108.1(3) 107.9(3) Rb(1)O(4 ^{xi})O(5 ^{xii})O(5 ^{xii})O(6 ^{xii})O(6 ^{xii})O(7 ^{xiv})O(7 ^{xv})O(8 ^{xvi})O(8 ^{xvi})O(8 ^{xvi})O(8 ^{xvi})	2.487(6) 1.504(5) 110.8(3) = 3.432(5) = 2.984(5) = 2.984(5) = 3.263(5) = 3.263(5) = 3.490(5) = 3.490(5) = 3.082(5) = 3.082(5)	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3) 107.9(3) i: - ii: - iii: y	1.519(5) 108.1(3) 107.9(3) Rb(1)O(4 ^{xi})O(5 ^{xii})O(5 ^{xii})O(6 ^{xii})O(6 ^{xii})O(7 ^{xiv})O(8 ^{xvi})O(8 ^{xvi})O(8 ^{xvi})O(8 ^{xvi})	2.487(6) 1.504(5) 110.8(3) = 3.432(5) = 2.984(5) = 2.984(5) = 3.263(5) = 3.263(5) = 3.490(5) = 3.490(5) = 3.082(5)) = 3.082(5) 2.22 - 2	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	i: - ii: - ii: - iii: y iv: -	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(4 ^{xi}) -O(5 ^{xii}) -O(6 ^{xii}) -O(6 ^{xii}) -O(7 ^{xiv}) -O(7 ^{xv}) -O(8 ^{xvi}) -O(8 ^{xvi}) -V Symmetry code x y x -y y x - y x + y x + y	2.487(6) 1.504(5) 110.8(3) = 3.432(5) = 2.984(5) = 2.984(5) = 3.263(5) = 3.263(5) = 3.490(5) = 3.490(5) = 3.082(5) = 3.082(5)	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3) 107.9(3) i: - ii: - iii: y iv: - v: x	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(4 ^{xi}) -O(5 ^{xii}) -O(6 ^{xii}) -O(6 ^{xii}) -O(7 ^{xv}) -O(8 ^{xvi}) -O(8 ^{xvi}) -O(8 ^{xvi}) -Ox Symmetry code x	2.487(6) 1.504(5) 110.8(3) = 3.432(5) = 2.984(5) = 2.984(5) = 3.263(5) = 3.263(5) = 3.490(5) = 3.490(5) = 3.082(5) = 3.082(5) = -z z -z ½-z	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3) 107.9(3) i: — ii: — iii: y iv: — v: x vi: y vi: y vi: 3	1.519(5) 108.1(3) 107.9(3) Rb(1)O(4 ^{xi})O(5 ^{xii})O(5 ^{xii})O(6 ^{xiii})O(6 ^{xiii})O(7 ^{xiv})O(8 ^{xvi})O(8 ^{xvi})O(8 ^{xvi})O(8 ^{xvi})	$\begin{array}{c} 2.487(6) \\ 1.504(5) \\ 110.8(3) \end{array}$ $= 3.432(5) \\ = 3.432(5) \\ = 2.984(5) \\ = 3.263(5) \\ = 3.263(5) \\ = 3.490(5) \\ = 3.490(5) \\ = 3.082(5) \\ = 3.082(5) \end{array}$ $\begin{array}{c} -2 \\ z \\ -z \\ \frac{1}{6} + z \end{array}$	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3) 107.9(3) ii: — iii: y iv: — v: x vi: y vii: $\frac{1}{3}$	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(5 ^{xii}) -O(5 ^{xii}) -O(6 ^{xiii}) -O(6 ^{xiii}) -O(7 ^{xiv}) -O(8 ^{xvii}) -O(8 ^{xvii}) Symmetry code x	$\begin{array}{c} 2.487(6) \\ 1.504(5) \\ 110.8(3) \\ = 3.432(5) \\ = 2.984(5) \\ = 2.984(5) \\ = 3.263(5) \\ = 3.263(5) \\ = 3.490(5) \\ = 3.490(5) \\ = 3.082(5) \\ = 3.082(5) \\ \end{array}$	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3) 107.9(3) i: - ii: - iii: y iv: - v: x vi: y vii: 1/3 viii: 1/3 ix: 1/3	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(4 ^{xi}) -O(5 ^{xii}) -O(6 ^{xiii}) -O(6 ^{xiii}) -O(7 ^{xv}) -O(8 ^{xvi}) -O(8 ^{xvi}) -O(8 ^{xvi}) -X - y $x - y$ $x + y$ $x - y$ $x + y$ $x + y$ $x - y$ $x + y - x$ $x + y - y$ $x + y$ $x + y - y$ $x + y$	$\begin{array}{c} 2.487(6) \\ 1.504(5) \\ 110.8(3) \\ = 3.432(5) \\ = 2.984(5) \\ = 2.984(5) \\ = 3.263(5) \\ = 3.263(5) \\ = 3.490(5) \\ = 3.490(5) \\ = 3.082(5) \\ = 3.082(5) \\ = 2.20 \\ -2.20 \\ $	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3) 107.9(3) i: - ii: - iii: y iv: - v: x vi: y vii: 1/3 viii: 1/3 ix: 1/3	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(4 ^x) -O(5 ^{xii}) -O(5 ^{xii}) -O(6 ^{xii}) -O(7 ^{xiv}) -O(8 ^{xvi}) -O(8 ^{xvi}) -O(8 ^{xvi}) Symmetry code $x -y \\ y x - y \\ -x + y -x \\ -y x \\ + x - y & \frac{3}{3} - x \\ + y - \frac{1}{3} - x \\ - \frac{1}{3} - \frac{1}{3} - x \\ + \frac{1}{3} - \frac{1}{3} - x \\ -\frac{1}{3} - \frac{1}{3} - \frac{1}{3} - x \\ -\frac{1}{3} - \frac{1}{3} - \frac{1}{3} - x \\ -\frac{1}{3} - \frac{1}{3} $	2.487(6) $1.504(5)$ $110.8(3)$ $= 3.432(5)$ $= 2.984(5)$ $= 2.984(5)$ $= 3.263(5)$ $= 3.263(5)$ $= 3.490(5)$ $= 3.082(5)$ $= 3.082(5)$ $= -z$ z $-z$ $= -z$ z $+ y$ $= -z$ $= + y$	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3) 107.9(3) ii: — iii: y iv: — v: x vi: y vii: $\frac{1}{3}$ vii: $\frac{1}{3}$ x: $\frac{3}{3}$	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(4 ^x i) -O(5 ^{xii}) -O(6 ^{xiii}) -O(6 ^{xiii}) -O(7 ^{xiv}) -O(8 ^{xvi}) -O(8 ^{xvi}) -O(8 ^{xvi}) -Y x - y x + y - x -y + x - y + x - y + x - y + x + y + x + y + x + x + x + x + x + x + x + x + x + x	$2.487(6)$ $1.504(5)$ $110.8(3)$ $= 3.432(5)$ $= 2.984(5)$ $= 2.984(5)$ $= 3.263(5)$ $= 3.263(5)$ $= 3.490(5)$ $= 3.082(5)$ $= 3.082(5)$ $= -z$ z $-z$ $\frac{1}{2} - z$ $\frac{1}{6} + z$ $\frac{1}{6} - z$ $+ y$ $-\frac{1}{6} + z$ y $-\frac{1}{6} + z$	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3) 107.9(3) i: - ii: - iii: y iv: - v: x vi: y vii: 1 ii: y ix: 1 ix	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(4 ^{xi}) -O(5 ^{xii}) -O(5 ^{xiii}) -O(6 ^{xiii}) -O(6 ^{xiii}) -O(7 ^{xv}) -O(8 ^{xvi})	$\begin{array}{c} 2.487(6) \\ 1.504(5) \\ 110.8(3) \\ = 3.432(5) \\ = 3.432(5) \\ = 2.984(5) \\ = 2.984(5) \\ = 3.263(5) \\ = 3.263(5) \\ = 3.490(5) \\ = 3.082(5) \\ = 3.082(5) \\ = 2.25 \\ -2.25 \\ $	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3) 107.9(3) i: - ii: - iii: - iii: y iv: - v: x vi: y vii: 3 xi: 3 xi: 3 xi: 3 xii: 3 xii: 3 xii: 3 xii: 3	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(4 ^x) -O(5 ^x) -O(5 ^x) -O(6 ^x) -O(7 ^x) -O(8	$\begin{array}{c} 2.487(6) \\ 1.504(5) \\ 110.8(3) \\ = 3.432(5) \\ = 3.432(5) \\ = 2.984(5) \\ = 2.984(5) \\ = 3.263(5) \\ = 3.263(5) \\ = 3.490(5) \\ = 3.082(5) \\ = 3.082(5) \\ = 3.082(5) \\ = 2 \\ -z \\ z \\ -z \\ -z \\ -z \\ -z \\ -z \\ -$	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3) 107.9(3) i: - ii: - ii: y iv: - v: x vi: y vii:	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(4 ^x) -O(5 ^x) -O(5 ^x) -O(6 ^x) -O(7 ^x) -O(8	$\begin{array}{c} 2.487(6) \\ 1.504(5) \\ 110.8(3) \\ = 3.432(5) \\ = 2.984(5) \\ = 2.984(5) \\ = 3.263(5) \\ = 3.263(5) \\ = 3.490(5) \\ = 3.082(5) \\ = 3.082(5) \\ = 3.082(5) \\ = -z \\ $	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3) 107.9(3) i:	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(5 ^{xii}) -O(5 ^{xii}) -O(6 ^{xiii}) -O(6 ^{xiii}) -O(7 ^{xiv}) -O(8 ^{xvi})	$\begin{array}{c} 2.487(6) \\ 1.504(5) \\ 110.8(3) \\ = 3.432(5) \\ = 3.432(5) \\ = 2.984(5) \\ = 3.263(5) \\ = 3.263(5) \\ = 3.490(5) \\ = 3.082(5) \\ = 3.082(5) \\ = 3.082(5) \\ = 3.082(5) \\ \end{array}$	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3) 107.9(3) i: - ii: - iii: - iii: y vi: x vi: y viii: 1 xi: 3 xi: 3 xi: 3 xi: 3 xii: 3 xiv: 3 xvi: 1 xvii: 1	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(5 ^{xii}) -O(5 ^{xii}) -O(6 ^{xiii}) -O(6 ^{xiii}) -O(6 ^{xiii}) -O(8 ^{xvii}) -O(8 ^{xvii}) -O(8 ^{xvii}) Symmetry code $\begin{array}{ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} 2.487(6) \\ 1.504(5) \\ 110.8(3) \\ = 3.432(5) \\ = 3.432(5) \\ = 2.984(5) \\ = 2.984(5) \\ = 3.263(5) \\ = 3.263(5) \\ = 3.490(5) \\ = 3.082(5) \\ = 3.$	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)
	O(4 ^{vi}) O(7 ^{viii})	113.4(3) 108.1(3) 107.9(3) i: - ii: - iii: y iv: - v: x vi: y vii: is: is: is: is: is: is: is: is: is:	1.519(5) 108.1(3) 107.9(3) Rb(1)-O(4 ^x) -O(5 ^{xii}) -O(5 ^{xii}) -O(6 ^{xiii}) -O(6 ^{xiii}) -O(7 ^{xiv}) -O(8 ^{xvi})	$\begin{array}{c} 2.487(6) \\ 1.504(5) \\ 110.8(3) \\ = 3.432(5) \\ = 3.432(5) \\ = 2.984(5) \\ = 3.263(5) \\ = 3.263(5) \\ = 3.490(5) \\ = 3.082(5) \\ = 3.082(5) \\ = 3.082(5) \\ = 3.082(5) \\ \end{array}$	2.476(2.538(1.544(Rb(2)-O(1x ^{viii} -O(3x ^v) -O(6x ^{vii}) -O(6x ^{viii} -O(7x ^{iv}) -O(10)	6) 5) 1 = 3.133(5) = 2.959(5) = 3.163(5) 1 = 3.457(5) = 3.301(5) = 3.409(5)

