solution (x = 0.14; 0.29; 0.43) are established in the temperature range 1220–1305°C. It is shown that in the reaction zone of interacting NiO, (Ni, Zn)O, or ZnO with Fe₂O₃ the ferrite phase crystallizes only on iron oxide. The distribution of the Fe, Ni, and Zn concentrations over the reaction layer thickness using electron probe and X-ray spectrum analysis is obtained. The interdiffusion coefficients over the investigated temperature range calculated in the (Ni, Zn, Fe)O and ferrite phases change from (0.8–7.0) × 10⁻⁹ to (1.0–12.0) × 10⁻¹⁰ cm²/sec, respectively. The interaction of (Ni, Zn)O with Fe₂O₃ takes place by the mechanism of interaction of interdiffusion of Fe³⁺, Fe²⁺, and Ni²⁺, Zn²⁺ along with a current of Zn²⁺ ions and electrons or oxygen ions directed to the ferrite/Fe₂O₃ interface.

Phase Equilibrium Relations in the Binary systems LiPO₃-CeP₃O₉ and NaPO₃-CeP₃O₉. M. RZAIGUI AND N. KBIR ARIGUIB,* Laboratoire de Physico-Chimie Minérale, Ecole Normale Supérieure, 43, Rue de la Liberté, Le Bardo, Tunis. The LiPO₃-CeP₃O₉ and NaPO₃-CeP₃O₉ systems have been investigated for the first time by DTA, X-ray diffraction, and infrared spectroscopy. Each system forms a single 1: 1 compound. LiCe(PO₃)₄ melts in a peritectic reaction at 980°C. NaCe(PO₃)₄ melts incongruently, too, at 865°C. These compounds have a monoclinic unit cell with the parameters: a = 16.415(6), b = 7.042(6), c = 9.772(7) Å, $\beta = 126.03(5)$ °, Z = 4, space group C2/c for LiCe(PO₃)₄ and a = 9.981(4), b = 13.129(6), c = 7.226(5) Å, $\beta = 89.93(4)$ °, Z = 4, space group $P2_1/n$ for NaCe(PO₃)₄. It is established that both compounds are mixed polyphosphates with chain structure of the type |M| $M_{11}^{(1)}$ (PO₃)₄|₈ $M_{11}^{(2)}$: alkali metal, $M_{11}^{(2)}$: rare earth.

Phase Relations in the Ternary W-Mo-O System. T. EKSTRÖM, E. SALJE, AND R. J. D. TILLEY,* School of Materials Science, University of Bradford, Bradford BD7 1DP, West Yorkshire, United Kingdom. The phases in the ternary W-Mo-O system have been determined using X-ray diffraction and electron microscopy. Series of mixed crystals occur for the fully oxidized compounds $W_xMo_{1-x}O_3$. Slightly reduced crystalline samples consist of CS phases containing {102} CS planes which are ordered when the Mo content is high. These latter have overall compositions $(W_xMo_{1-x})_n$ O_{3n-1} , with n increasing with increasing W-content from 9 to 16. More substantially reduced crystals show less tendency to form mixed crystals. In samples of overall composition near to M $O_{2.90}$ phase separation occurs into {102}-containing CS phases which are molybdenum rich and {103} CS phases which are tungsten rich. The tungsten oxides $WO_{2.82}$ and $W_{18}O_{49}$ seem to contain little or no Mo. These results are summarized in a phase diagram.

The Structure of Cubic YbZr F_7 . M. POULAIN AND B. C. TOFIELD,* Materials Development Division, AERE Harwell, Oxon, OXII ORA, United Kingdom. The structure of primitive-cubic YbZr F_7 has been determined using X-ray and neutron diffraction techniques. A unit cell (a=4.07 Å, space group Pm3m) contains one formula unit of Yb_{0.5}Zr_{0.5}F_{3.5}, with no ordering of cations, in materials prepared by rapid quenching from 1000°C. Metal and fluorine displacements from ideal sites are in accord with results previously obtained on Zr_{0.8}Yb_{0.2} F_{3.2}O_{0.3}. The separation between F-F pairs bridging neighboring metal ions is similar to those observed in other complex zirconium fluorides. The metal displacements, metal-fluorine distances and fluorine-fluorine distances are discussed with respect to the formation and stability of disordered fluorine-excess ReO₃-type phases. These materials are intermediate in character between phases such as monoclinic YbZrF₇, with perfect order on both metal and nonmetal sublattices, and ZrF₄-based glasses, where there is disorder on the metal as well as on the fluorine sublattice. No ordering effects are observed on heating to near 200°C, but near 400°C there is a slow transformation to the monoclinic YbZrF₇ structure.