

## Nature of Th-F Bonding in Crystalline and Glassy States Using EXAFS and XANES

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A binary glass of thorium and hafnium tetrafluorides (with dopant concentrations of LaF<sub>3</sub>) has been investigated using EXAFS and XANES analysis of the L<sub>III</sub> edge spectra of thorium. Results of EXAFS analysis indicate that there is no change in the number of nearest neighbors of thorium ions. However, the intensity of the white line in XANES is significantly higher in the glass. This increase in intensity has been explained semiquantitatively in terms of enhanced covalency of Th-F bonding. Since the coordination number of Th<sup>4+</sup> remains constant in both crystalline and glassy states, covalency enhancement seems to be a general characteristic of glassy state of ionic materials. © 1984 Academic Press, Inc.

### 1. Introduction

It is generally known (1) that the more covalently bonded materials are easy glass formers. In an earlier investigation of chemical shifts (2) it was suggested that the extent of covalency is higher compared to their parent crystalline states. We have also reported (3) that the white-line intensities of L<sub>III</sub> and L<sub>II</sub> edges of Nd<sup>3+</sup> are quite sensitive to small variations in the nature of the bonding. However, in the case of a NdF<sub>3</sub>-BeF<sub>2</sub> glass (3) the coordination number of Nd<sup>3+</sup> was also found to be lower in the glass than in crystalline NdF<sub>3</sub>.

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ThF<sub>4</sub>-HfF<sub>4</sub> glasses (4) are unique in that they are formed from isovalent tetrahalides. In such glasses it is conceivable that the local coordinations of cations do not change. If the coordination numbers do not change and still variations in absorption intensities at L-edges occur, it would be possible to consider such variations as incidental to glassy state and hence to the more covalent nature of bonding in the glassy state.

In this study of the Extended X-ray Absorption Fine Structures (EXAFS) and X-ray Absorption Near Edge Structure (XANES) associated with the L<sub>III</sub> edge of Th<sup>4+</sup> ions (5) (16,300.0 eV) in crystalline ThF<sub>4</sub> and a glass of composition 0.33 ThF<sub>4</sub>

$\cdot 0.6 \text{ HfF}_4 \cdot 0.07 \text{ LaF}_3$ , we have determined the local coordination of  $\text{Th}^{4+}$  in the glass using the phase parameters ( $\text{Th}-\text{F}_4$  pairs) determined from the EXAFS analysis of crystalline  $\text{ThF}_4$  and have compared the normalized white-line intensities of the  $L_{\text{III}}$  edge XANES of  $\text{Th}^{4+}$  in the  $\text{ThF}_4$  and the glass. We have found that the coordination number of  $\text{Th}^{4+}$  remains equal to 8 in the glass also while the absorption intensity is  $\sim 12\%$  higher in the glass. We attribute the intensity increase to the presence of electrons in valence  $6d$  levels of  $\text{Th}^{4+}$  which result from an increase of covalency of  $\text{Th}-\text{F}$  bonding. Such repopulation of valence  $6d$  levels is shown to affect the matrix element characterizing to  $L_{\text{III}}$  edge transition through a screening effect in a manner consistent with the observed trend.

## 2. Experimental

The glasses used in this investigation were prepared by heating the metal oxides in the presence of excess ammonium bifluoride according to the reaction  $(\text{Hf}, \text{Th})\text{O}_{2(\text{s})} + 4\text{NH}_4\text{F} \cdot \text{HF}_{(\text{s})} \rightarrow (\text{Hf}, \text{Th})\text{F}_{4(\text{s})} + 2\text{H}_2\text{O}_{(\text{g})} + 4\text{NH}_4\text{F}_{(\text{g})}$  at about  $400^\circ\text{C}$  then fusing at  $850\text{--}900^\circ\text{C}$  to form a glass.

