

BRIEF COMMUNICATIONS

IR and Polarized Raman Spectra of $(\text{NH}_3)_2(\text{CH}_2)_2\text{HPO}_4$

DAIZY PHILIP AND G. ARULDHAS*

*Department of Physics, University of Kerala, Kariavattom,
Trivandrum-695 581, India*

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The IR and polarized Raman spectra of $(\text{NH}_3)_2(\text{CH}_2)_2\text{HPO}_4$ have been recorded and analyzed. In the crystal, symmetry of the HPO_4 ion is found to be reduced from C_{3v} to C_1 . The observed splittings indicate that the site symmetry and the correlation field effects are appreciable in the crystal. Strong vibrational coupling between the P-O(H) stretching and POH out-of-plane bending modes is observed. Ethylene diammonium groups are crystallographically different in the unit cell. © 1989 Academic Press, Inc.

Introduction

Ethylenediamine forms crystals of the very stable ethylenediammonium phosphate, $(\text{NH}_3)_2(\text{CH}_2)_2\text{HPO}_4$, when mixed with orthophosphoric acid in the stoichiometric ratio and evaporated slowly at room temperature. The compound crystallizes in the monoclinic system $P2_1/a$ with four formula units per unit cell (*I*). A detailed vibrational analysis of its IR and Raman spectra has not been done so far. In the present investigation, the IR and single-crystal Raman spectroscopy studies have been taken up in order to understand the nature of the HPO_4 ion and to elucidate the distribution of vibrational energy levels in the crystal lattice.

Experimental

Raman spectra (Fig. 1) of the single crystal were recorded for the A_g and B_g species

in the Stokes region of the 5145-Å line using a SPEX RAMALOG 1401 double monochromator equipped with a Spectra Physics Model 165 Ar^+ laser. A Polytec FIR 30 interferometer was used to record the far-infrared spectrum in the region 50–500 cm^{-1} with the sample in polyethylene. The IR spectrum in the region 200–4000 cm^{-1} was obtained on a Perkin-Elmer 983 spectrophotometer with the sample in KBr and nujol. The spectrum in the 50–4000 cm^{-1} is shown in Fig. 2.

Discussion

In the crystal, the HPO_4 ions occupy site C_1 of lower symmetry than the free ion symmetry C_{3v} . This may lead to the activation of inactive modes along with shifting and splitting of the internal modes. The factor group analysis predicts 237 normal modes and they are distributed as

$$\overline{237} = 60 A_g + 60 B_g + 59 A_u + 58 B_u.$$

* To whom correspondence should be addressed.

