

CCCLXXVI.—*Preparation of the 5-Halogenoresorcinols.*

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THE three 5-halogenoresorcinols have been prepared by converting the 3-halogeno-5-amino-anisoles (this vol., p. 2078) into the 5-halogeno-3-hydroxyanisoles and hydrolysing these.

5-Halogeno-3-hydroxyanisoles.—The 3-halogeno-5-aminoanisole (14 g.) or its sulphate (20 g.) in a mixture of sulphuric acid (15 g.) and water (150 c.c.) was diazotised at 0.5°, the excess of nitrous acid destroyed by carbamide, and the clear solution added gradually to a vigorously boiling mixture of sulphuric acid (120 g.) and water (150 c.c.) through which a current of steam was passing. The 5-halogeno-3-hydroxyanisole was only slowly volatilised. The voluminous distillate was made just alkaline, concentrated to small bulk, and acidified; the product then separated as an oil which solidified.

The 5-halogeno-3-hydroxyanisoles all reduce boiling Fehling's solution very slowly, and in aqueous solution give a faint bluish-pink coloration with ferric chloride.

5-Chloro-3-hydroxyanisole is readily soluble in water, alcohol, benzene, or glacial acetic acid, and crystallises from light petroleum or water in colourless needles, m. p. 99° (Found: Cl, 22.45. $C_7H_7O_2Cl$ requires Cl, 22.4%). *5-Chloro-2:4:6-tribromo-3-hydroxyanisole* is obtained from glacial acetic acid in colourless needles, m. p. 110° (0.0884 g. gave 0.1574 g. of silver halides. Calc., 0.1581 g.), and *5-bromo-3-hydroxyanisole* from water in colourless needles, m. p. 102° (Found: Br, 39.2. $C_7H_7O_2Br$ requires Br, 39.4%); the latter is less readily soluble in water than the chloro-analogue. *2:4:5:6-Tetrabromo-3-hydroxyanisole* crystallises from glacial acetic acid in colourless needles, m. p. 120° (Found: Br, 72.5. $C_7H_4O_2Br_4$ requires Br, 72.7%). *5-Iodo-3-hydroxyanisole* sublimes in colourless needles, m. p. 90° (Found: I, 50.7. $C_7H_7O_2I$ requires I, 50.8%). It is almost insoluble in water or light petroleum, but dissolves readily in alcohol or benzene. *2:4:6-Tribromo-5-iodo-3-hydroxyanisole* crystallises from dilute acetic acid in colourless needles, m. p. 125° (0.0991 g. gave 0.1621 g. of silver halides. Calc., 0.1626 g.).

5-Halogenoresorcinols.—The 5-halogeno-3-hydroxyanisole (7 g.) was boiled under reflux for 3 hours with hydriodic acid (60 c.c.; *d* 1.6), the solution decolorised by addition of sodium thio-sulphate, and the 4-halogenoresorcinol extracted with ether. The product was stirred in cold 5% aqueous sodium hydroxide (80 c.c.)

to remove sulphur and recovered by acidification and extraction with ether (yields, 5—6 g.).

The 5-halogenoresorcinols all reduce boiling Fehling's solution and give pronounced bluish-purple colorations with ferric chloride. 5-Chloro- and 5-iodo-resorcinols differ from 5-bromoresorcinol in that their monohydrates are much more stable.

5-Chlororesorcinol.—The crude product from the foregoing hydrolysis crystallised from benzene in colourless needles, m. p. 67° , of the monohydrate (Found: Cl, 21.5. $C_6H_5O_2Cl \cdot H_2O$ requires Cl, 21.8%). When this was sublimed in a vacuum, microcrystalline needles of the anhydrous substance were obtained, m. p. 117° (Found: Cl, 24.4. $C_6H_5O_2Cl$ requires Cl, 24.5%). The anhydrous substance is readily soluble in all the usual solvents except light petroleum, in which it is insoluble, and rapidly absorbs water, regenerating the hydrate (Found: loss in a vacuum over concentrated sulphuric acid, 10.8. Calc.: H_2O , 11.1%). *5-Chloro-2:4:6-tribromoresorcinol* crystallises from glacial acetic acid or dilute alcohol in colourless needles, m. p. 143° (0.1539 g. gave 0.2762 g. of silver halides. Calc., 0.2851 g.). The low m. p. and the poor analytical result probably indicate the presence of a hydrate. *5-Bromoresorcinol* crystallises from benzene in colourless needles, m. p. 87° (Found: Br, 42.4. $C_6H_5O_2Br$ requires Br, 42.3%). These remain unchanged in m. p. after sublimation in a vacuum, but on recrystallisation from water large prisms of the monohydrate, m. p. about 79° , are obtained (Found: Br, 39.2. Calc. for monohydrate: Br, 38.6%). *2:4:5:6-Tetrabromoresorcinol* crystallises from glacial acetic acid or dilute alcohol in colourless needles, m. p. 212° (Claassen, *Ber.*, 1878, **11**, 1440, gives m. p. 163° ; Benedikt, *Monatsh.*, 1880, **1**, 366, gives m. p. 167°) (Found: Br, 75.0. Calc.: Br, 75.1%). *5-Iodoresorcinol* crystallises from benzene in colourless needles, m. p. 92.3° (Found: I, 49.9. $C_6H_5O_2I \cdot H_2O$ requires I, 50.0%). On sublimation in a vacuum, colourless needles were obtained which still retained water but melted at 105 — 113° (Found: I, 52.0. $C_6H_5O_2I$ requires I, 53.8%). Keeping over concentrated sulphuric acid or crystallisation from dry benzene in presence of phosphorus pentoxide failed to dehydrate the product completely. *2:4:6-Tribromo-5-iodoresorcinol* separates from glacial acetic acid in colourless needles, m. p. 214° (0.0793 g. gave 0.1339 g. of silver halides. Calc., 0.1340 g.).

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