Samples for EXAFS and XANES investigation were prepared by making fine powders ( $-400$  mesh) of the materials with Duco cement and casting the mull into films between glass slides. The procedure is discussed in detail elsewhere (6). The concentration of  $\text{Th}^{4+}$  in each such film is controlled such that it corresponds to about two absorption lengths at the  $L_{\text{III}}$  edge. Spectra were recorded at CHESS (Cornell High-Energy Synchrotron Source) facility at Cornell with CESR (Cornell Electron Storage Ring) operating at an electron energy of  $\sim 5.2$  GeV and injection current of  $\sim 15$  mA. The X-ray beam (C2 spectrometer) was monochromated with a channel-cut Si (220) crystal. Spectra were scanned with the C2 spectrometer at CHESS. The

$\text{Th}L_{\text{III}}$  edge was calibrated using  $\text{Pb}L_{\text{I}}$  edge (5) ( $15,860.8$  eV) from Pb metal. Detailed procedures for obtaining normalized XANES are given elsewhere (7). The smooth background in the EXAFS region was curve-fitted with a cubic spline using four equal segments and the background in the pre-edge region was determined by a linear regression analysis of the first 10 raw data points. The edge jump,  $S$ , was determined as the difference between the extrapolated values of the post-edge and pre-edge background at  $E - E_{\text{b}} = 0$ .  $E_{\text{b}}$  is the  $L_{\text{III}}$  edge energy of the Th at  $16300$  eV. The near-edge spectra was normalized by dividing the intensities by  $S$ . The normalized EXAFS,  $\chi(k)$  was determined as

$$\chi(k) = \frac{\mu(k) - \mu_0(k)}{S \cdot M(k)}, \quad (1)$$

where  $\mu(k)$  is the observed absorption,  $\mu_0(k)$  is the extrapolated post-edge background absorption, and  $M(k)$  is the McMaster Correction (8). Wave vector  $k$  is equal to  $(1/\hbar)\sqrt{2m(E - E_{\text{b}})}$ . The  $\chi(k)$  was Fourier-transformed and the first peak in the region of  $0.70\text{--}3.00$  Å was inverse Fourier-transformed to isolate the EXAFS originating from the shell of  $\text{F}^-$  ions.

## 3. Results and Discussion

*EXAFS analysis.* The results of EXAFS analysis of crystalline  $\text{ThF}_4$  and the glass are presented in Figs. 1 and 2, respectively. Raw spectra are shown in Figs. 1a and 2a while  $\chi(k) \cdot k$  are presented in 1b and 2b. The Fourier transforms of  $\chi(k)$  (the radial structure function, r.s.f.) are presented in 1c and 2c, respectively, for crystalline  $\text{ThF}_4$  and the glass. The  $k$ -range of transforms was limited to  $9.25 \text{ \AA}^{-1}$  since the EXAFS signal was very weak and the spectra was noisy beyond  $9.25 \text{ \AA}^{-1}$ . The scattering amplitudes of  $\text{F}^-$  ions are known to be low beyond this region and its neglect would not affect the analysis adversely.

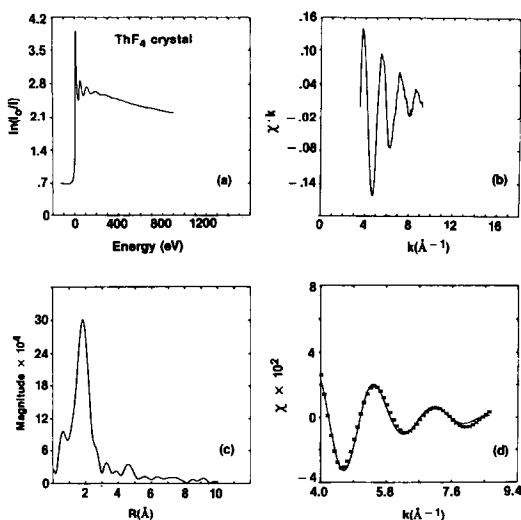


FIG. 1. (a) Room-temperature experimental scan of Th  $L_{III}$ -edge XANES and EXAFS in crystalline  $\text{ThF}_4$ . The energy scale is with respect to the Th  $L_{III}$ -edge energy (16,300 eV) taken as zero. (b) Normalized EXAFS as  $\chi(k) \cdot k$  versus  $k$ . (c) Fourier transform of (b). (d) Inverse transform (line) and simulated EXAFS (points) of the first shell in (c) in the region 0.7 to 3.0 Å.