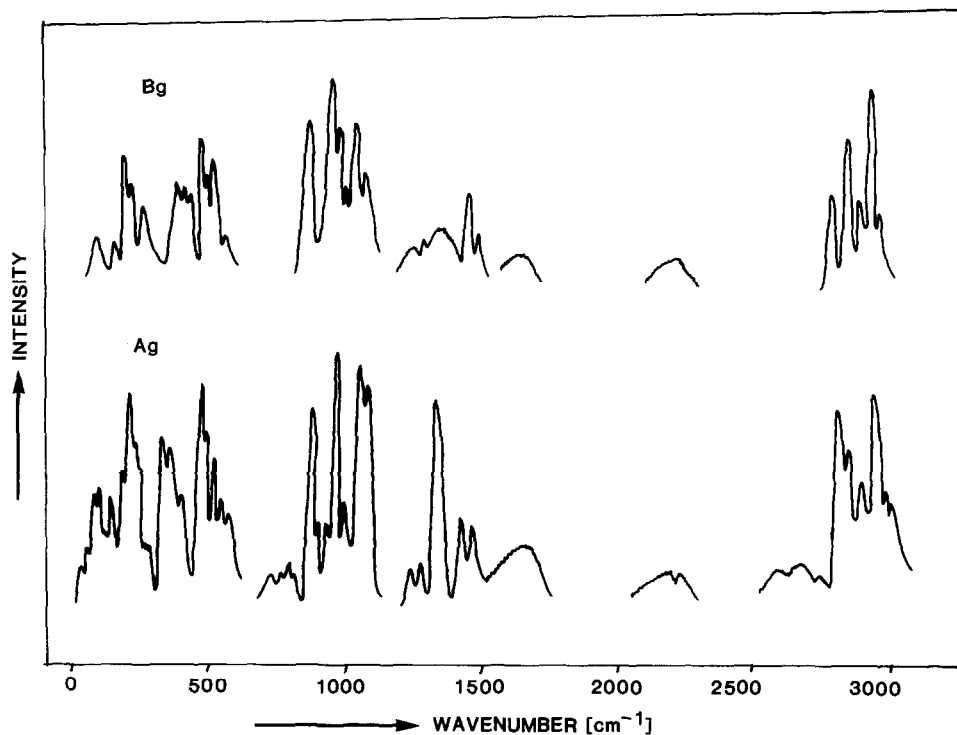


FIG. 1. Raman spectrum of $(\text{NH}_3)_2(\text{CH}_2)_2\text{HPO}_4$.

The free ion approach predicts three bands each for the stretching ($2A_1 + E$) and bending ($A_1 + 2E$) modes of the PO_4 part of the

HPO_4 ion in both Raman and IR spectra. However, the site symmetry approach leads to four and five bands, respectively,

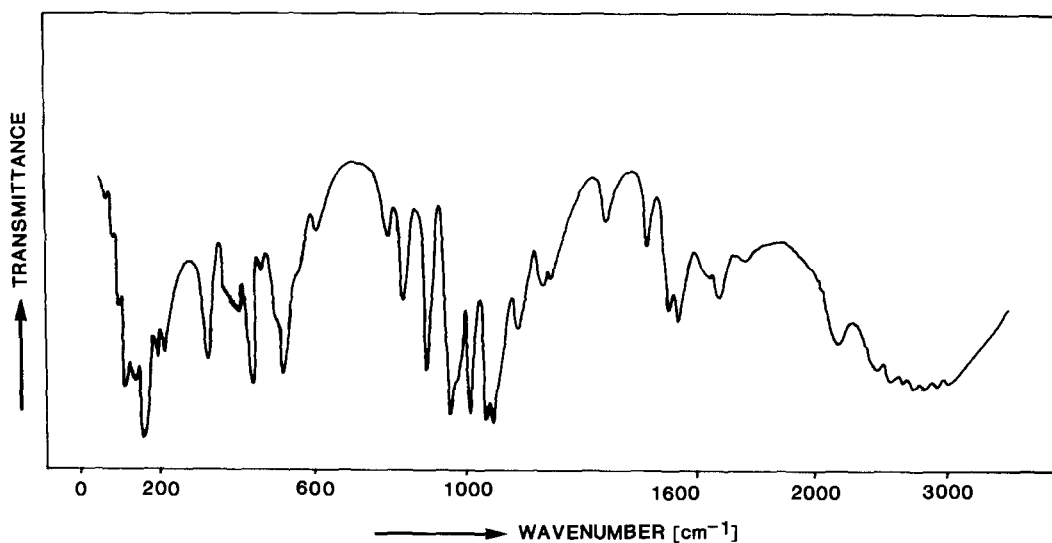


FIG. 2. IR spectrum of $(\text{NH}_3)_2(\text{CH}_2)_2\text{HPO}_4$.

TABLE I
VIBRATIONAL SPECTRAL DATA AND BAND ASSIGNMENTS (cm^{-1}) OF $(\text{NH}_3)_2(\text{CH}_2)_2\text{HPO}_4$

