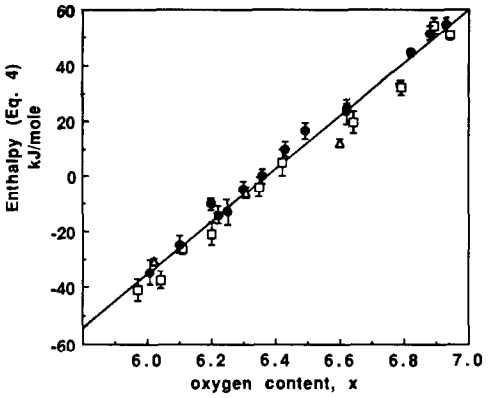

REVISED FIG. 2. Enthalpy of oxidation or reduction (Eq. (4)), kJ/mole. Open triangles, P. K. Davies samples; open squares, A. Jacobson samples; filled circles, E. Takayama-Muromachi samples.

temperature dependence of oxygen content and of -190 ± 20 kJ/mole obtained by acid calorimetry.

Volume **80,** Number 1 (1989), in the article "Syntheses and Characterization of Two Novel Inclusion Compounds: $AlAsO_4 \cdot 0.2(CH_3)_4NOH \cdot 0.3H_2O$ and $GaAsO_4 \cdot 0.2(CH_3)_4NOH \cdot 0.1H_2O$," by J. Chen and R. Xu, pages 149–151: The formulae $AlPO_4^{n-}$, $GaPO_4^{n-}$, $AlAsO_4^{1-}$, $AlAsO_4^{2-}$, and $GaAsO_4^{1-}$ should read $AlPO_4^{-n}$, $GaPO_4^{-n}$, $AlAsO_4^{-1}$, $AlAsO_4^{-2}$, and $GaAsO_4^{-1}$, respectively.