The inverse Fourier transform,  $\chi^F(k)$ , generated from the respective first peaks in r.s.f. are shown in Figs. 1d and 2d as solid lines. These were curve fitted to the theoretical EXAFS function (9)

$$\chi(k) = -\frac{1}{k} \sum_j A_j \sin[2r_j k + \Phi_j(k)], \quad (2)$$

where

$$A_j = (N_j/r_j^2) f_j(\pi, k) \exp(-2r_j/\lambda) \exp(-2\sigma_j^2 k^2), \quad (3)$$

$\Phi_j(k)$ ,  $f_j(\pi, k)$ , and  $\lambda$  are the phase shift, amplitude, and mean free path, respectively, of the scattered electron (scattering parameters) and  $N_j$ ,  $r_j$ , and  $\sigma_j$  are the number of neighbors, distances, and their mean spread, respectively (structural parameters), corresponding to the  $j$ th shell of neighbors. The phase shifts are expressed parametrically using the expression

$$\Phi(k) = p_0 + p_1 k + p_2 k^2 + p_3/k^3. \quad (4)$$

As pointed out earlier, EXAFS of crystalline  $\text{ThF}_4$  was analyzed in order to obtain  $\Phi(k)$  for the Th-F pairs.  $\text{ThF}_4$  is isostructural with  $\text{ZrF}_4^{10}$  and  $\text{Th}^{4+}$  is surrounded by 8  $\text{F}^-$  neighbors, but the Th-F distances are not known exactly. We estimated the distances using known range of Zr-F distances in  $\text{ZrF}_4$  and the difference in ionic radii (11) of  $\text{Th}^{4+}$  and  $\text{Zr}^{4+}$  ( $r_{\text{Th}}^{4+} = 1.20$ ;  $r_{\text{Zr}}^{4+} = 0.98$ ). The two extreme Th-F distances were thus taken to be 1.83 and 1.98 Å. Scattering amplitudes for  $\text{F}^-$  ions were taken from tabulations of Teo and Lee (12).

EXAFS curve fitting was performed using  $P$ 's and  $\sigma$  as adjustable parameters. Fitting was found unsatisfactory when a single average Th-F distance was used. The fitting was unsatisfactory even when the two extreme Th-F distances were used in two fluorine subshells of the  $4\text{F}^-$  ions each. A good fit could however be obtained using three subshells of the  $\text{F}^-$  ions with 3, 3, and 2  $\text{F}^-$  ions at 1.83, 1.91, and 1.98 Å, respectively. The phase parameters and  $\sigma^2$  values

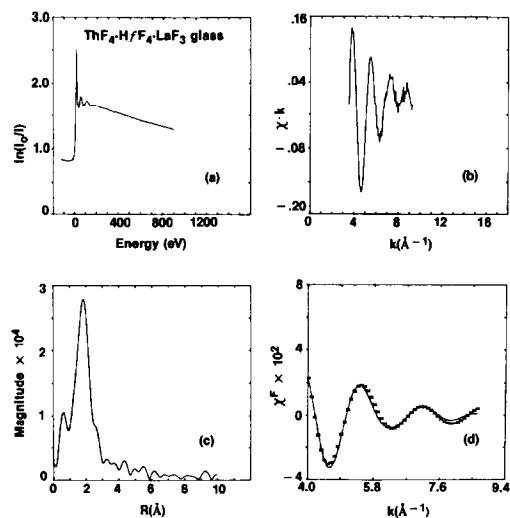


FIG. 2. (a) Room-temperature experimental scan of Th  $L_{III}$ -edge XANES and EXAFS in 0.33  $\text{ThF}_4 \cdot 0.6 \text{HfF}_4 \cdot 0.07 \text{LaF}_3$  glass. (b) Normalized EXAFS as  $\chi(k) \cdot k$  versus  $k$ . (c) Fourier transform of (b). (d) Inverse transform (line) and simulated EXAFS (points) of the first shell in (c) in the region 0.7 to 3.0 Å.

so determined for the three subshells are given in Table I. Corresponding fit is shown in Fig. 1d as crosses. It may be noted from Table I that the phase parameters for the three subshells are satisfactorily close. The structure of the glass was analyzed in the same  $k$  region using the phase shift parameters of Table I.  $N_j$ ,  $r_j$ , and  $\sigma_j$  were treated as adjustable parameters in the EXAFS fitting. A good fit (crosses in Fig. 2d) could be obtained in the case of glass also only with the use of three subshells of  $F^-$  ions. The structural parameters so obtained are given in Table II.

It may be seen that the total number of neighbors in the three subshells of  $Th^{4+}$  in glass is equal to 8.13 which is equal to the coordination number of  $Th^{4+}$  in crystalline  $ThF_4$  (=8) within 1.5%. We may therefore infer that the near-neighbor coordination number of  $Th^{4+}$  remains unaltered upon glass formation. The Th-F distances in the glass obtained in the fitting are also very nearly equal (within 0.01 Å) to those in crystalline  $ThF_4$  shown in Table I. Of interest also are the lower  $\sigma_j^2$  values of the two innermost subshells with respect to those of the crystal. This may be reflective of a covalency effect due to glass formation discussed below.

