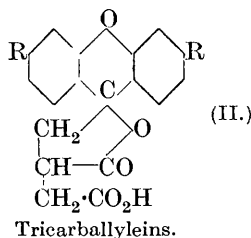
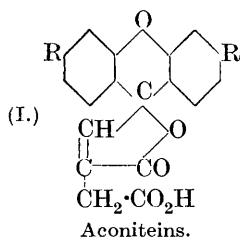


CCCXLV.—*The Effect of Unsaturated Chromophores on Pyronine Dyestuffs. Part I.*

By BRIG BEHARI DIKSHIT and JAMUNA DATT TEWARI.

SATURATED compounds behave as frictionless media for the passage of light waves in the visible and ultra-violet region of the spectrum, as is evident from the fact that such compounds show no absorption in these two regions. The entrance of unsaturation introduces friction into the molecule, causing it to absorb light waves in these two regions, and, in general, the converse holds. Any increase of unsaturation in the dye molecule enhances the colour of the compound. Examples of this type are found in dyes derived from succinic (Dutt and Thorpe, *J.*, 1924, **125**, 2524) and maleic acids (Lunge and Burckhardt, *Ber.*, 1884, **17**, 1598, 1600; Burekhardt, *Ber.*, 1885, **18**, 2864, 2878). In order to elucidate this point, dyes have been prepared by condensing various aromatic hydroxy- and amino-compounds with aconitic acid, and these are directly compared with the analogous dyes obtained from tricarballic acid, the saturated acid corresponding to aconitic acid. The resulting dyes will have the general formulæ (I) and (II). The free carboxyl



group has very little effect on the colour of the dyestuffs (Tewari and Dutt, *J. Indian Chem. Soc.*, 1928, **5**, 60).

Aconitic anhydride could, theoretically, have either of the struc-

tures (III) and (IV), but the former is the more stable of the two because the interposition of a double bond between the two carbon atoms brings their two carboxyl groups so near each other that the anhydride formation between them takes place with greatest ease



(Beesley, Ingold, and Thorpe, J., 1915, **107**, 1080; Ingold and Thorpe, 1919, **115**, 320; Kon, 1921, **119**, 810; Dhar and Dutt, *J. Indian Chem. Soc.*, 1927, **4**, 253).

The absorption maxima of the dyestuffs given in Table I clearly show that unsaturation in the molecule considerably enhances the intensity of the colour.

EXPERIMENTAL.

Phenol-, *resorcinol*-, *phloroglucinol*-, and *m*-*diethylaminophenol*-*aconiteins* and *-tricarbalyleins* were prepared and purified in the same way as the corresponding cinchomeroneins (Tewari, J., 1929, 1642) and iminazophthaleins (Tewari and Dutt, *J. Indian Chem. Soc.*, 1927, **4**, 202). The isolation and purification of catechol-*aconitein* and *-tricarbalylein* were effected as for catecholitaconein (Dhar and Dutt, *ibid.*, p. 248). These compounds, being insoluble in water, were extracted from the insoluble tin sulphide by means of alcohol.

The preparation and purification of the *m*-*dimethylaminophenol* compounds were the same as for the corresponding diethyl compounds (see above).

The condensation of *m*-*phenylenediamine* with these acids was done in an oil-bath at 150—160°, dry hydrogen chloride being used as a condensing agent. The dyestuff was extracted with glacial acetic acid, largely diluted with water, and precipitated from this acid solution by the addition of ammonia. It was recrystallised from acetic acid.

TABLE I.
Absorption maxima, μμ.

Phenol, etc.	Aconiteins.	Tricarbalyleins.
Phenol-	5530	5480
Resorcinol-	4930	4870
Phloroglucinol-	4940	4880
<i>m</i> -Diethylaminophenol-	5560	5500
<i>m</i> -Dimethylaminophenol-	5550	5480
Catechol-.....	4960	4920
<i>m</i> -Phenylenediamine-	4910	4840
Pyrogallol-	4950	4910
<i>o</i> -Cresol-	5540	5500
<i>m</i> -Cresol-	5530	5470
Tetrabromoresorcinol-	5490	5400
Tetraiodoresorcinol-	5600	5540

TABLE II.

