

free-ion factor is assumed. These results can be interpreted in terms of an extended spin density distribution due to covalent bonding in the  $\text{ReO}_6$  clusters.

*Les Phases  $\text{Na}_4\text{XO}_4$  ( $X = \text{Si, Ti, Cr, Mn, Co, Ge, Sn, Pb}$ ) et  $\text{K}_4\text{XO}_4$  ( $X = \text{Ti, Cr, Mn, Ge, Zr, Sn, Hf, Pb}$ ).* R. OLAZCUAGA, J. M. REAU, M. DEVALETTE, G. LE FLEM, AND P. HAGENMULLER. Service de Chimie Minérale Structurale de l'Université de Bordeaux I, C.N.R.S., 351 cours de la Libération, 33405 Talence, France. The isotypic  $\text{Na}_4\text{XO}_4$  ( $X = \text{Si, Ti, Cr, Mn, Co, Ge, Sn, Pb}$ ) and  $\text{K}_4\text{XO}_4$  ( $X = \text{Ti, Cr, Mn, Ge, Zr, Sn, Hf, Pb}$ ) phases crystallize in the triclinic system. Optical and magnetic properties of the chromium, manganese and cobalt compounds show that the transition element has a tetrahedral surrounding.

*First Order Magnetic Transition, Magnetic Structure and Vacancy Distribution in  $\text{Fe}_2\text{P}$ .* R. WÄPPLING, L. HÄGGSTRÖM, T. ERICSSON, S. DEVANARAYANAN, E. KARLSSON, B. CARLSSON, AND S. RUNDQVIST. Institute of Physics and Chemistry, Uppsala University, S-751 21 Uppsala 1, Sweden. The para to ferromagnetic transition in  $\text{Fe}_2\text{P}$  has been studied using Mössbauer spectroscopy. The magnetic hyperfine fields drop abruptly from about half of their saturation values to zero at 214.5°K indicating a first order transition. The isomer shifts show a discontinuous change at the transition point. For some samples the transition takes place over a wide temperature range, probably due to impurities and other imperfections in the samples. From the magnetic hyperfine fields at 15°K the magnetic moments can be deduced to be 1.14  $\mu\text{B}$  and 1.78  $\mu\text{B}$  for Fe(1) and Fe(2), respectively. An assignment of the components in the Mössbauer spectra to the two crystallographically nonequivalent iron positions has been made from the temperature variation of the spectra. The ordering of metal vacancies has been investigated by a Mössbauer study of a nonstoichiometric  $\text{Fe}_2\text{P}$  sample and by an X-ray diffraction study of a nonstoichiometric  $\text{Mn}_2\text{P}$  crystal.

*Mössbauer and X-Ray Studies of  $\text{Fe}_5\text{PB}_2$ .* L. HÄGGSTRÖM, R. WÄPPLING, T. ERICSSON, Y. ANDERSSON, AND S. RUNDQVIST. Institute of Chemistry and Institute of Physics, Uppsala University, S-751 21 Uppsala 1, Sweden. Mössbauer spectroscopic and X-ray crystallographic studies of the ternary compound  $\text{Fe}_5\text{PB}_2$  have been undertaken. It is suggested that in the nonstoichiometric compound part of the phosphorus positions are occupied by boron atoms. The Mössbauer data indicate that  $\text{Fe}_5\text{PB}_2$  is a simple ferromagnet with the tetragonal axis as the direction of easy magnetization, the moments being  $\sim 1.6 \mu\text{B}$  and  $\sim 2.2 \mu\text{B}$  for Fe(1) and Fe(2), respectively.

*Les Bronzes de Cobalt  $\text{K}_x\text{CoO}_2$  ( $x < 1$ ). L'Oxyde  $\text{KCoO}_2$ .* C. DELMAS, C. FOUASSIER, AND P. HAGENMULLER. Service de Chimie Minérale Structurale de l'Université de Bordeaux I, C.N.R.S., 351 cours de la Libération, 33405 Talence, France. The reaction between potassium and cobalt oxides leads to two bronze-type phases with formula  $\text{K}_x\text{CoO}_2$  ( $x = 0.50$  and  $x = 0.67$ ) and to a  $\text{KCoO}_2$  oxide with two allotropic forms. The lattice of  $\text{K}_x\text{CoO}_2$  is built up by sheets of  $(\text{CoO}_2)_n$  octahedra between which potassium ions are inserted with a prismatic coordination; they have a metallic behaviour. The structures of both  $\text{KCoO}_2$  varieties are related to the cristobalite type. The oxidation state of cobalt is discussed on the basis of structural, optical and magnetical data.