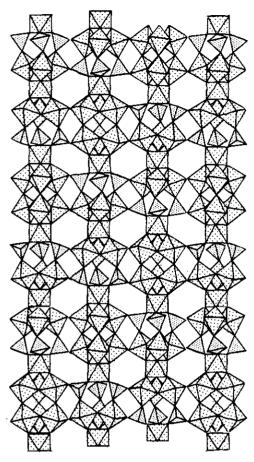


FIG. 3. The $[{\rm Nb_5P_2O_{23}}]_\pi$ layers parallel to (110), showing two kinds of hexagonal windows.

(Fig. 2). The phosphoniobate layers exhibit two kinds of blocks, " Nb_6O_{27} " and " $Nb_3P_3O_{24}$ " (Fig. 2a), whereas the double HTB layers of $Rb_{10}Nb_{29.2}O_{78}$ (Fig. 2b) can be described as built from " Nb_6O_{27} " units only.

The projections of one $[RbNb_2PO_8]_{\infty}$ layer along c_H (Fig. 5a), compared to those of the HTB $[RbNb_3O_8]_{\infty}$ layers observed in pyrochlore and in Rb₁₀Nb_{29 2}O₇₈ (Fig. 5b) and to the [RbNb₆O₁₅]_x layers observed in the latter (Fig. 5c), show the great similarity of the three kinds of layers. It appears clearly that the [RbNbO₈]_∞ layers exhibit HTB [NbO₃]_x chains waving along \mathbf{a}_{H} (Fig. 5a). However, their connection through PO₄ tetrahedra leads to seven-sided windows instead of the six-sided windows in pyrochlore or in Rb₁₀Nb_{29.2}O₇₈. The [RbNb₂PO₈]_x layers of this new phase exhibit a different arrangement from that of the phosphoniobate RbNb₂PO₈ (8). One indeed observes from the projection of RbNb₂PO₈ structure along **b** (Fig. 5d) that this structure is also closely related to the HTBs, but it forms two kinds of windows: hexagonal windows limited by five NbO₆ octahedra and one PO₄ and heptagonal windows formed by four octahedra and three tetrahedra, whereas for Rb₃Nb₅P₂O₁₉ one observes one kind of heptagonal window involving five octahedra and two tetrahedra.

