Magnetic and Electrical Properties in the Defect Perovskite System $La_{1-x}Na_xFeO_{3-\delta}$. HIROSHI YAMAMURA*, HAJIME HANEDA, SHIN-ICHI SHIRASAKI, AND KOU TAKADA, National Institute for Research in Inorganic Materials, 1-1, Namiki, Sakura-Mura, Niihari-Gun, Ibaraki, Japan. Magnetic and electrical properties were studied for the defect perovskite system $La_{1-x}Na_xFeO_{3-\delta}$. X-Ray diffraction showed that the materials with a composition range x=0.0-0.25 were monophasic perovskite-like compounds with orthorhombic symmetry. The present ferrite solid-solution showed a parasitic-ferromagnetic property, accompanying the antiferromagnetism. Both values of the ferromagnetic moment and the magnetic susceptibility increased with increase in Na content. Furthermore, the magnetization was strongly affected by oxygen partial pressure. These experimental facts seem to have a close relation with the oxygen vacancies. Electrical conductivity measurements suggested the presence of mixed valencies of Fe³⁺ and Fe⁴⁺ on a basis of the compositional dependence of the conductivity and its activation energy.

Etude structurale des composés $Fe_{0.25}TiSe_2$ et $Co_{0.25}TiSe_2$ à cristaux maclés. Surstructures et degré d'ordre des lacunes. Y. Arnaud and M. Chevreton*, Laboratoire d'Etude des Matériaux (ERA 602) Bt 303, INSA, 20 avenue Albert-Einstein, 69621 Villeurbanne Cédex, France. $Fe_{0.25}TiSe_2$ and $Co_{0.25}TiSe_2$ crystals have been studied. Meriedric twins are detected by means of X-ray diffraction, from extinction rules observed on the apparent cells that are multiples of the C6 hexagonal unit cell (a', c'). In $Fe_{0.25}TiSe_2$ [quasi-orthohexagonal monoclinic I true unit cell: $a \ (\approx a'3^{1/2} = 3.566(1) \times 3^{1/2}$; $b \ (\approx a') = 3.566(1)$; $c \ (\approx 2c') = 5.964(1) \times 2$ Å; $b \ \approx 90^\circ$ existence of a $b \ M_3 M_4$ ordering of vacancies, previously observed on powders, is confirmed by this study. In $b \ Co_{0.25}TiSe_2$, the true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$, the true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell is monoclinic $b \ Co_{0.25}TiSe_2$ he true unit cell $b \ Co_{0.25}TiSe_2$ he true unit cell $b \ Co_{0.25}TiSe_2$ he true unit cell $b \ Co_{0.25}TiSe_2$