Raman				Raman			
A_g	B_g	IR	Assignments	A_g	B_g	IR	Assignments
3016(m)			ν_s and ν_{as} NH_3^+	926(m)			ρ NH_3^+ and ρ CH_2
2983(m)	2983(m)	3018(w,br)		887(w)			P-O(H) stretch (A_1)
2966(vs)	2968(vs)			879(vs)	879(s)	901(vs)	
2902(s)	2921(m)	2940(w)	ν_{as} CH_2			829(s)	POH out-of-plane bend (A_2)
2862(s)	2876(s)			816(w)		792(m)	
2821(vs)	2818(m)	2850(w)	ν_s CH_2	778(w)			
2740(w,br)		2741(w,br)		762(w)		600(w)	Combinations
2676(w,br)		2665(w)	OH vibrations, overtones, and combinations	745(w)			
2600(w,br)	2224(w,br)	2590(w)		570(w)	558(w)	555(w)	δ_{as} PO_3 (E)
2200(w,br)		2490(w)		550(m)	530(s)	522(vs)	
2180(w,br)		2200(s,br)		519(s)		502(m)	δ_s PO_3 (A_1)
		1750(vw,br)					
1635(m,br)	1645(w,br)	1665(m)	δ_{as} NH_3^+	479(w)	482(m)		t NH_3^+
		1640(w)		466(vs)		451(w)	δ C-CN
		1565(s)	δ_s NH_3^+	462(w)	473(s)		
		1540(s)		407(m)	424(m)	430(s)	
1470(m)	1478(w)			384(s)	413(m)	410(s)	OPO(H) δ_s (E)
1445(m)	1455(m)	1473(m)	δ CH_2	375(s)	395(m)	330(s)	
1335(vs)	1338(m,br)			260(w,br)	250(m)	256(m)	
1282(w)	1280(w,br)	1365(m)	t twist and wag of CH_2	237(m)	202(m)	230(m)	t NH_3^+
				212(m)			
1249(w)	1252(w,br)	1224(w)	POH in-plane bend (A_2)	200(vs)	198(s)		R ?
		1200(m)					
		1140(m)	ρ NH_3^+			182(vs)	
1071(vs)	1066(m)	1080(vs)	ν_{as} PO_3 (E)	187(m)	154(w)	150(s,br)	T
1053(vs)	1056(s)	1060(vs)		144(s)		130(s)	
		1028(vs)	ν C-N	116(m,br)			ν O-H \cdots O
992(m)	995(w)	977(sh)		83(s)	88(w)	95(m)	T ?
963(vvs)	972(m)	966(vs)	ν_s PO_3 (A_1)	72(s)		75(w)	
	965(vs)			58(m)			δ O-H \cdots O
				55(w)			

Note. (vvs) Very very strong; (vs) very strong; (s) strong; (m) medium; (w) weak; (vw) very weak; (sh) shoulder; (s,br) strong broad; (m-br) medium broad; (w,br) weak broad; (δ) bending; (ν) stretching; (t) twisting; (ρ) rocking. Subscripts "as" and "s" refer to asymmetric and symmetric, respectively. R and T refer to the rotation and translation of the HPO_4 ion, respectively.

for these modes. The observation of more bands than this predicted number shows that both site symmetry and correlation field effects are appreciable in the crystal.

The PO_3 symmetric stretching mode (A_1) is expected to be the most intense and polarized in Raman spectrum. It splits into two components in A_g (963 and 992 cm^{-1}) and into three in B_g (965, 972, and 995 cm^{-1}) due to correlation field effect. In the IR spectrum it is also observed as two bands (966 and 977 cm^{-1}). Site symmetry effect is probably responsible for the observation of the PO_3 asymmetric stretch (E) as a doublet

in both Raman and IR. In compounds with HPO_4 ions, the polarized P-O(H) stretch (A_1) is expected around 860 cm^{-1} (2). Here, it is observed at 879 cm^{-1} in the Raman spectrum. However, in IR it is shifted to 901 cm^{-1} . The splitting of the F_2 stretching mode of the PO_4 ion into three very strong components in IR [1080 (E), 1060 (E) and 901 cm^{-1} (A_1)] confirms (3) the lowering of the symmetry of the HPO_4 ion from C_{3v} to C_1 in the crystal. The deformation vibrations are also identified (Table I).

Strong vibrational coupling between P-O(H) stretching and POH out-of-plane

bending (829 cm^{-1}) is evident (3) from the large intensity of the latter. The shift observed in IR for the P–O(H) stretching frequency also supports this. This vibrational coupling and correlation field effect are probably responsible for the splitting observed for these modes. The observation of triobands (4) for the HPO_4 ion in the regions $2800\text{--}2400$, $2350\text{--}1900$, and $1720\text{--}1600\text{ cm}^{-1}$ and the lowering of the NH_3^+ stretching frequency suggest that the system is strongly hydrogen bonded.

Doubling is observed for all the internal vibrations of the CH_2 group in Raman and also for the bending modes of NH_3^+ in IR with splittings of more than 20 cm^{-1} . This indicates the presence of crystallographically different ethylenediammonium groups

in the unit cell which is in agreement with the structural data (1).

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