

Preparation and Properties of WP_4

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Introduction

Many transition metal polyphosphides MP_4 have recently been prepared. VP_4 , CrP_4 , and MoP_4 are isomorphous and metallic conductors due to the formation of metal chains in their crystal structure (1, 2). Metal pairs are formed in TcP_4 and ReP_4 (3, 4). RuP_4 and OsP_4 have α - and β -temperature modifications (5, 6). There are no metal-metal bonds in the Ru and Os compounds. The α forms are isostructural with CdP_4 . MnP_4 and FeP_4 have also several modifications (6-10). The transition metals are divalent in these compounds and metal-metal interactions in their structures were explained by the number of d electrons (5). In the present investigation, WP_4 was prepared under high pressure.

Sample Preparation

Mixtures of tungsten and red phosphorous having various mixing ratios were reacted at 1000°C for 1 hr under 3 GPa in a cubic type high pressure vessel. Details of

the high pressure experiments have already been described (1, 11). An unknown phase was observed when the starting composition of P/W was more than 2. It was contaminated with WP_2 when $P/W < 4$ and with black phosphorous when $P/W > 4$. X-ray powder diffraction data are summarized in Table I. Black single crystals with metallic luster and with irregular shapes were obtained when the starting mixture of composition $P/W = 4$ was held at 1000°C for 1 hr and slowly cooled to 600°C over 6 hr and then quenched to room temperature under high pressure of 3 GPa.

Characterization and Discussion

The X-ray powder diffraction data of WP_4 could not be indexed assuming the compound to be isostructural with other MP_4 phases. It was indexed as tetragonal with the aid of Weissenberg photographs. The lattice parameters were refined to be $a = 5.702 \text{ \AA}$, $c = 9.352 \text{ \AA}$ by a least-squares procedure. The measured density was $6.2 \text{ g} \cdot \text{cm}^{-3}$ and the X-ray density was $6.72 \text{ g} \cdot \text{cm}^{-3}$ assuming four formula units.

Figure 1 shows the density of WP_4 compared with the values for other tungsten phosphides which can be obtained at ambi-

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TABLE I
X-RAY POWDER PATTERN OF WP₄

H K L	I _{obs}	d _{obs}	d _{calc}
0 0 2	vs	4.674	4.674
1 1 0	s	4.030	4.032
1 1 2	vvvs	3.057	3.054
2 0 0	m	2.852	2.851
2 1 1	w	2.459	2.460
2 0 2	s	2.433	2.434
1 1 4	s	2.022	2.022
2 2 0	m	2.017	2.016
2 1 3	vw	1.974	1.974
2 2 2	w	1.851	1.851
2 0 4	m	1.808	1.808
3 1 0	s	1.803	1.803
3 1 2	w	1.682	1.682
0 0 6	vw	1.559	1.559

ent pressure in a sealed quartz tube (12). The densities of the phosphides can be related to the amounts of phosphorous. WP₄ fits well into the curve in Fig. 1. All compounds in the figure, except black phosphorous and WP₄, can be obtained under an ambient pressure. These results suggest that WP₄ is metastable under normal pressure conditions. Decomposition of WP₄ to WP₂ and to WP was observed for a sample kept at ambient conditions for 3 months.

Electrical resistivity was measured on a pellet of polycrystalline material using the four-probe technique. WP₄ was a metallic conductor; $\rho_{285\text{ K}} = 1.42\ \Omega\text{cm}$, $\rho_{89\text{ K}} = 0.99\ \Omega\text{cm}$. Magnetic susceptibility was $1.73 \times 10^7\ \text{emu} \cdot \text{g}^{-1}$ in the range between room temperature and liquid nitrogen. WP₄ does not have exactly the same crystal structure as CrP₄ and MoP₄ although tungsten forms a VIa group with Cr and Mo in the periodic table. CrP₄ and MoP₄ have monoclinic lattices with $a = 5.191$, $b = 10.760$, $c = 5.771\ \text{\AA}$, $\beta = 110.65^\circ$ and $a = 5.313$, $b = 11.139$, $c = 5.820\ \text{\AA}$, $\beta = 110.64^\circ$ respectively. How-

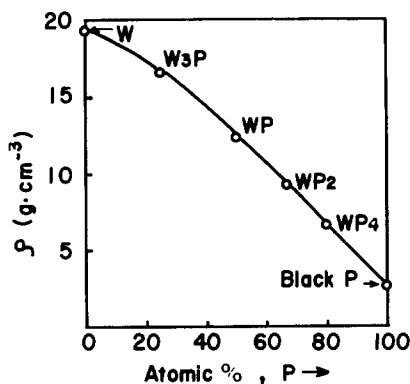


FIG. 1. Densities in the binary system W-P.

ever, WP₄ probably has W chains which are similar to those found in VP₄, CrP₄, and MoP₄.

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