	Appearance.	M. p.	Colour in alcohol, and effect of addition of alkali (or acid *).	Fluorescence.	Analysis, %.	
					Found.	Required.
					C. H.	C. H.
Aconiteïn.						
Phenol-	Colourless needles	250° (decomp.)	Light pink; deepened	—	66.8 4.1	66.4 4.3
Resorcinol-	Orange-yellow powder	190° (decomp.)	Orange-yellow; increased	Green	63.0 3.5	63.5 3.5
Catechol-	Dark needles	150°	Red; transient green	—	63.3 3.9	63.5 3.5
Phloroglucinol-	Brown powder	Blackens at 195°	Red; deepened	—	58.5 3.4	58.1 3.2
<i>m</i> -Diethyl-aminophenol-	Violet-red powder	112°	Bright pink; intensified †	Brown	(N, 6.6)	(N, 6.2)
<i>m</i> -Dimethyl-aminophenol-	Red powder	118° (decomp.)	Rose-red; intensified †	—	(N, 7.2)	(N, 7.1)
<i>m</i> -Phenylenediamine-	Dull red powder	248°	Deep yellow; intensified *	Green in acetic acid	(N, 15.8)	(N, 15.8)
<i>o</i> -Cresol-	Dark brown powder	195° (decomp.)	Nearly colourless; pink *	—	71.6 4.8	71.4 4.8
<i>m</i> -Cresol-	Orange-red powder	126° (decomp.)	Orange-yellow; pink *	—	71.0 4.7	71.4 4.8
Pyrogallol-	Black shining powder	> 300°	Red.	—	57.4 3.8	58.1 3.5
Tetrabromoresorcinol-	Brick-red crystalline powder	Decomp. above 170°	Red; intensified	—	(Br, 48.9)	(Br, 48.8)
Tetraiodoresorcinol-	Dark red powder	Decomp. above 187°	Dark red; intensified	—	(I, 60.3)	(I, 60.2)
Tricarballeyein.						
Phenol-	Dark brown powder	220° (decomp.)	Light orange; red	—	65.3 4.8	65.7 4.9
Resorcinol-	Bright yellow powder	207° (decomp.)	Yellow; intensified	Bright green	63.6 4.4	63.2 4.1
Catechol-	Dark black powder	180° (decomp.)	Red; transient green	—	63.2 4.3	63.2 4.1
Phloroglucinol-	Bright brown powder	119°	Bright red	—	57.9 3.6	57.8 3.8
<i>m</i> -Diethyl-aminophenol-	Scarlet powder	140° (decomp.; shrinks at 130°)	Deep rose-pink †	Brown	(N, 6.0)	(N, 6.2)
<i>m</i> -Dimethyl-aminophenol-	Scarlet powder	138°	Deep pink *	Brown	(N, 7.5)	(N, 7.1)
<i>m</i> -Phenylenediamine-	Brown powder	198—200° (decomp.; blackens at 190°)	Faint yellow; deepens *	—	(N, 15.7)	(N, 15.6)
<i>o</i> -Cresol-	Dark brown powder	168° (decomp.)	Dirty red; brilliant red in acetic acid; on dilution turns yellow	—	68.1 5.7	67.4 5.6
<i>m</i> -Cresol-	Black powder	220° (decomp.)	Deep red	—	67.6 5.4	67.4 5.6
Pyrogallol-	Scarlet powder	236° (decomp.; darkens at 195°)	Rose-red; dirty red	—	57.6 3.8	57.6 3.8
Tetrabromoresorcinol-	Dull red powder	187° (decomp.; darkens at 177°)	Pink; deep red	—	(Br, 48.4)	(Br, 48.9)
Tetraiodoresorcinol-	Scarlet powder	215° (decomp.)	Pink; deep red	Faint green	(I, 60.0)	(I, 60.0)

* Denotes effect of addition of acid.

Pyrogallol compounds were prepared and purified in the same way as in the case of the corresponding compound of itaconic acid (Dhar and Dutt, *loc. cit.*).

o-Cresol-aconiteïn and -tricarbalyleïn.—A mixture of 2 g. of the required acid and 2.5 g. of *o*-cresol was heated at 120—130° with a few drops of concentrated sulphuric acid for about 10 hours, till the melt became quite hard and brittle on cooling. The dyestuff was extracted by means of dilute caustic soda solution, reprecipitated as a thick red mass by gradual addition of hydrochloric acid, and crystallised from alcohol. The corresponding *m*-cresol compounds were similarly prepared and purified.

The bromine and iodine compounds of resorcinol-aconiteïn and -tricarbalyleïn were prepared in the same way as eosin and erythrosin from fluoresceïn.

The properties and analyses of these *compounds* are given in Table II.

In conclusion we wish to thank Dr. S. Dutt for his interest in this work.

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