*XANES intensities and covalency.* The normalized XANES spectra of the  $L_{III}$  edge of  $Th^{4+}$  in crystalline  $ThF_4$  and the glass are

TABLE I  
PHASE PARAMETERS FOR Th-F SCATTERING PAIRS  
FROM THE EXAFS OF  $L_{III}$  EDGE OF  $Th^{4+}$  IN  
CRYSTALLINE  $ThF_4$

| $N$ | $r$<br>(Å) | $\sigma^2$<br>(Å <sup>2</sup> ) | $P_0$  | $P_1$   | $P_2$  | $P_3$  |
|-----|------------|---------------------------------|--------|---------|--------|--------|
| 3.0 | 1.830      | 0.01103                         | 4.1263 | -1.2134 | 0.0293 | -29.42 |
| 3.0 | 1.910      | 0.00785                         | 4.3893 | -1.1760 | 0.0329 | 26.25  |
| 2.0 | 1.980      | 0.00026                         | 4.7721 | -1.1367 | 0.0356 | 96.23  |

Note. Standard deviation in curve fitting was found to be 6.6% of the maximum amplitude of  $\chi(k)$ .

TABLE II  
STRUCTURAL PARAMETERS OF  $0.33 ThF_4 \cdot 0.60 HfF_4$   
 $\cdot 0.07 LaF_3$  GLASS FROM EXAFS ANALYSIS  
OF  $L_{III}$  EDGE

| $N$<br>(Å) | $r$<br>(Å) | $\sigma^2$<br>(Å <sup>2</sup> ) | Standard deviation<br>of EXAFS fit |
|------------|------------|---------------------------------|------------------------------------|
| 2.89       | 1.83       | 0.0092                          | 7%                                 |
| 3.17       | 1.92       | 0.0065                          |                                    |
| 2.07       | 1.99       | 0.0005                          |                                    |

shown in Fig. 3. The white-line intensity (measured at the peak) is about 12% higher in the glass than in crystalline  $ThF_4$ . In order to understand this variation, we may note that the absorption coefficients are governed by the values of the transition dipole matrix element in the equation (13)

$$\mu = \frac{4\pi^2}{n} \frac{e^2}{\hbar c} \hbar\omega \sum_{ij} |M'_{ij}|^2 \delta(E_i - E_j - \hbar\omega). \quad (5)$$

In Eq. (5),  $M'_{ij} = \langle j|r_A|i \rangle$  is the dipole matrix element connecting the two states  $\langle j|$  and  $|i\rangle$ ;  $\delta$  is the Kronecker delta;  $E_i$  and  $E_j$  are the initial and final state energies;  $n$  is the refractive index and  $c$  is the velocity of light. For the  $L_{III}$  edge transition in  $Th^{4+}$  the initial and final states correspond to the core  $2p$  and valence  $6d$  levels. In the one-elec-

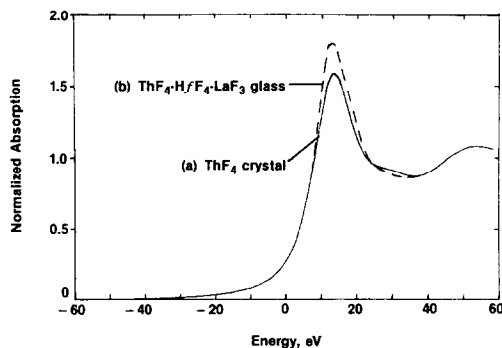


FIG. 3. Normalized  $Th L_{III}$ -edge XANES in (a) crystalline  $ThF_4$  and (b)  $0.33 ThF_4 \cdot 0.6 HfF_4 \cdot 0.07 LaF_3$  glass.

tron approximation, it has been shown that

$$M'_{ij} = \int_0^{\infty} U_{n,i}(r) \cdot r \cdot U_{n',l\pm 1} dr, \quad (6)$$

where  $U_{n,i}$  and  $U_{n',l\pm 1}$  are the initial and final state wave functions. Assuming the same density of empty  $6d$  states the ratio of the absorption intensities ( $\mu_g/\mu_c$ ), are the absorption coefficients of glass and crystal respectively, is given by

$$\frac{\mu_g}{\mu_c} = \frac{|M'_{i,j}(\text{glass})|^2}{|M'_{i,j}(\text{crystal})|^2}. \quad (7)$$