The double $[\mathrm{Nb_2PO_8}]_\infty$ layers delimit two kinds of large cages where the rubidium ions are located (Fig. 4). These cages are connected through hexagonal tunnels running along $[210]_H$. The Rb(1) ions, surrounded by 10 oxygen atoms, are located at the centers of the hexagonal window in the double layers (Fig. 6). The Rb(2) ions, which exhibit a sevenfold coordination, are located at the level of Nb(1) and Nb(2) at the intersection of two [210] tunnels, at the boundary between the $[\mathrm{Nb_2PO_8}]_\infty$ and $[\mathrm{NbO_3}]_\infty$ layers.

CONCLUSION

A phosphoniobate closely related to the pure octahedral structures of pyrochlore and Rb₁₀Nb_{29,2}O₇₈ has been synthesized for the first time. The similarities between these oxides suggest that many other mixed frameworks derived from this structural type should be generated. The ion

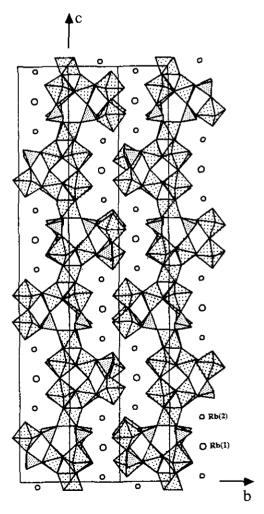


FIG. 4. The zig-zag tunnels running along ${\bf c}$ that contain the rubidium ions.

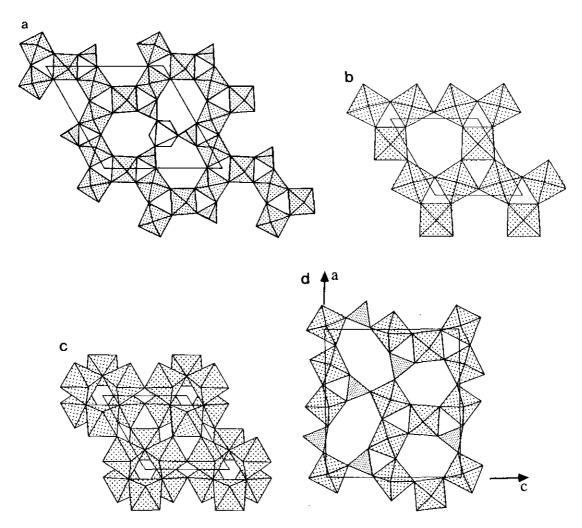


FIG. 5. (a) The $[RbNb_2PO_8]_x$ layer in $Rb_3Nb_5P_2O_{19}$ (the Rb atoms are omitted); (b) the HTB $[RbNb_3O_8]_x$ layer in pyrochlore and in $Rb_{10}Nb_{29,2}O_{78}$; (c) the $[RbNb_6O_{15}]_x$ layer in $Rb_{10}Nb_{29,2}O_{78}$; (d) the $[RbNb_2PO_8]_x$ layer in $RbNb_2PO_8$.

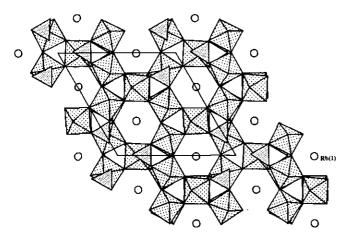


FIG. 6. The hexagonal windows containing Rb(1) in the $[RbNb_2PO_8]_{x}$ double layer.

exchange properties of this open structure will be investigated.

REFERENCES

- A. Leclaire, M. M. Borel, A. Grandin, and B. Raveau, J. Solid State Chem. 80, 12 (1989).
- M. M. Borel, A. Benabbas, H. Rebbah, A. Grandin, A. Leclaire, and B. Raveau, Eur. J. Solid State Inorg. Chem. 27, 525 (1990).
- M. M. Borel, A. Grandin, G. Costentin, A. Leclaire, and B. Raveau, Mater. Res. Bull. 25, 1155 (1990).
- 4. A. Magneli, Arkiv. Kemi 1, 213 and 269 (1949).
- B. Raveau, M. M. Borel, A. Leclaire, and A. Grandin, Inter. J. Modern Phys. B7, 4109 (1993).
- 6. A. Hussain and L. Kihlborg, Acta Crystallogr. Sect. A 32, 551 (1976).
- A. Benabbas, M. M. Borel, A. Grandin, A. Leclaire, and B. Raveau, J. Solid State Chem. 87, 360 (1990).
- 8. A. Leclaire, M. M. Borel, A. Grandin, and B. Raveau, J. Solid State Chem., in press.
- 9. C. Michel, A. Guyomarc'h, and B. Raveau, *J. Solid State Chem.* **25**, 251 (1978).