We may therefore investigate the effect of covalency using Eqs. (6) and (7). An increase of covalency in the glass is equivalent to a reverse transfer of a certain fraction of electrons from  $F^-$  ions to the central  $Th^{4+}$  ion. These electrons repopulate the now-shared  $6d$  valence shell of  $Th^{4+}$ . As a direct consequence of this repopulation of  $6d$  levels, the nuclear potential experienced by the  $6d$  levels is additionally screened by such electrons. Hence, the final states  $|j\rangle$  in  $M'_{i,j}$  are additionally screened as a consequence of increased covalency. Let us assume that  $\nu$  electrons are back-donated by  $F^-$  ions, then the fractional covalency is  $\nu/8$  or percentage covalency is  $(100 \nu/8)$ . The screening effect of electrons can be estimated by the use of Slater orbitals (14) in Eq. (6). Slater orbitals are of the form  $\Psi_n = r^{n^*-1} \exp[-(z-s)r/n^*]$ , where  $n^*$  is the effective quantum number and  $s$  is the Slater screening constant. The  $2p$  electron (initial state) in  $Th^{4+}$  experiences shielding due to seven other  $2s$  and  $2p$  electrons as well as two  $1s$  electrons. The  $6d$  electrons (final state) of  $Th^{4+}$  experiences shielding due to all the electrons (Slater-grouped electronic structure of  $Th^{4+}$  is  $1s^2; 2s^2; 2p^6; 3s^2; 3p^6; 3d^{10}; 4s^2; 4p^6; 4d^{10}; 4f^{14}; 5s^2; 5p^6; 5d^{10}; 5f^0; 6s^2; 6p^6; 6d^{10}; 6f^0; 7s^0$ ).

Thus the value of  $s$  in the Slater orbital expression for  $2p$  electrons is 4.15 and for the  $6d$  electrons is 86. The wave functions,  $U_{2,1}(r)$  and  $U_{6,2}(r)$  become  $r \exp[-(90 -$

$4.15)r/2]$  and  $r^{3.2} \exp[-(90 - 86)r/4.2]$ , respectively, for  $Th^{4+}$  in crystalline  $ThF_4$  assuming that the bonding in  $ThF_4$  is completely ionic. In the case of glass,  $U_{6,2}(r)$  becomes,  $r^{3.2} \exp[-(90 - 86 - 0.35\nu)r/4.2]$ . Using these wave functions in Eq. (7), we have

$$\frac{\mu_g}{\mu_c} = \frac{\left| \int_0^{\infty} r^{5.2} e^{-K_g r} \cdot dr \right|^2}{\left| \int_0^{\infty} r^{5.2} e^{-K_c r} \cdot dr \right|^2}. \quad (8)$$

Using standard methods of integration, we have

$$\frac{\mu_g}{\mu_c} = \frac{|\Gamma(6.2)/K_g^{6.2}|^2}{|\Gamma(6.2)/K_c^{6.2}|^2} = \left(\frac{K_c}{K_g}\right)^{12.4}. \quad (9)$$

In Eqs. (8) and (9),  $K_g = (4 - 0.35\nu)/4.2 + 85.85/2 = 43.8774 - 0.083\nu$  and  $K_c = 43.8774$ . Hence

$$\frac{\mu_g}{\mu_c} = \left[ \frac{43.8774}{43.8774 - 0.083\nu} \right]^{12.4} \sim 1 + 0.023\nu. \quad (10)$$

Equation (10), therefore clearly suggests that  $\mu_g > \mu_c$  for any positive value of  $\nu$  or any increase in covalency of bonding. In order that  $\mu_g/\mu_c = 1.12$ , as observed experimentally, it requires that  $\nu = 5.2$  which in turn amounts to 65% covalency. Such a value, we feel, is perhaps very high. However, the quantitative discrepancy is likely to originate from the fact that node-free Slater orbitals are only very approximate. Further, we are not at all clear of the ways in which the relativistic nature of the  $2p$  wave functions in  $Th^{4+}$  affect the above considerations.

The results discussed above suggest therefore, that even when the nearest neighbor coordination number of the cation ( $Th^{4+}$ ) does not change, the covalency of bonding to anions ( $F^-$  ions) increases in the glassy state. The results also imply that the increase in covalency is a concomitant effect of positional disordering in ionic materials.

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